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**ADAMAS UNIVERSITY**

**SCHOOL OF LIFE SCIENCE & BIOTECHNOLOGY**

**DEPARTMENT OF BIOLOGICAL SCIENCES**

**B.Sc. BIOCHEMISTRY (3 YEARS) PROGRAMME COURSE STRUCTURE &  
PROPOSED SYLLABUS  
TOTAL CREDITS –154**

ADAMAS UNIVERSITY, KOLKATA, SCHOOL OF LIFE  
SCIENCE AND BIOTECHNOLOGY

VISION OF THE UNIVERSITY

To be an internationally recognized university through excellence in inter-disciplinary education, research and innovation, preparing socially responsible well-grounded individuals contributing to nation building.

MISSION STATEMENTS OF THE UNIVERSITY

M.S 01: Improve employability through futuristic curriculum and progressive pedagogy with cutting-edge technology

M.S 02: Foster outcomes based education system for continuous improvement in education, research and all allied activities

M.S 03: Instill the notion of lifelong learning through culture of research and innovation

M.S 04: Collaborate with industries, research centers and professional bodies to stay relevant and up to date.

M.S 05: Inculcate ethical principles and develop understanding of environmental and social realities.

CHANCELLOR / VICE CHANCELLOR

ADAMAS UNIVERSITY, KOLKATA, SCHOOL OF LIFE SCIENCE  
AND BIOTECHNOLOGY

VISION OF THE SCHOOL

To achieve global standard and excellence in research on various interdisciplinary and multidisciplinary domains of biological sciences through biotechnological innovation along with producing global citizens as graduates by intensive teaching learning process who would be vanguard to sustainable societal development.

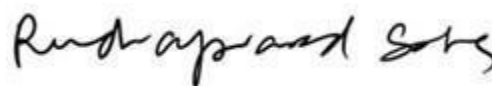
MISSION STATEMENTS OF THE SCHOOL

**M.S 01:** To disseminate knowledge of life science and biotechnology for scholarly progression, intellectual development and strive for innovation.

**M.S 02:** To enable latest skill sets in the domain of microbiology, biotechnology, biochemistry (biological sciences) with ability to evolve and engage in learn-unlearn and relearn, being a lifelong learner and use the knowledge to other multi-disciplinary programmes.

**M.S 03:** To establish state of art infrastructure and research ambiance in attracting the best minds to serve under the single roof of school of life science and biotechnology in undertaking scientific investigation of social relevance.

**M.S 04:** To inculcate values, culture along with scientific knowledge to foster the spirit of self- reliance and entrepreneurship development.



DEAN / SCHOOL CONCERNED

ADAMAS UNIVERSITY, KOLKATA SCHOOL OF LIFE SCIENCE AND  
BIOTECHNOLOGY

**VISION OF THE DEPARTMENT**

To achieve excellence in education and research in biochemistry for societal development through innovation and producing technologically sound graduates as global citizen fostering life-long learning.

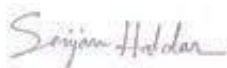
**MISSION STATEMENTS OF THE DEPARTMENT**

**M.S01:** Adopt and implement latest curriculum in biochemistry with futuristic approach and innovative pedagogy fostering knowledge, intellectual and skill development.

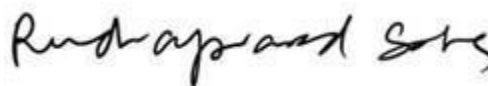
**M.S02:** To enable and enhance skill in biochemistry set through rigorous training and research through multidisciplinary approach.

**M.S03:** To cater professional and societal need of cutting-edge research in biochemistry through collaboration and industry-academic partnership.

**M.S04:** To inculcate values, culture along with knowledge about biochemistry to foster the spirit of self-reliance and entrepreneurship development.



**HOD**



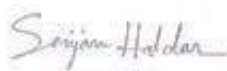
**DEAN / SCHOOL CONCERNED**

ADAMAS UNIVERSITY, KOLKATA SCHOOL OF LIFE SCIENCE AND  
BIOTECHNOLOGY

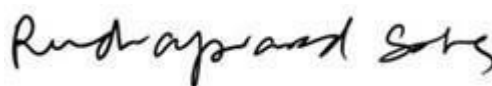
Name of the Programme: B.Sc. Biochemistry (Hons.)

**PROGRAMME EDUCATIONAL OBJECTIVES (PEO)**

- PEO 01** : Acquire basic theoretical and practical domain knowledge.
- PEO 02** : Acquainted with tools and technology related to the field of study.
- PEO 03** : Ability to identify research gaps, comprehend fundamentals, and specialize in the domain.
- PEO 04** : Develop as professional aspirants and sustainable learners.
- PEO 05** : Global outlook with imbibed human values.



**HOD**



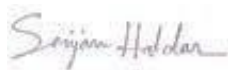
**DEAN / SCHOOL CONCERNED**

ADAMAS UNIVERSITY, KOLKATA SCHOOL OF LIFE SCIENCE AND  
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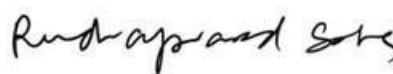
Name of the Programme: B.Sc. Biochemistry (Hons.)

GRADUATE ATTRIBUTE / PROGRAMME OUTCOME (PO)

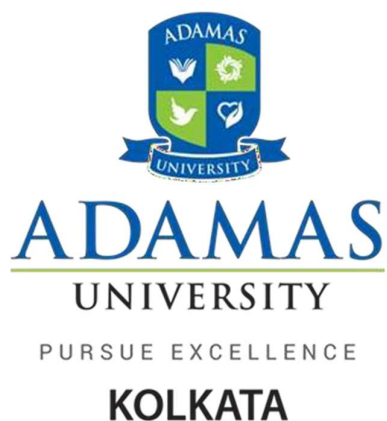
- GA 01/ PO 01: Fundamental Knowledge:** Strong fundamental knowledge in basic and applied field of biochemistry.
- GA 02/ PO 02: Critical Thinking:** Ability to correlate between courses and develop critical/logical thinking.
- GA 03/ PO 03: Skill:** Develop skill set related to biochemistry and allied fields.
- GA 04/ PO 04: Technical Knowledge:** Familiarized with classical as well as modern tools and techniques in biochemistry.
- GA 05/ PO 05: Logical Thinking:** Ability to identify scientific research gaps and problems pertaining to biochemistry and allied fields.
- GA 06/ PO 06: Problem identification ability:** Explore the acquired knowledge and skills of biochemistry to identify approaches for suitable solution.
- GA 07/ PO 07: Analytical Knowledge** Ability to retrieve biological data for a meaningful solution.
- GA 08/ PO 08: Career goals:** Decide upon career path, force the challenges and develop professional aspirations.
- GA 09/ PO 09: Teamwork:** Uphold integrity and collaborative approach in workplace.
- GA 10/ PO 10: Sustainable Development to environment:** To accept and implement learning towards sustainable development.
- GA 11/ PO 11: Development to society:** Practice ethical philosophies and systems in creating and partnering a progressive society.
- GA 12/ PO 12: Development to humanity:** Develop as global citizen to contribute in the greater benefits of humanity.



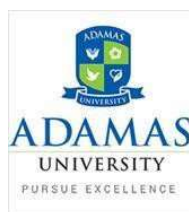
HOD



DEAN / SCHOOL CONCERNED



**ADAMASUNIVERSITY**  
**SCHOOL OF LIFE SCIENCE AND BIOTECHNOLOGY**  
**B.Sc. Biochemistry**  
**(Program Code: BIC3302)**  
**(2022-25)**



**ADAMAS UNIVERSITY, KOLKATA**

ADAMAS UNIVERSITY								
B.Sc.Biochemistry Semester-I								
Type of The Course	Old Course Code	Revised Course Code	CourseName	Contact Hours	L	T	P	Credit
CORE Theory	SBC31101	BIC11001	Molecules of Life	60	3	1	0	4
CORE Practical	SBC31201	BIC12002	Molecules of Life Lab	45	0	0	3	2
CORE Theory	SBC31103	BIC11003	Protein	60	3	1	0	4
CORE Practical	SBC31203	BIC12004	Protein Lab	45	0	0	3	2
AECCI	HEN31105	ENG11057	English Language and Literature	60	2	0	0	2
FOUNDATION	SBC31107	DGS11001	DesignThinking	30	2	0	0	2
Generic Elective 1(GE1*) Theory	SBN31105 /SBT31105 /SCY31105 /SPS31109	BOT11001 /ZOL11001 /CHM11151 /PSG13004	Any one of the following: Elective Botany I /Elective Zoology I/ Elective Chemistry I/ *Elective General Psychology	60	3	1	0	4
Generic Elective 1(GE1) Practical	SBN31205 /SBT31205 /SCY31205	BOT12002 /ZOL12002 /CHM12152	Corresponding Practical (anyone): Elective Botany I Lab/ Elective Zoology I Lab /Elective Chemistry I Lab/	45	0	0	3	2
Total								22

\*6 credit course for non-lab-based subject



ADAMAS UNIVERSITY								
B.Sc.BiochemistrySemester-II								
Type of the Course	Old Course Code	Revised Course Code	CourseName	Contact Hours	L	T	P	Credit
CORE Theory	SBC31102	BIC11005	Enzymes	60	3	1	0	4
CORE Practical	SBC31202	BIC12006	Enzymes Lab	45	0	0	3	2
CORE Theory	SBC31104	BIC11007	Human Physiology	60	3	1	0	4
CORE Practical	SBC31204	BIC12008	Human PhysiologyLab	45	0	0	3	2
AECCII Theory	SGY31106	EVS11112	Environmental Science	30	3	1	0	2
FOUNDATIO N	SBC31402	EIC11001	Venture Ideation	30	3	0	0	2
GenericEle ctive 2 <sup>#</sup> (GE2*) Theory	SBN31106 /SBT3110 6 /SPS3110 9	BOT11003 /ZOL1100 3 /PSG1300 7/CHM111 53	Any one of the following:  Elective Botany II /Elective ZoologyII/ *ElectiveApplied Psychology /Elective Chemistry II	60	3	1	0	4
Generic Elective 2 <sup>#</sup> (GE2*) Practical	SBC31206 /SBT3120 6	BOT12004 /ZOL1200 4/CHM121 54	Corresponding Practical (anyone):  Elective Botany II Lab /Elective Zoology II Lab/ Elective Chemistry Lab II	45	0	0	3	2
Total								22

\*6 credit course for non-lab-based subject





ADAMAS UNIVERSITY

B.Sc. Biochemistry Semester-V

Type of Course	Old Course Code	Revised Course Code	Course Name	Contact Hours Per Week	L	T	P	Credit
CORE Theory	SBC33101	BIC11026	Concept of Genetics	60	3	1	0	4
CORE Practical	SBC33201	BIC12027	Concept of Genetics Lab	45	0	0	3	2
CORE Theory	SBC33103	BIC11028	Gene Expression and Regulation	60	3	1	0	4
CORE Practical	SBC33203	BIC12029	Gene Expression and Regulation Lab	45	0	0	3	2
CORE Discipline Specific Elective I (DSEI) Theory	SBC33105 /SBC33107 /SBC33109/	BIC11030/ BIC11032/ BIC11034	Any one of the following: Bioinformatics/ Basic Microbiology/ Molecular Basis of non-infectious human diseases	60	3	1	0	4
CORE Discipline Specific Elective I (DSEI) Practical	SBC33205 /SBC33207 /SBC33209/	BIC12031/ BIC12033/ BIC12035	Corresponding Practical (any one): Bioinformatics Lab/ Basic Microbiology Lab/ Molecular Basis of non-infectious human diseases Lab	45	0	0	3	2
CORE Discipline Specific Elective II (DSEII) Theory	SBC33111/ SBC33113/ SBC33115	BIC11036/ BIC11038/ BIC11040	Any one of the following: Molecular Basis of Infectious Human Diseases/ Medical Microbiology/ Nutritional Biochemistry	60	3	1	0	4
CORE Discipline Specific Elective II (DSEII) Practical	SBC33211/ SBC33113 /SBC33115	BIC12037/ BIC12039/ BIC12041	Corresponding Practical (any one): Molecular Basis of Infectious Human Diseases Lab/ Medical Microbiology Lab/ Nutritional Biochemistry Lab	45	0	0	3	2
FOUNDATION	SBC33601	BIC14042	Industry Internship	-	-	-	-	2
Total								26

\*6 credit course for non-lab based subject

Note: Offer of DSE courses may vary from year to year based on availability of faculty

ADAMAS UNIVERSITY

B.Sc.Biochemistry Semester-VI

Type of the course	Old Course Code	Revised Course Code	Course Name	Contact Hours	L	T	P	Credit
CORE Theory	SBC33102	BIC11043	Genetic Engineering and Biotechnology	60	3	1	0	4
CORE Practical	SBC33202	BIC12044	Genetic Engineering and Biotechnology Lab	45	0	0	3	2
CORE Theory	SBC33104	BIC11045	Hormones: Biochemistry and Function	60	3	1	0	4
CORE Practical	SBC33204	BIC12046	Hormones Biochemistry and Function Lab	45	0	0	3	2
CORE Discipline Specific Elective III (DSE III)	SBC33106 / SBC33108 / SBC33110	BIC11047 / BIC11049 / BIC11051	Any one of the following: Plant Biochemistry / Advanced Cell Biology / Research Methodology*	60	3	1	0	4
CORE Discipline Specific Elective III (DSE III) Practical	SBC33206 / SBC33208	BIC12048 / BIC12050	Corresponding Practical (anyone): Plant Biochemistry Lab / Advanced Cell Biology Lab	45	0	0	3	2
	SBC33712	BIC15052	Dissertation	4	0	0	4	6
FOUNDATION	SBC33302	BIC15053	Seminar on Contemporary Research in Biochemistry					2
Total								26

**Total credit distribution semester-wise:**

Semester	I	II	III	IV	V	VI	Total
Credits	22	22	30	28	26	26	154

<b>BIC 11001</b>	<b>Molecules of Life (Theory)</b>	L	T	P	C
<b>Version 1.0</b>	<b>Contact Hours: 60</b>	3	1	0	4
<b>Pre-requisites/Exposure</b>	Knowledge of Organic Chemistry at 10+2 level				
<b>Co-requisites</b>	-				

### Course Objectives

1. To gain a deeper understanding structures of different types of carbohydrates.
2. To acquire the knowledge structures of different types of aminoacids.
3. To acquire the knowledge about the different classes of lipids.
4. To gain the knowledge about different types of nucleicacid.
5. To understand the properties of water and its essentiality in biochemistry.

### Course Outcomes

On completion of this course, the students will be able to

- CO1. **Describe** the structure and functions of different type of sugars.
- CO2. **Identify** different types of aminoacids.
- CO3. **Construct** the knowledge about structures of lipids.
- CO4. **Illustrate** the structure of DNA, RNA and their biological significances.
- CO5. **Demonstrate** the properties of water during the characterization of biomolecules.

