

ADAMAS UNIVERSITY

SCHOOL OF LIFE SCIENCE AND BIOTECHNOLOGY

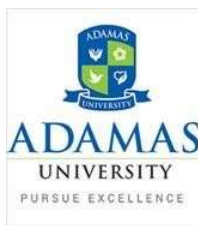
Department of Biological Sciences

M.Sc. Biochemistry (2 Years) Course Structure

Total Credits –88

(Program Code: BIC4201)

(2024-26)



VISION OF THE UNIVERSITY

To be an internationally recognized university through excellence in inter-disciplinary education, research and innovation, preparing socially responsible well-grounded individuals contributing to nation building.

MISSION STATEMENTS OF THE UNIVERSITY

M.S 01: Improve employability through futuristic curriculum and progressive pedagogy

with cutting-edge technology

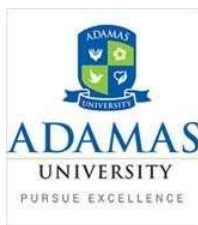
M.S 02: Foster outcomes based education system for continuous improvement in education, research and all allied activities

M.S 03: Instill the notion of lifelong learning through culture of research and innovation

M.S 04: Collaborate with industries, research centers and professional bodies to stay relevant and up-to-date

M.S 05: Inculcate ethical principles and develop understanding of environmental and social realities

CHANCELLOR / VICE CHANCELLOR



**ADAMAS UNIVERSITY, KOLKATA SCHOOL OF
LIFE SCIENCE AND BIOTECHNOLOGY
DEPARTMENT OF BIOLOGICAL SCIENCES**

VISION OF THE SCHOOL

To achieve global standard and excellence in research on various interdisciplinary and multidisciplinary domains of biological sciences through biotechnological innovation along with producing global citizens as graduates by intensive teaching learning process who would be vanguard to sustainable societal development.

MISSION STATEMENTS OF THE SCHOOL

M.S 01: To disseminate knowledge of life science and biotechnology for scholarly progression, intellectual development and strive for innovation.

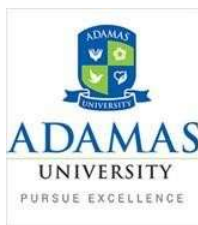
M.S 02: To enable latest skill sets in the domain of microbiology, biotechnology, biochemistry (biological sciences) with ability to evolve multidisciplinary functionality and engage in learn-unlearn and relearn, being a lifelong learner.

M.S 03: To establish state of art infrastructure and research ambiance in attracting the best minds to serve under the single roof of school of life science and biotechnology in undertaking scientific investigation of social relevance.

04: To inculcate values, culture along with scientific knowledge to foster the spirit of self-reliance and entrepreneurship development.

DEAN / SCHOOL CONCERNED

[Handwritten signature]



**ADAMAS UNIVERSITY, KOLKATA SCHOOL OF
LIFE SCIENCE AND BIOTECHNOLOGY
DEPARTMENT OF BIOLOGICAL SCIENCES**

VISION OF THE DEPARTMENT

To achieve excellence in education and research on biochemistry for societal development through innovation and producing technologically sound graduates as global citizen fostering life-long learning.

MISSION STATEMENTS OF THE DEPARTMENT

M.S 01: Adopt and implement latest curriculum in biochemistry with futuristic approach and innovative pedagogy fostering knowledge, intellectual and skill development.

M.S 02: To enable and enhance skill in biochemistry sets through rigorous training and research through multidisciplinary approach.

M.S 03: To cater professional and societal need of cutting-edge research in biochemistry through collaboration and industry-academic partnership.

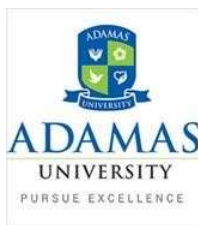
M.S 04: To inculcate values, culture along with knowledge about biochemistry to foster the spirit of self-reliance and entrepreneurship development.

Sanjay Haldar

Rudraprasad Saha

HOD

DEAN / SCHOOL CONCERNED



**ADAMAS UNIVERSITY, KOLKATA SCHOOL OF
LIFE SCIENCE AND BIOTECHNOLOGY
DEPARTMENT OF BIOLOGICAL SCIENCES**

Name of the Programme: M.Sc. in Biochemistry

PROGRAMME EDUCATIONAL OBJECTIVES (PEO)

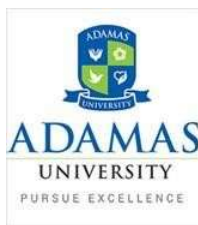
- PEO01** : Ability to do research, comprehend fundamentals and expertise in the domain.
- PEO 02** : Acquainted with modern tools and technology related to the field of
- PEO 03** : Ability to find routes of solution of existing scientific problems of the domain
- PEO04** : Develop as professional aspirants and sustainable learners.
- PEO05**

Srijan Haldar

Rudraprasad Saha

HOD

DEAN / SCHOOL CONCERNED



**ADAMAS UNIVERSITY, KOLKATA SCHOOL OF
LIFE SCIENCE AND BIOTECHNOLOGY
DEPARTMENT OF BIOLOGICAL SCIENCES**

Name of the Programme: M.Sc. Biochemistry

GRADUATE ATTRIBUTE / PROGRAMME OUTCOME (PO)

GA 01/ PO 01: Research and analysis -Develop research approaches to meet the scientific gaps on biochemistry and allied interdisciplinary or multidisciplinary fields.

GA 02/ PO 02: Academic excellence -Foster the knowledge and skills in biochemistry to identify and approach towards suitable solution.

GA 03/ PO 03: Data mining -Ability to salvage significant biological data for meaningful solution.

GA 04/ PO 04: Skill- Develop skill set related to biochemistry and allied fields

GA 05/ PO 05: Modern tools uses- Familiarized with latest and advanced tools and techniques of biochemistry.

GA 06/ PO 06: Development of solution- Investigate an existing problem to find a suitable solution, beneficial to the society.

GA 07/ PO 07: Diversity- Strong basic knowledge to support diversification in applied field of biochemistry.

GA 08/ PO 08: Professional Development- Ability to set career and professional goals based on a proper career planning process.

GA 09/ PO 09: Collaboration- Develop capacity to uphold integrity and collaborative approach in workplace.

GA 10/ PO 10: Sustainable Learning- To accept and implement changes in learning towards a sustainable development through learn, unlearn-relearn approach.

GA 11/ PO 11: Ethics- Practice ethical philosophies and systems in creating and partnering a progressive society.

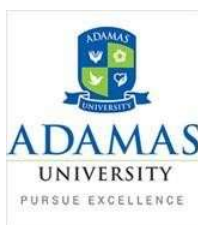
GA 12/ PO 12: Global perspectives-Develop as global citizen to contribute in the greater benefits of humanity.

Sanjay Haldar

HOD

Rudraprasad Saha

DEAN / SCHOOL CONCERNED



**ADAMAS UNIVERSITY, KOLKATA SCHOOL OF
LIFE SCIENCE AND BIOTECHNOLOGY
DEPARTMENT OF BIOLOGICAL SCIENCES**

Name of the Programme: M.Sc in Biochemistry

PROGRAMME SPECIFIC OUTCOME (PSO)

PSO 01: Students will be able to analyse the consequence of key principles of biochemistry in the living organisms at cellular and molecular level for supporting the life.

PSO 02: Students will be able to design the experimental work, interpret the data and present their work through oral and visual presentation.

PSO 03: Students will be able to focus on a scientific problem and to develop an original research proposal on it and also able produce substantial original research of significance and quality sufficient for publication and patent.

PSO 04: Students will be able to develop skills for establish themselves in the field like pclinical and forensic research labs, agriculture, pharmaceutical and biotechnology industries.

Sourajit Halder

Rudraprasad Saha

Semester-I							
Type of the Course	New Course Code	Course Name	Contact Hours Per Week	L	T	P	Credit
CORE (Theory)	BIC21501	Biomolecules and Biomolecular Interaction	3	3	0	0	3
CORE (Theory)	BIC21503	Biophysical Chemistry and Bio-analytical Techniques	3	3	0	0	3
CORE (Theory)	BIC21528	Plant Biochemistry	3	3	0	0	3
CORE (Theory)	BIC21507	Enzymology and Physiological Biochemistry	3	3	0	0	3
CORE (Theory)	BIC21541	Ecology and Evolution	3				3
CORE (Practical)	BIC22542	Biophysical Chemistry and Bio-analytical Techniques Lab	4	0	0	4	2
CORE (Practical)	BIC22557	Enzymology and plant biochemistry Lab	4	0	0	4	2
CORE (Theory)	BIC21534	Bioethics and Intellectual Property Rights	3	3	0	0	3
Foundation	BIC22570	Professional Development Course 1	1	0	0	1	1
Total			23	12	1	9	23

Semester-II							
Type of the Course	New Course Code	Course Name	Contact Hours Per Week	L	T	P	Credit
CORE (Theory)	BIC21511	Molecular Biology	3	3	0	0	3
CORE (Theory)	BIC21512	Recombinant DNA Technology	3	3	0	0	3
CORE (Theory)	BIC21543	Bioinformatics and Biostatistics	3	3	0	0	3
CORE (Theory)	BIC21544	Genomics and Proteomics	3	3	0	0	3
CORE (Practical)	BIC22545	Molecular Biology & Recombinant DNA Technology Lab	4	0	0	3	2
CORE (Practical)	BIC22546	Bioinformatics and Biostatistics Lab	4	0	0	3	2
CORE (Practical)	BIC22547	Genomic and Proteomics Lab	4	0	0	3	2
CORE (Theory) Discipline Specific Elective-I	BIC21517 BIC21518 / BIC21509 /BI C21 510	Any One of the following*: Cancer Biology / Nano biotechnology / Drug Design and Development/ Food and Dairy: Food Safety and Quality Control	3	3	0	0	3
Foundation	BIC22571	Professional Development Course 2	1	0	0	1	1
Total			27	15	0	10	22

Semester-III							
Type of the Course	Course Code	Course Name	Contact Hours Per Week	L	T	P	Credit
CORE (Theory)	BIC21522	Immunology	3	3	0	0	3
CORE (Theory)	BIC21536	Microbiology	3	3	0	0	3
CORE (Theory)	BIC21550 / BIC21551	Forensic Biology#/ Nutrition and toxicology*	3	3	0	0	3
CORE (Theory)	BIC21513	Bioenergetics and Metabolism	3	3	0	0	3
CORE (Practical)	BIC22523	Immunology Lab	4	0	0	3	2
CORE (Practical)	BIC22552 / BIC22553	Forensic Biology Lab#/ Nutrition and toxicology lab*	4	0	0	3	2
CORE (Practical)	BIC22549	Microbiology Lab	4	0	0	3	2
CORE (Theory) Discipline Specific Elective-II	BIC21554 /BIC21555/ BIC21532 / BIC21556 /BIC21558/ BIC24530	Any One of the following*: Applied toxicology*/ Environmental toxicology*/ Clinical Biochemistry / Advanced DNA Forensics#/ Advanced Forensic Chemistry#/ Research	3	3	0	0	3
FOUNDATION	BIC24535	Industry Internship	0	0	0	2	2
Foundation	BIC22572	Professional Development Course 3	1	0	0	1	1
Total			27	15	1	12	24

*Nutrition and Toxicology Specialization
#Forensic Biology Specialization

Type of the Paper	New Course Code	Theory/Practical	Contact Hours Per Week	L	T	P	Credit
CORE (Theory)	BIC25539	Comprehensive Viva		3	0	0	4
CORE (Theory)	BIC25540	Dissertation	15	0	0	15	15
Total			21	3		15	19

*Offering of DSE subjects will vary from year to year, subject to the availability of faculty

Total credit distribution semester -wise:

Semester	I	II	III	IV	Total
Credits	23	22	24	19	88

NB: 1 credit ~ 15 contact hours

BIC21501	Biomolecules & Biomolecular interaction (THEORY)	L	T	P	C
Version 1.0	Contact Hours - 45	3	0	0	3
Pre-requisites/Exposure	UG level knowledge of Biochemistry and Cell Biology				
Co-requisites	--				

Course Objectives

The cells of living organisms encompass thousands of biomolecules. From this course the students will identify the structure-function relationship of these biomolecules, and their importance with respect to maintenance and perpetuation of the living systems.

Course Outcomes

On completion of this course, the students will be able to

- CO1. Remembering-Recall the various types of weak interactions between the biomolecules and water.
- CO 2. Understanding-Comprehend how the polysaccharides and complex carbohydrates are made from the simple precursors.
- CO 3. Applying-Apply the structure-function relationships of the proteins and utilization of different techniques for elucidation of protein structure.
- CO 4. Analyzing-Describe a sequencing techniques and principle of NGS, and Biosynthesis of purine and pyrimidine.
- CO 5. Evaluate-Develop the concept of lipidomics and the processes of fatty acid oxidation and cholesterol biosynthesis to relate various interrelated physiological and metabolic events.

Catalog Description

The core-course of 'Biomolecules & Biomolecular interaction' will help to understand the structure and function of biomolecules: synthesis and properties of cellular macromolecules, basic properties of enzymes, principles of metabolism, bioenergetics, signal transduction, regulation of gene expression and function of biomolecules in cell structure and differentiation. All the lectures will be devoted on discussions of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as power point presentation, audio visual virtual lab session as per requirement. Students will perceive the basic concepts of the subject via exercise and discussions with the coordinator.

Course Content

Biomolecules & Biomolecular interaction (BIC 21501)

Unit 1 Bonding and interactions: Structure of atoms, molecules and chemical bonds, Stabilizing interactions (Van der Waals, electrostatic, hydrogen bonding, hydrophobic interaction etc.).

Unit 2 Carbohydrate: Classification, structure, general properties and functions of polysaccharides and complex carbohydrates; amino sugars, proteoglycans, glycoproteins and its significance. Hexose metabolism: pathways and energy metabolism. Metabolic labelling and glycomics.

Unit 3 Amino acids and Proteins: Structure and functional group properties, peptides and covalent structure of proteins, elucidation of primary and higher order structures, Ramachandran Plot, evolution of protein structure. Protein folding and its kinetics, chaperones and folding pathways, Structure determination using X-ray crystallography, CD- ORD, NMR and CryoEM. Overview of amino acid biosynthesis. Techniques and concepts in proteomics: LC-MS/MS and peptide mass fingerprinting.

Unit 4 Nucleic acids: Nucleic acids as genetic information carriers, Structure and function of various orders of nuclei acid organizations: forms and conformations, Sequencing techniques and principle of NGS. Denaturation of DNA. Biosynthesis of purine and pyrimidine.

Unit 5 Lipids: Classification, structure, properties and functions of fatty acids, essential fatty acids, fats, phospholipids, sphingolipids, cerebrosides, steroids, bile acids, prostaglandins, glycolipids. Fatty acid oxidation and cholesterol biosynthesis. Biosynthesis of saturated & unsaturated fatty acids and cholesterol. Lipidomics: sample preparation and analysis.

Textbook:

1. Nelson, D.L.; Cox, M.M. *Lehninger principles of biochemistry*. W.H. Freeman:2013.

Reference books:

- Campbell, MK (2012) *Biochemistry*, 7th ed., Published by Cengage Learning
- Campbell, PN and Smith AD (2011) *Biochemistry Illustrated*, 4th ed., Published by Churchill Livingstone
- Tymoczko JL, Berg JM and Stryer L (2012) *Biochemistry: A short course*, 2nd ed., W.H. Freeman
- Berg JM, Tymoczko JL and Stryer L (2011) *Biochemistry*, W.H. Freeman and Company Willey
- MJ, Sherwood, LM & Woolverton C J (2013) *Prescott, Harley and Klein's Microbiology* by. 9th Ed., McGraw Hill
- Voet, D. and Voet J.G (2004) *Biochemistry* 3rd edition, John Wiley and Sons
- *Biochemistry* by Jeremy M. Berg, John L. Tymoczko, Lubert Stryer, 2007
- *Fundamentals of Biochemistry: Life at the Molecular Level*, 4th Edition: Life at the Molecular Level by Voet, 2012
- *Biochemistry* by Lubert Stryer (8th Ed) 2015

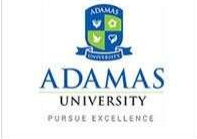
**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination
Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
Avg	3	3	2	3	3	-	-	3	-	-	1	3

Model Question Paper

Name:			
Enrolment No:			
Course: SBC21501 - BIOMOLECULES & BIOMOLECULAR INTERACTION(THEORY) Program:M.Sc.Biochemistry Time: 03Hrs. Semester:Odd 2020-21 Max. Marks:40			
Instructions: Attempt any four questions from Section A (each carrying 5 marks); any two questions from Section B (each carrying 10 marks).			
SECTION A (Attempt any Four questions) (5X4=20) (5X4=20)			
1.	Analyze the role of water in biological processes.	An	CO1
2.	Explain why is cellulose insoluble, while starch, which appears to have a very similar structure, is soluble? Identify the tools used to characterize the glycome.	U	CO2
3.	Illustrate why do proteins fold? What is peptide mass fingerprinting?	R	CO3
4.	Describe the principle of Next Generation Sequencing (NGS) technology.	U	CO4
5.	Develop a mass spectrometry-based lipid analysis protocol.	AP	CO5
SECTION B (Attempt any Two questions) (10X2=20)			
6.	What is the role of chaperones in protein folding? How do chaperones recognize unfolded proteins? Illustrate one of the pathways of chaperone-mediated protein folding in the cytosol. 3+2+5	U	CO3
7.	A sugar(C ₆ H ₁₀ O ₅) was treated by a method that reduces aldehyde groups and gave a product that was optically inactive. Assuming the sugar was D, identify the two possible structures of the product? Analyze the role of non- covalent interactions for determining the folding rate of two- state proteins. Explain regulation of glucokinaseactivity by glucokinase regulatoryprotein. 2+4+4	U,AN	CO1 CO2
8.	What is allowed region in Ramachandran plot? Which amino acid residue can occupy the greatest area in a Ramachandran plot? Identifythe purpose of a Ramachandran plot. Illustrate the role of glycoprotein in cell membrane. 2+1+3+4	AN,AP, U	CO3

9	<p>Which amino acid is required for both purine and pyrimidine synthesis? How much ATP is used in purine synthesis? Describe that small local variations in B-form DNA lead to a large variety of global geometries which can accommodate most DNA-binding protein motifs. Describe the steps of oxidation of odd-chain fatty acids.</p> <p style="text-align: center;">1+1+5+3</p>	AN,AP, U	CO4 CO5
---	---	-----------------	--------------------

BIC21503	Biophysical Chemistry & Bioanalytical Techniques (THEORY)	L	T	P	C
Version 1.0	Contact hours = 45	3	0	0	3
Pre-requisites/Exposure	BSc. level Biochemistry knowledge				
Co-requisites	-				

Course Objectives

To develop the skills of the application of basic and advanced techniques employed in quantitative and qualitative analysis of biomolecules. To be able to communicate and discuss the various methods available to purify and characterize biological molecules based on their physical and chemical properties. To be able to choose from the various methods available for purifying and characterizing biological molecules based on their physical and chemical properties. To provide scientific understanding of analytical techniques and detail interpretation of results.

Course Outcomes

On completion of this course, the students will be able to

- CO1. Comprehension of Thermodynamic Principles in Biological Systems:**
Students will be able to apply the laws of thermodynamics to biological systems, understanding how free energy, enthalpy, and entropy drive biological processes and membrane dynamics.
- CO2. Mastery of Water's Physicochemical Properties:**
Students will gain insights into the unique properties of water, including its ionic product, pH, and buffering capacity, and will understand their impact on enzyme catalysis and biological reactions.
- CO3. Proficiency in Quantum Chemistry and Spectroscopy:**
Students will be able to explain the fundamental concepts of quantum chemistry and electromagnetic radiation and apply techniques such as UV-visible, fluorescence, and IR spectroscopy to analyze biomolecules.
- CO4. Expertise in Advanced Separation Techniques:**
Students will develop a solid understanding of various separation techniques, including centrifugation, chromatography (TLC, HPLC, FPLC, and affinity chromatography), and ion-exchange chromatography, and apply these in experimental setups for biomolecular purification.
- CO5. Application of Radioactivity and Tracer Techniques in Biological Studies:**
Students will learn the principles of radioactivity and tracer techniques, mastering the use of radiotracers, isotope dilution, and radioimmunoassay in metabolic and distribution studies within biological systems.

Catalog Description

This course contains bioanalytical techniques along with their theory, working principle, common instrumentation and possible applications. This course will be equally beneficial to various scientific areas including, life science, chemical science, material science and environmental science. The information presented in this course will provide the student with valuable insight into the characterization and separation of biological macromolecules. By the end of this course, the student should be able to choose the correct method or combination of methods to characterize and separate biological macromolecules based on the physical and chemical properties of the molecules.

Course Content

Unit 1 (10 lecture hours).

Thermodynamics equilibrium; Laws of Thermodynamics and its application in biological processes; Concepts of enthalpy, entropy and free energy. Gibb's free energy; Bioenergetics; Application of thermodynamics in coupled reactions and biological systems.

Unit 2 (6 lecture hours).

Electromagnetic spectrum and transition energies. Quantum mechanical postulates, Operators, Eigenvalue and Eigenfunction, Schrodinger Equation, Set up of Hamiltonian in Particle in box, simple harmonic oscillator: energy quantization and wave functions, Approximate methods, Central concepts in spectroscopy. Scattering absorption and dispersion.

Unit 3 (8 lecture hours)

Spectroscopy I: Concept of electromagnetic radiations - UV, visible, IR, microwave region. Molecular Orbital theory: Bonding and antibonding; UV Visible Absorption Spectroscopy, Fluorescence Spectroscopy: Determination of Quantum yield of Fluorescence of a fluorophore and a protein, Emission of Protein, Quenching of emission of protein by acrylmide and KI, Protein Ligand Interaction.

Unit 4 (4 lecture hours).

Spectroscopy II Magnetic Resonance Spectroscopy, Basic principles and instrumentation in NMR Spectroscopy, Application to structure of biomolecules; Basics of ESR Spectroscopy and Application.

Unit 5 (4 lecture hours).

Method of conformational analysis and prediction of conformation: Structure determination using Circular Dichroism, Spectroscopy, X-ray diffraction.

Unit 6 (3 lecture hours).

Centrifugation: Principle of centrifugation and different types of centrifuge. Differential & density gradient centrifugation.

Unit 7 (7 lecture hours).

Chromatography Techniques—TLC, HPLC, HPTLC & FPLC, Size-exclusion Chromatography, Determination of void volume, Determination partition coefficient, Separation of two components in a sample. Affinity chromatography, Ion-exchange Chromatography. Nature of exchanger, capacity of column, Separation of amino acids.

Unit 8 (3 lecture hours).

Radioactivity--: Radioactive & stable isotopes; Units of radioactivity; Measurement of radioactivity; Measurement of stable isotopes; Falling drop method; Radiotracer techniques; Distribution studies; Isotope dilution technique.

Reference Books:

1. Lakowicz, J. R. (2006) Principles of Fluorescence Spectroscopy. 3rd edition. Springer.
2. Physical Chemistry for the Life Sciences. Peter Atkins, Julio de Paula by Peter Atkins, Julio de Paula, 2011

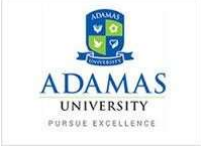
**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination
Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
Avg	3	3	2	3	3	-	-	3	-	-	1	3

Model Question Paper

Name:			
Enrolment No:			
Course: BIC21503 - Biophysical Chemistry & Bioanalytical Techniques (THEORY) Program: M.Sc. Biochemistry Time: 03 Hrs. Semester: Odd 2020-21			
		Max. Marks: 50	
Instructions: Attempt any four questions from Section A (each carrying 5 marks); any two questions from Section B (each carrying 10marks).			
SECTION A (Attempt any Four questions) (5X4=20) (5X4=20)			
1.	Explain two different models of enzyme action.	R	CO3
2.	Describe (i) Reaction orders and (ii) Carnot Engine.	U	CO1
3.	(i) Explain the Rate Law. (ii) Describe the factors that influence reaction rate	R	CO2
4.	(i) Explain Arrhenius equation? (ii) Interpret Transition State theory.	AP	CO2
5	At 1000°C, cyclobutane (C ₄ H ₈) decomposes in a first-order reaction, with the very high rate constant of 87 s ⁻¹ , to two molecules of ethylene (C ₂ H ₄). (i) If the initial C ₄ H ₈ concentration is 2.00M, find out the concentration after 0.010s? (ii) Find the fraction of C ₄ H ₈ that has decomposed in this time?	R,U,AN	CO5
SECTION B (Attempt any Three questions) (10X3=30)			
6.	Draw and explain Perrin-Jablonski diagram of fluorescence and phosphorescence.	R	CO3
7.	Describe the important characteristics of fluorophores? Explain quantum yield? Explain intrinsic fluorescence of proteins and peptides.	U	CO2 CO3
8.	Explain the electrophoresis process? Illustrate the differences between SDS and non-SDS electrophoresis.	R	CO1 CO5
9	Explain the mechanism of gel-exclusion chromatography. Describe how one can find out molecular weight of an unknown protein using gel-exclusion chromatography.	U,R	CO4

BIC21507	Enzymology and Physiological Biochemistry	L	T	P	C
Version 1.0	Contact Hours -45	3	0	0	3
Pre-requisites/Exposure	BSc. level Biology knowledge				
Co-requisites	-				

Course Objectives

1. To **classify** the enzymes according to the basis of their catalysed reactions.
2. To **analyse and evaluate** the kinetic behaviour of enzymes
3. To **develop** the concept and **determine** about different patterns of inhibitions of enzyme activity.
4. Will **build** the concept about the structures of active site of the enzymes and their mechanism of actions and their clinical application.
5. To **develop** the idea about regulation of enzyme activity.

Course Outcomes

On completion of this course, the students will be able to

CO1 Remembering-**Recall** the enzymes according to the basis of their catalyzed reactions.

CO2: Understanding-**Comprehend** the kinetic behaviour of enzymes.

CO3: Applying- **Apply** the concept and **determine** about different patterns of inhibitions of enzyme activity.

CO4: Analysing-**Build** the concept about the structures of active site of the enzymes and their mechanism of actions and their clinical application.

CO5 Evaluate-**Develop** the idea about regulation of enzyme activity.

Catalog Description

Nomenclature and classification of enzymes Holoenzyme, apoenzyme, cofactors, coenzyme, prosthetic groups, metallo enzymes, monomeric and oligomeric enzymes Activation energy and transition state theory, enzyme activity, specific activity, common features of active sites, enzyme specificity: types and theories Factors affecting enzyme activity, E, S, temp and pH Enzyme substrate complex: Concept of E-S complex, binding sites, active site, specificity, kinetics of enzyme activity Michaelis-Menten equation and its derivation Different plots for the determination of K_M and V_{max} and their physiological significance Two substrate reactions (random, ordered and ping pong mechanisms), enzyme inhibition, types of inhibition, determination of K_i , suicide inhibitor.