### Catalog Description

Life is a condition through which living organisms can be differentiated from non-living matters. For example growth, reproduction are the essential physiological phenomena of living organism. To support those physiological characteristics, some molecules are very much essential, which are considered as “molecules of life”. The learning of the detailed structures of those molecules are necessary for understanding the key of the life. So the course consists of the structure of carbohydrates (energy source), amino acids (structural unit of protein), water (a major part of protoplasm of the cell), lipids (building block of the cell) and nucleotides (responsible for construction of energy currency and genetic material).

### Course Content

#### Unit I: Carbohydrates and glycobiology

[ 12 Lecture hours]

Monosaccharides - structure of aldoses and ketoses, ring structure of sugars, conformations of sugars, mutarotation, anomers, epimers and enantiomers, structure of biologically important sugar derivatives, oxidation of sugars. Formation of disaccharides, reducing and nonreducing disaccharides. Polysaccharides – homo- and heteropolysaccharides, structural and storage polysaccharides. Structure and role of proteoglycans, glycoproteins and glycolipids (gangliosides and lipopolysaccharides). Carbohydrates as informational molecules, working with carbohydrates.

**UnitII: Lipids****[ 12 Lecturehours]**

Building blocks of lipids - fatty acids, glycerol, ceramide. Storage lipids - triacyl glycerol and waxes. Structural lipids in membranes – glycerophospholipids, galactolipids and sulpholipids, sphingolipids and sterols, structure, distribution and role of membrane lipids. Plantsteroids. Lipids as signals, cofactors and pigments.

**Unit III Aminoacids****[ 12 Lecturehours]**

Structure and classification, physical, chemical and optical properties of amino acids

**Unit IV Nucleic acids****[ 12 Lecturehours]**

Nucleotides - structure and properties. Nucleic acid structure – Watson-Crick model of DNA. Structure of major species of RNA - mRNA, tRNA and rRNA. Nucleic acid chemistry - UV absorption, effect of acid and alkali on DNA. Other functions of nucleotides - source of energy, component of coenzymes, second messengers.

**UnitV: Water****[ 12 Lecturehours]**

Unique properties, weak interactions in aqueous systems, ionization of water, buffers, water as a reactant and fitness of the aqueous environment.

**Reference Books**

1. Lehninger: Principles of Biochemistry (2013) 6th ed., Nelson, D.L. and Cox, M.M., WH Freeman and Company (New York), ISBN:13: 978-1-4641-0962-1 / ISBN:10:1-42923414-8.
2. Biochemistry (2011) 4th ed., Donald, V. and Judith G.V., John Wiley & Sons Asia Pvt. Ltd. (New Jersey), ISBN:978-1180-25024.
3. Biochemistry, Lubert Stryer, 8th Edition.
4. Organic Chemistry, Vol 1 & 2., I.L. Finar
5. Chemistry of Nucleic acids, Adams.
6. Organic Chemistry, Nasipuri.
7. Biochemical Calculations, Irwin Segel

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination**  
**Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

### Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and Pos		
	Course Outcomes (COs)	Mapped Program Outcomes
<b>CO1</b>	<b>describe</b> the structure and functions of different type of sugars.	<b>PO1, PO2, PO6</b>
<b>CO2</b>	<b>identify</b> different types of amino acids.	<b>PO1, PO2, PO3, PO4</b>
<b>CO3</b>	<b>illustrate</b> the knowledge about structures of lipids.	<b>PO1, PO2, PO5, PO7</b>
<b>CO4</b>	<b>explain</b> the structure of DNA, RNA and their biological significances.	<b>PO1, PO2, PO8, PO9</b>
<b>CO5</b>	<b>demonstrate</b> the properties of water during the characterization of biomolecules.	<b>PO1, PO2, PO6, PO7</b>

		Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking	Problem Identification Ability	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to Environment	Development to Society.	Development to Humanity
Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
BIC11001	Molecules of Life	3	3	1	1	2	2	2	1	1	-	-	-

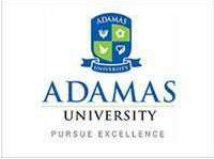
1=weakly mapped

2= moderately mapped

3=strongly mapped



## Model Question Paper

<b>Name:</b>  <b>Enrolment No:</b>	 <p style="font-size: small; margin: 0;">ADAMAS UNIVERSITY PURSUE EXCELLENCE</p>		
<b>Course: BIC 11001 –Molecules of Life</b>			
<b>Program: B.ScBiochemistry</b> <b>Semester: Odd 2020-21</b>		<b>Time: 03 Hrs.</b> <b>Max. Marks:40</b>	
<b>Instructions:</b> Attempt any three questions from <b>Section A</b> (each carrying 5 marks); any <b>Two Questions</b> from <b>Section B</b> (each carrying 10 marks).			
<b>Section A</b>			
1.	<b>Discuss</b> role of phospholipids in membrane constitution. (R)	<b>5</b>	<b>CO3</b>
2.	<b>Compare</b> dextrorotatory and levorotatory sugars. (U)	<b>5</b>	<b>CO1</b>
3.	<b>Define</b> syn and anti configuration of nucleotides. (R)	<b>5</b>	<b>CO4</b>
4.	How will you <b>evaluate</b> rancidity of a fat sample? (U)	<b>5</b>	<b>CO3</b>
<b>SECTION B (Attempt any Two Questions)</b>			
5.	<b>Explain</b> mutarotation. <b>Justify</b> its occurrence in aqueous medium. (An)	<b>10</b>	<b>CO1</b>
6.	<p>a) Histidine and lysine both are basic amino acids, still the nature of their titration curves are different, that is also reflected in their biological significance also-justify. (An)</p> <p>b) An aqueous solution of D-galactose has specific rotation of <math>+80.2^{\circ}</math> after standing for some hours. The specific rotation of <math>\alpha</math>-D-galactose and <math>\beta</math>-D-galactose are <math>+150.7^{\circ}</math> and <math>+52.8^{\circ}</math> respectively. <b>Evaluate</b> the proportions of <math>\alpha</math> and <math>\beta</math>-D galactose in the mixture. (An)</p> <p>c) <b>Distinguish</b> between dextrin and dextran. (R)</p>	<b>4</b> <b>4</b> <b>2</b>	<b>CO2</b> <b>CO1</b> <b>CO1</b>

7.	<p><b>1. a)</b> The saponification number of a sample of butter fat is 250 and iodine number of the butter fat is 68. Calculate the average molecular weight in the present triacylglycerols and the number of double bonds present in the butter fat sample. (Hint: atomic weight of I is=126.9)</p> <p><b>b)</b> Write down the structure and IUPAC name of following fatty acid:  Ricinoleic acid (18:1<math>\Delta^9</math>C-8 D-hydroxylated)</p> <p><b>c)</b> Design an experiment: To determine the Reichert-Meissl and Polenske number from a same fat sample. (Ap)</p>	<p style="text-align: center;">4 2 4</p>	<p style="text-align: center;"><b>CO3</b></p>
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<b>BIC12002</b>	<b>Molecules of Life Lab(Practical)</b>	L	T	P	C
<b>Version 1.0</b>	<b>Contact Hours: 45</b>	0	0	3	2
<b>Pre-requisites/Exposure</b>	Knowledge of Organic Chemistry at 10+2 level				
<b>Co-requisites</b>	Theory of Molecular biology				

### Course Objectives

1. To understand the presence of different types of carbohydrates in different samples.
2. To understand the presence of amino acids in a solution.
3. To characterize the lipids
4. To gain the knowledge about different types of nucleic acids.
5. To understand the properties of water.

### Course Outcomes

On completion of this course, the students will be able to

- CO1. **Describe** the chemical behaviour of different type of sugars
- CO2. **Identify** physical and chemical properties of amino acids
- CO3. **Develop** the knowledge about characteristics of lipids.
- CO4. **Illustrate** the chemical behavior of DNA and RNA for distinguishing them from each other.
- CO5. **Define** the properties of water during the characterization of biomolecules.

### Catalogue Description

Life is a condition through which living organisms can be differentiated from non-living matters. For example growth, reproduction are the essential physiological phenomena of living organism.

To support those physiological characteristics, some molecules are very much essential, which are considered as "molecules of life". The learning of the detailed structures of those molecules are necessary for understanding the key of the life. So the course consists of the identification of carbohydrates (energy source), amino acids (structural unit of protein), water (a major part of protoplasm of the cell), lipids (building block of the cell) and nucleotides (responsible for construction of energy currency and genetic material) by several qualitative tests.

## Course Content

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1. Safety measures in laboratories and Properties of Water. [5hours]
2. Preparation of normal and molar solutions, Preparation of Buffer. [5hours]
3. Qualitative Analysis of Carbohydrates. [5hours]
4. Qualitative Analysis of Proteins. [10hours]
5. Qualitative Analysis of Lipids. [10 hours]
6. Assay of Salivary Amylase. [10hours]
7. Estimation of vitamin C. [10hours]
8. Nucleic acid Estimation. [5 hours]

## Reference Books

1. Introduction to Practical Biochemistry: by Sawhney and Singh Biochemistry (2011) 4<sup>th</sup> ed.,
2. Advanced Practical Chemistry; Subhas Chandra Das
3. Organic Chemistry, Vol 1 & 2., I.L. Finar.
4. Chemistry of Nucleic acids, Adams
5. Organic Chemistry, Nasipuri.
6. Biochemical Calculations, Irwin Segel

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination**  
**Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

## Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and Pos		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	<b>describe</b> the chemical behaviour of different type of sugars	PO1, PO2, PO3, PO4, PO7
CO2	<b>explain</b> physical and chemical properties of amino acids	PO1, PO2, PO3, PO4, PO5, PO7
CO3	<b>demonstrate</b> the knowledge about characteristics of lipids	PO1, PO2, PO4, PO6, PO7, PO9
CO4	<b>illustrate</b> the chemical behavior of DNA and RNA for distinguishing them from each other.	PO1, PO2, PO5, PO6, PO7, PO9
CO5	<b>define</b> the properties of water during the characterization of biomolecules	PO1, PO2, PO3, PO4, PO5, PO6, PO7

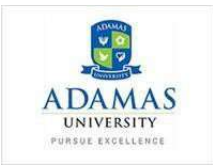
Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
BIC12002	Molecules of Life Lab	3	3	3	3	2	3	3	-	2	-	-	-
		Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking	Problem Identification Ability	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to Environment	Development to Society.	Development to Humanity

1=weakly mapped

2= moderately mapped

3=strongly mapped

## Model Question Paper

<b>Name:</b>  <b>Enrolment No:</b>		 <p style="font-size: small; margin: 0;">ADAMAS UNIVERSITY PURSUE EXCELLENCE</p>	
<b>Course: BIC12002 –Molecules of Life Lab</b>			
<b>Program: B.ScBiochemistry</b> <b>Semester:Odd 2020-21</b>	<b>Time: 03 Hrs.</b> <b>Max. Marks:40</b>		
<b>Instructions:</b> Attempt any two questions from <b>Section A</b> (each carrying 10 marks); <b>Section B</b> (each carrying 10 marks) is Compulsory.			
<b>Section A (Attempt any Two)</b>			
1.	<b>Identify</b> the molecules from the given sample A and B. (Ap) and Write the results along with the experimental procedure in detail. (An)	4 4 2	<b>CO1, CO2, CO3, CO4</b>
2.	<b>Estimate</b> the quantity of amino acid from the given sample (Ap) and detail the experimental procedure, observation and inference. (An)	4 4 2	<b>CO2</b>
3.	<b>Estimate</b> the quantity of Vitamin C from the given sample (Ap) and detail the experimental procedure, observation and inference. (An)	4 4 2	<b>CO5</b>
<b>SECTION B (Compulsory)</b>			
4.	<b>Viva-Voce</b> (U/An/Ap/R)	<b>10</b>	<b>CO1, CO2, CO3, CO4, CO5</b>
5.	<b>Laboratory Note Book</b> (U/An/Ap/Ev)	<b>10</b>	<b>CO1, CO2, CO3, CO4, CO5</b>

<b>BIC11003</b>	<b>PROTEIN (THEORY)</b>	<b>L</b>	<b>T</b>	<b>P</b>	<b>C</b>
<b>Version 1.0</b>	<b>Contact Hours - 60</b>	3	1	0	4
<b>Pre-requisites/Exposure</b>	PLUS TWO (12 <sup>th</sup> ) LEVEL BIOLOGY				
<b>Co-requisites</b>	--				

### Course Objectives

- 1.To provide students the basic understanding of structure and properties of amino acids, peptides and polypeptides.
- 2.To provide in depth knowledge of structural and functional diversity of proteins.
- 3.To describe protein purification techniques along with other biophysical techniques characterization.
- 4.To give general overview of protein databases and some special proteins.

### Course Outcomes

On completion of this course, the students will be able to

- CO1. **Explain** various structural components of amino acids, peptides and polypeptides.  
CO2. **Summarise** structural features of protein and compare their functional aspects.  
CO3. **Describe** different techniques for protein purification.  
CO4: **Illustrate** the role of different protein databases.  
CO5: **Compare** the structure and function of some special proteins.

### Catalogue Description

The core-course of ‘protein’ will help to understand the classification, structure and properties of amino acids. This course is a step-by-step journey from the basic to modern concepts of protein biochemistry. Furthermore, the application of biophysical methods along with bioinformatics databases have also been included for a thorough understanding of the subject. All the lectures will be devoted on discussions of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as power point presentation, audio visual virtual lab session as per requirement. The tutorials will enable the students with problem-solving ability led by the course coordinator. Students will perceive the basic concepts of the subject via exercise and discussions with the coordinator.

## **Course Content**

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### **Protein (BIC11003)**

#### **Unit 1-Introduction to amino acids, peptides and proteins: [6 Lecture Hours]**

Amino acids and their properties - hydrophobic, polar and charged. Biologically important peptides - hormones, antibiotics and growth factors. Multimeric proteins, conjugated proteins and metalloproteins. Diversity of function

#### **Unit 2-Extraction of proteins for downstream processing and Separation techniques: [6 Lecture Hours]**

Solubilization of proteins from their cellular and extracellular locations. Use of simple grinding methods, homogenization, ultrasonication, French press and centrifugation. Ammonium sulphate fractionation, solvent fractionation, dialysis and lyophilization. Ion-exchange chromatography, molecular sieve chromatography, hydrophobic interaction/reverse phase chromatography, affinity chromatography, HPLC and FPLC

#### **Unit 3-Characterization of proteins: [6 Lecture Hours]**

Determination of purity, molecular weight, extinction coefficient and sedimentation coefficient, IEF, SDS-PAGE and 2-D electrophoresis.