Course Content

1. Enzyme classification, isoenzymes, multienzyme; factors affecting enzyme activities; feedback and allosteric inhibition. Purification and characterization of enzymes. Single enzymes (end product inhibition) and metabolic pathways, feedback inhibition (aspartate transcarbamoylase).
[6 hours lecture] E
2. Enzyme kinetics: One substrate reactions, effect of pH, temperature and inhibitions. Two substrate reactions. Theory, order analysis, pre-steady state kinetics, stopped flow technique, Relaxation methods.
[7 hours lecture] M
3. Mechanism of enzyme action: Theoretical background, Factors leading to rate enhancement of enzyme catalyzed reactions: Acid-base catalysis, proximity and orientation effects, covalent catalysis, strain or distortion and change in environment. Experimental approaches of determination of enzyme mechanism: Kinetics studies, detection of intermediates, X-ray crystallographic studies, Chemical modification of amino acid side chain and affinity labeling, site directed mutagenesis. Examples of chymotrypsin, triose phosphate isomerases, aldolase etc.
[7 hours lecture] Z
4. Control of enzyme activity: Control of activities of single enzyme: inhibitor molecules, availability of substrate or cofactor. Product inhibition. Control by changes in covalent structure of enzymes.
[7 hours lecture] Z
5. Enzyme activation and phosphorylation dephosphorylation ligand induced changes: Allosteric enzymes, Theoretical models, Hill equation, Adair equation, M.W.C. and K.N.F. Models, usefulness of the models. Significance of allosteric and cooperative behavior in enzymes.
[7 hours lecture] C
6. Control of metabolic pathways: Amplification of signals, substrate cycles and Interconvertible enzyme cycles.
7. Multienzyme complex: Properties, pyruvate dehydrogenase system, (*E. COLI* and mammalian), Tryptophan synthetase, multienzyme complex from *E. coli*, fatty acid synthetase, glycogen particle.
[4 hours lecture] E
8. Enzyme turnover: Kinetics of enzyme turnover. Measurement of enzyme turnover, K_s and K_d . Correlation between the rates of enzyme turnover and structure and function of enzymes. Mechanism of enzyme degradation. Significance of enzyme turnover. [3 hours lecture]
9. Clinical aspects of enzymology: LDH isozymes, SGOT, SGPT, creatine kinase, alpha amylase, phosphatase, inborn errors.
[4 hours lecture]

Reference Books

1. Lehninger: Principles of Biochemistry (2013) 6th ed., Nelson, D.L. and Cox, M.M., W.H. Freeman and Company (New York), ISBN:13: 978-1-4641-0962-1 / ISBN:10:1-42923414-8.
2. Textbook of Biochemistry with Clinical Correlations (2011) 7th ed., Devlin, T.M., John Wiley & Sons, Inc. (New York), ISBN:978-0-470-28173-4.

Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination
Examination Scheme:

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

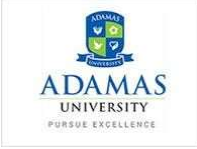
CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
Avg	3	3	2	3	3	-	-	3	-	-	1	3

1=weakly mapped

2= moderately mapped

3=strongly mapped

Model Question Paper

Name: Enrolment No:		 ADAMAS UNIVERSITY PURSUE EXCELLENCE	
Course: BIC21504 Enzymology Program: M.SCBiochemistry Semester: Even2019-20		Time: 03Hrs. Max. Marks:50	
Instructions: Attempt any four questions from Section A (each carrying 5 marks); any two Questions from Section B (each carrying 10 marks).			
Section A (Attempt any FOUR) (5X4=20)			
1	What is active site of enzyme? Give its function?()	U	CO2
2	What is allosteric regulation?	R	CO3
3	What are the difference between sequential and symmetry model of allosteric regulation	R	CO1
4	What are the advantages of allosteric regulation?	U	CO3
5	Why an enzyme having an allosteric regulation show a sigmoid curve instead of a regular hyperbolic curve?	AN, U	CO4
Section B (3X10=30)			
6	Which of these two cases is allosteric regulation? i. 'Phosphorylation of an amino acidsomewhere other than the active site' ii. 'The non-covalent binding of cAMP somewhere other than the activesite'	AN, R	CO3
7	Differentiate between apoenzyme and holoenzyme. What is induced fit Model? Give its significance.	AP, U	CO5
8	What is activation energy? How is it lowered? Explain the limitation of key and lock model.	U,R	CO2

BIC21528	PLANT BIOCHEMISTRY (THEORY)	L	T	P	C
Version 1.0	Contact Hours - 45	3	0	0	3
Pre-requisites/Exposure	FUNDAMENTALS OF PLANT BIOCHEMISTRY FROM GRADUATION				
Co-requisites	--				

Course Objectives

- To provide students the basic understanding of plant cell structure with emphasis to some special organelles.
- To provide wholesome knowledge on plant specific biochemical pathways like photosynthesis and nitrogen metabolism.
- Elaborating roles of phytohormones and secondary metabolites in growth and development of plants.
- General overview of plant tissue culture.

Course Outcomes

On completion of this course, the students will be able to

1. Knowledge: Define and explain the basic principles of plant biochemistry including photosynthesis, respiration, and nitrogen metabolism.
2. Comprehension: Interpret and explain the impact of various abiotic and biotic stressors on plant physiology and biochemistry.
3. Application: Apply knowledge of plant transport processes to describe how nutrients and other molecules are transported within plants.
4. Analysis: Analyze the molecular mechanisms underlying plant stress responses at the biochemical level.
5. Evaluation: Critically evaluate the current research on plant stress responses and propose future directions for investigation.

Catalog Description

The core-course of 'plant biochemistry' deals with the modern aspects of plant biochemistry. This course deals with plant cellular structure with emphasis to special organelles related to plant cells. It also includes topics related to plant specific biochemical pathways like photosynthesis, respiration and nitrogen fixation. Furthermore, it deals with the roles of phyto-hormones and secondary metabolites in plant growth and development. It also encompasses the very important industrially important plant tissue culture technique. All the lectures will be devoted on discussions of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as power point presentation, audio visual virtual lab session as per requirement. The tutorials will enable the students with problem-solving ability led by the course coordinator. Students will perceive the basic concepts of the subject via exercise and discussions with the coordinator.

Course Content

Course Content

- 1) Stress Physiology: Plant responses to abiotic & biotic stresses.
- 2) Secondary metabolites: Biosynthesis of nitrogenous compounds, phenols, and terpenes & their roles.
- 3) Solute Transport & Photoassimilate Translocation: Mechanisms of unloading & loading of photoassimilates, transpiration, translocation, transport & uptake of macromolecules, solutes & ions via phloem & xylem, across membranes, and through cells.
- 4) Sensory Photobiology: Biological clocks, photoperiodism, stomatal movement, mechanism of action, function & structure of phytochromes, phototropin & cryptochromes.
- 5) Plant Hormones: Mechanism of action & physiological effects, transport & breakdown, storage, biosynthesis.
- 6) Nitrogen Metabolism: Amino acid biosynthesis, ammonium & nitrate assimilation.
- 7) Respiration & Photorespiration: Photorespiratory pathway, alternate oxidase, ATP synthesis & plant mitochondrial electron transport, citric acid cycle.
- 8) Photosynthesis: CO₂ fixation-CAM, C₄, and C₃ pathways, photoprotective mechanisms, mechanisms of electron transport, light-harvesting complexes.

REFERENCE BOOKS:

1. Plant Biochemistry (2008), Caroline Bowsher, Martin Steer, Alyson Tobin, Garland Science ISBN 978-0-8153-4121-5
2. Biochemistry and molecular Biology of plant-Buchanan. (2005) 1 edition. Publisher: I K International. ISBN-10: 8188237116, ISBN-13: 978-8188237111.
3. Plant Biochemistry by P.M Dey and J.B. Harborne (Editors) (1997) Publisher: Academic Press ISBN-10: 0122146743, ISBN-13: 978-0122146749

Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination
Examination Scheme:

Components	Class Assessment	End Term
Weightage (%)	50	50

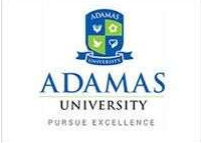
Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
Avg	3	3	2	3	3	-	-	3	-	-	1	3

1=weakly mapped

2= moderately mapped

3=strongly mapped

Name:			
Enrolment No:			
Course: BIC21528 – PLANT BIOCHEMISTRY (THEORY) Program: M.Sc. Biochemistry Semester: Odd 2019-20			
		Time: 03Hrs.	Max. Marks:50
Instructions: Attempt any four questions from Section A (each carrying 5 marks); any two questions from Section B (each carrying 10 marks).			
SECTION A (Attempt any Four questions) (5X4=20) (5X4=20)			
1.	What are peroxisomes? Explain its role in plant cell.	U,R	CO1
2.	Summarize the light reactions in photosynthesis. What is cyclic photophosphorylation?	U,R	CO2
3.	What are the genes involved in nitrogen fixation in plants? Summarize nitrogen uptake and assimilation in plant cells briefly.	U,R	CO3
4.	Compare cell suspension culture and protoplast culture. What is totipotency?	U,R	CO5
5.	Explain the molecular pathway of auxin response. What is the precursor of auxin?	U,R	CO4
SECTION B (Attempt any 3 questions) (10X3=30)			
6.	Illustrate the technique of post-harvest technology with its potential application. How ethylene regulates fruit ripening? What are the role of GA class of hormones? Which one is called stress hormone?	AP, U,R	CO4
7.	Explain dark reactions in photosynthesis. Describe the CAM pathway in detail. Compare between C3 & C4 pathway.	AN, U	CO2,CO3
8.	Illustrate the role of nitrogenase complex in nitrogen fixation. What is the role of vacuoles in plant cell? What is somatic embryogenesis and somaclonal variation? Mention its significance.	AP, R	CO1 CO3 CO5

BIC21541	Ecology and Evolution (THEORY)	L	T	P	C
Version 1.0	Contact Hours - 45	3	0	0	3
Pre-requisites/Exposure	UG LEVEL BIOLOGY				
Co-requisites	--				

Course Objectives

1. Provide students with the scope to develop knowledge base covering all attributes of the environment and enable them to attain scientific/technological capabilities to find answers to the fundamental questions before the society with regards to human action and environmental effects with due diligence.
2. Enhance the ability to apply this knowledge and proficiency to find solutions relating to environmental and ecological concerns of varied dimensions of present times through research activities.
3. Provide with a direction and technical capability to carry on collaborative endeavour, and decisionmaking.
4. Help graduates appreciate requirement of framing environmental policy guidelines.
5. Motivate graduates to appreciate that they are an integral stakeholder in the environmental management of India irrespective of their future jobs or working.

Course Outcomes

Upon completion of this course, students will be able to:

1. Remember: Recall and identify key concepts and terms related to ecology and evolution such as population dynamics, succession, evolutionary genetics, theories of evolution, and population genetics.
2. Understand: Explain the fundamental principles of ecology and evolution, including the mechanisms of natural selection, genetic drift, and gene flow.
3. Apply: Apply ecological and evolutionary concepts to analyze and interpret data related to population dynamics, genetic variation, and evolutionary processes.
4. Analyze: Critically evaluate scientific literature on ecology and evolution to draw conclusions and generate hypotheses related to population dynamics and evolutionary patterns.
5. Create: Design and conduct research projects to investigate ecological and evolutionary patterns in natural populations.

Catalog Description

This course covers ecological and evolutionary principles on population, community, ecosystem and biodiversity. The very nature of ecology and evolution requires students to view role of evolutionary process on modern human life. All the lectures will be devoted on discussions of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as power point presentation, audio visual virtual lab session as per requirement. The tutorials will familiarize the students with practical problem-solving techniques led by the course coordinator. Students will strongly grab the basic concepts of the subject via exercise and discussions with the coordinator.

Course Content

DIVERSITY OF LIFE FORMS AND ECOLOGICAL PRINCIPLES

3 Lecture Hours

1. **Principles and methods of taxonomy:** Concepts of species and hierarchical taxa, biological nomenclature, classical and quantitative methods of taxonomy of plants, animals and microorganisms.

3 Lecture Hours

2. **Levels of structural organization:** Unicellular, colonial and multicellular forms; levels of organization of tissues, organs and systems; comparative anatomy.

3 Lecture Hours

3. **Outline classification of plants, animals and microorganisms:** Important criteria used for classification in each taxon; classification of plants, animals and microorganisms; evolutionary relationships among taxa.

3 Lecture Hours

4. **Natural history of Indian subcontinent:** Major habitat types of the subcontinent, geographic origins and migrations of species; common Indian mammals, birds; seasonality and phenology of the subcontinent.

3 Lecture Hours

5. **Organisms of health and agricultural importance:** Common parasites and pathogens of humans, domestic animals and crops.

3 Lecture Hours

6. **The Environment:** Physical environment; biotic environment; biotic and abiotic interactions.

3 Lecture Hours

7. **Habitat and niche:** Concept of habitat and niche; niche width and overlap; fundamental and realized niche; resource partitioning; character displacement.

4 Lecture Hours

8. **Population ecology:** Characteristics of a population; population growth curves; population regulation; life history strategies (*r* and *K* selection); concept of metapopulation – demes and dispersal, interdemographic extinctions, age structured populations.

4 Lecture Hours

9. **Species interactions:** Types of interactions, interspecific competition, herbivory, carnivory, pollination, symbiosis.

3 Lecture Hours

10. Community ecology: Nature of communities; community structure and attributes; levels of species diversity and its measurement; edges and ecotones.

3 Lecture Hours

11. Ecological succession: Types; mechanisms; changes involved in succession; concept of climax.

4 Lecture Hours

12. Ecosystem: Structure and function; energy flow and mineral cycling (CNP); primary production and decomposition; structure and function of some Indian ecosystems: terrestrial (forest, grassland) and aquatic (fresh water, marine, estuarine).

3 Lecture Hours

13. Biogeography: Major terrestrial biomes; theory of island biogeography; biogeographical zones of India.

1

3 Lecture Hours

14. Conservation biology: Principles of conservation, major approaches to management, Indian case studies on conservation/management strategy (Project Tiger, Biosphere reserves).

Reference books:

1. Diversity of Life: The Five Kingdoms by Lynn Margulis, 1992
2. The Diversity of Living Organisms by Richard Stephen Kent Barnes, 2009
3. Ecology by Michael L. Cain, William D. Bowman, 2008
4. Fundamentals of Ecology by Odum and Barrett, 2005
5. Biodiversity: an introduction by Kevin J. Gaston, 2004

Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:

Components	Class Assessment	End Term
Weightage (%)	50	50

Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

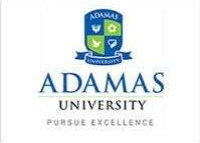
CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
Avg	3	3	2	3	3	-	-	3	-	-	1	3

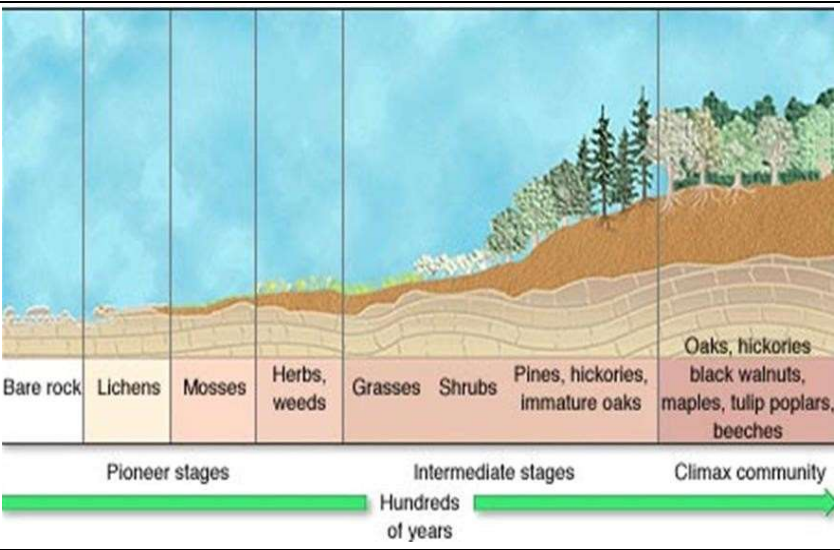
1=weakly mapped

2= moderately mapped

3=strongly mapped

Model Question Paper

Name:			
Enrolment No:			
Course: BIC21535 – DIVERSITY OF LIFE FORMS AND ECOLOGICAL PRINCIPLES Program: M.Sc. Biochemistry Semester: Odd2020-21			
		Time: 03Hrs. Max. Marks: 40	
Instructions: Attempt any three questions from Section A (each carrying 4 marks); any four questions from Section B (each carrying 7 marks).			
SECTION A (Attempt any Four questions) (5X4=20)			
1.	a) What is a clade? (R) b) Write the difference between α -taxonomy & β -taxonomy? (U)	U	CO5
2.	a) What are the differences between habitat and niche? (U) b) Explain it with an example. How do you explain G.F. Gause's classical experiment with two different paramecium species, <i>Paramecium aurelia</i> & <i>Paramecium caudatum</i> ? (An)	U, AN	CO1
3.	a) What are the factors that determine a species as threatened? (R) b) How can the loss of one species lead to the extinction of another? (U)	R, U	CO4
4.	Write all the categories according to level of organization of tissues in a descending order. (An)	AN	CO3
SECTION B (Attempt any Two questions) (10X2=20)			
5.	a) Suppose a metapopulation consists of 2 population patches, A & B, in which A is occupied and B is unoccupied. If 150 individuals from population patch A (total population size 600) is migrated to patch B, then evaluate what will be the colonization/recolonization co-efficient (c) of that metapopulation? Consider growth rate of that metapopulation or dP/dt is equal to 0 & extinction co-efficient (e) is 0.25. (Eva) b) Why the pyramid of energy can never be inverted? Give a suitable explanation for it. (An)	AN, C R	CO1 CO2
6.	a) Explain how species diversity of an area is reduced by the invasion of an alien species. (An) b) What are Biogeochemical Cycles of Biosphere? (R) c) State two advantages of in situ conservation. (U)	AN, R, U	CO4

7.	<p>a)Name and describe the type of succession in the following picture.(An)</p> 		<p>CO2 CO1</p>																								
8.	<p>a)The following table shows the number of individuals of each species found in two communities. Using the table calculate the species diversity, species richness and species dominance of both the communities.(Ap)</p> <p>b) Also you have to clearly mention in which community the species diversity & species richness is higher.(U)</p> <p>COMMUNITY 1 COMMUNITY 2</p> <table border="1" data-bbox="300 1207 1136 1719"> <thead> <tr> <th>Species Name</th> <th>Species Number</th> <th>Species Name</th> <th>Species Number</th> </tr> </thead> <tbody> <tr> <td>Orthoptera (Species 1)</td> <td>6</td> <td>Orthoptera (Species 1)</td> <td>25</td> </tr> <tr> <td>Orthoptera (Species 2)</td> <td>5</td> <td>Orthoptera (Species 2)</td> <td>2</td> </tr> <tr> <td>Lepidoptera</td> <td>9</td> <td>Lepidoptera</td> <td>17</td> </tr> <tr> <td>Hymenoptera</td> <td>21</td> <td>Hymenoptera</td> <td>12</td> </tr> <tr> <td>Coleoptera</td> <td>12</td> <td>Coleoptera</td> <td>5</td> </tr> </tbody> </table>	Species Name	Species Number	Species Name	Species Number	Orthoptera (Species 1)	6	Orthoptera (Species 1)	25	Orthoptera (Species 2)	5	Orthoptera (Species 2)	2	Lepidoptera	9	Lepidoptera	17	Hymenoptera	21	Hymenoptera	12	Coleoptera	12	Coleoptera	5	<p>AP,U</p>	<p>CO1 CO5</p>
Species Name	Species Number	Species Name	Species Number																								
Orthoptera (Species 1)	6	Orthoptera (Species 1)	25																								
Orthoptera (Species 2)	5	Orthoptera (Species 2)	2																								
Lepidoptera	9	Lepidoptera	17																								
Hymenoptera	21	Hymenoptera	12																								
Coleoptera	12	Coleoptera	5																								

<p>9.</p>	<p>a) Explain the following graph with respect to prey-predator relationship.(An)</p> <div style="border: 1px solid black; width: 300px; height: 100px; margin: 20px auto;"></div> <p>b) basic any</p> <p>the types of productivity and the organism responsible? (U)</p> <p>Mention the requirement for ecosystem to function and sustain. Name</p>	<p>AN,U</p>	<p>CO2 CO1</p>
-----------	---	-------------	--------------------

BIC21534	BIO-ETHICS AND INTELLECTUAL PROPERTY (THEORY)	L	T	P	C
Version 1.0	Contact Hours - 90	3	0	0	3
Pre-requisites/Exposure	Basic Knowledge of Biology, application of biotechnology and concept of innovation.				
Co-requisites	--				

Course Objectives

- To provide the students with understanding of components and process of obtaining protection using IPR.
- It will also discuss various aspects of bioethics
- To study the scope of entrepreneurship development using biotechnology and imbibe skills.

Course Outcomes

On completion of this course, Students will be able to

- CO 1 Remembering: Recall various components of IPR and/or evaluate the feasibility of an invention/ innovation to be protected through IPR.
- CO 2 Understanding: comprehend various ethical issues pertaining to biotechnological aspects.
- CO 3 Applying: classify biosafety levels.
- CO 4 Analyzing: Analyse skills for entrepreneurship through biotechnological innovation.
- CO5 Evaluate: Evaluate different aspects of biosafety and IPR.

Catalog Description

The core-course of bioethics, IPR and biological patent is a core course that discusses various concepts of IPR along with its background, history and method of obtaining them. This is a fundamental course that would help students to be aware of the legal protection of innovation and innovative products. Several bio-ethical concepts are also discussed to provide critical appraisal on various biological processes. The scope of entrepreneurship utilizing biotechnological ideas are also dealt in this course.

Course Content

BIO-ETHICS, INTELLECTUAL PROPERTY RIGHTS & BIOLOGICAL PATENT

(SBC52102)

Unit I. Intellectual Property Right (IPR)

2. Concept and provisions of IPR

Patents, Trademarks, Copyright, Conditional information, Breeder's right. Patent; importance, types, scope, criteria, applying for a patent. Protection of Biotechnological inventions. Patent infringement-meaning, scope, litigation, case studies and examples

3. Agreements and Treaties---History of GATT & TRIPS Agreement; Madrid Agreement; Hague Agreement; WIPO Treaties; Budapest Treaty; PCT

Unit II. Safety in Biotechnology

Introduction to Biological Safety Cabinets; Primary Containment for Biohazards; Biosafety Levels of Specific Microorganisms; Recommended Biosafety Levels for Infectious Agents and Infected Animals; Biosafety guidelines, Overview of Biotechnology Regulations and relevant International Agreements including Cartagena Protocol.

Unit III. Bioethics

Biotechnology information, communication and public perception, Future prospects of consumers and social acceptance. Case studies

Unit IV. Bio-entrepreneurship

Support mechanism for entrepreneurship in India; Leadership skills; Managerial skills; Team building; teamwork;. Taking decision on starting a venture; Assessment of feasibility of a given venture/new venture; Approach a bank for a loan; Sources of financial assistance; Making a business proposal/Plan for seeking loans from financial institution and Banks. Information technology for business administration, E-business setup and management.

Suggested Books:

1. The Ethics of Biotechnology by Jonathan Morris, 2005
2. Understanding Bioethics and the Law: The Promises and Perils of the Brave New World of Biotechnology by Barry R. Schaller, 2007
3. Nexus of Law and Biology: New Ethical Challenges by Barbara Ann Hocking, 2009
4. Intellectual Property and Biotechnology: Biological Inventions by Matthew Rimmer, 2008
5. An Introduction to Ethical, Safety and Intellectual Property Rights Issues in Biotechnology by [Padma Nambisan](#), 2017
6. Biotechnology Entrepreneurship by Craig Shimasaki, 2014

Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination
Examination Scheme:

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

Relationship between the Course Outcomes (COs) and Program Outcomes (POs)


CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	3	3	3	2	3	3	3	3	2	3	3	2
CO2	3	3	3	2	3	3	3	3	2	3	3	3
CO3	3	3	3	2	3	3	3	3	2	3	3	2
CO4	3	3	3	2	3	3	3	3	2	3	3	2
CO5	3	3	3	2	3	3	3	3	2	3	3	2
Avg	3	3	3	2-	3	3	3	3	2	3	3	3

1=weakly mapped

2= moderately mapped

3=strongly mapped

Model Question Paper

Name:			
Enrolment No:			
Course: – BIC21534BIO-ETHICS and INTELLECTUAL PROPERTY RIGHTS (THEORY) Program: M.Sc. Biochemistry Time: 03Hrs. Semester:Odd 2020-21 Max. Marks:50			
Instructions: Attempt any four questions from Section A (each carrying 5 marks); any two questions from Section B (each carrying 10 marks).			
SECTION A (Attempt any Four questions) (5X4=20)			
1.	Define IPR and Mention its components	U	CO1
2.	Analyze the ethical issues of using GM crops	R	CO2
3.	Which category of Biosafety is required to work with COVID:19? Mention the facilities required in such lab.	U	CO3
4.	Identify and enlist the skill-sets required to become an Entrepreneur.	U	CO4
5	Write a short role on Infringement of Patent	R	CO5
SECTION B (Attempt any 3 questions) (10X3=30)			
6.	Discuss the origin of WIPO. Why an International organization like WIPO is required? Mention the administrative components of WIPO. The logo of a company is protected through IPR: Justify and mention the benefits.	R,AP	CO1
7.	A person has invented a new method of doing non-invasive treatment of removing kidney stone in human. He had applied for patent but his patent was rejected. Justify the decision of Controller of Patents for such decision. Mention any other criterion for an invention being non-patentable. Write a note on the types of patent application. Mention the validity of a patent in terms of duration.	AP, CR	CO1,CO2
8.	Name one convention related to Biosafety of biodiversity. Describe major amendments of that convention. Mention the source of finance for a start-up and the method to approach them.	CR,AN	CO3, CO4

Course Title	Biophysical Chemistry and Bio-analytical Techniques Lab	L	T	P	C
CourseCode	BIC22542	0	0	4	2
Contact Hours	60				
Pre-requisites/Exposure	12 th level English + B.Sc Biology discipline				

Course Objectives

1. To gain a deeper understanding on analytical biochemistry techniques
2. To gain a deeper understanding on biophysical chemistry techniques

Course Outcomes

On completion of this course, the students will be able to

Catalogue Description

Familiarize students with the specific characteristics of a laboratory of analytical biochemistry & biophysical chemistry.

To know the analytical methods commonly used in the clinical laboratory. Know how can contribute the clinical laboratory to assess the health status of individuals. At the end of the course the student will know the techniques and applications of molecular biology and biochemistry. Emphasis on current techniques and structure/function relationships of biological macromolecules. This course covers the tools and techniques by which biological molecules are isolated, separated, identified, and analyzed. Detailed discussion of experimental methods for macromolecule purification and characterization is included.

The Introductory Biochemistry course covers fundamental biochemical and molecular biological laboratory techniques, supporting concepts, and data analysis. The aims of this course are 1. To provide students with practical knowledge and hands-on experience with some of the most common experimental methods used in biochemical and molecular biological research, and 2. To introduce students to the fundamentals of scientific writing. Methods include reagent preparation, proper use of instrumentation, biochemical analysis,

Course Outcomes

CO1. Remembering: Recall different aspects of biophysical chemistry and techniques.

CO2. Understanding: Comprehend different concepts of biophysical chemistry and techniques

CO3. Applying: Apply the knowledge of biophysical chemistry in different physical bioinstruments.