#### **Unit 4-Covalent structure of proteins: [6 Lecture Hours]**

Organization of protein structure into primary, secondary, tertiary and quaternary structures. N-terminal and C-terminal amino acid analysis. Sequencing techniques - Edman degradation. Generation of overlap peptides using different enzymes and chemical reagents. Disulfide bonds and their location. Mass spectrometric analysis, tandem MS. Solid phase peptide synthesis

#### **Unit 5-Three dimensional structures of proteins: [6 Lecture Hours]**

Nature of stabilizing bonds - covalent and non-covalent. Importance of primary structure in folding. The peptide bond - bond lengths and configuration. Dihedral angles  $\psi$  and  $\phi$ . Helices, sheets and turns. Ramachandran map. Techniques used in studying 3-D structures - X-ray diffraction and NMR. Motifs and domains. Tertiary and quaternary structures. Structures of myoglobin and haemoglobin



### **Unit 6-Protein folding and conformational diseases: [8 Lecture Hours]**

Denaturation and renaturation of Ribonuclease A. Introduction to thermodynamics of folding and molten globule. Assisted folding by molecular chaperones, chaperonins and PDI. Defects in protein folding. Diseases –Alzheimer's and Prion based.

### **Unit 7-Introduction to protein structure databases: [8 Lecture Hours]**

Protein sequence and structure databases (PDB). Use of sequence and domain information. Viewing protein structures using *in silico* tools.

### **Unit 8-Myoglobin, haemoglobin and Specialized proteins -[9 Lecture Hours]**

Antibodies and actin-myosin motors

Oxygen binding curves, influence of 2,3-BPG, CO<sub>2</sub> and Cl<sup>-</sup>. Hill plot. Cooperativity between subunits and models to explain the phenomena - concerted and sequential models. Haemoglobin disorders.

### **Unit 9-Membrane proteins: [6 Lecture Hours]**

Integral and membrane associated proteins. Hydrophathy plot to predict transmembrane domains. Significance of membrane proteins - bacteriorhodopsin.

#### **Textbook:**

1. Lehninger: Principles of Biochemistry (2013) 6<sup>th</sup> ed., Nelson, D.L. and Cox, M.M., W.H. Freeman and Company (New York), ISBN:13: 978-1-4641-0962-1 / ISBN:10:1-4292-3414-8.

#### **Reference books:**

1. Physical Biochemistry (2009) 2nd ed., Sheehan, D., Wiley-Blackwell (West Sussex), ISBN: 9780470856024 / ISBN:9780470856031.

2. The Tools of Biochemistry (1977; Reprint 2011) Cooper, T.G., Wiley India Pvt. Ltd. (New Delhi), ISBN:978-81-265-3016-8.

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination**  
**Examination Scheme:**

<b>Components</b>	<b>Mid Term</b>	<b>Attendance</b>	<b>Class Assessment</b>	<b>End Term</b>
<b>Weightage (%)</b>	<b>20</b>	<b>10</b>	<b>30</b>	<b>40</b>

**Relationship between the Course Outcomes (COs) and Program Outcomes (POs)**

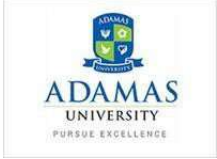
<b>Mapping between COs and POs</b>		
	<b>Course Outcomes (COs)</b>	<b>Mapped Program Outcomes</b>
<b>CO1</b>	<b>explain</b> various structural components of amino acids, peptides and polypeptides	<b>PO1</b>
<b>CO2</b>	<b>summarise</b> structural features of protein and compare their functional aspects.	<b>PO1, PO2</b>
<b>CO3</b>	analyse and <b>choose</b> between different techniques for protein purification.	<b>PO1, PO2, PO3, PO4, PO7</b>
<b>CO4</b>	<b>illustrate</b> the role of different protein databases	<b>PO1, PO2, PO5, PO6, PO7</b>
<b>CO5</b>	<b>compare</b> the structure and function of some special proteins	<b>PO1, PO2, PO4, PO5, PO7</b>

		Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking	Problem Identification Ability	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to Environment	Development to Society.	Development to Humanity
Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
<b>BIC11003</b>	<b>PROTEIN (THEORY)</b>	3	3	3	3	2	2	3	-	-	-	-	-

1=weakly mapped

2= moderately mapped

3=strongly mapped

<b>Name:</b>			
<b>Enrolment No:</b>			
<b>Course: BIC11003 - PROTEIN (THEORY)</b>		<b>Time: 03Hrs.</b>	
<b>Program: B.Sc. Biochemistry</b>		<b>Max. Marks: 40</b>	
<b>Semester: Odd 2019-20</b>			
<b>Instructions:</b>			
Attempt any <b>four</b> questions from <b>Section A</b> (each carrying 5 marks); any <b>two</b> questions from <b>Section B</b> (each carrying 10 marks).			
<b>SECTION A ( Attemptany Four questions)</b>			
1.	What is peptide bond? <b>Explain</b> with a suitable reaction. (R)	2+3	CO1
2.	<b>Classify</b> amino acids based on their side chains. 'W' stands for which amino acid? (U)	4+1	CO2
3.	What are the four levels of protein structure? <b>Differentiate</b> between alpha helix and beta sheet. (R)	4	CO3
4.	<b>Enlist</b> 3 important techniques for protein purification. Also mention the roles SDS and PEG in protein purification. (R)	2+3	CO3
5.	<b>Explain</b> the biochemical cause behind the development of thalassemia and sickle cell anemia. (U)	3+2	CO1
<b>SECTION B (Attempt any Two questions)</b>			
6.	<b>Illustrate</b> the technique of X-ray crystallography. Add a note on integral proteins. What is the reason behind Bohr effect? Name of metalloprotein. (R)	4+2+3+1	CO3
7.	Name two protein databases. <b>Explain</b> 'salting out' phenomena in detail. What is bacteriorhodopsin? <b>Compare</b> hemoglobin and myoglobin with respect to oxygen carrying property. (U)	2+4+2+2	CO1 CO2
8.	<b>Illustrate</b> the role of proline as helix breaker. What are hydrophathy plots? <b>Compare</b> between motifs and domain with example. Where do you find triple helix? Explain briefly. (U)	2+2+3+3	CO1 CO2
9.	<b>Outline</b> the principles of Edman degradation and solid phase peptide synthesis. <b>Analyse</b> the roles of different chromatographic techniques by briefly describing their principle. (An)	2+3+3+2	CO4 CO3

<b>BIC12004</b>	<b>PROTEIN LAB (PRACTICAL)</b>	<b>L</b>	<b>T</b>	<b>P</b>	<b>C</b>
<b>Version 1.0</b>	<b>Contact Hours - 45</b>	0	0	3	2
<b>Pre-requisites/Exposure</b>	PLUS TWO LEVEL BIOLOGY				
<b>Co-requisites</b>	Amino Acid: structure and function				

### Course Objectives

1. To provide students with hands-on training in the quantitative studies related to protein biochemistry.
2. Students will need to become proficient with terms, techniques, and applications.

### Course Outcomes

On completion of this course, the students will be able to

- CO1. **Explain** the basis of spectrophotometry for the determination of protein content.
- CO2. **Demonstrate** the principles of different protein estimation techniques.
- CO3. **Compare** between different protein purification techniques.
- CO4. **Illustrate** modern tools like spectrophotometers, colorimeters, micropipettes etc.
- CO5. **Develop** the knowledge of different modern tools and techniques in biological samples from everyday life.

### Catalog Description

The core course “protein lab” is a practical paper which has been designed to provide the knowledge of qualitative estimation of protein molecules. It deals with all the basic theories and modern techniques of protein estimation. Students will be able to understand the basic concepts of spectrophotometers and other techniques and will practice hands-on all of them. They will also learn to compare and use these methods for practical purpose. All the lectures will be devoted on discussions of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as powerpoint presentation, audiovisual virtual lab session as per requirement. The tutorials will familiarize the students with practical problem-solving techniques led by the course coordinator. Students will strongly grab the basic concepts of the subject via exercise and discussions with the coordinator.

### Course Content

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#### PROTEIN LAB

1. Spectrophotometry: Theory, Instrumentation and Application. [5 LectureHours]
2. Estimation of proteins using UV absorbance. [5 LectureHours]
3. Estimation of proteins using Biuret method. [5 LectureHours]
4. Microassay of proteins using Lowry method. [5 LectureHours]
5. Estimation of proteins using Bradford Method. [5 LectureHours]
6. Isoelectric focusing of casein. [10 LectureHours]
7. Estimation of protein content in yeast cells. [10 LectureHours]
8. SDS-PAGE analysis of proteins. [10 LectureHours]

## SUGGESTED READINGS

1. Lehninger: Principles of Biochemistry (2013) 6<sup>th</sup>ed., Nelson, D.L. and Cox, M.M., W.H. Freeman and Company (New York), ISBN:13: 978-1-4641-0962-1 / ISBN:10:1-4292-3414-8.
2. Physical Biochemistry (2009) 2<sup>nd</sup>ed., Sheehan, D., Wiley-Blackwell (West Sussex), ISBN: 9780470856024 / ISBN:9780470856031.
3. The Tools of Biochemistry (1977; Reprint 2011) Cooper, T.G., Wiley India Pvt. Ltd. (New Delhi), ISBN:978-81-265-3016-8.

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

## Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and Pos		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	<b>explain</b> the basis of spectrophotometry for the determination of protein content.	PO1, PO2, PO3, PO4
CO2	<b>demonstrate</b> the principles of different protein estimation techniques.	PO1, PO4, PO9, PO10
CO3	<b>compare</b> between different protein purification techniques.	PO1, PO2, PO3, PO4
CO4	<b>illustrate</b> their concept on modern tools like spectrophotometers, colorimeters, micropipettes etc.	PO1, PO3, PO5, PO6, PO7
CO5	<b>develop</b> the knowledge of different modern tools and techniques in biological samples from everyday life.	PO1, PO4, PO5, PO7, PO8


Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
		Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking	Problem Identification Ability	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to Environment	Development to Society.	Development to Humanity
<b>BIC12004</b>	PROTEIN LAB (PRACTICAL)	3	2	3	2	2	1	2	1	1	1	-	-

1=weakly mapped

2= moderately mapped

3=strongly mapped

## Model Question Paper

<b>Name:</b>  <b>Enrolment No:</b>		 <p style="font-size: small; margin: 0;">ADAMAS UNIVERSITY PURSUE EXCELLENCE</p>	
<b>Course: BIC12004 – PROTEIN LAB (PRACTICAL)</b>			
<b>Program: B.Sc. Biochemistry</b> <b>Semester: Odd 2019-20</b>	<b>Time: 03Hrs.</b> <b>Max. Marks:40</b>		
<b>Instructions:</b> Attempt any <b>two</b> questions from <b>Section A</b> (each carrying 10 marks); <b>Section B</b> is Compulsory (carrying 10 marks).			
<b>Section A ( Attemptany Two)</b>			
1.	a) Write the principle behind spectrophotometer. b) Draw a standard curve with the supplied data and determine the unknown protein concentration. (An)	<b>4</b> <b>6</b>	<b>CO1</b> <b>CO2</b>
2.	a) Explain the basic theory of SDS-PAGE. (U)	<b>4</b> <b>6</b>	<b>CO3</b> <b>CO4</b>
	b) Determine the unknown protein concentration using the supplied samples with the help of Biuret reaction. (Ap)		
3.	a) Write the principle behind ion-exchange chromatography.(R) b) Determine the unknown protein concentration using the supplied samples with the help of Lowry assay. (An)	<b>4</b> <b>6</b>	<b>CO3</b> <b>CO2</b>
	<b>SECTION B is compulsory</b>		
4.	<b>Viva-voce (U/An/Ap/R/Ev)</b>	<b>10</b>	<b>CO1</b> <b>CO2</b> <b>CO3</b> <b>CO4</b> <b>CO5</b>
5.	<b>Practical copy (U/Ap/Ev)</b>	<b>10</b>	<b>CO1</b> <b>CO2</b> <b>CO3</b>

Course Code ENG11057	Course Name English Language and Literature	L	T	P	Credit
		2	0	0	2
Course Outcome	Course Outcome Statement				
CO1	Have a basic understanding of communication processes and to know the practical implications and its challenges at the work place.				
CO2	Understand the practical uses of English grammar and to use grammar correctly and unambiguously.				
CO3	Be acquainted with some texts of English Literature and develop their reading and comprehension skills.				
CO4	Develop the ability to identify difficult sounds, words and phrases to support listening comprehension and be familiar with the various strategies of reading and develop the ability to read texts with fluency, understanding and competence.				
CO5	Acquire fluency in speaking English in order to carry out effective professional communication.				
CO6	Be familiar with different formats of business communication like reports, letters, CVs and other technical writings.				
<b>Course Content:</b> <b>Unit 1: Communication</b> a) Theory and Types of Communication b) Verbal and Non-verbal Communication					



- c) **Barriers and Strategies of Communication**
- d) **Workplace and Telephone Communication**

**Unit 2: Grammar and Syntax:**

- a) **Tense**
- b) **Parts of Speech**
- c) **Articles**
- d) **Prepositions**
- e) **Sentence-Making**
- f) **Voicechange**
- g) **Synonyms and antonyms**
- h) **One- Word Substitutions**

**Unit 3: Literature: Reading and Textual analysis**

- a) **Close Reading: Short Story: “The Gift of the Magi”: by O’Henry**
- b) **Paraphrasing: Poem: “Stopping by Woods on a Snowy Evening”: Robert Frost**
- c) **Summary: Non-fiction: Extracts from *The Great Derangement: Climate Change and The Unthinkable* by Amitav Ghosh**
- d) **Reading Comprehension**
- e) **Interpreting Graphs and Charts**

**Unit 4: Speaking skills**

- a) **Introduction**
- b) **Interpersonal Communication**
- c) **Group Discussion**
- d) **Interview**

**Unit 5: Writing Skills**

- a) **Composition**
- b) **Letter writing- CV and application letter**
- c) **Report Writing**
- d) **Memo-Writing**
- e) **Note-making**
- f) **Business Communication**

**Unit 6: Listening skills**

- a) **Active Listening**
- b) **Types of Listening**
- c) **Listening Exercises**
- d) **Reading Exercises: Comprehension**

**Text Book:**

1. **Kaul Asha. *Effective Business Communication*. PHI Learning Pvt Ltd. 2014.**
2. **Wren and Martin. *High School Grammar And Composition*. S. Chand, 1995.**
3. **Lewis, Norman. *Word Power Made Easy*. Anchor: 2014.**
4. **Riordan, Daniel G & Pauley Steven A. : *Technical Report Writing Today*. 2004.**
5. **Hamp-Lyons and Heasley, B. *Study Writing; A Course in Written English. For Academic and Professional Purposes*, Cambridge Univ. Press, 2006.**
6. **Quirk R., Greenbaum S., Leech G., and Svartik, J. *A Comprehensive Grammar of the English language*, Longman: London, 1985.**
7. **Gupta, A. *English Reading Comprehension*. Ramesh Publishing House, 2009.**
8. **Balasubramaniam, T. *A Textbook of English Phonetics for Indian Students*. Macmillan: 2012.**

DGS11001	Design Thinking	L	T	P	C
Version 1.0	Contact hours-30 lecture hours	2	0	0	2
Pre-requisites/Exposure	Knowledge of analyzing society problems and product usage problems and a zeal to improve the current situation, in addition to knowing to using laptop/computers, internet, social media interaction, file sharing and uploading, email and communication etiquettes.				
Co-requisites	-				

### Course Objectives

1. To enable students to acquire knowledge, imagination and be more assertive on opinions on problems in society.
2. To enable students to learn basics of research, data collection, analysis, brainstorming and find solutions to issues.
3. To make them understand Design Thinking methodologies to problems in field of study and other areas as well.
4. To help students to understand future Engineering positions with scope of understanding dynamics of working between inter departments of a typical OEM.

### Course Outcomes

On completion of this course, the students will be able to

- CO1. **Examine** design thinking concepts and principles
- CO2. **Illustrate** the methods, processes, and tools of design thinking
- CO3. **Assess** the Design Thinking approach and model to real world scenarios
- CO4. **Demonstrate** the role of primary and secondary research in the discovery stage of design thinking

### Catalogue Description

Design thinking course is a completely online course offered to the first year UG programs across all streams. This course is designed to help understand the steps followed in the process of designing a solution to a problem.

### Course Content

#### Unit I: 4 Lecture Hours

**WHAT IS DESIGN THINKING:** Designers seek to transform problems into opportunities. Through collaboration, teamwork, and creativity, they investigate user needs and desires on the way to developing human-centered products and/or services. This approach is at the very heart of design thinking.

**UnitII: 4 LectureHours**

**THE DESIGN THINKING MODEL:**A tool that helps guide you along a design thinking path. The model does this by providing a series of activities that that will help you effectively design a product, service or solution to a user’s need. The model presents the approach as a process, allowingustolookateachstep–orphase–alongthejourneytothedevelopmentofafinaldesign.

**UnitIII: 4 LectureHours**

**PHASE 1: DISCOVER:** Begin the design thinking process with the Discover phase, where you will identify the specific problem your design is intended to solve, as well as important usability aspects from those who will use your design. Discovery can be performed through a variety of different research methods which you will learn in this module.

**UnitIV: 4 LectureHours**

**PHASE2:DEFINE:**IntheDefinephase, youcometounderstandthe problem. Weoftenreferto this as framing the problem. You can do this by using a variety of tools, including storytelling, storyboarding, customer journey maps, personas, scenarios, andmore.

**UnitV: 4 LectureHours**

**PHASE 3: DEVELOP:** Turn your attention to solving the problem. In this phase you brainstorm custom creative solutions to the problems previously identified and framed. To do this, you conceptualize in any way that helps, putting ideas on paper, on a computer, or anywhere whereby they can be considered and discussed.

**UnitVI: 4 LectureHours**

**PHASE 4: DELIVER:** This phase is all about testing and building concepts. Here you take allof the ideas that have been discussed to this point and bring them a little closer to reality by building aconcept;somethingthatmakes iteasier forausertoexperiencea design.Thisconceptisreferred to as aprototype.

**UnitVII: 4 LectureHours**

**PHASE 5: ITERATE:** You will test the prototype of your design solution, collecting and acting on feedback received. These actions may mean minor or major revisions to your design, and are repeated as often as necessary until a solution is reached. Tools such as focus groups and questionnaires are used to help you collect feedback that can help with your final design.

**UnitVIII: 2 LectureHours**

**BEYOND DESIGN THINKING:** The Design Thinking Model is a tool that helps guide you along a design thinking path. The model does this by providing a series of activities that that will help you effectively design a product, service or solution to a user’s need. The model presents the approach as a process, allowing us to look at each step – or phase – along the journey to the development of a finaldesign.

**Reference Books**

1. Brown, Tim. “What We Can Learn from Barn Raisers.” Design Thinking: Thoughts by Tim Brown. Design Thinking, 16 January 2015. Web. 9 July 2015.
2. Knapp, Jake. “The 8 Steps to Creating a Great Storyboard.” Co.Design. Fast Company Inc., 21 Dec. 2013. Web. 9 July 2015.

3. Van der Lelie, Corrie. "The Value of Storyboards in the Product Design Process." Journal of Personal and Ubiquitous Computing 10.203 (2006): 159–162. Web. 9 July 2015.[PDF].
4. Millenson, Alisson. "Design Research 101: Prototyping Your Service with a Storyboard." Peer Insight. Peer Insight, 31 May 2013. Web. 9 July 2015.

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

**Relationship between the Course Outcomes (COs) and Program Outcomes (POs)**

Mapping between COs and Pos		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	<b>examine</b> design thinking concepts and principles	PO1, PO5, PO11
CO2	<b>illustrate</b> the methods, processes, and tools of design thinking	PO1, PO2, PO7, PO11
CO3	<b>construct</b> the Design Thinking approach and model to real world scenarios.	PO1, PO2, PO4, PO7
CO4	<b>examine</b> the role of primary and secondary research in the discovery stage of design thinking	PO1, PO5, PO7

		Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking Ability	Problem Identification	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to Environment	Development to Society	Development to Humanity
Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
DGS11001	Design Thinking	3	2	-	1	2	-	3	-	-	-	2	-

- 1=weakly mapped  
 2= moderately mapped  
 3=strongly mapped

**Model Question Paper**

**ADAMAS UNIVERSITY  
 SCHOOL OF ENGINEERING AND TECHNOLOGY  
 END-SEMESTER EXAMINATION: JULY 2020**

Name of the Program: B. SC.Semester:I      Stream: BIOCHEMISTREY  
 PAPER TITLE:DesignThinking  
 PAPER CODE: DGS11001  
 MaximumMarks:40      Time duration: 3hours  
 Total No ofquestions:12      Total No of Pages:01

**Instruction for the Candidate:**

1. At top sheet, clearly mention Name, Univ. Roll No., Enrolment No., Paper Name & Code, Date ofExam.
2. All parts of a Question should be answered consecutively. Each Answer should start from a freshpage.
3. Assumptions made if any, should be stated clearly at the beginning of youranswer.

<b>Section A (Answer All the Questions) (5 x 1 = 5)</b>			
1.	List the steps involved in Design Thinking.	U	CO1
2.	Enumerate the basic elements of Design Thinking.	U	CO2
3.	Define Napkin Pitch.	R	CO3
4.	What is Assumption testing?	R	CO4
5.	Give the principles of Ethnography.	U	CO2
<b>SECTION B (Attempt any Three Questions) (3 x 5 = 15)</b>			
4.	Briefly explain the importance of ethnography in design thinking?	U	CO2
5.	What are the successive steps for concept development?	Ap	CO3
6.	Elucidate the different types of concept development strategies.	Ap	CO3
7.	Explainwith Example: surface keys for Assumption Testing.	Evaluate	CO4
<b>SECTION (Answer Any Two Questions) (2 x 10 = 20)</b>			
8.	Explain in detail about importance of prototyping in Design Thinking.	U	CO4
9.	Write animportance of involving stakeholders in developing new concepts and Plan for conducting experiments within short time and inexpensively.	Create	CO3
10.	Distinguishbetween design thinking and visualization of a problem.	An	CO1

<b>BOT11001</b>	<b>ELECTIVE BOTANY – I (Theory)</b>	<b>L</b>	<b>T</b>	<b>P</b>	<b>C</b>
<b>Version 1.0</b>	<b>Contact Hours - 60</b>	3	1	0	4
<b>Pre-requisites/Exposure</b>	12 <sup>th</sup> with Biology as one subject				
<b>Co-requisites</b>	-				

### Course Objectives:

1. Students will be able to develop basic knowledge in plant science and also be able to explain various aspects of plant growth and development of lower group of plants;
2. Students will be able to understand intrinsic mechanism of plant growth and developments and their correlation with surrounding ecosphere as well biosphere;
3. Students will be able to develop fundamental knowledge and can be implemented techniques practices of economic important plant.

### Course Outcomes

On completion of this course, the students will be able to

**CO1. Illustrate** in detail phyecology in various aspects and also implementation of algal biotechnology for commercial purposes.

**CO2. Explain** and Categorize in detail various divisions of Fungi and their commercial importance as well harmful effects.

**CO3. Demonstrate** basic knowledge and economic importance of Lichen.

**CO4. Summarize** the basic concept of plant pathology and also can be able implement this knowledge in applied fields.

**CO5. Develop** fundamental knowledge and can be implemented this skill for cultivation practices and commercial uses of economic important plant.

### Catalogue Description

Elective Botany I lecture course covers a vast range of basic plant science including various branches of Botanical subjects. The course takes a broader approach and covers many aspects of lower group of plants as well physiological phenomena of all the plants community. Moreover, this curriculum covers cultivation practices of economic importance plants. Classroom activities will be designed to encourage students to play an active role in the construction of their own knowledge and in the design of their own learning strategies. We will combine traditional lectures with other

active teaching methodologies using digital platforms, such as analysis of video scenes and debates. Students will be encouraged to actively take part in all group activities and to give an oral group

presentation. Students will be expected to interact with media resources, such as, web sites, videos, DVDs, and newspapers etc.

**CourseContent:**

**(60hours)**

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**UNITI**

**(14hours)**

*Algae:* Introduction; habitat and range of thallus structure in algae; Principles of classification and outline classification of Lee (2009) up to divisions, Diagnostic characters of important algal family's examples, Algal genetics, Economic importance of algae, Bioreactor. cultivation of algae and its prospects.

**UNITII**

**(12hours)**

*Fungi:* Introduction; habitat, Basic classifications, Diagnostic characters of important families with examples, Fungal genetics, Economic importance of fungi, Pathogenic and poisonous fungi.

**UNITIII**

**(5hours)**

*Lichen:* Habitat and thallus structures; economic importance.

**UNITIV****(10hours)**

*Plant Pathology:* Plant Diseases: Introduction and Definition; concepts of parasitism and saprophytism, Koch's postulates, Classification of plant diseases based on symptoms; Factors influencing infection, colonization and development of symptoms, Genetic basis of disease resistance and pathogenicity: gene for gene hypothesis; breeding for disease resistance, Brief idea about symptoms; disease cycles and control measures of: Loose smut of wheat, Citrus canker, Late blight of potato, Rust of wheat, Brown spot of Rice & Alternaria blight of Brassica.

**UNITV****(15hours)**

*Plant physiology:* Transport in plants water and mineral uptake, Transpiration- Mechanism of stomatal movement, significance; Photosynthesis- types of photosystem, significance, cycles; Plant Growth regulators.

**UNITVI****(4hours)**

*Economic Botany:* Introduction, Method of cultivation, processing and utilities of the products of the following: Rice, Tea, Jute and Brassica.

**Text Books**

1. Phycology by Robert Edward Lee
2. Introduction to Fungi by John Webster
3. Plant Pathology by G.N. Agrios
4. Plant Physiology by Lincoln Taiz, Eduardo Zeiger

**Reference Books**

1. College Botany Vol. II By Gangulee and Kar
2. Studies in Botany Vol I & II by J.N. Mitra, D. Mitra, S.K. Chaudhuri

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40



### Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and POs		
	Course Outcomes (COs)	Mapped Program Outcomes
<b>CO1</b>	<b>develop</b> basic knowledge in plant science and also be able to explain various aspects of plant growth and development of lower group of plants	<b>PO1, PO4, PO8</b>
<b>CO2</b>	<b>explain</b> the categories in detail various divisions of Fungi and their commercial importance as well harmful effects.	<b>PO1, PO2, PO4</b>
<b>CO3</b>	<b>demonstrate</b> basic knowledge and economic importance of Lichen	<b>PO1, PO2, PO3, PO7</b>
<b>CO4</b>	<b>illustrate</b> basic knowledge in plant pathology and also be able to explain various genetic aspect of plant disease as well their impact plant diseases and their controlling measures.	<b>PO1, PO3, PO5, PO9</b>
<b>CO5</b>	<b>summarize</b> in plant physiology and also be able to explain various physiological reactions as well their impact on plant growth and development	<b>PO1, PO2, PO4, PO5</b>
<b>CO6</b>	<b>develop</b> fundamental knowledge and can be implemented this skill for cultivation practices and commercial uses of economic important plant	<b>PO1, PO2, PO6, PO9</b>


		Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking	Problem Identification Ability	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to Environment	Development to Society.	Development to Humanity
Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
<b>BOT11001</b>	ELECTIVE BOTANY - I (THEORY)	3	3	1	2	2	1	1	-	1	-	-	-

1=weakly mapped

2= moderately mapped

3=strongly mapped

### Model Question Paper

<b>Name:</b>  <b>Enrolment No:</b>	 <b>ADAMAS</b> UNIVERSITY <small>PURSUE EXCELLENCE</small>		
<b>Course: BOT11001-ELECTIVE BOTANY - I(THEORY)</b> <b>Program: B.Sc. Biochemistry</b> <b>Semester: Odd 2019-20</b>			
<b>Time: 03Hrs.</b> <b>Max. Marks: 40</b>			
<b>Instructions:</b> Attempt any <b>four</b> questions from <b>Section A</b> (each carrying 5 marks); any <b>two</b> questions from <b>Section B</b> (each carrying 10 marks).			
<b>SECTION A (Attempt any Four questions)</b>			
1.	Member of Cyanophyceae algae shows much resemblances with photosynthesizing bacteria (cyanobacteria)- justify under the light of evolution. (U)	5	CO4
2.	Differentiate Heterothallism and Homothallism in Fungi. (U)	5	CO2
3.	Write a short note on economic importance of Lichen. (R)	5	CO3
4.	State the most important reactor use for commercial algal production. (R)	5	CO1
5	Mention the key differences between Transpiration and Evaporation. (R)	5	CO5
<b>SECTION B (Attempt any Two questions)</b>			
6.	Differentiate plant and animal growth regulators. Mention the role of Ethylene on fruit ripening. Why ABA called as stress hormone? (U)	3+5+2	CO5
7.	Explain in brief oil extraction process from Brassica. Mention the cultivation practices of summer rice in India. (R)	4+6	CO6
8.	Elaborately discuss the prospect of using algae in nutraceutical industry. (Ap)	10	CO1
9	(a) Define Koch's postulates. (b) Briefly explain the disease cycles and control measures of Loose smut of wheat. (An)	3+4+3	CO4

<b>BOT12002</b>	Lab-Elective Botany I (Practical)	<b>L</b>	<b>T</b>	<b>P</b>	<b>C</b>
<b>Version 1.0</b>	<b>Contact Hours - 45</b>	0	0	3	2
<b>Pre-requisites/Exposure</b>	PLUS TWO LEVEL BIOLOGY				
<b>Co-requisites</b>	Theory of Elective Botany I				

### Course Objectives:

Upon completion of this course, students should be able to:

1. Demonstrate a basic understanding of developmental terms and mechanisms of different plants.
2. Utilize laboratory techniques to design and carry-out experimental studies related to botany.
3. Convey and discuss experimental results via written assignments.