CO4. Analyzing: Analyze different experiments of biophysical chemistry and techniques.

CO5.Evaluate: Critically analyze data from different experiments.

Content Unit description

Topic	Contact hours
Demonstration of analytical instruments (principles and applications) available in the Department as well as in USIC of VU.	4 hrs + 4 hrs
Methods of cell breakage.	4 hrs
Estimation of total protein, carbohydrate of a bacterial cell.	4 hrs
Estimation of carbohydrate of a bacterial cell.	4 hrs
Estimation of DNA and RNA of a bacterial cell.	4 hrs
Chromatography: Paper, TLC for sugar/lipid/amino acid.	4 hrs
Determination of activity of amylase, protease. Effect of pH, temperature on enzyme activity; Enzyme kinetics.	4 hrs + 4 hrs + 4 hrs
Determination of MW of protein by PAGE.	4 hrs
Demonstration of 2D-gelelectrophoresis and Gel documentation system.	4 hrs
Techniques for purifying and characterizing Proteins and Enzymes	4 hrs
Idea of all analytical techniques like Electrophoresis, Liquid Chromatography, Column Chromatography for protein analysis.	4 hrs + 4 hrs

Books & Other Resources

Text Book(s)	
T1	Introduction To Practical Biochemistry by Plummer DT, 2006
T2	Biochemistry (Lippincott Illustrated Reviews Series) by R. Harvey
T3	Practical Physiological Chemistry: A Book Designed for Use in Courses in Practical Physiological Chemistry in Schools of Medicine and of Science (Classic Reprint) by Philip Bovier Hawk, 2017

T=Text Book

Modes of Examination:

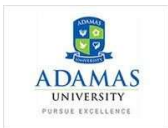
Assignment/Quiz/Project/Presentation/Written Exam Examination Scheme:

Components	Internal	Attendance	Mid Term	End Term
Weightage (%)	30	10	20	40

Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
CO1	3	3	3	2	3	3	3	3	2	3	3	2
CO2	3	3	3	2	3	3	3	3	2	3	3	3
CO3	3	3	3	2	3	3	3	3	2	3	3	2
CO4	3	3	3	2	3	3	3	3	2	3	3	2
CO5	3	3	3	2	3	3	3	3	2	3	3	2
Avg	3	3	3	2-	3	3	3	3	2	3	3	3

1=weakly mapped
2=moderately mapped
3=strongly mapped

Model Question Paper			
Name: Enrolment No:			
Course: SBC 51202 – Analytical Biochemistry Lab (PRACTICAL) Program: M.Sc. Biotechnology Semester: Odd 2020-21		Time: 03 Hrs. Max. Marks: 50	
Instructions: Attempt any three questions from Section A (each carrying 10 marks); Section B is compulsory (carrying 10 marks).			
Section A (Attempt any Two)			
1.	a) Identify the instruments shown in the picture. b) Write the operating principle and uses of this instrument. (2+4+4)	Ap/R	CO1
2.	a) Identify sample A using specific reagent provided. b) Write principle of this procedure and interpret the result (2+4+4)	Ap/U	CO3
3.	a) Identify sample B using specific reagent provided. b) Write principle of this procedure and justify the result (5+5)	Ap/Ev	CO3

4.	You want to purify two proteins with identical molecular weights. Design suitable chromatography technique to execute this process. Interpret the result. (5+5)	Cr/Ap	CO2 CO5
	SECTION B is compulsory		
5.	Viva-voce (10)	U/An/Ap/R/Ev	CO1 CO2 CO3 CO4 CO5
6.	Practical copy (10)	U/Ap/Ev	CO1 CO2 CO3 CO4

R=Remember; U=understand; Ap=Application; Ev=Evaluation; Cr=Create; An=Analysis

BIC22557	Enzymology and plant biochemistry Lab	L	T	P	C
Version 1.0	Contact Hours 60	0	0	4	2
Pre-requisites/Exposure	BSc. level Biology knowledge				
Co-requisites	-				

Course Objectives

1. To provide students with hands-on training in the field of enzymology & plant biochemistry.
2. To provide in depth knowledge of enzymology.
3. Students will become more proficient with different practical applications of plant biochemistry (e.g. plant tissue culture).

Course Outcomes

CO1. Remembering- Recall different aspects of enzymology and plant biochemistry experiments

CO2. Understanding- Comprehend background knowledge of plant enzyme assay.

CO3. Application- Apply various techniques for separating plant pigments.

CO4. Analyzing- Analyze data from plant biochemistry experiments to draw conclusions.

CO5. Evaluate- Evaluate the validity and reliability of experimental results in enzymology and plant biochemistry

Catalog Description

The discipline specific course “enzymology and plant biochemistry lab” is a practical paper which has been designed to provide the knowledge of different aspects of plant biochemistry. It will provide biochemical & molecular understanding of important physiological processes in plants. Students will be able to understand tissue culture and other techniques and will practice hands-on all of them. Students will comprehend different assay systems of plant enzymes and estimation of secondary metabolites. All the lectures will be

devoted on discussions of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as power point presentation, audio visual virtual lab session as per requirement.. Students will strongly grab the basic concepts of the subject via exercise and discussions with the coordinator.

Course Content

ENZYMOLGY AND PLANT BIOCHEMISTRY LAB

1. Induction of hydrolytic enzymes proteinases /amylases/lipase during germination. (15 Lectures)
2. Extraction and assay of Urease from Jack bean. (10 Lectures)
3. Estimation of carotene/ascorbic acid/phenols/tannins in fruits and vegetables. (10 Lectures)
4. Temp & pH dependence of enzymes. (5 Lectures)
5. Study of enzyme kinetics. (5 Lectures)
6. Culture of plants (explants). i)MS media preparation. ii)Callus culture. iii)Suspension culture. (15 Lectures)

SUGGESTED READINGS

1. Plant Biochemistry (2008), Caroline Bowsher, Martin steer, Alyson Tobin, Garland science ISBN 978-0-8153-4121-5

2. Biochemistry and molecular Biology of plant-Buchanan. (2005) 1 edition. Publisher: I

Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:

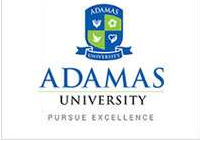
Components	Class Assessment	End Term
Weightage (%)	50	50

Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	3	3	2	3	3	3	3	3	1	3	3	3
CO2	3	3	2	3	3	3	3	3	1	3	3	3
CO3	3	3	2	3	3	3	3	3	1	3	3	3
CO4	3	3	2	3	3	3	3	3	1	3	3	3
CO5	3	3	2	3	3	3	3	3	1	3	3	3
Avg	3	3	2	3	3	3	3	3	1	3	3	3

1=weakly mapped
2= moderately mapped
3=strongly mapped

Model Question Paper

Name:			
Enrolment No:			
Course: SBC52207 – PLANT BIOCHEMISTRY LAB (PRACTICAL)			
Program: M.Sc. Biochemistry			
Time: 03 Hrs.			
Semester: Even 2019-20			
Instructions: Attempt any two questions from Section A (each carrying 10 marks); Section B is Compulsory (carrying 10 marks).			
Section A (Attempt any Two)			
1.	a) Write the principle of extraction of urease from jack bean.(U) b) Determine the effect of temperature on a membrane.(Ap)	4 6	CO1 CO2
2.	a) Explain the basic theory of tissue culture.(U) b)Determine the rate of oxygen evolution with respect to light intensity.(Ap)	4 6	CO3 CO4
3.	a) Write the principle behind column chromatography.(U) b)Demonstrate the presence of amylase in germinating seed with a simple experiment.(Ap)	4 6	CO3 CO2
SECTION B is compulsory			
4.	Viva-voce (U/An/Ap/R/Ev)	10	CO1 CO2 CO3 CO4 CO5
5.	Practical copy(U/Ap/Ev)	10	CO1 CO2 CO3

BIC22570	Professional Development Course-1 (Practical)	L	T	P	C
Version 1.0	Contact Hours - 30	0	0	1	1
Pre-requisites/Exposure	PLUS B.SC LEVEL SCIENCE				
Co-requisites	--				

Catalog Description: This professional development course aims to help you discover and achieve your goals by focusing on organization and action. You'll learn techniques to enhance goal-setting, communication, self-motivation, and a positive attitude, empowering you to maximize your performance both academically and professionally.

Course Syllabus:

The syllabus for Professional Development Course-I for senior students

(1st Semester- 3rd Semester for P.G students)

1. Introduction to Pre-Placement Training.
2. Resume Building & Cover Letter Writing.
3. Interview Skills.
4. Aptitude and Technical Skills.
5. Group Discussion and Communication Skills.
6. Personal Branding and Online Presence.
7. Professional Skills.
8. Industry Insights and Company Presentations.
9. Career Guidance for competitive entrance exams and Job Search Strategies
10. Mock Tests and Assessments.

Course learning outcomes:

CO1: Remembering-Students will be able to create professional resumes and cover letters tailored to specific job applications, demonstrating effective resume-building techniques.

CO2: Understanding-Students will analyze various interview scenarios to identify key strategies for successfully navigating different types of interview questions and formats.

CO3: Applying-Students will apply their aptitude and technical skills to solve real-world problems through mock tests and assessments, showcasing their problem-solving abilities.

CO4: Analysing-Students will evaluate their personal branding and online presence, making necessary adjustments to enhance their professional image on platforms like LinkedIn.

CO5: Evaluate-Students will demonstrate effective communication skills in group discussions, presentations, and professional interactions, ensuring clear and confident expression of ideas.

CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	-	3	3	1	3	3	3	3	-	3	2	2
CO2	-	3	3	1	3	3	3	3	-	3	2	2
CO3	-	3	3	1	3	3	3	3	-	3	2	2
CO4	-	3	3	1	3	3	3	3	-	3	2	2
CO5	-	3	3	1	3	3	3	3	-	3	2	2
Avg	-	3	3	1	3	3	3	3	-	3	2	2

Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination

Examination Scheme:

Components	CA	End Term
Weightage (%)	50	50

BIC21511	Molecular Biology (THEORY)	L	T	P	C
Version 1.0	Contact Hours - 45	3	0	0	3
Pre-requisites/Exposure	BSc. Level Biochemistry Knowledge				
Co-requisites	--				

Course Objectives

- To provide students basic idea about organization of prokaryotic and eukaryotic genome.
- It will also provide in depth knowledge about DNA replication and repair mechanism.
- To deliver detail mechanism of RNA synthesis and different RNA processing events.
- To provide students different methods of protein synthesis, protein transport mechanism.

Course Outcomes

On completion of this course, the students will be able to:

CO1: Remembering: Recall the fundamental principles, techniques, and terminology of molecular biology.

CO2: Understanding: Explain the key concepts and processes involved in molecular biology, such as gene expression, DNA replication, and genetic engineering.

CO3: Applying: Apply molecular biology techniques and methodologies to analyse genetic information, conduct experiments, and solve problems in the field.

CO4: Analysing: Analyse experimental data and scientific literature to draw conclusions and make informed decisions in molecular biology research.

CO5: Evaluating: Evaluate the significance and implications of molecular biology research findings, and assess the ethical considerations and potential applications of mole

Catalog Description

The core-course of 'Molecular Biology' will help to define fundamental difference between prokaryotic and eukaryotic genome organization. This course includes comprehensive approach through studying the DNA replication and repair mechanism and different RNA species and their mode of action. Furthermore, the implication of different protein modification and transport of proteins inside the cell will also be illuminated. All the lectures will be devoted on discussions of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as power point presentation, audio visual virtual lab session as per requirement. The tutorials will enable the students with problem-solving ability led by the course coordinator. Students will perceive the basic concepts of the subject via exercise and discussions with the coordinator.

Course	Content
--------	---------

MOLECULAR BIOLOGY

- 1. Genome organization:** Organization of bacterial genome. Structure of eukaryotic chromosomes, Chromatin organization & packaging. Heterochromatin and Euchromatin; DNA reassociation kinetics (Cot curve analysis); Repetitive and unique sequences; Satellite DNA; DNA melting and buoyant density; Nucleosome phasing; DNase I hypersensitive regions; DNAmethylation [Lectu
- 2. DNA replication and repair:** Unit of replication, enzymes involved, replication origin and replication fork, fidelity of replication, extrachromosomal replicons, DNA damage and repair mechanisms in prokaryotes and eukaryotes. [Lecture hours 6]
- 3. RNA synthesis and processing:** RNA world and RNA replication; Transcription factors and machinery, formation of initiation complex, transcription activators and repressors, RNA polymerases, capping, elongation and termination, RNA processing, RNA editing, splicing, polyadenylation, structure and function of different types of RNA, RNA transport. [Lectu
- 4. Protein synthesis and processing:** Ribosome, formation of initiation complex, initiation factors and their regulation, elongation and elongation factors, termination, genetic code, aminoacylation of tRNA, tRNA-identity, aminoacyl tRNA synthetase, translational proof-reading, translational inhibitors, post-translational modification of proteins. [Lectu
- 5. Protein localization:** Chaperones, SRP, translocons, protein transport, ubiquitination
- 6. Molecular Evolution:** Concepts of neutral evolution, molecular divergence and molecular clocks; molecular tools in phylogeny, classification and identification; protein and nucleotide sequence analysis; origin of new genes and proteins; gene duplication and divergence. Speciation; allopatricity and sympatricity; convergent evolution; sexual selection; co-evolution. [Lectu
- 7. Recombination:** Homologous and non-homologous; Site specific recombination; Chi sequences in prokaryotes; Gene targeting; Gene disruption; FLP/FRT and Cre/Lox recombination. [Lectu

Reference Books:

1. Molecular biology of gene by J. D. Watson
2. Biochemistry by L. Stryer 4th edition
3. Fundamentals of biochemistry by D. Voet, J. Voet and C.W. Pratt
4. Molecular cell biology 4th ed. Lodish B., Zipursky Matsudaira, Ball.

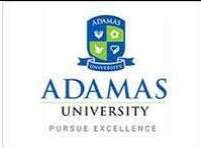
Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:

Components	Class Assessment	End Term
Weightage (%)	50	50

Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
Avg	3	3	2	3	3	-	-	3	-	-	1	3

1=weakly mapped
2= moderately mapped
3=strongly mapped

Name: XXXXXXXXXX			
Enrolment No:			
Course: BIC21509 – Molecular Biology (THEORY) Program: M.Sc. Biochemistry Semester: Even2020-21		Time: 03Hrs. Max. Marks:50	
Instructions: Attempt any four questions from Section A (each carrying 5 marks); any two questions from Section B (each carrying 10 marks).			
SECTION A (Attempt any Four questions) (5X4=20) (5X4=20)			
1.	What is the difference between prokaryotic and eukaryotic genome? Explain DNA melting temperature	U	CO1
2.	Define Cot curve. Describe significance of satellite DNA.	R	CO2
3.	Classify different RNA species based on their i) structure and ii) function.	R	CO3
4.	Define genetic code. What are the start codon and stop codon?	U	CO4
5.	Define natural evolution and co-evolution.	U	CO5
SECTION B (Attempt any Three questions) (10X3=30)			
6.	Calculate the weight in grams of a double-helical DNA molecule stretching from the Earth to the moon (~320,000 km). The DNA double helix weighs about 1×10^{-18} g per 1,000 nucleotide pairs; each base pair extends 3.4 Å. Explain why the absorption of UV light by double-stranded DNA increases (the hyperchromic effect) when the DNA is denatured.	AN,AP	CO2
7.	How is the helical structure of a long and fully base-paired (except at the end) hairpin in RNA different from that of a similar hairpin in DNA?	AN,AP	CO1 CO2
8.	Protein A has a binding site for ligand X with a K_d of 10^{-6} M. Protein B has a binding site for ligand X with a K_d of 10^{-9} M. Which protein has a higher affinity for ligand X? Explain your reasoning. Convert the K_d to K_a for both proteins.	AN,AP,R	CO4
9.	A team of biochemists uses genetic engineering to modify the interface region between hemoglobin subunits. The resulting hemoglobin variants exist in solution primarily as $\alpha\beta$ dimers (few, if any, $\alpha_2\beta_2$ tetramers form). Are these variants likely to bind oxygen more weakly or more tightly? Explain your answer.	AN,R	CO4 CO5

BIC21512	Recombinant DNA Technology	L	T	P	C
Version 1.0	Contact Hours: 45	3	0	0	3
Pre-requisites/Exposure	Knowledge of Molecular Biology of B.Sc Level				
Co-requisites	-				

Course Objectives

- To conceptualize the characteristics of recombinant DNA.
- To acquire the knowledge about restriction and modification system and cloning vectors
- To acquire the knowledge about amplification of gene in *in vitro* system.
- To gain the knowledge about techniques for analysis of gene expression.
- To understand the transcriptomics, genomics and their application in recombinant DNA technology.

Course Outcomes

On completion of this course, the students will be able to

CO1. Remember the basic principles of Recombinant DNA Technology, including the methods used for DNA extraction, purification, and manipulation.

CO2. Understand the process of cloning DNA fragments and recombinant DNA technology.

CO3. Apply various techniques such as restriction enzyme digestion, gel electrophoresis, and polymerase chain reaction (PCR) in the laboratory to create recombinant DNA constructs.

CO4. Analyze and interpret the results of experiments conducted in the Recombinant DNA Technology lab, including the identification of recombinant DNA clones.

CO5. Evaluate the ethical considerations and potential applications of Recombinant DNA Technology in various fields such as healthcare, agriculture, and biotechnology.

Catalog Description

Recombinant DNA (rDNA) molecules are DNA molecules formed by laboratory methods of genetic recombination (such as molecular cloning) to bring together genetic material from multiple sources, creating sequences that would not otherwise be found in the genome. Recombinant DNA is the general name for a piece of DNA that has been created by combining at least two fragments from two different sources. Recombinant DNA is possible because DNA molecules from all organisms share the same chemical structure, and differ only in the nucleotide sequence within that identical overall structure. Recombinant DNA molecules are sometimes called chimeric DNA, because they can be made of material from two different species, like the mythical chimera. R-DNA technology uses palindromic sequences and leads to the production of sticky and blunt ends. The DNA sequences used in the construction of recombinant DNA molecules can originate from any species. For

example, plant DNA may be joined to bacterial DNA, or human DNA may be joined with fungal DNA. In addition, DNA sequences that do not occur anywhere in nature may be created by the chemical synthesis of DNA, and incorporated into recombinant molecules. Using recombinant DNA technology and synthetic DNA, literally any DNA sequence may be created and introduced into any of a very wide range of living organisms. Proteins that can result from the expression of recombinant DNA within living cells are termed recombinant proteins. When recombinant DNA encoding a protein is introduced into a host organism, the recombinant protein is not necessarily produced. Expression of foreign proteins requires the use of specialized expression vectors and often necessitates significant restructuring by foreign coding sequences. Recombinant DNA differs from genetic recombination in that the former results from artificial methods in the test tube, while the latter is a normal biological process that results in the remixing of existing DNA sequences in essentially all organisms. However, the goal of this paper to analyse the artificially created recombinant DNA and expression of their genes.

Course Content

Unit I: Basics of DNA Cloning [9 Lecture Hours] Simple cloning and cloning using linkers and adaptors. Cloning into various kinds of vectors – plasmids, phages lambda and M13, phagemids, cosmids, P1 phage, PACs, BACs and YACs. Selection and screening of clones.

Unit II: Methods of DNA and protein analysis [9 Lecture Hours] Agarose, polyacrylamide and pulsed field gel electrophoresis of DNA. Southern and Northern Blotting. Radiolabelling probes. Isolation and purification of DNA. RFLP analysis. DNA fingerprinting and its application in forensics, in disease diagnosis and in identification of strains. Native PAGE, SDS-PAGE and two-dimensional PAGE analysis of proteins. Western Blotting analysis.

Unit III: Polymerase Chain Reaction [9 Lecture Hours] Concept of PCR and various thermophilic enzymes used in PCR. Gradient PCR versus Touchdown PCR. Designing primers. Cloning PCR products. Long PCR, Inverse PCR, Vector PCR, RT-PCR, 5' and 3' RACE, qPCR, Real Time PCR using SYBR Green, Scorpion primers and TaqMan probes, MOPAC, Multiplex PCR, Differential Display PCR, RAPD fingerprinting of micro-organisms, Ligation Chain Reaction, Overlap PCR, Rolling Circle Amplification Technology.

Unit IV: Construction of cDNA and genomic DNA libraries [9 Lecture Hours] Vectors used in the construction of cDNA versus genomic DNA libraries. Steps and enzymes involved in the construction of cDNA versus genomic DNA libraries. Screening libraries by colony hybridization and colony PCR. Screening expression libraries. Enriching for clones in cDNA libraries by positive selection and subtractive hybridization. Identifying genes in complex genomes by direct selection of cDNA and exon trapping.

UNIT V: Transcriptional analysis of gene expression and transcriptomics [9 Lecture Hours]

Gene expression analysis by Northern Blotting, RT-PCR, EST analysis and the use of reporter genes. Enzymatic and bioluminescent reporters. Reporters used in protein localization and trafficking studies. Promoter analysis – deletion analysis and linker scanning analysis coupled to reporter assays, mapping transcriptional start sites by S1 nuclease mapping, primer extension studies or 5' RACE. Transcriptome analysis by DD-PCR and EST analysis, DNA microarrays (cDNA arrays and oligo arrays), Serial Analysis of Gene Expression (SAGE).

Reference Books

1. Recombinant DNA: Genes and Genomes - a Short Course by James D. Watson, 2006
2. Principles of Gene Manipulation and Genomics by Sandy Primrose and Twyman, 2006
3. From genes to genomes concepts and applications of DNA technology by Jeremy Wdale and Malcolm von Scrantz, 2011
4. Molecular Biotechnology: Principles and Applications of Recombinant DNA by Bernard Glick 2009
5. Genomes 3 by T.A. Brown, 2006

Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:

Components	Class Assessment	End Term
Weightage (%)	50	50

Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

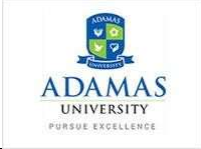
CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
Avg	3	3	2	3	3	-	-	3	-	-	1	3

1=weakly mapped

2= moderately mapped

3=strongly mapped

Model Question Paper

Name:			
Enrolment No:			
Course: BIC21510 –Recombinant DNA Technology Program: M.Sc Biochemistry Semester: Even2020-21		Time: 03 Hrs. Max. Marks: 50	
Instructions: Attempt any three questions from Section A (each carrying 5 marks); any Two Questions from Section B (each carrying 10 marks).			
SECTION A (Attempt any Four questions) (5X4=20)			
1.	Discuss Serial Analysis of Gene Expression.	R	CO4
2.	Distinguish cloning vector and expression vector.	R	CO2
3.	Define melting temperature (T _m) with respect to DNA. Evaluate the theoretical T _m of the primer sequence: 5'GGATTCAGAGAGGATCC3'	U	CO3
4.	Distinguish between Native-PAGE and SDS-PAGE.	R	CO5
SECTION B (Attempt any 3 questions) (10X3=30)			
5.	How can you use the AFLP-PCR analysis for paternity test? Briefly describe . A student researcher overexpresses an exogenous protein in cell culture and wants to determine if that protein is in fact, overexpressed. Which blotting technique would best demonstrate that this protein is expressed in these cells? Schematically represent the indirect detection of protein in western blot technique.	AN,AP,U	CO1 CO4
6.	Describe 5' RACE technique utility with suitable illustration. Which platform of DNA sequencing uses a 'sequence by synthesis' approach? Describe briefly with suitable diagram. Write two major advantages of SMRT sequencing over illumine sequencing. Which of the following technique can be used to detect DNA-protein interaction within cell? a) ChIP and ChIP-	AN,U,AP,R	CO3 CO4 CO5

	seq b) Footprinting c) EMSA d) RACE		
7.	Discuss the steps of genomic DNA and cDNA library construction. Describe two advantage of creating a cDNA library compared to genomic library?	4 6	CO1

BIC22543	Bioinformatics, & Biostatistics (THEORY)	L	T	P	C
Version 1.0	Contact hours = 45	3	0	0	3
Pre-requisites/Exposure	BSc. level Biology knowledge				
Co-requisites	-				

Course Objectives

- To provide those students with an introductory level knowledge to Biostatistics, Bioinformatics & Computer Applications .
- It will also provide in depth knowledge of biostatistics.
- Elaborating the database and biological database
Explore the knowledge of modern methods of Bioinformatics such as Microarray experiment, Clustering of microarray data, Principal component analysis.

Course Outcomes

On completion of this course, the students will be able to

- CO1 Remembering- Recall** biostatistics techniques.
- CO2 Understanding- Comprehend the** biological database and their role in bioinformatics.
- CO3 Applying- Apply** the knowledge of Cluster analysis; Phylogenetic clustering, Sequence Comparison.
- CO4 Analysing- Analyse** modern methods of Bioinformatics such as Microarray experiment, Clustering of microarray data, Principal component analysis
- CO5 Evaluate- demonstrate the** structure based application in bioinformatics, protein structure prediction through homology modeling and current research activities in the field of bioinformatics

Catalog Description

The core-course will help to understand the introductory level knowledge to biostatistics, bioinformatics & computer applications. This course is an beginning to the biostatistics, the application of different bioinformatics methods to biological data analysis, biological database and some current research activities in the field of bioinformatics. Furthermore, the possible applications of this knowledge in biostatistics, bioinformatics & computer applications would also be illuminated. All the lectures will be devoted on discussions of basic theories and advanced topics, focusing on practical implementation of knowledge.

Classes will be conducted by lecture as well as power point presentation, audio visual virtual lab session as per requirement. The tutorials will enable the students with problem-solving ability led by the course coordinator. Students will perceive the basic concepts of the subject via exercise and discussions with the coordinator.

Course Content

Bioinformatics [20 hrs]

1. Introduction to Bioinformatics and Computational Biology with historical background, major developments.
2. Biological databases, data query and data mining; Boolean operators; Problems and Applications to biological problems.
3. Nucleic acid sequence analysis, alignment, similarity searches including remote similarity searches, secondary structure element, motifs.
4. Protein sequence analysis; alignment, similarity searches including remote similarity searches, secondary structure elements, motifs

Biostatistics [25 hrs]

1. Introduction: Applications of statistics in biology, definitions (populations, samples), Introduction to probability theory, Basic concepts, definitions to understand probability and sampling; Defining sample space, computing probability.
2. Random variables and probability distributions: Discrete random variables, Bernoulli random variable, binomial distribution, Poisson distribution with examples
Continuous random variables, Normal random variable, other continuous distributions, Central limit theorem
3. Arithmetic and other means, median, mode; when to use each measure of location
Measures of spread: Variance and Standard Deviation, Standard Error.
4. Framework for statistical analyses Framing hypothesis, The scientific method; deduction and induction; The Hypothetico-deductive method; Testing hypothesis, Significance and p-values;
5. Data Analyses: Computing sums of squares, standard error of differences between means, T-test, Regression, Fitting data to a linear model; Variances and co-variances; least-square parametric estimates; Hypothesis test with regression; Assumptions, Analyses of variance, ANOVA and Partitioning of Sum of Squares, Assumptions; Hypothesis tests with ANOVA; Constructing F-Ratios; ANOVA tables, Analyses of categorical data, Two-way contingency tables; Chi-square test.