### Course Outcome

CO1: **Develop** the the skill of handling of laboratory instruments –Microscope, Autoclave, Incubator, centrifuge, Analytical balance, pH Meter, Colorimeter, Water bath, Distillation plant, Laminar Air Flow operation etc.

CO2: **compare** between monocot and dicot plants.

CO3: **Illustrate** plant pigments.

CO4: **Demonstrate** amount of water absorption, retention and transpiration.

CO5: **Explain** percent, normal, molal and molar solutions of any compound.

### Catalogue Description:

Botany is a rapidly expanding field of biology integrating concepts from plant science, genetics, molecular biology, cell biology, physiology, ecology and evolution. This course investigates the cellular and molecular processes involved in regulating plant development. Topics to be covered include experimental approaches to understand the principle of laboratory instruments involved in botanical studies with their mechanism, study the anatomy and developmental stages of different plants with a comparative understanding. Laboratory work will include a demonstration based learning of plant water relationship and phenomenon like guttation and also the separation of photosynthetic pigments by paper chromatography.

**Course Content:****Paper name:** Lab-Elective Botany I**Paper code:** BOT12002

1.	Acquaintance with laboratory instruments –Microscope, Autoclave, Incubator, centrifuge, Analytical balance, pH Meter, Colorimeter, Water bath, Distillation plant, Laminar Air Flow operation etc.	<b>8 hours</b>
2.	Study of the following genera and their identification: <i>Oscillatoria</i> , <i>Scytonema</i> , <i>Oedogonium</i> , <i>Chara</i> , <i>Ectocarpus</i> , <i>Polysiphonia</i> . (vegetative and reproductive structures).	<b>8 hours</b>
3.	Study of the following genera and their identification: <i>Rhizopus</i> , <i>Penicillium</i> , <i>Ascobolus</i> , <i>Agaricus</i> , <i>Polyporus</i> .	<b>8 hours</b>
4.	Identification of specimens with diseases prescribed in the theoretical syllabus: Loose smut of wheat, Citrus canker, Late blight of potato, Rust of wheat & Brown spot of Rice.	<b>7 hours</b>
5.	Chemical separation of photosynthetic pigments by paper chromatography.	<b>4 hours</b>
6.	Preparation of percent, normal, molal and molar solutions of any compound.	<b>4 hours</b>
7.	Comparison of imbibitions of starchy, proteinaceous and fatty seeds.	<b>4 hours</b>
8.	Determination of amount of water absorption, retention and transpiration.	<b>4 hours</b>
9.	Demonstration of Guttation.	<b>2 hours</b>
10.	Study of Monocot root.	<b>4 hours</b>
11.	Study of Dicot root.	<b>4 hours</b>
12.	Study of leaves anatomy of different plants.	<b>3 hours</b>

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination  
Examination Scheme:**

<b>Components</b>	<b>Mid Term</b>	<b>Attendance</b>	<b>Class Assessment</b>	<b>End Term</b>
<b>Weightage (%)</b>	<b>20</b>	<b>10</b>	<b>30</b>	<b>40</b>

**Relationship between the Course Outcomes (COs) and Program Outcomes (POs)**

<b>Mapping between COs and Pos</b>		
	<b>Course Outcomes (COs)</b>	<b>Mapped Program Outcomes</b>
<b>CO1</b>	<b>explain</b> the handling of laboratory instruments – Microscope, Autoclave, Incubator, centrifuge, Analytical balance, pH Meter, Colorimeter, Water bath, Distillation plant, Laminar Air Flow operation etc.	<b>PO1, PO2, PO3</b>
<b>CO2</b>	<b>compare</b> between monocot and dicot plants.	<b>PO1, PO2, PO4</b>
<b>CO3</b>	<b>Illustrate</b> plant pigments.	<b>PO1, PO2, PO3, PO4</b>
<b>CO4</b>	<b>Demonstrate</b> amount of water absorption, retention and transpiration	<b>PO1, PO3, PO6, PO9</b>
<b>CO5</b>	<b>explain</b> percent, normal, molal and molar solutions of any compound.	<b>PO1, PO3, PO4, PO8,</b>

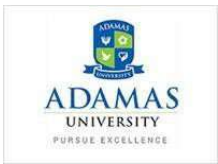
Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
		Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking	Problem Identification Ability	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to Environment	Development to Society.	Development to Humanity
<b>BOT12002</b>	Lab- Elective Botany 1	3	2	3	2	-	1	-	1	1	-	-	-

1=weakly mapped

2= moderately mapped

3=strongly mapped

## Model Question Paper

<b>Name:</b>  <b>Enrolment No:</b>			
<b>Course: BOT12002 I LAB Elective Botany I (PRACTICAL)</b> <b>Program: B.Sc. Biochemistry</b> <span style="float: right;"><b>Time: 03Hrs.</b></span> <b>Semester: Even 2020-21</b> <span style="float: right;"><b>Max. Marks: 40</b></span>			
<b>Instructions:</b>  Attempt any <b>two</b> questions from <b>Section A</b> (each carrying 10 marks); <b>Section B</b> is Compulsory (carrying 10 marks).			
<b>Section A ( Attemptany Two)</b>			
1.	a) <b>Identify</b> specimen A under microscope and write your conclusion. (Ap)  b) <b>Classify</b> simple tissue in plants. (An)	6  4	<b>CO2</b>
2.	a) <b>Identify</b> specimen B under microscope and write your conclusion. (Ap)  b) <b>Classify</b> different type of plants. (An)	6  4	<b>CO1</b>
3.	a) <b>Identify</b> specimen C. (Ap)  b) <b>Write</b> about the guttation. (U)	5  5	<b>CO4</b>
4.	a) <b>Identify</b> the model of dissection of cockroach D. (Ap)  b) <b>Draw</b> the monocot root. (Ap)	6  4	<b>CO3</b>  <b>CO5</b>
<b>SECTION B is compulsory</b>			
5.	<b>Viva-voce</b> (U/An/Ap/R/Ev)	<b>10</b>	<b>CO1, CO2, CO3, CO4, CO5</b>
6.	<b>Practical copy</b> (U/Ap/Ev)	<b>10</b>	<b>CO1, CO2, CO3</b>

<b>ZOL11001</b>	<b>ELECTIVE ZOOLOGY I (THEORY)</b>	<b>L</b>	<b>T</b>	<b>P</b>	<b>C</b>
<b>Version 1.0</b>	<b>Contact Hours - 60</b>	3	1	0	4
<b>Pre-requisites/Exposure</b>	PLUS TWO LEVEL BIOLOGY				
<b>Co-requisites</b>	--				

### **Course Objectives**

1. To provide those students with some biology background with an introduction to zoology and the study of animals. This course is designed for students of any major, but will especially benefit biology majors, as well as secondary science education majors.
2. It will also provide an informative elective for 5-8 math/science education majors.
3. Gathering information about other organisms' structure and function, and how it compares to human beings, enables us to live a more knowledgeable, involved, and environmentally aware life in a science-conscious age.

### **Course Outcomes**

On completion of this course, the students will be able to

- CO1. **Comprehend** the fundamentals of animal sciences, which helps them to understand the complex relations among various living organisms.
- CO2. **Illustrate** and analyze the course of evolution: i.e. how complex multicellular organisms develop from unicellular cells and correlate with other fields of biology.
- CO3. **Demonstrate** and compare the basis of life processes in the non-chordates and chordates which helps them to identify the economically important organisms.
- CO4. **Compare** between acoelomate, pseudo-coelomate and coelomate.
- CO5. **Develop** as lifelong learner about vertebrate and invertebrate organisms which contribute in greater benefit of humanity.

### **Catalogue Description**

Elective zoology I course will help to understand the behavior structure and evolution of animals. This course includes diverse approaches by studying animals and develop a better understanding of how we, ourselves, function and interact with the world around us. All the lectures will be devoted on discussions of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as power



point presentation, audio visual virtual lab session as per requirement. The tutorials will familiarize the students with practical problem-solving techniques led by the course coordinator. Students will strongly grab the basic concepts of the subject via exercise and discussions with the coordinator.

<b>Course</b>	<b>Content</b>
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**ELECTIVE ZOOLOGY I**

**GE1: DIVERSITY OF LIFE FORMS (THEORY)(SEMESTER I)**

**Unit1: Protista** **3 LectureHours**

General characters of Protozoa; Life cycle of *Plasmodium*; Conjugation in *Paramoecium*.

**Unit2: Porifera** **3 LectureHours**

General characters and canal system in Porifera.

**Unit3: Radiata** **3 LectureHours**

General characters of Cnidarians and polymorphism.

**Unit4: Acoelomates** **3 LectureHours**

General characters of Helminthes; Life cycle of *Taeniasolium*.

**Unit 5: Pseudocoelomates** **3 LectureHours**

General characters of Nemathehelminthes; Parasitic adaptations.

**Unit 6: Coelomate Protostomes** **4 LectureHours**

General characters of Annelida; Metamerism.

**Unit7: Arthropoda** **4 LectureHours**

General characters; Social life in insects; Communication in Honey Bees.

**Unit8: Mollusca** **4 Lecture Hours**

General characters of mollusca; Torsion in gastropoda.

**Unit 9: Coelomate Deuterostome** **4 LectureHours**

General characters of Echinodermata; Water Vascular system in Starfish.

**Unit10. Protochordata** **4 LectureHours**

Salient features

**Unit11.Pisces** **5 LectureHours**

Outline of classification; Parental care in Fish.

**Unit12.Amphibia** **5 Lecture Hours**

General characters; Outline ofclassification;Paedogenesis.

**Unit13.Reptilia** **5 Lecture Hours**

Amniotes; Origin of reptiles; Outline of classificationin reptiles.

**Unit14.Aves** **5 LectureHours**

General characters; Flight adaptations

**Unit15.Mammalia** **5 LectureHours**

Outline of classification; Dentition in mammals.

**Reference Books**

1. Barnes, R.D. (1992). Invertebrate Zoology. Saunders College Pub.USA.
2. Campbell & Reece (2005). Biology, Pearson Education, (Singapore) Pvt.Ltd.
3. Kardong, K. V. (2002). Vertebrates Comparative Anatomy. Function and Evolution. Tata McGraw Hill Publishing Company. NewDelhi.
4. Ruppert, Fox and Barnes (2006) Invertebrate Zoology. A functional Evolutionary Approach 7th Edition, ThomsonBooks/Cole
5. Raven, P. H. and Johnson, G. B. (2004). Biology, 6th edition, Tata McGraw Hill Publications. NewDelhi

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

<b>Components</b>	<b>Mid Term</b>	<b>Attendance</b>	<b>Class Assessment</b>	<b>End Term</b>
<b>Weightage (%)</b>	<b>20</b>	<b>10</b>	<b>30</b>	<b>40</b>

### Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and POs		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	<b>comprehend</b> the fundamentals of animal sciences, which helps them to understand the complex relations among various living organisms.	PO1, PO2, PO6
CO2	<b>illustrate</b> the course of evolution: i.e. how complex multicellular organisms develop from unicellular cells and correlate with other fields of biology.	PO1, PO2, PO6, PO9
CO3	<b>demonstrate</b> the basis of life processes in the non-chordates and chordates which helps them to identify the economically important organisms.	PO1, PO2, PO5, PO8
CO4	<b>compare</b> between acoelomate, pseudocoelomate and coelomate.	PO1, PO2, PO3
CO5	<b>develop</b> as life-long learner about vertebrate and invertebrate organisms which contribute in greater benefit of humanity.	PO1, PO11, PO12

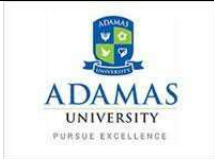
		Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking	Problem Identification Ability	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to Environment	Development to Society.	Development to Humanity
Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
ZOL11001	ELECTIVE ZOOLOGY I (THEORY)	3	3	1	-	1	2	-	1	1	-	1	1

1=weakly mapped

2= moderately mapped

3=strongly mapped

## Model Question Paper

<b>Name:</b>  <b>Enrolment No:</b>	 <b>ADAMAS</b> UNIVERSITY <small>PURSUE EXCELLENCE</small>		
<b>Course: ZOL11001 - ELECTIVE ZOOLOGY I (THEORY)</b> <b>Program: B.Sc.Biochemistry</b> <b>Semester: Odd2020-21</b>			
<b>Time: 03Hrs.</b> <b>Max. Marks:40</b>			
<b>Instructions:</b> Attempt any <b>three</b> questions from <b>Section A</b> (each carrying 4 marks); any <b>four</b> questions from <b>Section B</b> (each carrying 7 marks).			
<b>SECTION A (Attempt any Three questions)</b>			
1.	<b>Describe</b> the social structure of a termite colony.	<b>4</b>	<b>CO5</b>
2.	<b>What</b> are the adaptations acquired by the round worms to maintain their parasitic nature in their habitat inside human host? (R)	<b>4</b>	<b>CO2</b>
3.	Maternal care is more often provided in mammals' – <b>Why?</b> (An)	<b>4</b>	<b>CO4</b>
4.	Will it be possible to propose that annelids evolve from <i>platyhelminthes</i> according to con or fission theory? <b>Give justification</b> against your answer. (An)	<b>1</b> <b>3</b>	<b>CO3</b>
<b>SECTION B (Attempt any Four questions)</b>			
5.	a) <b>How</b> do Gastropods lose their symmetry? <b>Explain</b> the process with diagram. (Ap) b) <b>What</b> do you understand by chiasmata? (U)	<b>2</b> <b>3</b> <b>2</b>	<b>CO2</b>
6.	a) <b>What</b> is the difference between monogenetic and digenetic life cycle? (U) b) <b>Discuss</b> the process of sexual reproduction of malarial parasite. (U)	<b>3</b> <b>4</b>	<b>CO1</b> <b>CO2</b> <b>CO4</b>
7.	a) <b>Discuss</b> the process of sexual reproduction in Paramecium with proper diagram. (Cr) b) <b>Why</b> pinacocyte and choanocyte cells are there in the canal system of <i>Scypha</i> ? (An)	<b>5</b> <b>2</b>	<b>CO1</b> <b>CO2</b>
8.	a) 'Parental care is defined as "any form of parental behavior that appears likely to increase the fitness of the parent's offspring"' – <b>Explain</b> this phrase in your own words. (Ev) b) <b>What</b> do you mean by parental care? Give example. (U)	<b>4</b> <b>2</b> <b>1</b>	<b>CO5</b> <b>CO2</b>
9.	a) <b>List</b> anatomical modification in bird's that enhances flight. (R) b) <b>Write</b> down two characteristic features of order chelononia. <b>What</b> is notochord? <b>Name</b> one agnathan chordate. (U)	<b>3</b> <b>2</b> <b>1</b> <b>1</b>	<b>CO4</b>

ZOL12002	ELECTIVE ZOOLOGY I LAB (PRACTICAL)	L	T	P	C
Version 1.0	Contact Hours - 45	0	0	3	2
Pre-requisites/Exposure	PLUS TWO LEVEL BIOLOGY				
Co-requisites	Elective Zoology I				

### Course Objectives

1. To provide students with hands-on activities designed to encourage interest in the field of Zoology, as well as promote greater understanding of the concepts presented in lecture.
2. Students will need to become proficient with terms, techniques, and applications.

### Course Outcomes

On completion of this course, the students will be able to

- CO1. **Identify**, classify and compare between non chordate and chordate specimens.
- CO2. **Explain** the ecological importance of these both chordate and non chordate specimens.
- CO3. **Compare** between the digestive, reproductive and nervous system of cockroach.
- CO4. **Demonstrate** the course of evolution: i.e. how complex multicellular organisms develop from unicellular cells and correlate with other fields of biology.
- CO5. **Illustrate** the knowledge of using different modern tools and techniques in the field of biology which will help in their further academics.

### Catalogue Description

Elective zoology I Practical (LAB) is the overall about diversity of life forms which include identification, classification and ecological importance of unicellular organisms to multicellular organisms, non chordate to chordate, in the course of evolution. This course covers laboratory techniques and study of animal phyla. The very nature of zoology lab requires students to view animals in different settings and to identify and describe many of the animal phyla. All the lectures will be devoted on discussions of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as power point presentation, audio visual virtual lab session as per requirement. The tutorials will familiarize the students with practical problem-solving techniques led by the course coordinator. Students will strongly grab the basic concepts of the subject via exercise and discussions with the coordinator.

## Course Content

### ELECTIVE ZOOLOGY LAB I

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#### DIVERSITY OF LIFE FORMS (PRACTICAL)

1. Identification and Classification of Any these of the following:

##### 15 Lecture Hours

- a. **Non-chordate specimens:** *Scypha*, *Obelia*, *Sea-anaemone*, *Ascaris*, *Hirudinaria*, *Scorpion*, *Bombyxmori*, *Achatina*, *Loligo*, *Starfish*, *Balanoglossus*.

##### 15 Lecture Hours

- b. **Chordate specimens:** *Branchiostoma*, *Petromyzon*, *Scolidon*, *Lates*, *Axolotllarva*, *Tylostotriton*, *Gekko*; *Hemidactylus*, *Turtle*, *Naja*, *Chiroptera*.

##### 15 Lecture Hours

2. Ecological Note – On any of the specimens in Exercise No1.

##### 15 Lecture Hours

3. Models of dissection of Cockroach - Cockroach: Digestive, Reproductive, Nervous System.

#### Reference Books

1. Kardong, K. V. (2002). *Vertebrates Comparative Anatomy. Function and Evolution*. Tata McGraw Hill Publishing Company. NewDelhi.
2. Ruppert, Fox and Barnes (2006) *Invertebrate Zoology. A functional Evolutionary Approach* 7th Edition, ThomsonBooks/Cole
3. Raven, P. H. and Johnson, G. B. (2004). *Biology*, 6th edition, Tata McGraw Hill Publications. NewDelhi

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

### Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and Pos		
	Course Outcomes (COs)	Mapped Program Outcomes
<b>CO1</b>	<b>Identify</b> , classify and compare between non chordate and chordate specimens.	<b>PO1, PO2, PO3</b>
<b>CO2</b>	<b>explain</b> the ecological importance of these both chordate and non chordate specimens.	<b>PO1, PO3</b>
<b>CO3</b>	<b>compare</b> between the digestive, reproductive and nervous system of cockroach.	<b>PO1, PO2, PO3, PO4, PO8</b>
<b>CO4</b>	<b>demonstrate</b> the course of evolution: i.e. how complex multicellular organisms develop from unicellular cells and correlate with other fields of biology.	<b>PO1, PO3, PO6, PO9</b>
<b>CO5</b>	<b>develop</b> the knowledge of using different modern tools and techniques in the field of biology which will help in their further academics.	<b>PO1, PO4, PO8</b>

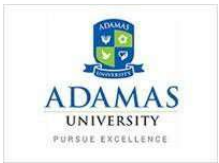
		Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking	Problem Identification Ability	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to Environment	Development to Society.	Development to Humanity
Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
<b>ZOL12002</b>	ELECTIVE ZOOLOGY LAB I (PRACTICAL)	3	2	3	2	-	1	-	1	2	-	-	-

1=weakly mapped

2= moderately mapped

3=strongly mapped

## Model Question Paper

<b>Name:</b>  <b>Enrolment No:</b>			
<b>Course: ZOL12002 – ELECTIVE ZOOLOGY I LAB (PRACTICAL)</b> <b>Program: B.Sc. Biochemistry</b> <span style="float: right;"><b>Time: 03Hrs.</b></span> <b>Semester: Even 2020-21</b> <span style="float: right;"><b>Max. Marks: 40</b></span>			
<b>Instructions:</b> Attempt any <b>two</b> questions from <b>Section A</b> (each carrying 10 marks); <b>Section B</b> is Compulsory (carrying 10 marks).			
<b>Section A ( Attemptany Two)</b>			
1.	a) <b>Identify</b> specimen A.(Ap) b) <b>Classify</b> its taxonomical position and <b>write</b> about its identifying characters. (An)	2 4 4	CO1 CO4
2.	a) <b>Identify</b> specimen B.(Ap) b) <b>Classify</b> its taxonomical position and <b>write</b> about its identifying characters. (An)	2 4 4	CO1 CO4
3.	a) <b>Identify</b> specimen C.(Ap) b) <b>Write</b> about its ecological importance.(U)	5 5	CO2
4.	a) <b>Identify</b> the model of dissection of cockroach D.(Ap) b) <b>Draw</b> the model and <b>write</b> about the functions of the endocrine gland in this model. (Ap)	2 4 4	CO3 CO5
<b>SECTION B is compulsory</b>			
5.	<b>Viva-voce</b> (U/An/Ap/R/Ev)	10	CO1 CO2 CO3 CO4 CO5
6.	<b>Practical copy</b> (U/Ap/Ev)	10	CO1 CO2 CO3



<b>BIC11005</b>	<b>Enzymes(Theory)</b>	L	T	P	C
<b>Version 1.0</b>	<b>Contact Hours: 60</b>	3	1	0	4
<b>Pre-requisites/Exposure</b>	Class 12 level Biology knowledge				
<b>Co-requisites</b>	Amino acid: Structure & Function				

### Course Objectives

1. To understand the enzymes according to the basis of their catalyzed reactions.
2. To gain knowledge about the kinetic behaviour of enzymes.
3. To generate the concept and determine about different patterns of inhibitions of enzyme activity.
4. To understand the concept about the structures of active site of the enzymes and their mechanism of actions and their clinical application.
5. To acquire the idea about regulation of enzyme activity.

### Course Outcomes

On completion of this course, the students will be able to

CO1: **classify** the enzymes according to the basis of their catalysed reactions.

CO2: **evaluate** the kinetic behaviour of enzymes.

CO3: **develop** the concept and determine about different patterns of inhibitions of enzyme activity.

CO4: **illustrate** the concept about the structures of active site of the enzymes and their mechanism of actions and their clinical application.

CO5: **develop** the idea about regulation of enzyme activity.

### Catalogue Description

Nomenclature and classification of enzymes Holoenzyme, apoenzyme, cofactors, coenzyme, prosthetic groups, metallo enzymes, monomeric and oligomeric enzymes Activation energy and transition state theory, enzyme activity, specific activity, common features of active sites, enzyme specificity: types and theories Factors affecting enzyme activity, E, S, temp and pH Enzyme substrate complex: Concept of E-S complex, binding sites, active site, specificity, kinetics of enzyme activity Michaelis-Menten equation and its derivation Different plots for the determination of  $K_M$  and  $V_{max}$  and their physiological significance Two substrate reactions (random, ordered and ping pong mechanisms), enzyme inhibition, types of inhibition, determination of  $K_i$ , suicide inhibitor.

## **Course Content**

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### **Unit I Introduction to enzymes and Features of enzyme catalysis [10 lecture hours]**

Nature of enzymes - protein and non-protein (ribozyme). Cofactor and prosthetic group, apoenzyme, holoenzyme. IUBMB classification of enzymes. Factors affecting the rate of chemical reactions, collision theory, activation energy and transition state theory, catalysis, reaction rates and thermodynamics of reaction. Catalytic power and specificity of enzymes (concept of active site), Fischer's lock and key hypothesis, Koshland's induced fit hypothesis.

### **Unit II Enzyme kinetics [10 lecture hours]**

Relationship between initial velocity and substrate concentration, steady state kinetics, equilibrium constant - monosubstrate reactions. Michaelis-Menten equation, Lineweaver-Burk plot, Eadie-Hofstee and Hanes plot.  $K_m$  and  $V_{max}$ ,  $K_{cat}$  and turnover number. Effect of pH, temperature and metal ions on the activity of enzyme.

### **Unit III Bisubstrate reactions [10 lecture hours]**

Types of bi-reactions (sequential-ordered and random, ping-pong reactions). Differentiating bi-substrate mechanisms (diagnostic plots, isotope exchange).

### **Unit IV Enzyme inhibition [10 lecture hours]**

Reversible inhibition (competitive, uncompetitive, non-competitive, mixed and substrate). Mechanism based inhibitors - antibiotics as inhibitors.

### **Unit V Mechanism of action of enzymes [10 lecture hours]**

General features - proximity and orientation, strain and distortion, acid-base and covalent catalysis (chymotrypsin, lysozyme). Metal activated enzymes and metalloenzymes, transition state analogues.

### **Unit V Regulation of enzyme activity [10 lecture hours]**

Control of activities of single enzymes (end product inhibition) and metabolic pathways, feedback inhibition (aspartate transcarbamoylase), reversible covalent modification phosphorylation (glycogen phosphorylase). Proteolytic cleavage - zymogen. Multienzyme complex as regulatory enzymes. Occurrence and isolation, phylogenetic distribution and properties (pyruvate dehydrogenase, fatty acyl synthase) Isoenzymes - properties and physiological significance (lactate dehydrogenase).

## Unit VI Involvement of coenzymes in enzyme catalysed reactions and Applications of enzymes. [10 lecture hours]

TPP, FAD, NAD, pyridoxal phosphate, biotin, coenzyme A, tetrahydrofolate, lipoic acid.

Application of enzymes in diagnostics (SGPT, SGOT, creatine kinase, alkaline and acid phosphatases), enzyme immunoassay (HRPO), enzyme therapy (Streptokinase). Immobilized enzymes.

### Reference Books

1. Lehninger: Principles of Biochemistry (2013) 6th ed., Nelson, D.L. and Cox, M.M., W.H. Freeman and Company (New York), ISBN:13: 978-1-4641-0962-1 / ISBN:10:1-42923414-8.
2. Textbook of Biochemistry with Clinical Correlations (2011) 7th ed., Devlin, T.M., Jhn Wiley & Sons, Inc. (New York), ISBN:978-0-470-28173-4.

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

### Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and Pos		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	<b>Classify</b> the enzymes according to the basis of their catalysed reactions.	PO1, PO2, PO6
CO2	<b>evaluate</b> the kinetic behaviour of enzymes	PO1, PO2, PO3, PO6
CO3	<b>Develop</b> the concept and <b>determine</b> about different patterns of inhibitions of enzyme activity.	PO3, PO5, PO7, PO8
CO4	<b>illustrate</b> the concept about the structures of active site of the enzymes and their mechanism of actions and their clinical application	PO1, PO3, PO5, PO9
CO5	<b>Develop</b> the idea about regulation of enzyme activity	PO1, PO7, PO8, PO10

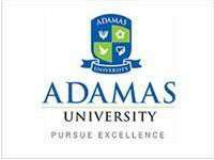
Course Code	Course Title	PO 1	PO 2	PO 3	PO 4	PO 5	PO 6	PO 7	PO 8	PO 9	PO10	PO11	PO12
<b>BIC11005</b>	Enzymes	3	1	3	-	2	2	2	1	-	-	1	-
		Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking Ability	Problem Identification	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to Environment	Development to Society	Development to Humanity

1=weakly mapped

2= moderately mapped

3=strongly mapped

## Model Question Paper

<b>Name:</b> <b>Enrolment No:</b>			
<b>Course: BIC11005 Enzymes</b> <b>Program: B.Sc. Biochemistry</b> <b>Semester: Even 2019-20</b>		<b>Time: 03Hrs.</b> <b>Max. Marks: 40</b>	
<b>Instructions:</b> <b>Attempt any three questions from Section A (each carrying 4 marks); any four Questions from Section B (each carrying 7 marks).</b>			
<b>Section A ( Attempt any Three)</b>			
1	What is active site of enzyme? Give its function. (R)	4	CO2
2	What is allosteric regulation? (U)	4	CO3
3	What are the difference between sequential and symmetry model of allosteric regulation. (R)	4	CO1
4	What are the advantages of allosteric regulation? (An)	4	CO3
5	Why an enzyme having an allosteric regulation show a sigmoid curve instead of a regular hyperbolic curve? (Ap)	7	CO4
6	Which of these two cases is allosteric regulation? (An) i. 'Phosphorylation of an amino acid somewhere other than the active site' ii. 'The non-covalent binding of cAMP somewhere other than the activesite'	7	CO3
7	Differentiate between apoenzyme and holoenzyme. What is induced fit Model? Give its significance. (U)	4+3	CO5
8	Prove $K_m = (S) \text{ at } \frac{1}{2} V_{max}$ . (An)	7	CO3
9	What is activation energy? How is it lowered? Explain the limitation of key and lock model. (R)	3+4	CO2

<b>BIC12006</b>	<b>Enzymes Lab (Practical)</b>	L	T	P	C
<b>Version 1.0</b>	<b>Contact Hours: 45</b>	0	0	3	2
<b>Pre-requisites/Exposure</b>	Class 12 level Biology knowledge				
<b>Co-requisites</b>	Enzyme				

### Course Objectives

1. Conceptualizing the activity and specific activity of an enzyme.
2. Evaluation of the  $V_{max}$  and  $K_m$  value of an enzyme.
3. Determination the pH optima and temperature optima of an enzyme.
4. Analysis of the  $K_i$  value of an inhibitor of enzyme.
5. Determination of the activity of an enzyme through coupled assay.

### Course Outcomes

On completion of this course, the students will be able to

CO1: **Determine** the activity and specific activity of an enzyme.

CO2: **Assess** the  $V_{max}$  and  $K_m$  value of an enzyme.