Text Books:

1. Bioinformatics: Sequence and Genome Analysis by David W. Mount, 2004
2. Introduction to Bioinformatics by Arthur M. Lesk, 2002
3. Biostatistics: A Foundation for Analysis in the Health Sciences by Wayne W. Daniel, 2004
4. Computational Biology by David Fenyo, 2010
5. Statistical Methods by Statistical Methods by William G. Cochran, George W. Snedecor 1972.
6. Let us C – Kanetkar.

Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination
Examination Scheme:

Components	Class Assessment	End Term
Weightage (%)	50	50

Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

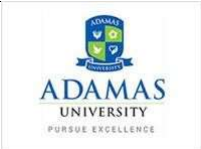
1=weakly mapped

2= moderately mapped

3=strongly mapped

CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
Avg	3	3	2	3	3	-	-	3	-	-	1	3

Model Question Paper

Name:			
Enrolment No:			
Course: BIC22525- Bioinformatics, Computational Biology & Biostatistics (THEORY) Program: M.Sc. Biochemistry Time: 03Hrs. Semester: Even 2019-20 Max. Marks: 50			
Instructions: Attempt any four questions from Section A (each carrying 5 marks); any two questions from Section B (each carrying 10 marks).			
SECTION A (Attempt any Four questions) (5X4=20) (5X4=20)			
1.	What is Great mean ? Explain Scatter plot.	R	CO1
2.	Demonstrate the role of Database development in Bioinformatics	U	CO2
3.	Explain the process of X-ray crystallography for 3D structure determination of protein	R	CO3
4.	Enlist 3 features of PDB and 2 features of NDB.	U	CO4
5	Demonstrate the Box Plot and Histogram with diagram.	R	CO5
SECTION B (Attempt any 3 questions) (10X3=30)			
6.	What is Phylogenetic Tree? Draw a label diagram of a phylogenetic tree? Explain different types of the Phylogenetic Tree.	AN,U,R	CO3
7.	What is microarray? Illustrate the microarray using flowchart and normalization of microarray data	U,AP	CO1 CO2
8.	Some trees are having the following heights: 150 cm, 200 cm, 250 cm, 300 cm, 350 cm, 400 cm, 450 cm, 500 cm Calculate the mean, variance and standard deviation	AN	CO4 CO5
9	“Hidden Markov Model (HMM) is used in modelling of eukaryotic gene, two sequences analysis and multiple sequences analysis. Explain this statement	AP	CO4 CO5

BIC21544	Genomics and Proteomics (THEORY)	L	T	P	C
Version1.0	ContactHours -45	3	0	0	3
Pre-requisites/Exposure	BasicknowledgeofGeneticsand ProteinBiochemistry				
Co-requisites	--				

CourseObjectives:

1. The course will provide an introduction to what genomics, proteomics andnanobiotechnologyisandwhyitisimportantinthecurrentcontextofbiologicalscience.
2. The course will give an overview about the application are asofgenomics and proteomics ,with a focus on the topics that will be taught in the course.
3. Thecoursewillbeabletodescribethebasicsofnanotechnologyandtheirapplicationsinnanobiotechnology

Courseoutcome:

Thestudentswillbeable

- CO1. **Remembering**-To **recall** the importan tconcept of Omic stechnologies, with emphasis on genomics and proteomics.
- CO2. **Understanding**-To **comprehend** the information and key technological developments that enabled modern genomics and proteomics studies.
- CO3. **Applying**-To **assess** the advanced genomics and proteomics technologies and techniques innanobiosciences.
- CO4. **Analysing**-To **explain** the different types of genome variation and their relationship to human diseases.
- CO5. **Evaluate**-To **evaluate** the biological systems information relating to genes, proteins andcellular structures which can be used tomodel living cells, and even tocreate newsynthetic cells.

Course Description:

The main aim of this module is to provide an understanding about the genomics and proteomicstechniques and their applications in biological sciences. The subject deals with a rapidly evolving scientific are athat introduces students into genomes, proteomes, databases and nanobiotechnology that store various data about genes, proteins, genomes and proteomes. Students would learn about genomics , proteomics and nano biotechnology and offer basic knowledge of genome sequencing, major differences between prokaryotic and eukaryotic genomes, basic proteomics and its applications. Students would gain skills in comparative evolutionary, human genomics and functional genomics.The acquired knowledge during the course would be helpful to those students who want to work incorefacilitiesandcommercialbiologicalandmedicallaboratoriesaswellasintheirpostgraduate studies.

Course Content:

Genomics and Proteomics

Unit I. Introduction

15 Lecture hours

Structural organization of genome in Prokaryotes and Eukaryotes; Organelle DNA-mitochondrial, chloroplast; DNA sequencing-principles and translation to large scale projects; Recognition of coding and non-coding sequences and gene annotation; Tools for genome analysis-RFLP, DNA fingerprinting, RAPD, PCR, Linkage and Pedigree analysis-physical and genetic mapping.

Unit II. Genome sequencing projects

15 Lecture

hours Microbes, plants and animals; Accessing and retrieving genome project information from web; Comparative genomics, Identification and classification using molecular markers-16S rRNA typing/sequencing, ESTs and SNPs.

Unit III. Proteomics

10 Lecture hours

Protein analysis (includes measurement of concentration, amino-acid composition, N-terminal sequencing); 2-D electrophoresis of proteins; Microscale solution isoelectric focusing; Peptide fingerprinting;

Unit IV Quantitative Proteomics

LC/MS-MS for identification of proteins and modified proteins; MALDI-TOF; SAGE and Differential display proteomics, Protein-protein interactions, Yeast two hybrid system.

Unit V. Functional genomics and proteomics

5 Lecture

hours Analysis of microarray data; Protein and peptide microarray-based technology; PCR-directed protein *in situ* arrays; Structural proteomics

Suggested Books

1. Berg, J.M., Tymoczko, J.L. and Stryer, L. (2006). Biochemistry. VI Edition. W. H. Freeman and Co.
2. Buchanan, B., Gruissem, W. and Jones, R. (2000) Biochemistry and Molecular Biology of Plants. American Society of Plant Biologists.
3. Nelson, D.L., Cox, M.M. (2004) Lehninger Principles of Biochemistry, 4th Edition, W. H. Freeman and Company, New York, USA.
4. Hopkins, W.G. and Huner, P.A. (2008) Introduction to Plant Physiology. John Wiley and Sons.

Modes of Examination: Assignment/Quiz/Project/Presentation/Written

ExamExaminationScheme:

Components	Assessment	EndTerm
Weightage(%)	50	50

Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
Avg	3	3	2	3	3	-	-	3	-	-	1	3

BIC22545	Molecular Biology & Recombinant DNA Technology Lab (Practical)	L	T	P	C
Version 1.0	Contact Hours - 60	0	0	4	2
Pre-requisites/Exposure	BSc. Level Biochemistry Knowledge				
Co-requisites	--				

Course Objectives

- To provide students basic idea about how to isolate plasmid DNA from bacteria and genomic DNA from plant or liver cells.
- It will also illustrate how to measure DNA concentration using Spectrophotometer.
- To perform PCR amplification and analyze DNA by Agarose gelelectrophoresis.
- To provide students how to do restriction digestion, transformation and cloning of plasmid DNA.

Course Outcomes

On completion of this course, the students will be able to:

CO1. Remember the basic principles of Recombinant DNA Technology, including the methods used for DNA extraction, purification, and manipulation.

CO2. Understand the process of cloning DNA fragments and recombinant DNA technology.

CO3. Apply various techniques such as restriction enzyme digestion, gel electrophoresis, and polymerase chain reaction (PCR) in the laboratory to create recombinant DNA constructs.

CO4. Analyze and interpret the results of experiments conducted in the Recombinant DNA Technology lab, including the identification of recombinant DNA clones.

CO5. Evaluate the ethical considerations and potential applications of Recombinant DNA Technology in various fields such as healthcare, agriculture, and biotechnology.

The practical course of 'Molecular Biology Lab' will help to hands on experience on isolation of plasmid DNA from bacteria and genomic DNA from plant or liver cells. This course includes comprehensive approach to perform PCR amplification of desired gene and analysis by Agarose gel electrophoresis. Furthermore, recombinant biotechnology techniques like restriction digestion, transformation and cloning of plasmid DNA will also be illuminated. The hands-on experience will enable students to enrich in experimental protocol. The tutorials will enable the students with problem-solving ability led by the course coordinator. Students will perceive the basic concepts of the subject via exercise and discussions with the coordinator.

Course Content

MOLECULAR BIOLOGY LAB

1. Isolation of DNA from E. coli/ liver/ plant/ plasmid [8 Lecture Hours]
2. Determination of base composition (spectrophotometry) [8 Lecture Hours]

3. Agarose gel electrophoresis of DNA[8 LectureHours]
4. PCR amplification of desired gene[8 LectureHours]
5. Restriction digestion and ligation of DNA, Endonuclease mapping of DNA[8 Lecture Hours]
6. Transduction[6 LectureHours]
7. Transformation[7 LectureHours]
8. Expression analysis[7 LectureHours]


Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	50	50

Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	3	3	2	3	3	3	3	3	1	3	3	3
CO2	3	3	2	3	3	3	3	3	1	3	3	3
CO3	3	3	2	3	3	3	3	3	1	3	3	3
CO4	3	3	2	3	3	3	3	3	1	3	3	3
CO5	3	3	2	3	3	3	3	3	1	3	3	3
Avg	3	3	2	3	3	3	3	3	1	3	3	3

1=weakly mapped
2= moderately mapped
3=strongly mapped

Name:			
Enrolment No:			
Course: BIC22513 – Molecular Biology Lab (Practical) Program: M.Sc. Biochemistry Semester: Even2020-21		Time: 03Hrs. Max. Marks:40	
SECTION A (Compulsory) (10X2=20)			
1.	Isolate DNA from E. Coli. Measure DNA concentration.	AN,AP	CO1
2.	PCR amplify a desired DNA. Run on Agarose gel.	AN,AP	CO2
SECTION B (Compulsory) (10X2=20)			
3.	Lab Notebook	AN,AP,U, R	CO1 CO2
4.	Viva voce	AN,AP,U, R	CO1 CO2 CO3

BIC21546	Bioinformatics and Biostatistics Lab(PRACTICAL)	L	T	P	C
Version1.0	ContactHours-60	0	0	4	2
Pre-requisites/Exposure	UGLEVELBIOLOGY				
Co-requisites	--				

Course Objectives

1. To provide students with hands-on activities designed to encourage interest in the field of Bioinformatics, as well as promote greater understanding of the concepts presented in lecture.
2. Students will need to become proficient with terms, techniques, and applications.

Course Outcomes

On completion of this course, the students will be able to

1. Knowledge:

- Identify different databases used in bioinformatics and computational biology for storing biological information

2. Comprehension:

- Compare and contrast different genome annotation tools and their capabilities

3. Application:

- Construct phylogenetic trees based on molecular data to infer evolutionary relationships

4. Analysis:

- Analyze RNA and protein structure prediction results to understand molecular structure-function relationships

5. Synthesis:

- Design and implement a bioinformatics and computational biology project using appropriate tools and methods

Catalogue Description

Bioinformatics Lab (Practical) is the overall Learn and apply the knowledge of using different modern tools and techniques in the field of Bioinformatics. This course covers laboratory techniques describes different modern practical methods related to Bioinformatics such as genes and genomes, sequence alignment of DNA and proteins, basic programming using python, predict protein structure-function and phylogenetic tree. All the lectures will be devoted on discussions of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by

lecture as well as powerpoint presentation, audio visual virtual session as per requirement. The tutorials will familiarize the students with practical problem-solving techniques led by the course coordinator. Students will strongly grab the basic concepts of the subject via exercise and discussions with the coordinator.

Course Content

Bioinformatics Lab(BIC21540)[12 hrs.each experiment]

1. Retrieving genomes, identifying of and annotating genes, databases.
2. Sequence Alignment of DNA and Proteins.
2. Applying UNIX, basic programming using python:
3. Predicting protein structure-function
4. Building phylogenetic tree
5. Protein structure-homology modelling and docking.

Suggested reading:

1. Essentials of Bioinformatics, Xin Xiong, Cambridge
2. Bioinformatics: Sequence and Genome Analysis by David W. Mount, 2004.

Modes of Evaluation: Quiz/Assignment/presentation/extempore/Written Examination Examination Scheme:

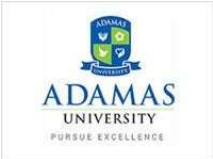
Components	Class Assessment	End Term
Weightage(%)	50	50

Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	3	3	2	3	3	3	3	3	1	3	3	3
CO2	3	3	2	3	3	3	3	3	1	3	3	3
CO3	3	3	2	3	3	3	3	3	1	3	3	3
CO4	3	3	2	3	3	3	3	3	1	3	3	3
CO5	3	3	2	3	3	3	3	3	1	3	3	3
Avg	3	3	2	3	3	3	3	3	1	3	3	3

1=weakly mapped
 2=moderately
 mapped3=strongly mapped

Model Question Paper

Name: Enrolment No:			
Course: BIC21540–Bioinformatics Lab(PRACTICAL) Program: M.Sc.Biochemistry Semester: Even 2019-20			
Instructions: Attempt any two questions from Section A (each carrying 10 marks); Section B is Compulsory (carrying 10 marks).			Time: 03 Hrs. Max. Marks: 50
Section A (Attempt any 3)			
1.	a) Retrieving one gene from different five	Ap/An	CO1
	b) Design a table using species name, gene accession no and gene length. (5+5)		CO5
2.	a) Retrieving Five Proteins structure of SARS-CoV-2 from PDB with PDB ID as a basic step for homology modelling. Write about the method b) Design a table using Protein name, PDB ID and description structure. (2+4+4)	Ap/An	CO4CO5
3.	a) Perform Sequence Alignment of one Proteins from different five species .b) Write about the method and its importance. (4+3+3)	Ap/R/An	CO2
4.	a) Draw one phylogenetic tree of CRP protein from different six species. b) Explain its methodology and result. (4+3+3)	Ap/U	CO4CO5
	SECTION B is compulsory		

5.	Viva-voce(10)	U/An/Ap/R/Ev	CO1 CO2 CO3 CO4 CO5
6.	Practicalcopy(10)	U/Ap/Ev/Cr	CO1 CO2 CO3 CO4

BIC22547	Genomics and Proteomics (PRACTICAL)	Lab	L	T	P	C
Version1.0	ContactHours -45		0	0	4	2
Pre-requisites/Exposure	BasicknowledgeofGeneticsand ProteinBiochemistry					
Co-requisites	--					

CourseObjectives:

1. Theobjectivesofthiscourseistoprovideintroductoryknowledgeconcerninggenomics,proteomics andtheirapplicationsinbioscience today.
2. The course will give an overview about the basic techniques and different softwaresused in areas of genomics and proteomics for their applications in current areas ofresearch.
3. The course will be providing some important insights about designing nanomaterialsandtheirapplicationinseveralareas ofbioscience.

Courseoutcome:

1. Remembering: Recall the basic concepts and terminology related to genomics and proteomics, including genome sequencing, genome-wide screening, and genome editing.
2. Understanding: Explain the principles and techniques used in genome analysis tools like CHIP-SEQ, RNA-SEQ etc., and proteome anlysis tools such as 2D-gel electrophoresis and mass spectrometry for proteomics research.
3. Applying: Apply knowledge of genomics and proteomics to interpret and analyze data from genome sequencing and proteomics experiments (MS-data).
4. Analyzing: Analyze and compare different methods of genome analysis and proteomics techniques for differential proteomics studies.
5. Creating: Design and implement experiments using genome analysis tools and proteomics techniques to investigate biological processes and functions at the molecular level.

CourseDescription:

The main aim of this module is to provide an understanding about the genomics and proteomicstechniques and their applications in biological sciences. The subject deals with a rapidly evolvingscientificareathatintroducesstudentsintogenomes,proteomes,databasesandanobiotechnologythats torevariousdataaboutgenes,proteins,genomesandproteomes.Studentswouldlearnaboutgenomics,proteomics andnanobiotechnologyandofferbasicknowledgeofgenomesequencing,majordifferencesbetweenprokaryotic andeukaryotic

genomes, basic proteomics and its applications. Students would gain skills in applied nanoscience, comparative, evolutionary, human genomics and functional genomics. The acquired knowledge during the course would be helpful to those students who want to work in core facilities and commercial biological and medical laboratories as well as in their postgraduate studies.

Course Content:

1. Bacterial DNA extraction from different sources. 4 Lecture hour
2. PCR amplification of DNA using 16srRNA primers. 6 Lecture hour
3. Comparative analysis of sequencing result for phylogenetic analysis using suitable software. 6 Lecture hour
4. SDS-PAGE of isozymes. 8 Lecture hour
5. SDS-PAGE analysis of serum proteins. 8 Lecture hour
6. Western blot analysis of actin and tubulin. 8 Lecture hour
7. Biosynthesis of Fe/Ag based bionanoparticles. 10 Lecture hour
8. Detection of antimicrobial properties of bionanoparticles. 10 Lecture hour

Text Books

1. Primrose, S.B., Twyman, R.M., Primrose, S.B., & Primrose, S.B. (2006). Principles of Gene Manipulation and Genomics.
2. Malden, MA: Blackwell Pub. 2. Liebler, D.C. (2002). Introduction to Proteomics: Tools for the New Biology.
3. Totowa, NJ: Humana Press. 3. Campbell, A.M., & Heyer, L.J. (2003). Discovering Genomics, Proteomics, and Bioinformatics. San Francisco: Benjamin Cummings.
4. Bionanotechnology by David S. Goodsell, 2004, Wiley Publications.

Modes of Examination: Assignment/Quiz/Project/Presentation/Written Exam Examination Scheme:

Components	MidTerm	EndTerm
Weightage(%)	50	50

CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	3	3	2	3	3	3	3	3	1	3	3	3
CO2	3	3	2	3	3	3	3	3	1	3	3	3
CO3	3	3	2	3	3	3	3	3	1	3	3	3
CO4	3	3	2	3	3	3	3	3	1	3	3	3
CO5	3	3	2	3	3	3	3	3	1	3	3	3
Avg	3	3	2	3	3	3	3	3	1	3	3	3

1=weaklymapped

2= moderatelymapped

3=stronglymapped

BIC22519	DSE-I CANCER BIOLOGY(THEORY)	L	T	P	C
Version 1.0	Contact Hours - 45	3	0	0	3
Pre-requisites/Exposure	B.Sc. LEVEL BIOLOGY				
Co-requisites	--				

Course Objectives

- Students will **understand** the structures and purposes of basic carcinogenic components especially organic and inorganic carcinogens
- Students will **understand** how cancer cells sabotage the normal metabolomics of a healthy cell and these cellular components are used to generate and utilize energy in cells
- Students will **understand** the cellular components underlying cell division and molecular basis of carcinogenesis
- Students will **apply** their knowledge of cancer biology to selected examples of changes or losses in cell function. These can include responses to environmental or physiological changes, or alterations of cell function brought about by mutation.

Course Outcomes

On completion of this course, the students will be able to

CO1. Remembering: Recall the fundamental principles of cancer genetics

CO2. Understanding: Demonstrate a thorough comprehension of the genetic mechanisms underlying cancer development and the role of developmental genetics in cancer progression.

CO3. Applying: Apply basic laboratory techniques and methodologies commonly used in cancer and developmental genetics research, such as polymerase chain reaction (PCR), DNA sequencing, and gene expression analysis.

CO4. Analyzing: Analyze experimental data obtained from laboratory experiments, interpret results, and draw conclusions regarding the genetic basis of cancer and developmental disorders.

CO5. Evaluating: Critically evaluate research articles and scientific literature related to cancer .

Catalog Description

The core-course of ‘Cancer Biology’ will help to understand the classification, structure and function of different carcinogenic compounds affecting animals. This course includes comprehensive approach through studying molecular mechanism of carcinogenesis, onset and progression of cancer in humans. It also includes the role of virus as carcinogenic agents. Furthermore, the application of virus and other carcinogens in carcinogenesis, therapeutics and gene delivery would also be illuminated. All the lectures will be devoted on discussions of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as power point presentation, audio visual virtual lab session as per requirement. The tutorials will enable the students with problem-solving

ability led by the course coordinator. Students will perceive the basic concepts of the subject via exercise and discussions with the coordinator.

Course Content

Cancer Biology (BIC22517)

Unit I: Fundamentals of cancer biology (7 HOURS)

Regulation of Cell cycle, Apoptosis, Mutations that cause changes in signal molecules, effects on receptor, signal switches, tumour suppressor genes, Modulation of cell cycle-in cancer, Different forms of cancers, Diet and cancer.

Unit II: Mutations; Oncogenes and Tumor suppressor genes (7 HOURS)

Nonsense, missense and point mutations; Intragenic and Intergenic suppression; Frameshift mutations; Physical, chemical and biological mutagens; Transposition - Transposable genetic elements in prokaryotes and eukaryotes; Mechanisms of transposition; Role of transposons in mutation; Viral and cellular oncogenes; Tumor suppressor genes from humans; Structure, function and mechanism of action of pRB and p53 tumor suppressor proteins; Activation of oncogenes and dominant negative effect; Suppression of tumor suppressor genes; Oncogenes as transcriptional activators.

Unit III: Principles of carcinogenesis (7 HOURS)

Chemical Carcinogenesis, Metabolism of Carcinogenesis, Natural History of Carcinogenesis, Targets of Chemical Carcinogenesis, Principles of Physical Carcinogenesis, X-Ray radiation – Mechanism of radiation Carcinogenesis.

Unit IV: Principles of molecular cell biology of cancer (7 HOURS)

Oncogenes, Identification of Oncogenes, Retroviruses and Oncogenes, detection of Oncogenes, Growth factor and Growth factor receptors that are Oncogenes. Oncogenes / Proto Oncogenes activity. Growth factors related to transformations.

Unit V: Principles of cancer metastasis (7 HOURS)

Clinical significances of invasion, heterogeneity of metastatic phenotype, Metastatic cascade, Basement membrane disruption, Three step theory of invasion, Proteinases and tumour cell invasion.

Unit VI: New molecules for cancer therapy (10 HOURS)

Different forms of therapy, Chemotherapy, Radiation Therapy, Detection of Cancers, Prediction of aggressiveness of Cancer, Advances in Cancer detection.

Reference book:

1. Molecular Biology of Cancer: Mechanisms, Targets, and Therapeutics by Lauren Pecorino, 2016
2. The Biology of Cancer by Robert A. Weinberg, 2006

3. Introduction to the Cellular and Molecular Biology of Cancer by Margaret A. Knowles, 2005

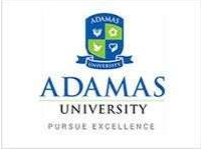
Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination
Examination Scheme:

Components	Class Assessment	End Term
Weightage (%)	50	50

Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
Avg	3	3	2	3	3	-	-	3	-	-	1	3

1=weakly mapped
2= moderately mapped
3=strongly mapped

Name:			
Enrolment No:			
Course: BIC22517 – DSE I CANCER BIOLOGY (THEORY) Program: M.Sc. Biochemistry Semester: Odd 2020-21		Time: 03Hrs. Max. Marks:50	
Instructions: Attempt any four questions from Section A (each carrying 5 marks); any two questions from Section B (each carrying 10marks).			
SECTION A (Attempt any Four questions) (5X4=20) (5X4=20)			
1.	What is Capsid? Explain various symmetry of viral particles.	R, U	CO1
2.	Classify virus based on their genetic material. What is the role of ICTV in viral classification?	R	CO2
3.	Describe four different stages of cancer and find the differences between them.	U	CO3
4.	Enlist 3 viruses that is associated with Cancer. Also mention the role of virus in etiology of cancer.	U	CO4
5.	A person was suffering from Estrogen-induced breast cancer. What medicinal options do the doctor have to treat him to prevent cancer cell multiplication. Explain the mechanisms involved.	R	CO5
SECTION B (Attempt any 3 questions) (10X3=30)			
6.	Design an experiment to locate p53 protein in a given sample from a cancer patient. What would be your preferred sample? Illustrate the headful mechanism. Add a note on mutated p53.	AN,AP	CO3
7.	What is the basis of EMT? Illustrate various stages of EMT. What is seed & Soil Hypothesis? How would you differentiate between EMT & MET?	AN,AP	CO1 CO2
8.	Mention any three ways of dietary management of cancer. Which foods are most suitable if we want to prevent the onset of cancer in a given tissue? What all techniques can be employed to inoculate cancer cells in mice? Illustrate the role of tumor suppressor gene? What is a papilloma virus?	AN,AP,R	CO1 CO2
9.	“A person was suffering from hepatocellular carcinoma and at that stage metastasis took place. Justify whether the patient is in preliminary or advanced stage of cancer. Which organ of his body is affected? Predict the viral infection that might be a etiological factor. Is the viral infection sole reason for cancer: comment. Mention the hallmarks of cancer.	AN,AP,U, R	CO4 CO5

BIC22520	DSE-I Nanobiotechnology (THEORY)	L	T	P	C
Version 1.0	Contact Hours - 45	3	0	0	3
Pre-requisites/Exposure	BSc. Level Chemistry and Biochemistry Knowledge				
Co-requisites	--				

Course Objectives

1. To provide students basic idea about electron microscopy, Cryo-electron microscopy and scanning electron microscopy.
2. It will also provide in depth knowledge about biological nanomaterials and engineered nanomaterials.
3. To deliver detail mechanism microfabrication and nanofabrication.
4. To provide students recent advancement nano-biotechnology and its impact in society.

Course Outcomes

On completion of this course, the students will be able to:

CO 1 Remembering-Recall the fundamental mechanism of electron microscopy, Cryo-electron microscopy and scanning electron microscopy.

CO 2 Understanding-Comprehend different mode of action of biological nanomaterials and engineered nanomaterials.

CO 3 Applying-Apply microfluidics and nano-fluidics and its importance in drug discovery.

CO 4 Analyzing-Analyze detail mechanism microfabrication and nanofabrication.

CO 5 Evaluate-Interpret recent advancement nano-biotechnology and its impact in society.

Catalog Description

The core-course of 'Nano-biotechnology' will help to define fundamental mechanism of electron microscopy, Cryo-electron microscopy and scanning electron microscopy. This course includes comprehensive approach through studying different mode of action of biological nanomaterials and engineered nanomaterials. Furthermore, the implication of recent advancement nano-biotechnology and its impact in society will also be illuminated. All the lectures will be devoted on discussions of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as power point presentation, audio visual virtual lab session as per requirement. The tutorials will enable the students with problem-solving ability led by the course coordinator. Students will perceive the basic concepts of the subject via exercise and discussions with the coordinator.