CO3: **Evaluate** the pH optima and temperature optima of an enzyme.

CO4: **Determine** the  $K_i$  value of an inhibitor of enzyme.

CO5: **Illustrate** the activity of an enzyme through coupled assay.

### Catalogue Description

To provide the students with detailed knowledge in enzyme activity and kinetics, their mechanism of action and regulation and about the way of enzyme application and exploitation. The course include deepening knowledge in the areas of purification and isolation of enzymes, classification of enzymes and cofactors, kinetics and regulation of enzymes and their applications in industry, therapeutics and diagnosis.

## Course Content

1. Determination of Enzyme activity of ALP. [8 lecture hours]
2. Determination of Enzyme activity of ALP at various Temperatures. [8 lecturehours]
3. Determination of Enzyme activity of ALP at various pH. [8 lecturehours]
4. Determination of Enzyme activity of ALP in animal sample lab 1.[10 lecture hours]
5. Determination of Enzyme activity of ALP at animal sample lab 2. [10 lecture hours]
6. Assay of Salivary Amylase at different pH and temperature. [6 lecturehours]
7. Assay of Plant Amylase. [10 lecture hours]

## Reference Books

1. Lehninger: Principles of Biochemistry (2013) 6th ed., Nelson, D.L. and Cox, M.M., W.H.Freeman and Company (New York), ISBN:13: 978-1-4641-0962-1 / ISBN:10:1-42923414-8.
2. Textbook of Biochemistry with Clinical Correlations (2011) 7th ed., Devlin, T.M., John Wiley & Sons, Inc. (New York), ISBN:978-0-470-28173-4.

## Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

## Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and Pos		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	<b>Discuss</b> the Elements, Tools & Methods of Construction Management	PO1, PO2, PO9
CO2	<b>Illustrate</b> the Fundamentals of Network Analysis to Schedule a Project	PO1, PO3, PO4, PO7
CO3	<b>Demonstrate</b> Schedule for Time and Cost of a Construction Project	PO4, PO5, PO6, PO7
CO4	<b>Explain</b> the Type and Capacity of Construction Equipment Required for the Project Site	PO5, PO7, PO8, PO9
CO5	<b>Illustrate</b> Organizational Structure and Safety Procedures to the Project Site	PO1, PO3, PO4, PO7, PO9

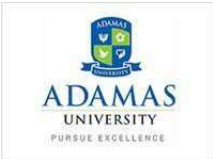
		Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking Ability	Problem Identification	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to Environment	Development to Society	Development to Humanity
Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
<b>BIC12006</b>	Enzymes Lab	3	1	3	3	2	1	3	1	3	-	-	-

1=weakly mapped

2= moderately mapped

3=strongly mapped

### Model Question Paper

<b>Name:</b>			
<b>Enrolment No:</b>			
<b>Course: BIC12006 Enzymes Lab</b>			
<b>Program: B.SC Biochemistry</b>	<b>Time: 03Hrs.</b>		
<b>Semester: Even 2019-20</b>	<b>Max. Marks: 40</b>		
1	Measure the activity of alkaline phosphatase. (Ap)	<b>10</b>	<b>CO2</b>
2	Calculate the amount of enzyme from the data collected from the experiment. (U)	<b>10</b>	<b>CO1</b>
3	Note book	<b>10</b>	<b>CO1, CO2, CO3</b>
4	Viva Voce	<b>10</b>	<b>CO1, CO2, CO3</b>



<b>BIC11007</b>	<b>Human Physiology (Theory)</b>	L	T	P	C
<b>Version 1.0</b>	<b>Contact Hrs: 60</b>	3	1	0	4
<b>Pre-requisites/Exposure</b>	Knowledge of Biology at 10+2 level				
<b>Co-requisites</b>	-				

### Course Objectives

1. To gain a deeper understanding about the organization of buffering system of body fluid.
2. To acquire the knowledge anatomical and physiological organization of respiratory and circulatory system of human body.
3. To acquire the knowledge about the biochemical basis of food digestion.
4. To gain the knowledge about biochemical basis of kidney function.
5. To understand the biochemical properties of the excitable tissue.

### Course Outcomes

On completion of this course, the students will be able to

- CO1. **Develop** the concept about homeostasis and organization of body fluid compartments.  
CO2. **Illustrate** their knowledge biochemistry in the respiratory and cardiovascular physiology.  
CO3. **Explain** their knowledge biochemistry in the digestive and renal physiology.  
CO4. **Demonstrate** the behaviour of excitable tissue like nerve and muscle.  
CO5. **Develop** their concept about determination of sex and sexual differentiation

### Catalogue Description

Physiology is the scientific study of functions and mechanisms in a living system. As a sub-discipline of biology, physiology focuses on how organisms, organ systems, individual organs, cells, and biomolecules carry out the chemical and physical functions in a living system. Central to physiological functioning are biophysical and biochemical processes, homeostatic control mechanisms, and communication between cells. Physiological state is the condition of normal function of human body and other living system. These normal functions are response of some specific biochemical consequences of the living system. So how to biochemical reason is translated into physiological response, that is the goal of this paper.

## Course Content

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- 1. Homeostasis and the organization of body fluid compartments:** Intracellular, extracellular and interstitial fluid. Homeostasis, control system and their components. Plasma as an extracellular fluid, RBC, molecular mechanism of blood coagulation, role of vitamin K in coagulation, anticoagulant and fibrinolytic systems. Anemias, polycythemia, haemophilia and thrombosis.  
[10 lecture hours]
- 2. Cardiovascular physiology:** Pressure, flow and resistance. Anatomy of heart. Physiology of the cardiac muscle, automaticity of the cardiac muscle contraction, excitation-contraction coupling, relationship between cardiac cycle, heart sound, ventricular volumes and the ECG, control of cardiac function and output. The arterial system, venous system, the microcirculation and mechanics of capillary fluid exchange. Control of blood flow to the tissues. Portal circulations. Arterial pressure and its regulation. Hypertension, congestive heart disease, atherosclerosis and myocardial infarction.  
[10 lecture hours]
- 3. Respiration:** Organization of the pulmonary system. Mechanism of respiration, pulmonary ventilation and related volumes, pulmonary circulation. Principles of gas exchange and transport. Regulation of respiration. Pulmonary oedema and regulation of pleural fluid. Hypoxia, hypercapnea, pulmonary distress, emphysema, ARDS.  
[10 lecture hours]
- 4. Renal physiology:** Anatomy of the kidney and the nephron. Regulation of renal blood flow. Cell biology of the Bowman's capsule. Physiology of glomerular filtration and GFR. Tubular processing of the glomerular filtrate. Micturition reflex and voluntary control of micturition. Regulation of ECF electrolyte and water content, blood volume and long term blood pressure. Blood buffer systems, renal and pulmonary control of blood pH, renal clearance. Assessment of kidney function. Acidosis and alkalosis. Glomerular nephritis, renal failure, dialysis and diuretics.  
[10 lecture hours]
- 5. Gastrointestinal and hepatic physiology:** Histology of the gastrointestinal tract. Propulsion and motility of food and digested material. Enteric reflexes, secretory functions of the gastrointestinal tract, digestion and absorption of macro and micronutrients. Peptic ulcer, Sprue, celiac disease, IBD, regurgitation, diarrhoea and constipation. Anatomy of the hepatic lobule and blood flow into the liver. Formation and secretion of bile. enterohepatic cycle, reticuloendothelial system, metabolic importance of liver. Liver function tests. Jaundice, liver cirrhosis and fatty liver.  
[10 lecture hours]
- 6. Musculoskeletal system:** Bone structure and formation. Physiology of muscle contraction in striated and non-striated muscle.  
[8 lecture hours]
- 7. Reproductive physiology:** Sex determination and differentiation. Development of female and male genital tracts. Spermatogenesis, capacitation and transport of sperm, blood-testis barrier. Ovarian function and its control. Uterine changes, fertilization and implantation. Placenta as a feto-maternal unit, gestation and parturition.  
[2 lecture hours]

## Reference Books

- Vander's Human Physiology (2008) 11th ed., Widmaier, E.P., Raff, H. and Strang, K.T., McGraw Hill International Publications (New York), ISBN:978-0-07-128366-3.

2. Harper's Biochemistry (2012) 29<sup>th</sup> ed., Murray, R.K., Granner, D.K., Mayes and PA, Rodwell, V.W., Lange Medical Books/McGraw Hill. ISBN:978-0-07-176-576-3.

3. Textbook of Medical Physiology (2011) 10<sup>th</sup> ed., Guyton, A.C. and Hall, J.E., Reed Elsevier India Pvt. Ltd. (New Delhi). ISBN:978-1-4160-4574-8.

4. Fundamental of Anatomy and Physiology (2009), 8<sup>th</sup> ed., Martini, F.H. and Nath, J.L., Pearson Publications (San Francisco), ISBN: 10:0-321-53910-9 / ISBN: 13:978-0321-53910-6. Chemistry of Nucleic acids, Adams.

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

**Relationship between the Course Outcomes (COs) and Program Outcomes (POs)**

Mapping between COs and POs		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	<b>develop</b> the concept about homeostasis and organization of body fluid compartments.	PO1, PO2, PO3, PO5
CO2	<b>demonstrate</b> their knowledge biochemistry in the respiratory and cardiovascular physiology.	PO1, PO2, PO3, PO6
CO3	<b>illustrate</b> their knowledge biochemistry in the digestive and renal physiology.	PO1, PO2, PO3, PO5, PO7
CO4	<b>explain</b> the behaviour of excitable tissue like nerve and muscle.	PO1, PO2, PO3, PO4, PO6
CO5	<b>develop</b> their concept about determination of sex and sexual differentiation.	PO1, PO2, PO4, PO5, PO6

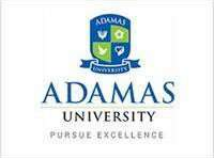
		Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking	Problem Identification Ability	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to Environment	Development to Society.	Development to Humanity
Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
<b>BIC11007</b>	Human Physiology	3	3	3	2	2	3	1	-	1	-	-	-

1=weakly mapped

2= moderately mapped

3=strongly mapped

### Model Question Paper

<b>Name:</b>  <b>Enrolment No:</b>			
<b>Course: BIC11007 –Human Physiology</b> <b>Program: B.ScBiochemistry</b> <span style="float: right;"><b>Time: 03 Hrs.</b></span> <b>Semester:Even 2020-21</b> <span style="float: right;"><b>Max. Marks:40</b></span>			
<b>Instructions:</b> Attempt any three questions from <b>Section A</b> (each carrying 4 marks); any <b>Two Questions</b> from <b>Section B</b> (each carrying 10 marks).			
<b>Section A</b>			
1.	<b>Describe</b> the steps of erythropoiesis in red bone marrow. (R)	<b>5</b>	<b>CO2</b>
2.	<b>Discuss</b> the role of TDF and SOX-9 genes in male sex determination. (U)	<b>5</b>	<b>CO5</b>
3.	If you will apply the second stimulus within the latent period and starting of period of contraction differently, then <b>what</b> will be the alteration in muscle	<b>3</b> <b>2</b>	<b>CO4</b>

	twitch?(Explain through the curve) (An)		
4.	<b>Describe</b> the physiological functions of glutamate and GABA. <b>What</b> will happen if the area 44, area 17 and area 39 is abolished from your cerebral cortex differently? (Ap)	3 2	CO4
<b>SECTION B (Attempt any Two Questions)</b>			
5.	<b>What is saltatory conduction? How</b> nerve impulse is transmitted along the length of the axon?(explain with graphical presentation). <b>Why</b> anterior hypothalamus is called centre of sweating and what is its relation with posterior hypothalamus?(explain with the case of fever) (An)	2 5 3	CO3
6.	<b>How</b> haemoglobin maintains the pH of blood plasma and red blood cell upon oxygenation and carboxygenation? <b>Why</b> the person with O+ blood group can donate only packed RBC to a person of A+ blood group, not whole blood, although O is the universal donor? <b>What</b> is thrombocytopenia? "The renal threshold value of urea is 15"- <b>what</b> does it mean? (R)	4 3 1 2	CO1 CO2
7.	What do you mean by obligatory urine volume. Write down its normal value. How hypertonic urine is formed by juxtamedullary nephrons during water deficit of the body? What is Bence Jones protein? Write down the pH value of urine. (R)	2 4 1 2 1	CO3

<b>BIC12008</b>	<b>Human Physiology Lab</b>	L	T	P	C
<b>Version 1.0</b>	<b>Contact Hours: 45</b>	0	0	3	2
<b>Pre-requisites/Exposure</b>	Knowledge of Biology at 10+2 level				
<b>Co-requisites</b>	Human Physiology				

### Course Objectives

1. To gain a deeper understanding about the organization of buffering system of body fluid.
2. To understand anatomical and physiological organization of respiratory and circulatory system of human body.
3. To analyse the biochemical basis of food digestion.
4. To analyse the biochemical basis of kidney function.
5. To gain knowledge about the biochemical properties of the excitable tissue.

### Course Outcomes

On completion of this course, the students will be able to

- CO1. **Determine** RBC and WBC count from blood.
- CO2. **Identify** the histological slides of different mammalian tissue.
- CO3. **Evaluate** the concentration of haemoglobin from blood.
- CO4. **Illustrate** and fractionate plasma proteins.
- CO5. **Evaluate** the activities of different types of LDH from serum.

### Catalogue Description

Physiology is the scientific study of functions and mechanisms in a living system. As a sub-discipline of biology, physiology focuses on how organisms, organ systems, individual organs, cells, and biomolecules carry out the chemical and physical functions in a living system. Central to physiological functioning are biophysical and biochemical processes, homeostatic control mechanisms, and communication between cells. Physiological state is the condition of normal function of human body and other living system. These normal functions are response of some specific biochemical consequences of the living system. So how to analyze biochemical reason which are responsible to translate into physiological response, that is the goal of this paper.

## Course Content

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1. Hematology-RBC and WBC counting. [10 LectureHours]
2. Estimation of haemoglobin. [10 LectureHours]
  3. Whole protein estimation from serum. [10 LectureHours]
4. Whole Carbohydrate estimation from serum. [10 LectureHours]
  5. Estimation of clotting time. [5 LectureHours]
  6. Estimation of Blood Pressure. [5 LectureHours]
7. Estimation of Hb. [5 LectureHours]
8. Case studies (Identification of Histological Slides). [5 Lecture Hours]

## Reference Books

1. Vander's Human Physiology (2008) 11<sup>th</sup> ed., Widmaier, E.P., Raff, H. and Strang, K.T., McGraw Hill International Publications (New York), ISBN:978-0-07-128366-3.
2. Harper's Biochemistry (2012) 29<sup>th</sup> ed., Murray, R.K., Granner, D.K., Mayes and PA, Rodwell, V.W., Lange Medical Books/McGraw Hill. ISBN:978-0-07-176-576-3.
3. Textbook of Medical Physiology (2011) 10<sup>th</sup> ed., Guyton, A.C. and Hall, J.E., Reed Elsevier India Pvt. Ltd. (New Delhi). ISBN:978-1-4160-4574-8.