Course Content**DSE 2 NANOTECHNOLOGY**

1. Course overview. Nanoscale Properties (Electrical, Optical, Chemical), Nanoscale visualization techniques: Electron microscopy (TEM, SEM, Cryo-SEM), Scanning probe microscopy (AFM, STM), Diffraction techniques (XRD, synchrotron) **[8 LectureHours]**
2. Bionanomaterials, Biological building blocks, Bio nanostructures (nanofibers, nanotubes, nanocellulose). Biological nanomachines: Ribosomes, Photosynthesis systems, Bio nanomotors. **[7 LectureHours]**
3. Engineered Nanomaterials: Carbon nanomaterials (fullerenes, graphene, nanotubes, nanofibers), Metal nanoparticles (synthesis, properties and applications) , Magnetic nanoparticles (synthesis, properties and applications), Quantum dots, liquid crystals, Nano porous materials (metallic, zeolite, MOFs) **[8 LectureHours]**
4. Microfabrication methods (photolithography, soft lithography, replication). Nanofabrication methods (Top-Down approaches). Nanotechnology by self-assembly (Bottom-Up approach): Principles, thermodynamics, interactions, properties, Supramolecular self-assembly, Protein nanotechnology DNA nanotechnology**[7 Lecture Hours]**
5. Microfluidics: surface tension, capillarity, Reynolds number, diffusion, viscosity. Nanofluidic: nanopores and nanocapillaries. Debye length, Diffusion in solid phase and drug delivery. **[8 LectureHours]**
6. Biological and medical microdevices: lab on chips, organ-on chips, Biosensors (fabrication, functionalization, applications). Nanotechnology safety and the environment. Impact of nanotechnology on society and industry **[7 LectureHours]**

Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:

Components	Class Assessment	End Term
Weightage (%)	50	50

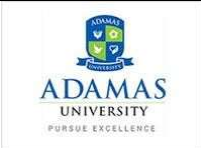
Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
Avg	3	3	2	3	3	-	-	3	-	-	1	3

1=weakly mapped

2= moderately mapped

3=strongly mapped

Name: XXXXXXXXXX			
Enrolment No:			
Course: BIC22518 – DSE-I Nano-biotechnology (THEORY) Program: M.Sc. Biochemistry Semester: Even2020-21			
		Time: 03Hrs. Max. Marks:50	
Instructions: Attempt any four questions from Section A (each carrying 5 marks); any two questions from Section B (each carrying 10 marks).			
SECTION A (Attempt any Four questions) (5X4=20)			
1.	Define Nanotechnology. Write different modes of classification of Nanomaterials.	2+3	CO1
2.	Explain Chemical Vapor Deposition of Carbon Nanotubes.	5	CO2
3.	Describe the challenges faced by Nanotechnology.	5	CO3
4.	Write short note on (i) Carbon fullerenes (ii) Carbon Nanotubes.	2+3	CO4
5	Analyze the effect of recent advancement of nano-biotechnology in society.	5	CO5
SECTION B (Attempt any 3 questions) (10X3=30)			
6.	Make short note on: (i) Atomic Force Microscopy (ii) Scanning Electron Microscopy	5+5	CO2
7.	Explain in detail Electrical, magnetic, optical, thermal, and mechanical properties of nanostructured materials.	2+2+2+2+2	CO1 CO2
8.	What is nanomedicine? How can nanomaterials be used for targeted drug delivery. Explain your answer.	3+2+5	CO4
9	Describe synthesis of Nanoparticles through Homogenous and Heterogenous nucleation.	5+5	CO4 CO5

BIC21509	DSE-I Drug design and development (THEORY)	L	T	P	C
Version 1.0	Contact Hours - 45	3	0	0	3
Pre-requisites/Exposure	BSc. Level Chemistry and Biochemistry Knowledge				
Co-requisites	--				

Course Objectives

- To provide students basic idea about classification of drugs, drug targets and drug action.
- It will also provide in depth knowledge about identification of drug targets; fundamentals of receptor-ligand interactions; concept of structure–activity relationship.
- To deliver detail information about pharmacophores, antagonist, agonist, prodrugs, pharmacokinetics and pharmacodynamics.
- To provide students recent advancement in drug metabolism.

Course Outcomes

On completion of this course, the students will be able to:

CO1 Remembering-Identify different stages of drug development process.

CO2 Understanding-Explain the principles and techniques of drug design and development.

CO3 Applying-Utilize computational tools and software for drug design.

CO4 Analysing-Evaluate the efficacy and safety of potential drug candidates.

CO5 Evaluate-Assess strategies for optimizing drug design process and design a comprehensive drug development plan for a specific target.

Catalog Description

The core-course of ‘Drug design and development’ will help to define fundamental knowledge about classification of drugs, drug targets and drug action. This course includes comprehensive approach through studying different drug targets; fundamentals of receptor- ligand interactions; concept of structure–activity relationship. Furthermore, the implication of recent advancement in drug metabolism and its impact in society will also be illuminated. All the lectures will be devoted on discussions of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as power point presentation, audio visual virtual lab session as per requirement. The tutorials will enable the students with problem-solving ability led by the course coordinator. Students will perceive the basic concepts of the subject via exercise and discussions with the coordinator.

Course Content

DSE-I Drug design and Development

1. Introduction: classification of drugs, drug targets and drug action, concepts of drug dosing, half-life, tolerance, potency, physical dependence, and therapeutic index (LD-50 & CD-50). [9 LectureHours]
2. Definition of terminology: pharmacophores, lead, antagonist, agonist, prodrugs, pharmacokinetics and pharmacodynamics. [9 LectureHours]
3. Drug discovery: Identification of drug targets; Fundamentals of receptor-ligand interactions; Concept of structure –activity relationship (SAR & QSAR). Pharmacogenomics and pharmacogenetics, Toxicogenomic; Metagenomics and drug development. [9 LectureHours]
4. Drug metabolism (Biotransformation of drugs): Definition, classification and mechanism of action of common antibiotics, antivirals, antifungals, local anti-infective drugs, Sulfa drugs, aspirin, paracetamol. [9 LectureHours]
5. Logic of Drug design and drug action: Drugs based on targeting enzyme inhibition, nucleic acids (Alkylating agents and intercalating agents), metabolic diseases and Endocrine function and psychopharmacological agents. Definition, classification and mechanism of action of commonly used drugs: Analgesics, nonsteroidal anti-inflammatory drugs (NSAIDs) as pain killers, Antineoplastic agents, Antihistamines, hormone drugs, steroidal drug, Cardiovascular drugs, anaesthetic drugs, antidepressants, antipsychotics, prostaglandin synthesis. Drug resistance and mechanism. [9 Lecture Hours]

References Books:

1. Madsen U, Krosggaard-Larsen P, Liljefors T (2002). Textbook of Drug Design and Discovery. Washington, DC: Taylor & Francis. ISBN978-0-415-28288-8.
2. Reynolds CH, Merz KM, Ringe D, eds. (2010). Drug Design: Structure- and Ligand-Based Approaches (1 ed.). Cambridge, UK: Cambridge University Press. ISBN978-0521887236.
3. Wu-Pong S, Rojanasakul Y (2008). Biopharmaceutical drug design and development (2nd ed.). Totowa, NJ Humana Press: Humana Press. ISBN978-1-59745-532-9.
4. R. B. Silverman & M. W. Holladay 'The Organic Chemistry of Drug Design and Drug Action', 3rd Edition; Burlington : Elsevier Science, 2014. ISBN:978-0-12-382030-3.
5. DenizEkinhi 'Medicinal Chemistry and Drug Design' Croatia, InTech, ISBN 978-953-51- 0513-8.
6. Madsen U, Krosggaard-Larsen P, Liljefors T (2002). Textbook of Drug Design and Discovery. Washington, DC: Taylor & Francis. ISBN978-0-415-28288-8.
7. Reynolds CH, Merz KM, Ringe D, eds. (2010). Drug Design: Structure- and Ligand-Based Approaches (1 ed.). Cambridge, UK: Cambridge University Press. ISBN978-0521887236.

Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination
Examination Scheme:

Components	Class Assessment	End Term
Weightage (%)	50	50

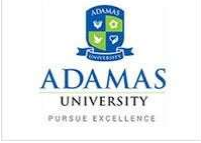
Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
Avg	3	3	2	3	3	-	-	3	-	-	1	3

1=weakly mapped

2= moderately mapped

3=strongly mapped

Name: XXXXXXXXXX			
Enrolment No:			
Course: BIC22519 – DSE-I Drug design and development (THEORY) Program: M.Sc. Biochemistry Time: 03Hrs. Semester: Even2020-21 Max. Marks:50			
Instructions: Attempt any four questions from Section A (each carrying 5 marks); any two questions from Section B (each carrying 10 marks).			
SECTION A (Attempt any Four questions) (5X4=20) (5X4=20)			
1.	What is QSAR? Discuss how Hansch developed a correlation between numerical descriptor of physicochemical properties and biological activities in his correlation analysis.	U,AN	CO1
2.	Describe receptor theories of drug action.	U	CO2
3.	What is the need for prodrug design? Which example define the benefits of prodrug over routine drugs?	U	CO3
4.	Give a brief overview of solid phase synthesis. Explain its utility in drug discovery.	R	CO4
5.	What are the various drug receptor interaction involved for drug activity? Explain the effect of voltage gated and ion channels in drug receptor interaction.	R	CO5
SECTION B (Attempt any 3 questions) (10X2=20) (10X2=20)			
6.	How will you approach to design an enzyme inhibitor based on understanding of binding pockets and active sites?	AN	CO2
7.	What are the statistical test used for the validation of QSAR equation? Explain all of them.	R	CO1 CO2
8.	Discuss i) Computer in drug design, ii) 3D pharmacophore, iii) Ab-initio methods.	R,AN	CO4
9.	Write the generic, empirical force field equation and explain the significance of various terms in energy calculation.	AP	CO4 CO5

BIC22510	DSE-I Food and dairy: food safety and quality control (THEORY)	L	T	P	C
Version 1.0	Contact Hours - 45	3	0	0	3
Pre-requisites/Exposure	BSc. Level Chemistry and Biochemistry Knowledge				
Co-requisites	--				

Course Objectives

- To provide students basic idea about Food as substrate for microorganisms and about the intrinsic and extrinsic factors affecting growth of microbes.
- It will also provide in depth knowledge about sources of food contamination and spoilage and also the principles of food spoilage.
- To deliver detail information about principles and methods of food preservation.
- To provide students Good Hygiene Practices, Sanitation in manufacture and retail trade.

Course Outcomes

On completion of this course, the students will be able to:

- CO 1 Remembering-Define** the fundamental knowledge about Food as substrate for microorganisms and about the intrinsic and extrinsic factors affecting growth of microbes.
- CO 2 Understanding-Identify** sources of food contamination and spoilage and also the principles of food spoilage.
- CO 3 Applying-Explain** principles and methods of food reservation.
- CO 4 Analysing-Analyze** Food-borne infections and intoxication.
- CO 5 Evaluate-Describe** Good Hygiene Practices, Sanitation in manufacture and retail trade.

Catalog Description

The core-course of 'Food and dairy: food safety and quality control' will help to define fundamental knowledge about Food as substrate for microorganisms and about the intrinsic and extrinsic factors affecting growth of microbes. This course includes comprehensive approach through studying sources of food contamination and spoilage and also the principles of food spoilage. Furthermore, the implication of recent advancement in principles and methods of food preservation will also be illuminated. All the lectures will be devoted on discussions of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as power point presentation, audio visual virtual lab session as per requirement. The tutorials will enable the students with problem-solving ability led by the course coordinator. Students will perceive the basic concepts of the subject via exercise and discussions with the coordinator.

Course Content

DSE-I FOOD AND DAIRY: FOOD SAFETY AND QUALITY CONTROL

- 1. Scope of food microbiology and biotechnology.** Food as substrate for microorganisms, intrinsic and extrinsic factors affecting the growth of microbes, important microorganisms in food (moulds, yeasts and bacteria) and their source (air, soil, water, plants and animals). [7 LectureHours]
- 2. Proximate composition of food.** Sources of food contamination and spoilage. Principles of food spoilage; spoilage of cereals, sugar products, vegetables, fruits, meat and meat products, milk and milk products, fish and sea food, poultry; spoilage of canned food; conventional and modern methods for detection of spoilage and characterization. [7 Lecture Hours]
- 3. Importance of food Preservation,** Principles and methods of food preservation - Physical (temperature, irradiation, drying, canning, processing for heat treatment-D, Z and F values) Chemical (Organic acids, food additives. Class I and Class II preservatives), Bio preservation. [6 LectureHours]
- 4. Food Packaging-** Types of packaging materials, properties and benefits. Other methods of preservation- curing, pickling, smoking, fermentation, addition of chemical preservatives, high pressure processing, hurdle technology. [6 LectureHours]
- 5. Food-borne infections and intoxication:** Bacterial- Brucella, Bacillus, Clostridium, Campylobacter, Escherichia, Listeria, Vibrio; Food intoxication- Botulism, Staphylococcal. Mycotoxins & their types – aflatoxins, ochratoxins, fumonisins, trichothecenes, zearalenone, ergot alkaloids. Laboratory testing procedures. Preventive measures. 6 LectureHours]
- 6. SCP-** Nutritional & therapeutic importance, Quorn and SCO and their Industrial production. Dairy food (cheese, srikhand). Production procedure of Kefir, Yogurt, Acidophilus milk; Probiotics, Prebiotics and Symbiotic. Nutraceuticals, functional food and their quality standards. Application of fungal pigments in food industry. 7 Lecture Hours]
- 7. Food and sanitation:** Good Hygiene Practices, Sanitation in manufacture and retail trade; food control agencies and their regulation, hazard analysis and critical control points (HACCP); GMP, quality control. Recent trends and development in food technologies in India. 6 LectureHours]

References Books:

1. Introduction to food biotechnology / Perry Johnson-Green. Latest edition.
2. James, M. J. Martin, J. Loessner, and David, A.G. (2006) Modern food microbiology (7th ed.)
3. John S Norak, Gerald M Sapers, Vijaya Kumar Juneja, Daniel K Gay. (2002), .Microbial Safety of Minimally Processed Foods. 1st Edition. CRC Press.

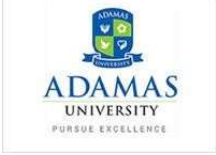
Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination
Examination Scheme:

Components	Class Assessment	End Term
Weightage (%)	50	50

Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
Avg	3	3	2	3	3	-	-	3	-	-	1	3

1=weakly mapped
2= moderately mapped
3=strongly mapped

Name:			
Enrolment No:			
Course: BIC22520 – DSE-I Food and dairy: Food safety and quality control (THEORY) Program: M.Sc. Biochemistry Semester: Even2020-21			
		Time: 03Hrs. Max. Marks:50	
Instructions: Attempt any four questions from Section A (each carrying 5 marks); any two questions from Section B (each carrying 10 marks).			
SECTION A (Attempt any Four questions) (5X4=20) (5X4=20)			
1.	Explain food sanitation. Define botulism.	U	CO1
2.	Discuss about different milk borne diseases.	R	CO2
3.	Describe food infection by Salmonella Typhimurium.	U	CO3
4.	Identify source of food spoilage microorganisms.	R	CO4
5.	Define Brucella ring test and test for mastitis.	U	CO5
SECTION B (Attempt any 3 questions) (10X3=30)			
6.	Explain the spoilage of meat and poultry products. Describe food prevention by canning.	R	CO2
7.	Give the steps involved in commercial sterilization process in industrial canning. Which are the important spoilage organisms of acid foods in cans?	R, AN	CO1 CO2
8.	Which organism spoils pasteurized milk, chicken, fish and bread?	R	CO4
9.	How will you determine the microbial loading in a food sample? What are practical rules for good sanitation?	U	CO4 CO5

BIC21522	Immunology	L	T	P	C
Version 1.0	Contact Hours - 45	3	0	0	3
Pre-requisites/Exposure	Graduate level degree in biology or relevant area				
Co-requisites	--				

Course Objectives:

1. To provide basic understanding of our immune system and its medical implication.
2. To provide basic understanding of the activation, mechanism and regulation of the immune system and Host pathogen interaction.
3. To understand how an altered signaling pathways of the immune system lead to Immune disorder.

Course Outcomes

On completion of this course, the students will be able to

CO1

- Recognize the components and steps of the immune response

CO2

- Explain the interaction between antigens and antibodies

CO3

- Apply knowledge of antigens and antibodies in diagnosing and treating immunological disorders

CO4

- Evaluate the effectiveness of different antigens in eliciting immune responses

CO5

- Assess the effectiveness of immune responses in combating infections and diseases

Course Description:

Immunology and Medical Biotechnology course will provide an advanced understanding of the principles and mechanisms of the immune system and immune responses in the context of infection, malignancy and immunological disorders. All the lectures will be devoted on discussions of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as power point presentation, audio visual virtual lab session as per requirement. The tutorials will familiarize the students with

practical problem-solving techniques led by the course coordinator. Students will strongly grab the basic concepts of the subject via exercise and discussions with the coordinator.

Course Content:

IMMUNOLOGY (BIC21522)

Unit I 10 LectureHours

Immunology- fundamental concepts and anatomy of the immune system

Components of innate and acquired immunity; Phagocytosis; Complement and Inflammatory responses; Haematopoiesis; Organs and cells of the immune system- primary and secondary lymphoid organs; Lymphatic system; Lymphocyte circulation; Lymphocyte homing; Mucosal and Cutaneous associated Lymphoid tissue. (MALT&CALT); Mucosal Immunity; Antigens - immunogens, haptens; Major Histocompatibility Complex - MHC genes, MHC and immune responsiveness and disease susceptibility, HLA typing

Unit II 10 LectureHours

Immune responses generated by B and T lymphocytes

Immunoglobulins-basic structure, classes & subclasses of immunoglobulins, antigenic determinants; Multigene organization of immunoglobulin genes; B-cell receptor; Immunoglobulin superfamily; Principles of cell signalling; Basis of self –non-self-discrimination; Kinetics of immune response, memory; B cell maturation, activation and differentiation; Generation of antibody diversity; T-cell maturation, activation and differentiation and T-cell receptors; Functional T Cell Subsets; Cell-mediated immune responses, ADCC; Cytokines-properties, receptors and therapeutic uses; Antigen processing and presentation- endogenous antigens, exogenous antigens, non-peptide bacterial antigens and super-antigens; Cell-cell co-operation, Hapten-carrier system

Unit III 10 LectureHours

Antigen-antibody interactions

Precipitation, agglutination and complement mediated immune reactions; Advanced immunological techniques - RIA, ELISA, Western blotting, ELISPOT assay, immunofluorescence, flow cytometry and immunoelectron microscopy; Surface Plasmon resonance, Biosensor assays for assessing ligand – receptor interaction, CMI techniques- lymphoproliferation assay, Mixed lymphocyte reaction, Cell Cytotoxicity assays, Apoptosis, Microarrays, Transgenic mice, Gene knock outs

Unit IV 15 LectureHours

Clinical Immunology

1. Pathogenic infection: Normal human microflora. Recognition and entry processes of different pathogens like bacteria viruses into animal and plant host cells, Virulence factors and pathogenicity islands, alteration of host cell behaviour by pathogens. Immunity to Infection : Bacteria, viral, fungal and parasitic infections (with examples from each group) Hypersensitivity – Type I-IV; Autoimmunity; Types of autoimmune diseases; Mechanism and role of CD4+ T cells; MHC and TCR in autoimmunity; Treatment of autoimmune diseases; Transplantation – Immunological basis of graft rejection; Clinical transplantation and immunosuppressive therapy; Tumor immunology – Tumor antigens; Immune response to tumors and tumor evasion of the immune system, Cancer immunotherapy; Immunodeficiency-Primary immunodeficiency, Acquired or secondary immunodeficiency.

2. Vaccine technology: Active and passive immunization; Live, killed, attenuated, sub unit vaccines; Role and properties of adjuvants, recombinant DNA and protein based vaccines, plant-based vaccines, reverse vaccinology; Peptide vaccines, conjugate vaccines; Antibody genes and antibody engineering- chimeric and hybrid monoclonal antibodies; Catalytic antibodies and generation of immunoglobulin genelibraries.

Reference Books

Suggested Books:

1. Kuby Immunology by Judy Owen, Jenni Punt, Sharon Stranford,2013
2. Roitt's Essential Immunology (Essentials) by Ivan M. Roitt,2016
3. Medical Microbiology & Immunology by Warren Levinson,2004
4. Basic and Clinical Immunology by MarkPeakman

Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:

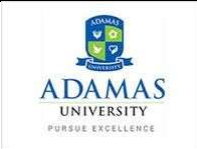
Components	Class Assessment	End Term
Weightage (%)	50	50

Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
Avg	3	3	2	3	3	-	-	3	-	-	1	3

1=weakly mapped
2= moderately mapped
3=strongly mapped

Model Question Paper

Name:		
Enrolment No:		
Course: BIC21522 – IMMUNOLOGY		Time: 03Hrs. Max. Marks:50
Program: M.Sc. Biochemistry		
Semester: Odd 2020-2021		
Instructions:		
Attempt all the questions from Section A (each carrying 4 marks); all the questions from Section B (each carrying 6 marks).		
SECTION A (Attempt all the questions)		
1.	<p>(i) All of the following are true of antigen EXCEPT which one of the following? (U)(R)</p> <p>A. They contain epitopes. B. They contain antigenic determinants. C. They can elicit an immune response. D. They contain paratopes</p> <p>(ii) A 27-year-old housewife presents to her family physician with a 3month history of fatigue, and a facial rash which is aggravated by sun exposure. Laboratory tests reveal anemia of 10-gm/dl (normal is 12-14) and immune testing reveals auto-antibodies against DNA. What is the most likely diagnosis? (An) (Ap)(R)</p> <p>A. Rheumatoid Arthritis B. Myasthenia Gravis C. Systemic Lupus Erythematosus D. Graves' Disease</p> <p>(iii) A Type IV hypersensitivity reaction is characterized by: (U) (R)</p> <p>A. Neutrophil infiltrate B. Participation by complement C. T lymphocyte infiltration D. Cytotoxic antibody</p> <p>(iv) The major role of the complement system is to work in conjunction with (U)</p> <p>A. antibodies to lyse cells via the C8 and C9 components B. the major histocompatibility complex for cell recognition C. antibodies to opsonize cells D. the T-cell receptor for production of lymphokines</p>	<p>U</p> <p>AN</p> <p>U</p> <p>U</p> <p>AP</p>
		CO1, CO2, CO3, CO5

2.	Explain , how does immune specificity fit with non-specific cytokines? List the functions of cytokine? (U) (R)	AP	CO2, CO3			
3.	Demonstrate the underlying mechanism of Grave’s disease. Use diagram to explain . Discuss about mechanism of hypersensitive reaction in ABO blood incompatibility. (U)(Ev)	U,R	CO1			
4.	Define polygenism and polymorphism in context of MHC. How is polymorphism beneficial for the population? (R) (U)	U	CO2, CO4			
SECTION B (Attempt any 3 questions) (10X3=30)						
5.	(a) Distinguish between attenuation and inactivation. (2)(An) (b) Discuss the drawbacks of passive immunization. (3)(Cr) (c) Why MMR vaccine is not given before 12 to 15 months of age? List the disadvantages of attenuated bacterial or viral vaccines. (2+3)(R)	R,U,	CO4			
6.	You have prepared knockout mice with mutations in the genes that encode various complement components. Each knockout strain cannot express one of the complement components listed across the top of the table below. Predict the effect of each mutation on the steps in complement activation and on the complement effector functions indicated in the table using the following symbols: NE – no effect; D- process/function decreased but not abolished; A- process/function abolished. Justify your answer for each. (Ev) (Cr)	AP	CO3, CO4			
		Complement component knocked out				
Function		C1q	C4	C3	C5	Factor B
Formation of classical pathway C3 convertase						

BIC 21536	MICROBIOLOGY	L	T	P	C
Version 1.0	Contact Hours - 45	3	0	0	3
Pre-requisites/Exposure	Graduate in any discipline of Biology				
Co-requisites	--				

Course Objectives:

1. To gain a deeper understanding of the scope, evolution, history and developments in the field of Microbiology
2. To be able to distinguish between cellular structures of prokaryotes and eukaryotes
3. To appreciate microbial diversity in the world
4. To be able to discern between applications of Microbiology in diverse areas

Course Outcomes

On completion of this course, the students will be able to

CO1.

Remembering-Recall and analyse the contributions of various microbiologists in shaping the field of Microbiology

CO2.

Understanding-Compare and contrast structures and functions between prokaryotic and eukaryotic cells

CO3.

Analysing-Explore the diversity of microbial world

CO4.

Applying-Enlist and describe the scope of microbiology

CO5.

Evaluate-Evaluate the different aspects of microorganisms.

Catalogue Description:

Introduction to Microbiology and Microbial Diversity introduces learners and students to the exciting world of Microbiology and covers history, scope, applications in the field of Microbiology.

Course Content:

Unit 1 Introduction to Microbiology (15 h)

Definition of Microbes; Categories of Microbes; Evolution and classification of Microbes; Overview of history of Microbiology: Biogenesis and abiogenesis. Contributions of Redi, Spallanzani, Needham, Pasteur, Lister, Koch, Jenner and Flemming. Scope of Microbiology. Notable contributions in the development of Microbiology: i) Spontaneous generation (abiogenesis). ii) Biogenesis. iii) Germ Theory of Disease. iv) Koch's Postulates. Development of the field of soil microbiology: Contributions of Martinus W. Beijerinck, Sergei N. Winogradsky, Selmán A. Waksman. Establishment of fields of medical microbiology and immunology through the work of Paul Ehrlich, Elie Metchnikoff, Edward Jenner. Technological Microbiology and contributions of Ananda Chakraborty and patenting. Role of Warner Arber, Hamilton Smith, Daniel Nathans in the discovery of restriction enzymes. Contributions of Kary Mulis and Carl Woese.

Unit 2 Prokaryotic cell structure and function (15 h)

Structure and function; Cell envelope: Plasma membranes; Cell Wall and types. Components external

to cell envelope: Capsule, Slime Layer, S Layer, Pili, Fimbriae and Flagella. Components internal to the Cell envelope: Cytoplasmic matrix, Inclusion bodies, Ribosome; Bacterial chromosomes and plasmids; Bacterial endospores and their formation.

Unit 3 Diversity of Microbial World (25 h)

A. Systems of classification

Binomial Nomenclature, Whittaker’s five kingdom and Carl Woese’s three kingdom classification systems and their utility. Difference between prokaryotic and eukaryotic microorganisms

B. General characteristics of different groups: Acellular microorganisms (Viruses, Viroids, Prions) and Cellular microorganisms (Bacteria, Algae, Fungi and Protozoa) with emphasis on distribution and occurrence, morphology, mode of reproduction and economic importance.

• **Algae**

History of phycology with emphasis on contributions of Indian scientists; General characteristics of algae including occurrence, thallus organization, algae cell ultra-structure, pigments, flagella, eyespot food reserves and vegetative, asexual and sexual reproduction. Different types of life cycles in algae with suitable examples: Haplobiontic, Haplontic, Diplontic, Diplobiontic and Diplohaplontic life cycles. Applications of algae in agriculture, industry, environment and food.

• **Fungi**

Historical developments in the field of Mycology including significant contributions of eminent mycologists. General characteristics of fungi including habitat, distribution, nutritional requirements, fungal cell ultra-structure, thallus organization and aggregation, fungal wall structure and synthesis, asexual reproduction, sexual reproduction, heterokaryosis, heterothallism and parasexual mechanism. Economic importance of fungi with examples in agriculture, environment, industry, medicine, food, biodegradation and mycotoxins.

• **Protozoa**

General characteristics with special reference to Amoeba, Paramecium, Plasmodium, Leishmania and Giardia

Unit 4 An overview of Scope of Microbiology (5h)

Different applications of microbiology in various industries related to food, agriculture, chemical and fuels, environment, medical and materials. ,

Text Books

T1. Willey, J.M.; Sherwood, L.; Woolverton, C.J. *Prescott's microbiology*. McGraw-Hill: 2016

Reference Books

R1

R2

Modes of Examination: Assignment/Quiz/Project/Presentation/Written Exam

Examination Scheme:

Components	Internal	Mid Term	End Term
Weightage (%)	30	20	50

Course Outcomes for BIC 21528

Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

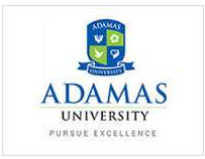

CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
Avg	3	3	2	3	3	-	-	3	-	-	1	3

1=weakly mapped

2= moderately mapped

3=strongly mapped

Model Question Paper

Name: Enrolment No:			
Course: BIC21528 -MICROBIOLOGY Program: M.Sc. BIOCHEMISTRY Semester: Odd 2020-21		Time: 03 Hrs.	
Instructions: Attempt all questions from Section A (each carrying 4 marks) and all questions from Section B (each carrying 6 marks).			
Section A (Attempt all questions)			
1.	How do the components of bacterial envelope differ from the components of archaeal cell walls? (R, An)	4	CO2
2.	Contrast between differential and selective medium giving an example of each. (R, U)	4	CO3
3.	List down the Koch's postulates. (R, U)	4	CO1
4.	Discuss how the piece of equipment shown in the image below was used to disprove the theory of spontaneous generation. (E, An)	4	CO1
			
5.	Why did agar offer an improvement over gelatin for the growth of microorganisms? (E)	4	CO1
SECTION B (Attempt all questions)			
6.	Write names of two microbes that harbor multiple chromosomes. What are the major types of bacterial plasmids? Mention an example each type of plasmid. (R, U)	6	CO2
7.	Illustrate the scientific method (include observation, hypothesis, experimental design, results and interpretation of results) as applied to Jenner's experiment on vaccination. (An, C)	6	CO1
8	How would you visualize flagella in a compound microscope? Outline the different arrangements of flagella with an example of each. (E, R)	6	CO6
9	Schematically differentiate between flagella of gram positive and Gram negative bacteria. (R,	6	CO3

	An)		
10	Discuss the four major categories of medically relevant fungi using a table (R, U)	6	CO4

BIC 22549	Microbiology Lab	L	T	P	C
Version 1.0	Contact Hours - 60	0	0	4	2
Pre-requisites/Exposure	UG level Biology				
Co-requisites	--				

Course Objectives:

1. To provide hands on training on how to work safely in a Microbiology lab
2. To acquaint working principle of different instruments in a Microbiology lab
3. To make students learn on how to properly handle and care for lab microscope
4. To make students learn about growing and isolating microbes in the laboratory
5. Hands on training of various staining procedure including simple, negative and gram staining.

Course Outcomes

On completion of this course, the students will be able to

CO1: Remembering: To Identify different bacterial and viral species based on their unique characteristics.

-

CO2. Understanding: To Interpret the results of various laboratory tests used to identify bacteria and viruses.

-

CO3. Applying: To Perform aseptic techniques and handle microbial cultures safely in a laboratory setting. Apply molecular techniques such as PCR and gene cloning in the study of microbial genetics.

CO4. Analysing: To Critically evaluate the significance of microbial interactions in various environments.

-

CO5. Evaluate: Evaluate the reliability and validity of experimental results obtained in the laboratory. Design experiments to investigate specific research questions in bacteriology, virology, and microbial genetics.

Catalogue Description:

Introduction to Microbiology and Microbial Diversity Lab introduces learners and students to the exciting world of Microbiology lab and covers lab safety, instrumentation used in Microbiology lab, microscope handling and staining techniques to visualize microbes in the lab.

Course Content:

1. Lab safety

(5h)

2. To know the principles and mode of operation of various instruments in Microbiology lab including Microscope, Laminar air flow, autoclave, biological incubator, weighing balance, pH meter

(5h)

3. Microbiological media preparation

(3

h)

4. Aseptic techniques

5. Environmental sampling of microbes (3 h)
- h)
6. Isolation of microbes using streak plate, and pour plates. (3)
- h)
7. Enumeration of microbes using spread plate method. (3)
8. Simple staining (3h)
9. Negative staining (3h)
10. Fungal staining (3h)
11. Use of temporary mounts to study *Spirogyra* and *Chlamydomonas* and *Volvox* (3h)
12. Demonstration of permanent mounts/photographs of *Amoeba*, *Entamoeba*, *Paramecium* and *Plasmodium* (5h)
- (5h)

Text Books

T1. Cappucino J and Sherman N (2010). Microbiology: A Laboratory Manual, 9th edition, Pearson Education

Modes of Examination: Assignment/Quiz/Project/Presentation/Written Exam

Examination Scheme:

Components	Internal	End Term
Weightage (%)	50	50

Course Outcomes

Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

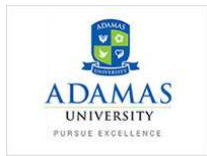
CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	3	3	2	3	3	3	3	3	1	3	3	3
CO2	3	3	2	3	3	3	3	3	1	3	3	3
CO3	3	3	2	3	3	3	3	3	1	3	3	3
CO4	3	3	2	3	3	3	3	3	1	3	3	3
CO5	3	3	2	3	3	3	3	3	1	3	3	3
Avg	3	3	2	3	3	3	3	3	1	3	3	3

1=weakly mapped

2= moderately mapped

3=strongly mapped

Model Question Paper

Name: Enrolment No:			
Course BIC 22548 –Microbiology Lab Program: M.Sc. BIOCHEMISTRY Time: 03 Hrs. Semester: Odd 2020-21			
Instructions: Attempt all questions from Section			
Section A (Attempt all questions)			
1.	State the principle and method of quadrant streaking. Determine if streaking has been proper in plate marked ___? Identify whether the provided culture is pure? (R, An)	15	CO3
2.	State the principle and method of gram staining. Identify if the given microbial strain is gram positive or negative by performing gram staining. Mention the shape and arrangement of the given sample. (R, Ap)	15	CO4
3.	Lab Notebook and Viva (R, U, An, Ap, E. C)	20	CO1, CO2, CO3, CO4

BIC21550	FORENSIC BIOLOGY (THEORY)	L	T	P	C
Version 1.0	Contact Hours - 45	3	0	0	3
Pre-requisites/Exposure	UNDERSTANDING OF BASIC BIOLOGY				
Co-requisites	--				

Course Objectives

To provide students the basic understanding of forensic biology.
 It will also provide in depth knowledge of forensic science.
 Elaborating biophysical and biochemical techniques for forensics.
 General overview of forensic genetics and advanced DNA forensics.

Course Outcomes

On completion of this course, the students will be able to

CO1.Remembering-Students will be able to **recall** various fundamental components of forensic biology.

CO2Understanding-.Students will be able to **summarise** tools and techniques of forensic biology.

CO3.Analysing-Students will be able to **analyse** different administration and organizational setup.

CO4.Applying-Students will be able to **outline** the forensic genetics.

CO5.Evaluate-Students will be able to **explore** advanced DNA forensics.

Catalog Description

The core-course of 'forensic biology' will help to understand the fundamental components of forensic biology. This course is a step by step journey from the basic to modern concepts of forensic biology. Furthermore, students will be able to summarise tools and techniques of forensic biology. They will be able to outline the forensic genetics and advanced DNA forensics.

Course Content

Forensic Biology (BIC 21550)

I) ELEMENTARY FORENSIC SCIENCE: Definition of Forensic Science, The Role of the Forensic Laboratory, History and Development of Forensic Science in India & Abroad, Pioneers in Forensic Science, Multidisciplinary nature, Forensic Technology solving crimes with advanced technology, Forensic intelligence and Interviews. Administration and Organizational Setup: DFSS, CFSL, GEQD, SFSL, RFSL, MFSL, FPB, NICFS, CDTS, NCRB, BPR&D, Qualifications and duties of Forensic Scientists Academic centres of education and research: Indian and Academy of Forensic Science, American Board of Forensic Odontology, Interpol and FBI, Australian Academy of Forensic Sciences.

II) GENERAL FORENSIC TOOLS AND TECHNIQUES: Meaning and Terminology of Instrumentation; Definition, Need of Instrumentation in Forensic Science, Qualitative and quantitative methods of analysis, Destructive and Non-Destructive Methods. Centrifugation Techniques, Basic principles of sedimentation. Theory and basic principles, setup and Forensic applications of Compound, Comparison, Fluorescence, Polarized, Stereo-zoom microscope. Electron Microscopy- Theory and basic principles of Electron Microscopy. Introductory Chromatography: Definition, Chromatographic Techniques, History of Chromatography, Theoretical principles of Chromatography. Forensic Toxicology, Serology & Microbiology.

III) FORENSIC GENETICS: Concepts of Human Genetics; DNA Profiling: Introduction, History of DNA Typing, molecular biology of DNA, variations, polymorphism, SNPs.

IV) ADVANCED DNA FORENSICS:

DNA Extraction-Organic and Inorganic extraction, Comparison of Extraction methods, Commercial kits DNA typing systems- RFLP analysis, PCR amplifications, sequence polymorphism. Analysis of SNP, YSTR, Mitochondrial DNA, Ancient DNA typing, Evaluation of results. Forensic Significance of DNA profiling, New and future technologies: DNA chips, SNPs and limitations of DNA profiling.

Textbook:

Forensic Biology By Richard Li. 2nd Edition, CRC Press. Taylor & Francis Group.
Essential Forensic Biology, 3rd edition. Alan Gunn. ISBN: 978-1-119-14140-2. WILEY.


Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination

Components	Class Assessment	End Term
Weightage (%)	50	50

Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
Avg	3	3	2	3	3	-	-	3	-	-	1	3

1=weakly mapped
2= moderately mapped
3=strongly mapped

Name:			
Enrolment No:			
Course: BIC 21550 – FORENSIC BIOLOGY (THEORY)			
Program: M.Sc. Biochemistry		Time: 03 Hrs.	
Semester: Odd			
Instructions:			
Attempt any four questions from Section A (each carrying 5 marks); any two questions from Section B (each carrying 10marks).			
SECTION A (Attempt any Four questions)			
1.	What is forensic intelligence? Explain with a suitable example.	2+3	CO1
2.	Classify Administration and Organizational Setup.	5	CO2
3.	Describe non-descriptive method of forensics.	5	CO3
4.	Enlist 3 important organizational set up.	5	CO3
5	Explain forensics toxicology and serology .	3+2	CO1
SECTION B (Attempt any Two questions)			

6.	Illustrate the technique of electron microscopy. Describe history of DNA forensics.	4+2+3+1	CO3
7.	Explain analysis of hair in forensic. Outline the methods of forensic sampling	2+4+2+2	CO1 CO2
8.	Illustrate the role of Forensic applications of Compound? Compare between motifs and domain with example. Where do you find triple helix? Explain briefly.	2+2+3+3	CO1 CO2
9	Outline the principles advanced DNA forensics. Analyse the roles of different chromatographic techniques by briefly describing their principle.	2+3+3+2	CO4 CO3

BIC 21551	NUTRITION AND TOXICOLOGY (THEORY)	L	T	P	C
Version 1.0	Contact Hours - 45	3	1	0	4
Pre-requisites/Exposure	BASIC UNDERSTANDING OF BIOCHEMISTRY				
Co-requisites	--				

Course Objectives

To provide students the basic understanding of nutrition, energy metabolism and toxicology. It will also provide in depth knowledge of functional aspects food and drug interactions with nutraceuticals. Elaborating dietary components of health and diseases. General overview of nutritional and toxicological assessment .

Course Outcomes

On completion of this course, the students will be able to

- CO1.Students will be able to recall various aspects of nutrition, energy metabolism and toxicology.
- CO2.Students will be able to **summarise** functional features of food and drugs.
- CO3.Students will be able to **analyse** and **choose** between food, drug, nutraceuticals and toxins.
- CO4.Students will be able to **apply** the role of different nutraceuticals.
- CO5.Students will be able to **evaluate** major reasons for disease and toxicology.

Catalog Description

The core-course of ‘Nutritional Biochemistry’ will help to understand the classification, structure and properties of foods and nutraceuticals. Nutritional biochemistry has also helped to reveal facts about how nutrients influence the growth, development, and function of cells and tissues. Therefore, studying the biochemistry of nutrition has a significant real-world impact. It has the potential to greatly influence the future of preventative and therapeutic strategies for mental and physical illness.

Course Content

Nutritional Biochemistry (BIC21551)

Unit I Introduction to Nutrition and Energy Metabolism

Defining Nutrition, role of nutrients. Unit of energy, Biological oxidation of foodstuff. measurement of energy content of food, Physiological energy value of foods, SDA. Measurement of energy expenditure, estimating energy requirements, BMR factors Recommended Nutrient Intakes (RNI) and Recommended Dietary Allowances for different age groups.

Unit II Dietary components and health

Review functions of carbohydrates, lipids, proteins and vitamins. Digestion, absorption. Their classification, sources, functions, digestion, absorption, utilization and storage. Deficiency diseases (Kwashiorkor, Scurvy, Rickets, Xerophthalmia etc.).Minerals (Ca,P,Fe etc.) absorption, importance and deficiency disease.

Unit III Assessment of Nutritional status

Anthropometric measurements; Z scores, BMI, skinfold, circumference ratios. Biochemical assessment; Basal metabolic panel, Comprehensive metabolic panel, CBC, Urine Analysis, Assessment of Anemia, ROS assessment, GTT and glycosylated Hb, Differential diagnosis of B12 and folate.

Unit IV Food, Drug interactions, Nutraceuticals and Toxicology

Nutrient interactions affecting ADME of drugs, Alcohol and nutrient deficiency, Anti-depressants, psychoactive drugs and nutrient interactions, Appetite changes with drug intakes and malnutrition. Food as medicine.

Unit V Toxicology

Classification of Food Toxicants. Food, Law and Safety. Principles of Toxicology I: Exposure, the Dose-Response Curve. Absorption, Distribution and Elimination of Toxicants. Biotransformation Reactions (Phase I & Phase II).Carcinogenesis, Mutagenesis, Teratogenesis. Organ Toxicity.Natural Toxins in Foods of Plant Origin I. Risk Assessment. Pesticides in Foods. Marine,Toxinns Poisonous Mushrooms , Mycotoxins ,Toxicants Resulting from Food Processing I , Food Additives I , Food Adulteration Pesticides .

Textbook:

1. Nutritional Biochemistry 1st Edition, ISBN: 978-93-90699-76-6, Nitya Publication. Dr.Renu Verma.
2. Lehninger: Principles of Biochemistry (2013) 6th ed., Nelson, D.L. and Cox, M.M., W.H. Freeman and Company (New York), ISBN:13: 978-1-4641-0962-1 / ISBN:10:1-4292-3414-8.

Reference books:

- 1.
- 2.

Physic
The T

Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination

Examination Scheme:


Components	Class Assessment	End Term
Weightage (%)	50	50

Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
Avg	3	3	2	3	3	-	-	3	-	-	1	3

1=weakly mapped
2= moderately mapped
3=strongly mapped



Name:			
Enrolment No:			
Course: Nutrition and toxicology (THEORY)			
Program: M.Sc. Biochemistry		Time: 03 Hrs.	
Semester: Odd 2019-20			
Instructions:			
Attempt any four questions from Section A (each carrying 5 marks); any two questions from Section B (each carrying 10marks).			
SECTION A (Attempt any Four questions)			
1.	What are toxins? Explain with a suitable reaction.	2+3	CO1
2.	Illustrate the concept of BMR. Explain the significance.	4+1	CO2
3.	Explain Kwashiorkor and Scurvy with full biochemical explanation.	4	CO3
4.	Enlist 3 important techniques for food analysis. What is the importance of urine analysis.	2+3	CO3
5	Explain the biochemical cause behind the development of thalassemia and sickle cell anemia.	3+2	CO1
SECTION B (Attempt any Two questions)			
6.	Explain dose-response curve. Name some marine toxicants and summarize their mode of action. Name one food additive extensively used in food industry.	4+2+3+1	CO3
7.	Summarize any four Anthropometric measurements which are useful for nutritional assessment.	2.5 x 4	CO1 CO2
8.	Illustrate the role of ADME studies for the development of drugs in industry. Explain the nutritional basis of different minerals in food.	5+5	CO1 CO2
9	Outline the principles of Edmann degradation and solid phase peptide synthesis. Analyse the roles of different chromatographic techniques by briefly describing their principle.	2+3+3+2	CO4 CO3

BIC 22552	FORENSIC BIOLOGY LAB	L	T	P	C
Version 1.0	Contact Hours - 45	0	0	4	2
Pre-requisites/Exposure	UNDERSTANDING OF BASIC BIOLOGY				
Co-requisites	--				

Course Objectives

To provide students hands on understanding of forensic biology.

It will also provide in depth practical knowledge of forensic science.

Elaborating biophysical and biochemical techniques for forensics.

General overview of forensic serology and advanced DNA forensics.

Course Outcomes

On completion of this course, the students will be able to

CO1.Students will be able to **recall** various fundamental components of forensic biology.

CO2.Students will be able to **understand** tools and techniques of forensic biology.

CO3.Students will be able to apply different DNA techniques.

CO4.Students will be able to **analyse** the forensic serology techniques.

CO5.Students will be able to **explore** advanced DNA forensics.

Catalog Description

The core-course of ‘forensic biology lab’ will help to understand the fundamental components of forensic biology. This course is a hands on journey from the basic to modern concepts of forensic biology. Furthermore, students will be able to summarise tools and techniques of forensic biology. They will be able to outline the forensic genetics and advanced DNA forensics.

Course Content

Forensic Biology Lab

1. To prepare gel plates for electrophoresis.

2. Organic extraction of DNA from blood.
3. Extraction of DNA from other body fluids.
4. Quantification of DNA
5. PCR for DNA samples.
6. Serology Test.

Textbook:

1. Forensic Biology By Richard Li. 2nd Edition, CRC Press. Taylor & Francis Group.
2. Essential Forensic Biology, 3rd edition. Alan Gunn. ISBN: 978-1-119-14140-2.WILEY.

F

E

Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:


Components	Class Assessment	End Term
Weightage (%)	50	50

Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	3	3	2	3	3	3	3	3	1	3	3	3
CO2	3	3	2	3	3	3	3	3	1	3	3	3
CO3	3	3	2	3	3	3	3	3	1	3	3	3
CO4	3	3	2	3	3	3	3	3	1	3	3	3
CO5	3	3	2	3	3	3	3	3	1	3	3	3
Avg	3	3	2	3	3	3	3	3	1	3	3	3

- 1=weakly mapped
 2= moderately mapped
 3=strongly mapped



Name:			
Enrolment No:			
Course: BIC22552 – FORENSIC BIOLOGY LAB (PRACTICAL)			
Program: M.Sc. Biochemistry		Time: 03 Hrs.	
Semester: Odd			
Instructions: Attempt any four questions from Section A (each carrying 5 marks); any two questions from Section B (each carrying 10marks).			
SECTION A (Attempt any Four questions)			
1.	Perform PCR identification of human DNA	10	CO1
2.	Perform microscopic observation of human hair	10	CO2
3.	Describe the importance of DNA forensics.	10	CO3
SECTION B (Attempt any Two questions)			
6.	Lab copy.	10	CO3
7.	Viva Voce.	10	CO1 CO2

BIC 22553	NUTRITIONAL AND TOXICOLOGY LAB (PRACTICAL)	L	T	P	C
Version 1.0	Contact Hours - 60	0	0	4	2
Pre-requisites/Exposure	BASIC KNOWLEDGE OF BIOCHEMISTRY				
Co-requisites	--				

Course Objectives

To provide students with hands-on training in the field of nutritional biochemistry.

To provide in depth knowledge of modern research on nutrition and toxicology.

Students will become more proficient with different practical applications nutrition and toxicology.

Course Outcomes

On completion of this course, the students will be able to

CO1.Remembering-Recall the basic principles of nutritional biochemistry.

CO2.Understanding-Understand biochemical aspects of nutrition and toxicology.

CO3.Applying-Applydifferent assay systems of vitamins, minerals and food adulterant.

CO4.Analysing-Analyse food samples and microbes.

CO5. Evaluate-Evaluate the knowledge of nutritional biochemistry to understand different practical applications.

Catalog Description

The discipline specific course “nutritional biochemistry lab” is a practical paper which has been designed to provide the knowledge of different aspects of nutritional biochemistry. It will provide biochemical & molecular understanding of important processes in nutrition. Students will be able to understand biochemical aspects of nutrition and toxicology. Students will comprehend different assay systems of vitamins, minerals and food adulterant. Apply the knowledge of nutritional biochemistry to understand different practical applications.Students will strongly grab the basic concepts of the subject via exercise and discussions with the coordinator.

Course Content

NUTRITIONAL AND TOXICOLOGY LAB

- 1.Estimation of vitamins. (15 Lectures)
- 2.Estimation of minerals. (10 Lectures)
- 3.Estimation of adulterant in food stuffs. (10 Lectures)
- 4.Tests on Heavy Metal Toxicity. (15 Lectures)
- 5Tests for microbial toxicity. (10 Lectures)

SUGGESTED READINGS

1.A Practical Handbook of Food and Nutrition (2019): ISBN-9789387195660; Shivalik Prakashan. Jayashree Mishra and Pravabati Guru.

2.Practical Forensic Medicine and Toxicology: ISBN: 9789388178846, KK Banerjee. CBS Publishers & Distributors. (2019)

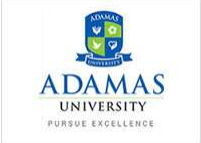
Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:

Components	Class Assessment	End Term
Weightage (%)	50	50

Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	3	3	2	3	3	3	3	3	1	3	3	3
CO2	3	3	2	3	3	3	3	3	1	3	3	3
CO3	3	3	2	3	3	3	3	3	1	3	3	3
CO4	3	3	2	3	3	3	3	3	1	3	3	3
CO5	3	3	2	3	3	3	3	3	1	3	3	3
Avg	3	3	2	3	3	3	3	3	1	3	3	3

1=weakly mapped
 2= moderately mapped
 3=strongly mapped

Name: Enrolment No:		 ADAMAS UNIVERSITY <small>PURSUE EXCELLENCE</small>	
Course: BIC– NURITIONAL BIOCHEMISTRY LAB (PRACTICAL)			
Program: M.Sc. Biochemistry Semester: Even 2019-20		Time: 03 Hrs.	
Instructions: Attempt any two questions from Section A (each carrying 10 marks); Section B is Compulsory (carrying 10 marks).			
Section A (Attempt any Two)			
1.	Write the principle of estimation of ascorbic acid. Perform the experiment.	10	CO1 CO2
2.	Determine different adulterant in food stuffs.	10	CO3 CO4
3.	a) Write the principle behind column chromatography. b) Demonstrate the presence of amylase in germinating seed with a simple experiment.	4 6	CO3 CO2
SECTION B is compulsory			
4.	Viva-voce (U/An/Ap/R/Ev)	10	CO1 CO2 CO3 CO4 CO5
5.	Practical copy(U/Ap/Ev)	10	CO1 CO2 CO3

BIC21513	BIOENERGETICS AND METABOLISM (THEORY)	L	T	P	C
Version 1.0	Contact Hours - 45	3	0	0	3
Pre-requisites/Exposure	GRADUATION IN BIOCHEMISTRY				
Co-requisites	FUNDAMENTAL KNOWLEDGE IN BIOENERGETICS AND METABOLISM				

Course

objectives

- To provide students the basic understanding of laws of thermodynamics in membrane metabolism.
- To provide in depth knowledge of carbohydratemetabolism.
- To outline details of lipidmetabolism.
- To discuss general overview of amino acidmetabolism.

Course Outcomes

On completion of this course, the students will be able to

1. Remembering: Recall the key concepts and principles of metabolism, including bioenergetics, photosynthesis, oxidative phosphorylation, carbohydrate metabolism, lipid metabolism, amino acid metabolism, and nucleic acid metabolism.
2. Understanding: Demonstrate an understanding of the different metabolic pathways utilized to generate energy, synthesize macromolecules, and regulate metabolic processes.
3. Applying: Apply knowledge of metabolism to analyze and interpret experimental data related to bioenergetics, photosynthesis, oxidative phosphorylation, carbohydrate metabolism, lipid metabolism, amino acid metabolism, and nucleic acid metabolism.
4. Analyzing: Evaluate the impact of different environmental factors on metabolism and predict how changes in these conditions can affect metabolic pathways in microorganisms.
5. Creating: Design experiments to investigate specific aspects of metabolism, formulate hypotheses, and propose innovative strategies for optimizing metabolic pathways for practical applications in the fields of biotechnology and medicine.

2.

Catalog Description

The core-course of 'bioenergetics and metabolism' deals with intrinsic laws of thermodynamics in the field of metabolism. The syllabus includes different types of anabolic and catabolic pathways as well

as their relation to our lives. Three main classes of biomolecules i.e. carbohydrates, lipids and protein metabolism have been included in the syllabus for a better understanding of life processes. All the lectures will be devoted on discussions of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as power point presentation, audio visual virtual lab session as per requirement. The tutorials will enable the students with problem-solving ability led by the course coordinator. Students will perceive the basic concepts of the subject via exercise and discussions with the coordinator.

Course Content

Bioenergetics and Metabolism (BIC21511)

Carbohydrate Metabolism [20 Lecture Hours]

1. Survey of metabolism: Carbon, oxygen, nitrogen cycle catabolism, use of mutants and isotopes in the study of metabolism, compartmentation, food chain and energyflow.
2. Glycolysis: Anaerobic pathway of glucose metabolism, two phases of glycolysis. Detailed study of all the reactions, entry of other carbohydrates in Glycolytic pathway, energy balance sheet regulation of glycolytic sequence by enzymes and hormones, alcoholic fermentation.
3. Gluconeogenesis, Reciprocal regulation of glycolysis and gluconeogenesis
4. Glycogen metabolism: Biosynthesis and degradation of glycogen and its regulation. Starch and cellulose biosynthesis.
5. Alternate pathways of carbohydrate metabolism: Pentose phosphate pathway, glyoxalate cycle, glucuronic acid cycle, inter conversion of hexoses, Pasteur effect.

Lipid Metabolism [25 Lecture Hours]

6. Biosynthesis of lipids: Requirements of carbon dioxide and citrate for biosynthesis, fatty acid synthase complex, regulation of biosynthesis. Biosynthesis of triglycerides, cholesterol and phospholipids.
7. Lipid metabolism: Fatty acid metabolism, Beta oxidation of saturated and unsaturated fatty acids, the phases of fatty acid oxidation, energetics of beta oxidation. Oxidation of fatty acids with odd number of carbon atoms, formation of ketone bodies, other types of fatty acid oxidation. Integration of carbohydrate and lipid metabolism.
8. Citric acid cycle: Aerobic pathway of glucose metabolism, historical background, details of the cycle, use of isotope for the study of citric acid cycle, interconversion of hexoses, Pasteur Effect.

Amino Acid Metabolism

9. Biosynthesis of amino acids: amino acid biosynthesis, precursor functions of amino acids, biosynthesis of aromatic amino acids, Histidine, glycine, serine, cysteine, methionine, threonine. Peptides, polyamines, Porphyrins, gamma glutamyl cycle, glutathione biosynthesis, nonribosomal protein biosynthesis.
10. Oxidative degradation of amino acids: Proteolysis, Transamination, oxidative deamination, acetyl CoA, Alpha ketogutarate, acetoacetyl CoA, succinate, fumarate and oxaloacetate pathway, decarboxylation, urea cycle, ammonia excretion.
11. Biosynthesis of Purine and pyrimidine nucleotides, Regulation, Biosynthesis of nucleotide coenzymes. Purine pyrimidine degradation.

Reference books:

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination
Examination Scheme:**

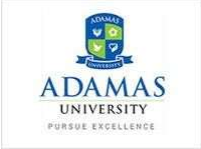
Components	Class Assessment	End Term
Weightage (%)	50	50

Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
Avg	3	3	2	3	3	-	-	3	-	-	1	3

1=weakly mapped
2= moderately mapped
3=strongly mapped



Name:			
Enrolment No:			
Course: BIC21511 -BIOENERGETICS AND METABOLISM (THEORY) Program: M.Sc. Biochemistry Time: 03Hrs. Semester: Even2019-20 Max. Marks:50			
Instructions: Attempt any four questions from Section A (each carrying 5 marks); any two questions from Section B (each carrying 10 marks).			
SECTION A (Attempt any Four questions) (5X4=20) (5X4=20)			
1.	What is cori cycle? Explain the fates of pyruvate.	U	CO2
2.	What do you understand by transamination and oxidative decarboxylation? Explain with proper reaction.	R	CO4
3.	Differentiate between beta oxidation of saturated and unsaturated fatty acids.	R	CO3
4.	Enlist different types of isotopes used in the dissection of a metabolic pathway.	U	CO3
5.	Outline the mechanism gluconeogenesis briefly with its regulatory steps.	AN	CO2
SECTION B (Attempt any 3 questions) (10X3=30)			
6.	Citric acid cycle is anaplerotic in nature- explain . What is the main enzyme in glycogen degradation? State its regulation. Explain the role of PP pathway in metabolism.	R	CO5,CO1
7.	What are the main regulatory enzymes in the purine and pyrimidine biosynthesis pathways? Explain with reactions. How many ATPs are generated from palmitoyl Co A when it undergoes beta oxidation?	R,AN	CO4 CO3
8.	Illustrate the role of HMGCofA reductase in cholesterol biosynthesis. Why statin group of drugs are used to treat hypercholesterolemia? Describe the glyoxalate cycle with its importance. Why it is a major target of drug development for pathogenic microorganisms.	AN,U,AP	CO2 CO5

BIC22523	IMMUNOLOGY LAB (PRACTICAL)	L	T	P	C
Version 1.0	Contact Hours - 45	0	0	3	2
Pre- requisites/Exposure	Concept of immunology at UG level				
Co-requisites	--				

Course Objectives

1. to demonstrate and interpret different antigen-antibody interactions.
2. to acquaint with various components of the immune system and apply this knowledge in immunodiagnostics.
3. to apply various immunological techniques for clinical and research purpose.
4. to quantify antigen/ antibody in different samples.
5. to identify and demonstrate host pathogen interaction.

Course Outcomes

On completion of this course, the students will be able to

- CO1. Remembering-Recall different antigen-antibody interactions.
- CO2. Understand-Understand different components of immune system in human system
- CO3. Applying-Apply different immunological techniques for research and clinical purposes.
- CO4. Analysing- Estimate amount of antigen/antibody present in different samples
- CO5. Evaluate-Evaluate different host pathogen interactions.

Course Description

The student will be able to use the knowledge obtained to perform and analyze different types of antigen-antibody interaction. Identification of different components of the immune system is possible with the concept obtained. Students will gain the ability to apply different immunological techniques for research and clinical purposes. All the experiments will be based on hands-on training in laboratory setup along with discussions of basic theories and advanced topics for practical implementation of knowledge. Classes will be conducted by hands-on lab training and/or audio visual virtual lab session as per requirement. Students will perceive the basic concepts of the subject via exercise and discussions with the coordinator.

Course Content

IMMUNOLOGY LAB (BIC22523)

1. To study morphological and staining characteristics of lymphocytes, neutrophils, monocytes, eosinophils, and basophils. **[8 Lecture Hours]**
2. To perform immunoelectrophoresis. **[8 Lecture Hours]**
2. To perform radial immunodiffusion assay. **[8 Lecture Hours]**
3. To perform rocket immunoelectrophoresis. **[8 Lecture Hours]**
4. To stain a tissue by immunohistochemical reaction **[8 Lecture Hours]**

5. To study quantitative precipitation assay. [10 LectureHours]
6. Gel Techniques; ELISA; SDS PAGE/Western blot[8 LectureHours]
7. To perform latex agglutination test [10 LectureHours]

Text Book(s)

2. Immunology Lab Manual by Wilmore Weberly,2015
3. Immunology methods manual - The comprehensive source book by Lefkovits. ,1996
4. Manual of clinical laboratory immunology by Rose NR,2002
5. Laboratory Immunology by BradshawLJ.1997

T1. Owen, J.A.; Punt, J.; Kuby, J.; Stranford, S.A. Kuby immunology. W.H. Freeman: 2013.

Reference books

R1. T1. Owen, J.A.; Punt, J.; Kuby, J.; Stranford, S.A. Kuby immunology. W.H. Freeman: 2013.

Modes of Examination: Assignment/Quiz/Project/Presentation/Written Exam

Examination Scheme:


Components	Mid Term	End Term
Weightage (%)	50	50

Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	3	3	2	3	3	3	3	3	1	3	3	3
CO2	3	3	2	3	3	3	3	3	1	3	3	3
CO3	3	3	2	3	3	3	3	3	1	3	3	3
CO4	3	3	2	3	3	3	3	3	1	3	3	3
CO5	3	3	2	3	3	3	3	3	1	3	3	3
Avg	3	3	2	3	3	3	3	3	1	3	3	3

- 1=weakly mapped
 2= moderately mapped
 3=strongly mapped



Name:			
Enrolment No:			
Course: BIC22525 – IMMUNOLOGY LAB Program: M.Sc. Biochemistry Semester: Odd 2020-21			
		Time: 03Hrs. Max. Marks:50	
Instructions: Answer the following questions”			
SECTION A (Attempt all questions) (10X2=20)			
1.	Summarize the working principle of DOT-ELISA.	Ap	CO1
2.	Perform the procedure for latex agglutination test for the given sample/s and evaluate the results <i>(Practical skill- 5, Method accuracy -5)</i>	AP	CO1, CO2
SECTION B (Attempt all questions) (10X2=20)			
6.	Lab note book	AP,AN,U,R	CO1, CO2, CO3, CO4. CO5
7.	Viva	AP,AN,U,R	CO1, CO2, CO3, CO4. CO5

BIC21554	Applied Toxicology (THEORY)	L	T	P	C
Version 1.0	Contact Hours - 45	3	0	0	3
Pre-requisites/Exposure	UG level knowledge of Biochemistry and Cell Biology				
Co-requisites	--				

Course Objectives

The study on toxic substances have clinical and environmental significance. From this course the students will identify various approaches of toxicity analysis and therapeutics.

Course Outcomes

On completion of this course, the students will be able to

- CO 1. Remembering-Recall the various types of clinical toxicity
- CO 2. Understanding-Explain biochemical mechanisms of toxicity
- CO 3. Analyzing-interpret toxic ingredients in food and cosmetics
- CO 4. Applying-Apply various routs and mechanisms of environmental toxicity
- CO 5. Evaluate-Develop the concept of toxicity analysis

Catalog Description

The elective course ‘Appled toxicology’ will help to understand the scope and dimensions of toxicology. All the lectures will be devoted on discussions of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as power point presentation, audio visual virtual lab session as per requirement. Students will perceive the basic concepts of the subject via exercise, case studies and discussions with the coordinator.

Course Content

Applied toxicology (BIC 21554)

UNIT-I Medical Toxicology: Mission of medical toxicology, Comparative toxicology, Human risk assessment, Toxicological database. Hepatic Toxicology Anatomy and physiology of the liver, Types of liver injury (acute and chronic), Hepatotoxic histological analysis of liver injury. Renal Toxicology Renal structure and functions, chemically induced renal, nephrotoxic potential of drugs and chemicals (with special reference to paraquat and amphotericin B) and their mechanism of action, histological analysis of renal injury. Pulmonary Toxicology, Lung toxicants (silica, asbestos, acid fumes) and their effects (acute and chronic) in occupational toxicology, pulmonary diseases and dysfunction (Asthma, COPD, and Lung cancer) by inhaled toxicants, in vitro and in vivo assessment of pulmonary injury, histological characterization of pulmonary injury. Reproductive Toxicology- heavy metals and genotoxicity. Neuro-toxiity: Pesticide and heavy metal induced. Neurotoxic chemicals found in food, drugs and environment.

UNIT-II: Biochemical Toxicology: Fundamentals of Toxicology and Dose- Response Relationships , Factors Affecting Toxic Responses: Disposition , Factors Affecting Toxic Responses, Biochemical Mechanisms of Toxicity, Toxicokinetics- Absorption, distribution, and storage of toxic chemicals. Metabolism of a toxicant. Biotransformation and detoxification reactions- Phase I and Phase II. Genotoxicity: Mechanisms and Methods Introduction and Importance of genotoxicity studies, classification of carcinogens, Mechanism of genotoxicity, Standard test battery for genotoxicity, In-Vitro testing methods, Ames test, Mammalian chromosome aberration test, In-Vivo genotoxicity testing methods: Laddering and tunnels assay, comet assay, micronuclei test.

UNIT III: Food and cosmetic toxicology: Food adulterants, contaminants and Food additives toxicity Agricultural and industrial contaminants in foods (pesticides residues in fruits and vegetables, metal contaminants such as lead, arsenic and mercury in foods), Food additives and its mode of action in packed food, classification and mechanism of toxicity of food additives with special reference to BHT (Butylated hydroxyl toluene) and BHA (Butylated hydroxyl anisole), Toxicants in food Enzyme inhibitors, antivitamins, glycoalkaloids, saponins, goitrogens, teratogens. Mycotoxins -Aflatoxin B1 and its metabolism, toxicity and preventive measures Food borne bacterial illness with reference to Staphylococcus aureus and Bacillus cereus. Cause, Treatment and prevention of food allergies-marine foods. Cosmetic toxicity Cosmetic induced disorders. Defense mechanism of skin against UV radiation, Agencies role in launching a cosmetic finish product, Toxicity of shampoos, conditioners, bleachers and dyes, Toxicities evaluation of cosmetic products.

UNIT- IV: Environmental toxicology: Pesticide Toxicity Classification and use of and toxicity of major groups of pesticides, Herbicide toxicity and toxicity of major groups of herbicides. Biomagnification of pesticides. Method of analysis. Heavy Metal Toxicity: Acute and Chronic Toxicity of Metals, Lead, Mercury, Arsenic, Cadmium, Chromium, Mechanism of heavy metal toxicity, Heavy Metal Toxicity Pathway, Oxidative damage by heavy metals, Genotoxicity of heavy metals, Ecotoxicology of Metals. Toxicology of Chemical Warfare Agents Chemical weapons, management of chemical warfare agents.

UNIT-V: Analytical toxicology: Qualitative Descriptions of Toxicity Exposure Limits Determination of LD50 and ED50, Units in Toxicology. Analysis of pesticides by Chromatography Method. Analysis of heavy metals by Spectrophotometric Methods. Determination of heavy metals by Atomic Absorption Spectrophotometer. Determination of heavy metals by Atomic Emission Spectrophotometer, Inductively Coupled Plasma Spectrophotometers, Determination of drug of abuse. Estimating genotoxicity: Immunoassay Techniques: ELISA.

Textbook:

P.K. Gupta, Fundamentals of toxicology, ELSEVIER 2016

Reference books:

Text book of modern toxicology, Ernest Hodgson, WILEY, 2004

Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:

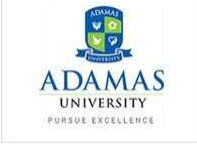
Components	Class Assessment	End Term
Weightage (%)	50	50

Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
Avg	3	3	2	3	3	-	-	3	-	-	1	3

1=weakly mapped
2= moderately mapped
3=strongly mapped

Model Question Paper

Name: Enrolment No:	 ADAMAS UNIVERSITY <small>PURSUE EXCELLENCE</small>		
Course: BIC21554 – APPLIED TOXICITY (THEORY) Program: M.Sc. (Biochemistry) Semester: Odd			
Time: 03Hrs. Max. Marks: 50			
Instructions: Attempt any four questions from Section A (each carrying 5 marks); any two questions from Section B (each carrying 10 marks).			
SECTION A (Attempt any Four questions) (5X4=20) (5X4=20)			
1.	Analyze arsenic toxicity.	An	CO1
2.	Explain why toxicological data base is required.	U	CO2
3.	Illustrate why do proteins folding gets affected by heavy metals.	R	CO3
4.	Describe the use of Next Generation Sequencing (NGS) technology in determining genotoxicity	U	CO4
5.	Develop a mass spectrometry-based toxin analysis protocol.	AP	CO5
SECTION B (Attempt any Two questions) (10X2=20)			
6.	Elaborate major groups of pesticides and explain their toxicity? How do herbicides recognize affect crop quality? Comprehend herbicide resistance as potential environmental hazard.	U	CO3
7.	Describe biomagnification and detoxification. Why amphotericin B) elicits nephrotoxicity? Elaborate laws for chemical decontamination. 2+4+4	U,AN	CO1 CO2
8.	Define and classify poisons? How presence of a poison can be detected in a food sample? Elaborate heavy metal toxicity. 2+1+3+4+1+5+3	AN,AP, U	CO4 CO5

BIC21555	Environmental Toxicology (THEORY)	L	T	P	C
Version 1.0	Contact Hours - 45	3	0	0	3
Pre-requisites/Exposure	UG level knowledge of Biochemistry and Cell Biology				
Co-requisites	--				

Course Objectives

The study on toxic substances environmental significance. From this course the students will identify various approaches of toxicity analysis and therapeutics.

Course Outcomes

On completion of this course, the students will be able to

CO 1. Remembering- Recall different toxic substances affecting environment

CO 2. Understanding-Understanding toxicity inflicted by pesticide, herbicide, and heavy metals

CO 3. Analyzing-interpret toxic ingredients in food

CO 4. Applying-Describe a toxic substances present in and cosmetics

CO 5. Evaluate-Develop the concept of toxicity as an occupational hazard

Catalog Description

The elective course 'Environmental toxicology' will help to understand the scope and dimensions of toxicological aspects of environmental studies. All the lectures will be devoted on discussions of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as power point presentation, audio visual virtual lab session as per requirement. Students will perceive the basic concepts of the subject via exercise, case studies and discussions with the coordinator.

Course Content

Environmental toxicology (BIC 21555)

Unit-I Introduction to environmental toxicology, Hazardous substances in environment sources. Properties of toxic chemical influencing their distribution and accumulation. Bioaccumulation and biomagnifications. Toxicokinetics- Absorption, distribution, and storage of toxic chemicals. Metabolism of a toxicant. Biotransformation and detoxification reactions- Phase I and Phase II. UNIT- II Pesticide Toxicity- Pesticide management, Toxic effects of pesticides: Pyrethroids, DDT, Organophosphate pesticides, Cyclodienes, Lindane, Carbamate, Method of analysis. Herbicide Toxicity- Herbicide Selectivity, mode of action, Benzoic acids (dicamba), Pyridines (picloram), Paraquat, Triazines (atrazine, cyanazine), Sulfonylureas (chlorsulfuron, tribenuron), Aryloxyphenoxypropionates (diclofop), Isoxamolidinones (clomazone). Heavy Metal Toxicity Acute and Chronic Toxicity of Metals, Lead, Mercury, Arsenic, Cadmium, Chromium, Mechanism of heavy

metal toxicity, Genotoxicity of heavy metals, Ecotoxicology of Metals . Heavy Metals in Medicine. Methods of analysis.

UNIT-III: Food toxicology: adulterants, contaminants and Food additives toxicity Agricultural and industrial contaminants in foods (pesticides residues in fruits and vegetables, metal contaminants such as lead, arsenic and mercury in foods), Food additives and its mode of action in packed food, classification and mechanism of toxicity of food additives with special reference to BHT (Butylated hydroxyl toluene) and BHA (Butylated hydroxyl anisole), Toxicants in food Enzyme inhibitors, antivitamins, glycoalkaloids, saponins, goitrogens, teratogens. Mycotoxins -Aflatoxin B1 and its metabolism, toxicity and preventive measures Food borne bacterial illness with reference to Staphylococcus aureus and Bacillus cerus.

UNIT-IV: Cosmetic toxicity: Cosmetic induced disorders such as acne, pruritis, nodules, papules etc. Defense mechanism of skin against UV radiation, Agencies role in launching a cosmetic finish product, Toxicity of shampoos, conditioners, bleachers and dyes, Toxicities evaluation of cosmetic products.

Unit- V Occupational toxicology: Occupational Hazards; Occupational Exposure, Occupational Disease; Occupational Dermatitis, Chloracne, Occupational Lung Diseases; COPD, Silicosis, Asbestosis, Asthama. Hazardous chemical substances and risk assessment of environmental chemical exposure. Rules and regulation governing release of hazardous chemicals into the environment.

Textbook:

P.K. Gupta, Fundamentals of toxicology, ELSEVIER 2016

Reference books:

Text book of modern toxicology, Ernest Hodgson, WILEY, 2004

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination
Examination Scheme:**

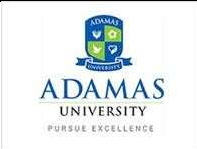
Components	Class Assessment	End Term
Weightage (%)	50	50

Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
Avg	3	3	2	3	3	-	-	3	-	-	1	3

1=weakly mapped
2= moderately mapped
3=strongly mapped

Model Question Paper

Name: Enrolment No:			
Course: BIC21555 – ENVIRONMENTAL TOXICITY (THEORY) Program: M.Sc. (Biochemistry) Semester: Odd			
Time: 03Hrs. Max. Marks: 50			
Instructions: Attempt any four questions from Section A (each carrying 5 marks); any two questions from Section B (each carrying 10 marks).			
SECTION A (Attempt any Four questions) (5X4=20) (5X4=20)			
1.	Analyze arsenic toxicity.	An	CO1
2.	Explain why toxicological data base is required.	U	CO2
3.	Illustrate why do proteins folding gets affected by heavy metals.	R	CO3
4.	Describe the use of Next Generation Sequencing (NGS) technology in determining genotoxicity	U	CO4
5.	Develop a mass spectrometry-based toxin analysis protocol.	AP	CO5
SECTION B (Attempt any Two questions) (10X2=20)			
6.	Elaborate major groups of pesticides and explain their toxicity? How do herbicides recognize affect crop quality? Comprehend herbicide resistance as potential environmental hazard.	U	CO3
7.	Describe biomagnification and detoxification. Why amphotericin B) elicits nephrotoxicity? Elaborate laws for chemical decontamination. 2+4+4	U,AN	CO1 CO2
8.	Define and classify poisons? How presence of a poison can be detected in a food sample? Elaborate heavy metal toxicity. Elaborate oxidative damage induced by heavy metals 2+1+3+4+1+5+3+1+3+4	AN,AP, U	CO3

BIC21556	Advanced DNA forensics (THEORY)	L	T	P	C
Version 1.0	Contact Hours - 45	3	0	0	3
Pre-requisites/Exposure	UG level knowledge of Biochemistry and Cell Biology				
Co-requisites	--				

Course Objectives

The study on toxic substances environmental significance. From this course the students will identify various approaches of toxicity analysis and therapeutics.

Course Outcomes

On completion of this course, the students will be able to

- CO 1. Remembering-Recall different genetic variation in population and its relevance in forensics
- CO 2. Understanding-Explain fundamentals of DNA sequencing and its relevance in forensics
- CO 3. Analysing-Interpret genomic variations at molecular level and its relevance in forensics
- CO 4. Applying-Apply statistical approach for DNA typing and its relevance in forensics
- CO 5. Evaluate-Evaluate the forensic significance of DNA profiling

Catalog Description

The elective course 'Advanced DNA forensics' will help to understand the scope and dimensions of state of the art DNA technologies in forensic science. All the lectures will be devoted on discussions of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as power point presentation, audio visual virtual lab session as per requirement. Students will perceive the basic concepts of the subject via exercise, case studies and discussions with the coordinator.

Course Content

Advanced DNA forensics (BIC 21556)

Unit-I Human Genetics, Alleles, Mutations and Population Genetics, The concept of Genetics polymorphism, Hardy-Weinberg Law. Pedigree analysis.

Unit-II DNA sequencing. Sequencing of DNA- Sanger sequencing. Approaches for next gen sequencing. Long and short read sequencing. Genome assembly and annotation. Human genome project and variation data bases- SNPdb and OMIM.

Unit-III DNA Profiling: Introduction, History of DNA Typing, polymorphism, DNA Extraction-Organic and Inorganic extraction, Comparison of Extraction methods, RFLP analysis, PCR amplifications, sequence polymorphism. Analysis of SNP, STR, Mitochondrial DNA, Ancient DNA typing, Evaluation of results. DNA figure printing.

Unit-IV DNA Statistics: frequency estimate calculations, interpretations, allele frequency determination, Paternity/Maternity index, Sibling index, Probability of match. Human Genome Project: Introduction, History, Goals, Benefits, Social, Ethical and Legal Issues DNA Forensic Databases

Unit-V Forensic Significance of DNA profiling: Applications in disputed paternity cases, child swapping, missing person's identity- civil immigration, veterinary, wildlife and agriculture cases, legal perspectives- legal standards for admissibility of DNA profiling, procedural and ethical concerns, status of development of DNA profiling in India and abroad. New and future technologies: DNA chips, SNPs and limitations of DNA profiling. Case studies.

Textbook:

P.K. Gupta, Fundamentals of toxicology, ELSEVIER 2016

Reference books:

Text book of modern toxicology, Ernest Hodgson, WILEY, 2004

Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Scheme:

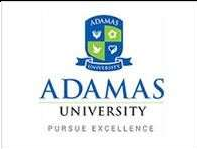
Components	Class Assessment	End Term
Weightage (%)	50	50

Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
Avg	3	3	2	3	3	-	-	3	-	-	1	3

1=weakly mapped
2= moderately mapped
3=strongly mapped

Model Question Paper

Name: Enrolment No:	 ADAMAS UNIVERSITY <small>PURSUE EXCELLENCE</small>		
<p>Course: BIC21554 – Advanced DNA forensics (THEORY)</p> <p>Program: M.Sc. (Biochemistry) Time: 03Hrs.</p> <p>Semester: Odd Max. Marks: 50</p> <p>Instructions: Attempt any four questions from Section A (each carrying 5 marks); any two questions from Section B (each carrying 10 marks).</p>			
SECTION A (Attempt any Four questions) (5X4=20) (5X4=20)			
1.	Define: multiple allele	An	CO1
2.	Explain gene pool and allele frequency	U	CO2
3.	Illustrate VNTR	R	CO3
4.	Describe the principle of Next Generation Sequencing (NGS) technology.	U	CO4
5.	Develop an array based analysis protocol.	AP	CO5
SECTION B (Attempt any Two questions) (10X2=20)			
6.	What is the role of dideoxynTPs in Sanger sequencing? How do massively parallel sequencing is accomplished? Illustrate the steps of sequence assembly.	U	CO3
7.	Explain DNA fingure printing. Mention the features on mtDNA. How mtDNA anlysis helps in forensics.- elaborate 2+4+4	U,AN	CO1 CO2
8.	Elaborate how DNA is extracted from forensic samples. Describe the use of PCR in forensics. If you need to determine paternal identity how can you implement DNA forensics? 2+1+3+4	AN,AP, U	CO3

BIC21558	Advanced forensic chemistry (THEORY)	L	T	P	C
Version 1.0	Contact Hours - 45	3	0	0	3
Pre-requisites/Exposure	UG level knowledge of Biochemistry and Cell Biology				
Co-requisites	--				

Course Objectives

The study on forensic chemistry, i.e. chemical approaches in forensic science. From this course the students will identify various approaches of toxicity analysis and therapeutics.

Course Outcomes

On completion of this course, the students will be able to

CO1.Remembering: Students will be able to recall complex DNA extraction, quantification, and amplification techniques, including next-generation sequencing (NGS) and mitochondrial DNA analysis.

CO2. Understanding: Students will be proficient in interpreting genetic evidence, including understanding statistical methods used to assess the significance of DNA match results.

CO3. Analysing: Students will analyze real-world case studies to evaluate the role of DNA evidence in legal contexts, including its implications for justice and wrongful convictions.

CO4. Applying: Students will articulate the ethical issues surrounding DNA forensics, including privacy concerns, informed consent, and the potential for misuse of genetic information.

CO5: Evaluate: Students will demonstrate an understanding of the legal standards and guidelines governing the collection, analysis, and presentation of DNA evidence in court.

Catalog Description

The elective course ‘Advanced forensic chemistry’ will help to understand the scope and dimensions of state of the art DNA technologies in forensic science. All the lectures will be devoted on discussions of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as power point presentation, audio visual virtual lab session as per requirement. Students will perceive the basic concepts of the subject via exercise, case studies and discussions with the coordinator.

Course Content

Advanced forensic chemistry (BIC 21558)

Unit I - Forensic Chemistry- Introduction. Trap cases, Preliminary analysis of evidence in trap cases. Alcoholic Beverages: Types of alcohols and analysis. Dyes: Scope & Significance of dyes in crime investigation, analysis of ink by TLC and UV visible spectrophotometry. Petroleum products and their adulterations. Analysis of petrol, kerosene, diesel.

Unit II- Forensic Toxicology Poisons- uses and origin. Types, routes of administration, toxicity, sign and symptoms of various poisons. Medico-legal aspects of poisoning cases. Pesticides: Different types and their formulations, identification of pesticides. Guidelines for collecting forensic evidences in poisoning cases at crime scene. Importance of Post mortem examination in poisoning cases. Sample preparation for the analysis of poisons in body tissues/fluids and analysis by various instrumental techniques.

Unit III- Narcotic Drugs and Psychotropic Substances. Scope and significance NDPS drugs in forensic science, NDPS Act, Classification and characterization of NDPS drugs, Sample preparation for analysis, Preliminary analysis of drugs, Drug laws and Reporting of drug cases, Drug abuse, Drug addiction and its problems.

Unit IV – Fire/Arson and Explosives Fire: Introduction to Fire & Arson, origin of fire, Chemistry of Fire, Firefighting operations, preservation of fire scene, collection of evidences, Analysis of fire debris, Case studies related to fire and Arson. Explosive and Explosion: Scope & significance of explosive analysis in forensic science, Types of explosives, deflagration and detonation, preliminary analysis of explosives.

Unit V: Case studies.

Textbook:

P.K. Gupta, Fundamentals of toxicology, ELSEVIER 2016

Reference books:

Text book of modern toxicology, Ernest Hodgson, WILEY, 2004
 Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Scheme:

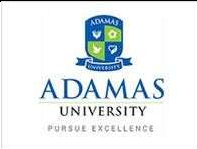
Components	Class Assessment	End Term
Weightage (%)	50	50

Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
Avg	3	3	2	3	3	-	-	3	-	-	1	3

1=weakly mapped
2= moderately mapped
3=strongly mapped

Model Question Paper

Name: Enrolment No:			
Course: BIC21558– Advanced forensic chemistry(THEORY) Program:M.Sc. (Biochemistry) Semester: Odd			
Time: 03Hrs. Max. Marks:50			
Instructions: Attempt any four questions from Section A (each carrying 5 marks); any two questions from Section B (each carrying 10 marks).			
SECTION A (Attempt any Four questions) (5X4=20) (5X4=20)			
1.	Define Trap cases. Mention analysis of evidence in trap cases.	An	CO1
2.	Explain how petroleum products are adulterated and how such adulteration might be detected.	U	CO2
3.	Illustrate why do proteins fold? What is peptide mass fingerprinting?	R	CO3
4.	Describe chemical analysis in postmortem for burn cases.	U	CO4
5.	Develop a method to analyze fire debris	AP	CO5
SECTION B (Attempt any Two questions) (10X2=20)			
6.	Define narcotics and psychedelics? Illustrate NDPS Act. How drug abuse can be confirmed? 3+2+5	U	CO3
7.	Elaborate collection and forensic analysis of tissue fluids. Discuss the medico-legal aspects of poisoning cases.	U,AN	CO1 CO2
8.	Elaborate how DNA is extracted from forensic samples. Describe the use of PCR in forensics. If you need to determine paternal identity how can you implement DNA forensics? 2+1+3+4	AN,AP, U	CO3

BIC21532	Clinical Biochemistry (THEORY)	L	T	P	C
Version 1.0	Contact Hours - 45	3	0	0	3
Pre-requisites/Exposure	BSc. Level Biochemistry Knowledge				
Co-requisites	--				

Course Objectives

- To provide students basic idea about instrumentation and automation in clinical biochemistry laboratories safety regulations.
- It will also provide in depth knowledge about different biochemical reactions that are used to determine different disease parameters.
- Outlining the types of specimen for biochemical analysis.
- To provide students different parameters like precision, accuracy, quality control, precautions and limitations that are used in clinical biochemistry.

Course Outcomes

On completion of this course, the students will be able to:

1. Remembering: Recall and describe the normal biochemical pathways in the human body
2. Understanding: Interpret the significance of abnormal biochemistry results in disease diagnosis and monitoring
3. Applying: Apply knowledge of biochemical pathways to analyze and interpret abnormal test results
4. Analyzing: Analyze and evaluate the correlations between biochemical changes and clinical manifestations of diseases
5. Evaluating: Critically evaluate the ethical implications of using biochemical tests in clinical practice

Catalog Description

The core-course of 'Clinical Biochemistry' will help to understand the basic idea about instrumentation and automation in clinical biochemistry. This course includes comprehensive approach through studying different biochemical reactions that are used to determine different disease parameters. Furthermore, the implication of precision, accuracy, quality control, precautions and limitations in different test results will also be illuminated. All the lectures will be devoted on discussions of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as power point presentation, audio visual virtual lab session as per requirement. The tutorials will enable the students with problem-solving ability led by the course coordinator. Students will perceive the basic concepts of the subject via exercise and discussions with the coordinator.

Course Content

DSE II: CLINICAL BIOCHEMISTRY

1. Disorders of Carbohydrate Metabolism – Diabetes mellitus, glucose and galactose tolerance tests, sugar levels in blood, renal threshold for glucose, factors influencing blood glucose level, glycogen storage diseases, pentosuria, galactosemia.
2. Disorders of Lipids – Plasma lipoproteins, cholesterol, triglycerides & phospholipids in health and disease, hyperlipidemia, hyperlipoproteinemia, Gaucher's disease, Tay-Sach's and Niemann-Pick disease, ketone bodies, Abetalipoproteinemia.
3. Inborn Errors of Metabolism – Phenylketonuria, alkaptonuria, albinism, tyrosinosis, maple syrup urine disease, Lesch-Nyhan syndrome, sickle cell anemia, Histidinemia.
4. Digestive diseases – Maldigestion, malabsorption, creatorrhoea, diarrhoea and steatorrhoea. Disorders of liver and kidney – Jaundice, fatty liver, normal and abnormal functions of liver and kidney. Inulin and urea clearance.
5. Electrolytes and acid-base balance – Regulation of electrolyte content of body fluids and maintenance of pH, reabsorption of electrolytes.
6. Diagnostic Enzymes – Enzymes in health and diseases. Biochemical diagnosis of diseases by enzyme assays – SGOT, SGPT, CPK, cholinesterase, LDH.
7. Abnormalities in Nitrogen Metabolism – Uremia, hyperuricemia, porphyria and factors affecting nitrogen balance.
8. Blood Clotting – Disturbances in blood clotting mechanisms – haemorrhagic disorders – haemophilia, von Willebrand's disease, purpura, Rendu-Osler-Werber disease, thrombotic thrombocytopenic purpura, disseminated intravascular coagulation, acquired prothrombin complex disorders, circulating anticoagulants.
9. Cancer – Cellular differentiation, carcinogens and cancer therapy

SUGGESTED READINGS

1. Medical Laboratory Technology - a Procedure Manual for Routine Diagnostic Tests Vol. I (2010), Mukherjee, K.L., Tata McGraw-Hill Publishing Company Limited (New Delhi). ISBN:9780070076594 /ISBN:9780070076631
2. Medical Laboratory Technology - a Procedure Manual for Routine Diagnostic Tests Vol. II (2010), Mukherjee, K.L., Tata McGraw-Hill Publishing Company Ltd. (New Delhi), ISBN:9780070076648.
3. Medical Biochemistry (2005) 2nd ed., Baynes, J.W. and Dominiczak, M.H., Elsevier Mosby Ltd. (Philadelphia), ISBN:0-7234-3341-0.
4. Experimental Biochemistry: A Student Companion (2005) Rao, B.S. and Deshpande, V., IK International Pvt. Ltd. (New Delhi), ISBN:81-88237-41-8.

Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination

Examination Scheme:

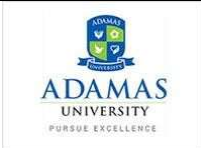
Components	Class Assessment	End Term
Weightage (%)	50	50

Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
Avg	3	3	2	3	3	-	-	3	-	-	1	3

1=weakly mapped
2= moderately mapped
3=strongly mapped

Question

Name: Enrolment No:			
Course: BIC21532 – Clinical Biochemistry (THEORY) Program: M.Sc. Biochemistry Semester: Odd 2019-20			
Time: 03Hrs. Max. Marks:50			
Instructions: Attempt any four questions from Section A (each carrying 5 marks); any two questions from Section B (each carrying 10 marks).			
SECTION A (Attempt any Four questions) (5X4=20)			
1.	Write short notes on galactosemia and glycogen storage disease.	U,R	CO1
2.	How can you measure blood glucose level? What is the normal range of blood sugar level?	U,R	CO2
3.	What is glycemic index? Describe various mechanisms for regulation of blood glucose.	U,R	CO3
4.	What is liver function test? Explain your answer.	U,R	CO4
5.	Explain the formation, function and clinical significance of thyroid hormones.	U,R	CO5
SECTION B (Attempt any 3 questions) (10X3=30)			
6.	Briefly demonstrate the importance of automation in a clinical laboratory.	U,R	CO2
7.	Do you think measuring blood sugar level in fasting condition is a true representation of sugar level for a diabetic patient? Explain. If not, what will be better technique to monitor blood sugar level for a diabetic patient? Why?	U,CR	CO1 CO2
8.	Describe the laboratory investigation of kidney disease. Discuss the potential pitfalls and how can you overcome the pitfalls.	AN	CO1 CO2
9.	What is Atherosclerosis? What are the risk factors for coronary artery disease? What is the link between smoking and heart disease? Explain your answer.	AN,AP	CO4 CO5

BIC21529	DSE-II Research Methodologies and GLP (THEORY)	L	T	P	C
Version 1.0	Contact Hours - 45	3	0	0	3
Pre-requisites/Exposure	BSc. Level Biochemistry Knowledge				
Co-requisites	--				

Course Objectives

- To provide students basic idea about how to understand and formulate a good research plan.
- It will also provide in depth knowledge about plagiarism and how to follow research ethics.
- To illustrate good, automated laboratory practice.
- To describe quality management system in a research lab.

Course Outcomes

On completion of this course, the students will be able to:

CO1. Remember: Recall and explain the fundamental principles of research methodologies in biochemistry.

CO2. Understand: Analyze and compare different research paradigms in biochemistry.

CO3. Apply: Design and implement appropriate study designs for biochemistry research projects.

CO4. Analyze: Evaluate the importance of Good Laboratory Practices (GLP) and work standards in biochemistry research.

CO5. Evaluate: Critically assess the validity and reliability of research findings in biochemistry.

Catalog Description

The core-course of 'Research Methodology and GLP' will help to define fundamental knowledge about how to understand and formulate a good research plan. This course includes comprehensive approach to understand plagiarism and to follow research ethics. Furthermore, the implications of good, automated laboratory practice and quality management system will also be illuminated. All the lectures will be devoted on discussions of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as power point presentation, audio visual virtual lab session as per requirement. The tutorials will enable the students with problem-solving ability led by the course coordinator. Students will perceive the basic concepts of the subject via exercise and discussions with the coordinator.

Course Content

DSE-II Research Methodology and GLP

1. Meaning of research problem, Sources of research problem, Criteria Characteristics of a good research problem, Errors in selecting a research problem, Scope and objectives of research problem. Approaches of investigation of solutions for research problem, data collection, analysis, interpretation, Necessary instrumentations. [**5 Lecture Hours**]
2. Effective literature studies approach, analysis Plagiarism, Research ethics. [**5 Lecture Hours**]
3. Effective technical writing, how to write report, Developing a Research Proposal, Format of research proposal, presentation and assessment by a review committee. [**5 Lecture Hours**]
4. Introduction to the WHO/TDR Handbook on GLP; Current Good Manufacturing Practices: [**5 LectureHours**]
5. Introduction, US Cgmp Part 210 and Part 211.EC Principles of GMP (Directive 91/356/EEC) Article 6 to Article 14 and WHO cGMP guidelines GAMP-5; Medical device and IVDs Global Harmonization Task Force (GHTF) Guidancedocs.
6. Introduction, USFDA GLP Regulations (Subpart A to Subpart K), Controlling the GLP inspection process, Documentation, Audit, goals of Laboratory Quality Audit, Audit tools, Future of GLP regulations, relevant ISO and Quality Council of India (QCI) Standards. [**5 LectureHours**]
7. Good Automated LaboratoryPractices:
8. Introduction to GALP, Principles of GALP, GALP Requirements, SOPs of GALP, Training Documentation,21 CFR Part 11, General check list of 21CFR Part 11, Software Evaluation checklist, relevant ISO and QCI Standards. [**5 LectureHours**]
9. Good DistributionPractices:
Introduction to GDP, Legal GDP requirements put worldwide, Principles,
10. Personnel, Documentation, Premises and Equipment, Deliveries to Customers, Returns, Self-Inspection, Provision of information, Stability testing principles, WHO GDP, USP GDP (Supply chain integrity), relevant CDSCO guidance and ISO standards[**5 LectureHours**]
11. Quality management systems: [**5 LectureHours**]
Concept of Quality, Total Quality Management, Quality by design, Six Sigma concept, Out of Specifications (OOS), Change control. Validation: Types of Validation, Types of Qualification, Validation master plan (VMP), Analytical Method Validation. Validation of utilities, [Compressed air, steam, water systems, Heat Ventilation and Air conditioning (HVAC)]and Cleaning Validation. The International Conference on Harmonization (ICH) process, ICH guidelines to establish quality, safety and efficacy of drug substances and products, ISO 13485, Sch MIII and other relevant CDSCO regulatory guidancedocuments.

References:

1. Montgomery, Douglas C. (2007) 5/e, Design and Analysis of Experiments(Wiley India).
2. Krishnswamy, K.N., Shivkumar, AppaIyer and Mathiranjana M. (2006) Management Research Methodology; Integration of Principles, Methods and Techniques(Pearson Education, NewDelhi)
3. Good Laboratory Practice Regulations, by Sandy Weinberg, Fourth Edition Drugsand the Pharmaceutical Sciences,Vol.168
4. Good Laboratory Practice Regulations, by Sandy Weinberg, Fourth Edition Drugsand the Pharmaceutical Sciences,Vol.16

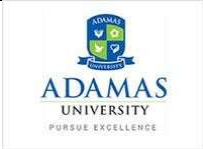
Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:

Components	Class Assessment	End Term
Weightage (%)	50	50

Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
Avg	3	3	2	3	3	-	-	3	-	-	1	3

- 1=weakly mapped
- 2= moderately mapped
- 3=strongly mapped

Name:			
Enrolment No:			
<p>Course: BIC21529 – DSE-II Research Methodology and GLP (THEORY) Program: M.Sc. Biochemistry Time: 03Hrs. Semester: Odd 2020-21 Max. Marks:50</p> <p>Instructions: Attempt all questions each carrying 20 marks</p>			
	Presentation of research in a seminar	AN,AP, R	ALL POs
	Question answer session	AP, CR	ALL POs

BIC24535	Industry Internship (Practical)	L	T	P	C
Version 1.0	Contact Hours	0	0	0	2
Pre-requisites/Exposure	BSc. Level Biochemistry Knowledge				
Co-requisites	--				

Course Objectives

- To provide students basic idea about work habits and attitudes necessary for job success.
- It will also illustrate the career alternatives prior to graduation.
- To develop communication, interpersonal and other critical skills in the job interview process.
- To provide students the ability to analyze interests and abilities in their field of study.

Course Outcomes

On completion of this course, the students will be able to:

- CO 1 Remembering-Recall work habits and attitudes necessary for job success.
- CO 2 Understanding-Understanding career alternatives prior to graduation.
- CO 3 Apply-Develop communication, interpersonal and other critical skills in the job interview process.
- CO 4 Analysing-Develop interests and abilities in their field of study.
- CO 5 Evaluate-Evaluate employment contacts leading directly to a full-time job following graduation from college.

Catalog Description

The practical course of 'Industry Internship' will help to develop work habits and attitudes necessary for job success. This course includes comprehensive approach to develop communication, interpersonal and other critical skills in the job interview process. Furthermore, interests and abilities in their field of study will also be illuminated. The practical experience will enable students to enrich in real-life scenario. The tutorials will enable the students with problem-solving ability led by the course coordinator. Students will perceive the basic concepts of the subject via exercise and discussions with the coordinator

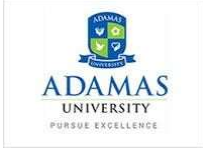
**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination
Examination Scheme:**

Components	Report submission	Presentation
Weightage (%)	50	50

Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	3	3	3	1	3	3	3	3	3	-	-	3
CO2	3	3	3	1	3	3	3	3	3	-	-	3
CO3	3	3	3	1	3	3	3	3	3	-	-	3
CO4	3	3	3	1	3	3	3	3	3	-	-	3
CO5	3	3	3	1	3	3	3	3	3	-	-	3
Avg	3	3	3	1	3	3	3	3	3	-	-	3

- 1=weakly mapped
- 2= moderately mapped
- 3=strongly mapped

Name:		
Enrolment No:		
Course BIC21533– Industry Internship (Practical)		
Program: M.Sc. Biochemistry		Time: 03Hrs.
Semester:Odd 2020-21		Max. Marks:50
Presentation on the work		

BIC21539	Comprehensive viva	L	T	P	C
Version 1.0		0	0	0	2
Pre-requisites/Exposure	Knowledge about the biochemistry at M.Sc level and contemporary research in Biochemistry				
Co-requisites	-				

Course Objectives

1. Defining and outlining a research area with a clear question
2. Identifying the leading issues
3. Sourcing the relevant information
4. Evaluating the evidence on all sides of a debate
5. Coming to a well-argued conclusion

Course Outcomes

On completion of this course, the students will be able to

- CO1. Remember-Recall their knowledge during their interview for biochemistry related jobs.
CO2. Understand-Explore their knowledge during their interview for biochemistry related research fields.
CO3. Apply- Develop the skill to conclude a scientific fact.
CO4. Analyse-Discuss about the biochemical data.
CO5. Evaluate-Establish himself/herself as a good biochemist in society.

Catalog Description

The objective of comprehensive viva-voce is to assess the overall knowledge of the student in the relevant field of Biochemistry acquired over 2 years of study in the postgraduate program

Course Content

1. Reading of Biochemistry Text books, very recent research papers from high impact journals containing biochemical research work and also performance of laboratory based research oriented experiments.

Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination
Examination Scheme:

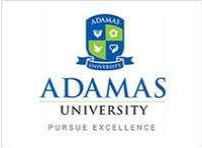
Components	Presentation
Weightage (%)	100

Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	3	3	3	1	3	3	3	3	3	-	-	3
CO2	3	3	3	1	3	3	3	3	3	-	-	3
CO3	3	3	3	1	3	3	3	3	3	-	-	3
CO4	3	3	3	1	3	3	3	3	3	-	-	3
CO5	3	3	3	1	3	3	3	3	3	-	-	3
Avg	3	3	3	1	3	3	3	3	3	-	-	3

- 1=weakly mapped
- 2= moderately mapped
- 3=strongly mapped

Model Question Paper

Name: Enrolment No:	 ADAMAS UNIVERSITY PURSUE EXCELLENCE
Course: BIC21538 –Comprehensive Viva Program:M.ScBiochemistry Semester: Even 2020-21 Time: 01 Hrs. Max. Marks:100 Instructions: Attempt any two questions from Section A (each carrying 10 marks); Section B (each carrying 10 marks)isCompulsory.	
Answer all asked questions	

BIC25540	Dissertation	L	T	P	C
Version 1.0		0	0	0	12
Pre-requisites/Exposure	Knowledge about the basic knowledge and contemporary research in Biochemistry				
Co-requisites	-				

Course Objectives

6. Defining and outlining a research area with a clear question
7. Identifying the leading issues
8. Sourcing the relevant information
9. Assessing its reliability and legitimacy
10. Evaluating the evidence on all sides of a debate
11. Coming to a well-argued conclusion

Course Outcomes

On completion of this course, the students will be able to

Upon completion of the Dissertation course, students will be able to:

1. **Conduct Independent Research:** Design and implement a research project that demonstrates the ability to identify, formulate, and address a significant problem using cutting-edge techniques.
2. **Apply Experimental Methods:** Utilize appropriate experimental methods to collect, analyze, and interpret data, ensuring methodological rigor and relevance to the field of study.
3. **Engage with Current Literature:** Critically review and synthesize existing literature to contextualize their research within the broader academic discourse, identifying gaps and potential contributions.
4. **Demonstrate Project Management Skills:** Effectively manage all phases of the dissertation project, from proposal development through to execution and final presentation, ensuring adherence to timelines and academic standards.
5. **Communicate Findings:** Present research findings clearly and effectively, utilizing appropriate formats (written, oral, and visual) to communicate complex ideas to diverse audiences.

Catalog Description

Dissertation allows students present their findings in response to a question or proposition that they choose themselves. The aim of the project is to test the independent research skills students have acquired during their time at university, with the assessment used to help determine their final grade. Although there is usually some guidance from your tutors, the dissertation project is largely independent.

Course Content

1. Reading of very recent research papers from high impact journals containing biochemical research work and also performance of laboratory based research oriented experiments.

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination
Examination Scheme:**

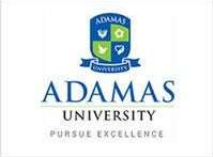
Components	Thesis	Presentation
Weightage (%)	50	50

Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

CO Number	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO 10	PO 11	PO 12
CO1	3	3	3	1	3	3	3	3	3	-	-	3
CO2	3	3	3	1	3	3	3	3	3	-	-	3
CO3	3	3	3	1	3	3	3	3	3	-	-	3
CO4	3	3	3	1	3	3	3	3	3	-	-	3
CO5	3	3	3	1	3	3	3	3	3	-	-	3
Avg	3	3	3	1	3	3	3	3	3	-	-	3

1=weakly mapped
2= moderately mapped
3=strongly mapped

Model Question Paper

Name: Enrolment No:	 ADAMAS UNIVERSITY PURSUE EXCELLENCE
Course: BIC25539 –Dissertation Program: M.Sc Biochemistry Semester: Even 2020-21 Time: 01 Hrs. Max. Marks: 50	
Presentation	

CO PO relationship

	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
BIC21501												
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
BIC21503												
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
BIC21507												
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
BIC21528												
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
BIC21541												
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
BIC22542												
CO1	3	3	2	3	3	3	3	3	1	3	3	3
CO2	3	3	2	3	3	3	3	3	1	3	3	3
CO3	3	3	2	3	3	3	3	3	1	3	3	3
CO4	3	3	2	3	3	3	3	3	1	3	3	3
CO5	3	3	2	3	3	3	3	3	1	3	3	3
BIC22527												
CO1	3	3	2	3	3	3	3	3	1	3	3	3
CO2	3	3	2	3	3	3	3	3	1	3	3	3

CO3	3	3	2	3	3	3	3	3	1	3	3	3
CO4	3	3	2	3	3	3	3	3	1	3	3	3
CO5	3	3	2	3	3	3	3	3	1	3	3	3
BIC22570												
CO1	-	3	3	1	3	3	3	3	-	3	2	2
CO2	-	3	3	1	3	3	3	3	-	3	2	2
CO3	-	3	3	1	3	3	3	3	-	3	2	2
CO4	-	3	3	1	3	3	3	3	-	3	2	2
CO5	-	3	3	1	3	3	3	3	-	3	2	2
BIC21511												
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
BIC21512												
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
BIC22543												
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
BIC21544												
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
BIC22545												
CO1	3	3	2	3	3	3	3	3	1	3	3	3
CO2	3	3	2	3	3	3	3	3	1	3	3	3
CO3	3	3	2	3	3	3	3	3	1	3	3	3
CO4	3	3	2	3	3	3	3	3	1	3	3	3
CO5	3	3	2	3	3	3	3	3	1	3	3	3
BIC21546												
CO1	3	3	2	3	3	3	3	3	1	3	3	3
CO2	3	3	2	3	3	3	3	3	1	3	3	3

CO3	3	3	2	3	3	3	3	3	1	3	3	3
CO4	3	3	2	3	3	3	3	3	1	3	3	3
CO5	3	3	2	3	3	3	3	3	1	3	3	3
BIC22547												
CO1	3	3	2	3	3	3	3	3	1	3	3	3
CO2	3	3	2	3	3	3	3	3	1	3	3	3
CO3	3	3	2	3	3	3	3	3	1	3	3	3
CO4	3	3	2	3	3	3	3	3	1	3	3	3
CO5	3	3	2	3	3	3	3	3	1	3	3	3
BIC22519												
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
BIC22520												
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
BIC21509												
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
BIC22510												
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
BIC21522												
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
BIC21536												
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3

CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
BIC22549												
CO1	3	3	2	3	3	3	3	3	1	3	3	3
CO2	3	3	2	3	3	3	3	3	1	3	3	3
CO3	3	3	2	3	3	3	3	3	1	3	3	3
CO4	3	3	2	3	3	3	3	3	1	3	3	3
CO5	3	3	2	3	3	3	3	3	1	3	3	3
BIC21550												
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
BIC21551												
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
BIC22552												
CO1	3	3	2	3	3	3	3	3	1	3	3	3
CO2	3	3	2	3	3	3	3	3	1	3	3	3
CO3	3	3	2	3	3	3	3	3	1	3	3	3
CO4	3	3	2	3	3	3	3	3	1	3	3	3
CO5	3	3	2	3	3	3	3	3	1	3	3	3
BIC22553												
CO1	3	3	2	3	3	3	3	3	1	3	3	3
CO2	3	3	2	3	3	3	3	3	1	3	3	3
CO3	3	3	2	3	3	3	3	3	1	3	3	3
CO4	3	3	2	3	3	3	3	3	1	3	3	3
CO5	3	3	2	3	3	3	3	3	1	3	3	3
BIC21513												
CO1	3	3	2	3	3	3	3	3	1	3	3	3
CO2	3	3	2	3	3	3	3	3	1	3	3	3
CO3	3	3	2	3	3	3	3	3	1	3	3	3
CO4	3	3	2	3	3	3	3	3	1	3	3	3
CO5	3	3	2	3	3	3	3	3	1	3	3	3
BIC22523												
CO1	3	3	2	3	3	3	3	3	1	3	3	3
CO2	3	3	2	3	3	3	3	3	1	3	3	3

CO3	3	3	2	3	3	3	3	3	1	3	3	3
CO4	3	3	2	3	3	3	3	3	1	3	3	3
CO5	3	3	2	3	3	3	3	3	1	3	3	3
BIC21554												
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
BIC21555												
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
BIC21556												
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
BIC21558												
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
BIC21532												
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
BIC21529												
CO1	3	3	2	3	3	-	-	3	-	-	1	3
CO2	3	3	2	3	3	-	-	3	-	-	1	3
CO3	3	3	2	3	3	-	-	3	-	-	1	3
CO4	3	3	2	3	3	-	-	3	-	-	1	3
CO5	3	3	2	3	3	-	-	3	-	-	1	3
BIC24535												
CO1	3	3	3	1	3	3	3	3	3	-	-	3
CO2	3	3	3	1	3	3	3	3	3	-	-	3

CO3	3	3	3	1	3	3	3	3	3	-	-	3
CO4	3	3	3	1	3	3	3	3	3	-	-	3
CO5	3	3	3	1	3	3	3	3	3	-	-	3
BIC21534												
CO1	3	3	3	2	3	3	3	3	2	3	3	2
CO2	3	3	3	2	3	3	3	3	2	3	3	3
CO3	3	3	3	2	3	3	3	3	2	3	3	2
CO4	3	3	3	2	3	3	3	3	2	3	3	2
CO5	3	3	3	2	3	3	3	3	2	3	3	2
BIC21539												
CO1	3	3	3	1	3	3	3	3	3	-	-	3
CO2	3	3	3	1	3	3	3	3	3	-	-	3
CO3	3	3	3	1	3	3	3	3	3	-	-	3
CO4	3	3	3	1	3	3	3	3	3	-	-	3
CO5	3	3	3	1	3	3	3	3	3	-	-	3
BIC25540												
CO1	3	3	3	1	3	3	3	3	3	-	-	3
CO2	3	3	3	1	3	3	3	3	3	-	-	3
CO3	3	3	3	1	3	3	3	3	3	-	-	3
CO4	3	3	3	1	3	3	3	3	3	-	-	3
CO5	3	3	3	1	3	3	3	3	3	-	-	3
BIC22570												
CO1	-	3	3	1	3	3	3	3	-	3	2	2
CO2	-	3	3	1	3	3	3	3	-	3	2	2
CO3	-	3	3	1	3	3	3	3	-	3	2	2
CO4	-	3	3	1	3	3	3	3	-	3	2	2
CO5	-	3	3	1	3	3	3	3	-	3	2	2
BIC22571												
CO1	-	3	3	1	3	3	3	3	-	3	2	2
CO2	-	3	3	1	3	3	3	3	-	3	2	2
CO3	-	3	3	1	3	3	3	3	-	3	2	2
CO4	-	3	3	1	3	3	3	3	-	3	2	2
CO5	-	3	3	1	3	3	3	3	-	3	2	2
BIC22572												
CO1	-	3	3	1	3	3	3	3	-	3	2	2
CO2	-	3	3	1	3	3	3	3	-	3	2	2
CO3	-	3	3	1	3	3	3	3	-	3	2	2
CO4	-	3	3	1	3	3	3	3	-	3	2	2
CO5	-	3	3	1	3	3	3	3	-	3	2	2
Total	555	615	450	540	615	270	270	615	105	225	320	591
Total Courses	48											
Average	2.3125	2.5625	1.875	2.25	2.5625	1.125	1.125	2.5625	0.4375	0.9375	1.333333	2.4625