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

## Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and POs		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	determine RBC and WBC count from blood.	PO1, PO2, PO3, PO7
CO2	identify the histological slides of different mammalian tissue.	PO1, PO2, PO6, PO7
CO3	evaluate the concentration of haemoglobin from blood.	PO1, PO3, PO6, PO7
CO4	analyse and fractionate plasma proteins.	PO1, PO3, PO5, PO6
CO5	evaluate the carbohydrates from serum.	PO1, PO3, PO4, PO5, PO9

Course Code	Course Title	
<b>BIC12008</b>	Human Physiology Lab	PO1 Fundamental Knowledge
		PO2 Critical Thinking
		PO3 Skill
		PO4 Technical Knowledge
		PO5 Logical Thinking
		PO6 Problem Identification Ability
		PO7 Analytical Knowledge
		PO8 Career Goals
		PO9 Team Work
		PO10 Sustainable Development to Environment
		PO11 Development to Society.
		PO12 Development to Humanity

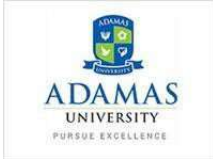
1=weakly mapped

2= moderately mapped

3=strongly mapped



## Model Question Paper

<b>Name:</b>			
<b>Enrolment No:</b>			
<b>Course: BIC12008 –Human Physiology Lab</b>			
<b>Program: B.ScBiochemistry</b>		<b>Time: 03 Hrs.</b>	
<b>Semester:Odd 2020-21</b>		<b>Max. Marks:40</b>	
<b>Instructions:</b>			
Attempt any two questions from <b>Section A</b> (each carrying 10 marks); any <b>Two Questions</b> from <b>Section B</b> (each carrying 10 marks) is <b>Compulsory</b> .			
<b>Section A ( Attemptany Three)</b>			
1.	<b>Identify</b> the specimen of the two given slides (Ap) , write down their specific characteristics (An)	<b>5</b> <b>5</b>	<b>CO1</b> <b>CO2</b>
2.	Estimate the albumin: globulin(A:G) from the supplied serum sample using the standard curve of BSA with biuret reagent(Ap),write down the principle, procedure, observation and inference.(An)	<b>10</b>	<b>CO4</b>
3.	Estimate the hemoglobin content from the supplied packed RBC sample (Ap),write down the principle, procedure, observation and inference.(An)	<b>10</b>	<b>CO3</b>
<b>SECTION B (Attempt any Two Questions)</b>			
5.	<b>Viva-Voce (U/An/Ap/R)</b>	<b>10</b>	<b>CO1, CO2,CO3, CO4,CO5</b>
6.	<b>Laboratory Note Book(U/An/Ap/Ev)</b>	<b>10</b>	<b>CO1, CO2,CO3, CO4,CO5</b>

<b>EVS 11112</b>	<b>Environmental Science</b>	<b>L</b>	<b>T</b>	<b>P</b>	<b>C</b>
<b>Version 1.1</b>	<b>Contact Hours – 30</b>	3	1	0	2
<b>Pre-requisites/Exposure</b>	Basic physics, chemistry, biology and mathematics				
<b>Co-requisites</b>	--				

## Course Objectives

1. To understand the intrinsic relation between humans and environment, our position in the ecosystem around us
2. To comprehend the significance of the biodiversity surrounding us.
3. To figure out the importance and need for energy resources, various sources of energy, renewable and non-renewable sources, conventional and unconventional sources.
4. To have basic concepts about sustainability, our dependence on nature and the consequences of overexploitation.
5. To enable students to appreciate the importance and how much we owe to the earth systems for our survival.
6. To have a basic concept about the types of pollution and mitigation procedures.
7. To have an overall idea about the environmental legal framework in our country and about the EIA and environmental audit procedures.

## Course Outcomes

On completion of this course, the students will be able to

CO 1: **Explain** multidimensional complex nature of environmental problems, various types of ecosystems, ecosystem dynamics, perceive and appreciate the surrounding nature.

CO 2: **Illustrate** the intrinsic relation between humans and environment, our position in the ecosystem around us, and importance of biodiversity.

CO 3: **Comprehend** the presence of various pollutants, their significance, and impacts, and develop the underlying concepts involved in various air pollution prevention and mitigation measures.

CO4: **Define** the routes of generation, classification, management, and environmental significance of solid waste.

CO 5: **Demonstrate** water chemistry, deduce the relationship between various water pollutants, and understand the principles of various water and wastewater treatment procedures.

CO 6: **Create** awareness and concern about importance of environmental resources and their damage and protection.

## **Catalogue Description**

To distinguish between various types of ecosystems, ecosystem dynamics, perceive and appreciate the surrounding nature and feel connected, develop the concept of innate relationship of humans and biodiversity, need for conservation and different conservation strategies. The students will be developed in a way so that they can spontaneously comprehend the importance of studying about the various air pollutants, their significance and impacts, and develop the underlying concepts involved in various air pollution prevention and mitigation measures, understand fundamental water chemistry, deduce the relationship between various water pollutants, and understand the principles of various water and wastewater treatment procedures.

They will understand the routes of generation, classification, management and environmental significance of solid waste, apply the basic concepts of waste management in their daily lives, understand the need of the 5Rs of waste management, importance of waste minimization.

## **Detailed syllabus**

### **Unit I: [5 Lecture Hours]**

Multidisciplinary nature of environmental sciences; scope and importance; need for public awareness; concept of sustainability and sustainable development

Forest resources: Function of forests, cause and effects of deforestation, case studies.

Water resources: distribution of water, hydrological cycle, use and over-utilization of surface and ground water, floods, drought, conflicts over water, dams-benefits and problems

Mineral resources: Use and exploitation, environmental effects of extracting and using mineral resources, case studies

Food Resources: World food problems and environmental concern, Food security, case studies

Energy resources: Concept of energy, SI Units of Work, Heat and Power, World energy use, Energy consumption pattern in India and U.S., Environmental aspects of energy utilization

Renewable and non-renewable sources; Fossil fuel: types, use and environmental impacts, Solar

energy: Solar Radiation – Passive and active solar systems – Flat Plate and Concentrating Collectors – Solar direct Thermal Application– Fundamentals of Solar Photo Voltaic

Conversion- advantages and disadvantages of Solar Power generation, Solar energy status in India; Wind Energy: site selection, Wind turbine: basic working principle and types, Wind

energy status in India, advantages and disadvantages of Wind Power generation; Hydroelectric power : How is it generated, advantages and disadvantages; Biomass energy: various types,

generations of biofuel, Biogas plants, Bio diesel; Geothermal Energy: source, advantages and disadvantages, Nuclear Power: nuclear fission, moderation of reaction, nuclear reactor:

pressurized water reactor, advantages and disadvantages

### **Unit II: Ecosystems [5 Lecture Hours]**

Concept of an ecosystem, Structure and function of an ecosystem, Producers, consumers and decomposers, Food chains, food webs and ecological pyramids, energy flow, ecological succession

### **Unit III: Biodiversity and its conservation [5 Lecture Hours]**

Level of Biodiversity: genetic, species and ecosystem diversity. Biogeographical classification of India, Values of biodiversity, Biodiversity at global, National and local levels, India as a mega-diversity nation, Biodiversity hotspots, Threats to Biodiversity, In-situ and Ex-situ conservation of Biodiversity

### **Unit – IV: Environmental Pollution and Waste Management [10 Lecture Hours]**

Environmental pollution: types, causes, effects and controls; Air, water, soil and noise pollution, marine pollution; case studies. Nuclear hazards and human health risks.

Sources and generation of solid wastes, their characterization, chemical composition and classification. Different methods of disposal and management of solid wastes (Hospital Wastes and Hazardous Wastes) Recycling of waste material. Waste minimization technologies. Hazardous Wastes Management and Handling Rules, 1989

### **Unit – V: Global Issues and Environmental Acts [5 Lecture Hours]**

Climate change, global warming, acid rain, ozone layer depletion, nuclear accidents. Environment Laws: Environment Protection Act; Air (Prevention & Control of Pollution) Act; Water (Prevention and control of Pollution) Act; Wildlife Protection Act; Forest Conservation Act. International agreements: Montreal and Kyoto protocols and Convention on Biological Diversity (CBD).

### **Text Books:**

1. Principles of Environmental Science, 4th edition by Cunningham, W.P. and Cunningham, M.A. (2002), Tata McGraw-Hill Publishing Company, New Delhi
2. Basic Environmental Engineering & Elementary Biology by Monidranath Patra and Rahul Kumar Singha, Aryan Publishing house
3. Introduction to Environmental Engineering and Science, by Masters, G.M., Prentice Hall of India, Second Indian Reprint.

### **Reference Books:**

- 1 Wastewater Engineering: Treatment and Reuse, 4th Edition, Metcalf and Eddy, Inc. McGraw-Hill, Inc., New York, 2002
- 2 Environmental Engineering”, Howard S. Peavy, Donald R. Rowe and George Tchobanoglous, McGraw-Hill Education (India) Private Limited, New Delhi

3 Introduction to Environmental Engineering, 2nd Ed. by Davis, M. L. and Cornwell D. A. McGraw Hill, Singapore.

4 Environmental Sciences: The Environment and Human Impact by Jackson, A.R.W. and Jackson, J.M., , Longman Publishers

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

**Relationship between the Course Outcomes (COs) and Program Outcomes (POs)**

Mapping between COs and POs		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	<b>Describe</b> multidimensional complex nature of environmental problems, various types of ecosystems, ecosystem dynamics, perceive and appreciate the surrounding nature.	PO1, PO2, PO10, PO12
CO2	<b>Explain</b> the intrinsic relation between humans and environment, our position in the ecosystem around us, and importance of biodiversity.	PO1, PO2, PO10, PO11, PO12
CO3	<b>Comprehend</b> the presence of various air pollutants, their significance, and impacts, and develop the underlying concepts involved in various air pollution prevention and mitigation measures.	PO8, PO10
CO4	<b>Define</b> the routes of generation, classification, management, and environmental significance of solid waste.	PO6, PO11, PO12
CO5	<b>Demonstrate</b> water chemistry, deduce the relationship between various water pollutants, and understand the principles of various water and wastewater treatment procedures.	PO5, PO6, PO10
CO6	<b>Create</b> awareness and concern about importance of environmental resources and their damage and protection.	PO10, PO11, PO12

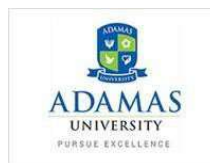
Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
<b>EVS 11112</b>	Environmental Science and Energy Resources	Fundamental Knowledge											
		Critical Thinking											
		Skill											
		Technical Knowledge											
		Logical Thinking											
		Problem Identification Ability											
		Analytical Knowledge											
		Career Goals											
		Team Work											
		Sustainable Development to Environment											
		Development to Society.											
		Development to Humanity											

1=weakly mapped

2= moderately mapped

3=strongly mapped

Name:  
Enrolment No:



Course: EVS 11112– Environmental science Program: B.Sc. Biochemistry Semester: Even 2020-21

Time: 03Hrs.  
Max. Marks: 40

**Instructions:**

Attempt any **five questions** from **Section A** (Each Carrying 2 Marks); any **four questions** from **Section B** (Each Carrying 5 Marks). Any **one question** from **Section C** (Carrying 10Marks).

**SECTION A (Answer any five questions) (5 x 2 = 10)**

1.	What information about any ecosystem are conveyed by ecological pyramids?	Remembering	CO1
2.	Give one example for each. a. Inverted pyramid of number; b. Inverted pyramid of biomass	Understanding	CO2
3.	A sample of sewage water has 4-day 20°C BOD value of 60% of the final. Find the rate constant per day.	Applying	CO5
4.	Mention few problems associated with large dams?	Remembering	CO6
5.	What are the different types of wind turbine?	Remembering	CO6
6.	What is PV cell? What are the diverse applications of solar energy unlike other renewable energy resources?	Remembering	CO3

**SECTION B (Attempt any four questions) (4 x 5 = 20)**

1.	What is electrostatic precipitator? What are the advantages of electrostatic precipitator? (1+2+2=5)	Understanding	CO3
2.	What is “Dobson unit”? What is ozone hole? How does it happen chemically in the stratosphere? (1+1+3=5)	Remembering	CO9
3.	What are the adverse effects of open dumping of municipal solid wastes on environment? How does sanitary landfill differ from open dumping? (3+2=5)	Analysing	CO4
4.	Describe the distribution of water resources.	Remembering	CO6
5.	What are the fundamental steps involved in an EIA? Draw a simple flowchart describing the steps that are followed in an EIA process in India. (2+3=5)	Remembering	CO8

**SECTION C (Attempt any one question) (1 x 10 = 10)**

1.	What are the various methods for water resources management? What is integrated solid waste management? How does it differ from the age-old conventional system of solid waste management in India? (5+3+2=10)	Applying	CO6
2.	How is photochemical smog formed? What are effects of photochemical smog? Discuss the factors affecting photochemical smog? (4+3+3=10)	Understanding	CO3

<b>IDP14001</b>	<b>Interdisciplinary Project</b>	<b>L</b>	<b>T</b>	<b>P</b>	<b>C</b>
<b>Version 1.0</b>		2	0	0	3
<b>Pre-requisites/Exposure</b>	PLUS TWO LEVEL BIOLOGY/SCIENCE				
<b>Co-requisites</b>	Basic subject knowledge of Semester I				

### Course Objectives

1. This will enable students to design and evaluate scientific investigations of interdisciplinary nature.
2. Students will learn to deduce evidence-based conclusions in collaboration with others.
3. Skill of presentation and scientific content writing will be improved.

### Course Outcomes

On completion of this course, the students will be able to

- CO1. **comprehend** novel interdisciplinary ideas to enrich their scientific interest.
- CO2. **explain** the theoretical and practical knowledge for identification, estimating different molecules and their applications.
- CO3. **compare** different techniques and analyze the results obtained.
- CO4. **evaluate** scientific investigations of interdisciplinary nature.
- CO5. **illustrate** evidence-based conclusions.

### Catalogue Description

The core-course of ‘interdisciplinary project’ will enable the students to nurture their research interest by compiling knowledge obtained different subjects during their education together with novel ideas. An idea about appropriate application of various skills for industrial and research purpose can be developed. With the potential to design and evaluate scientific investigations to the students, who will learn to comprehend conclusions based on experimental evidences. The entire literature review work and experimentation focuses on practical implementation of knowledge in a collaborative environment. Students will perceive the basic concepts of the subject via exercise and discussions with the mentor(s).



**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination  
Examination Scheme:**

Components	Project Thesis	Presentation
Weightage (%)	50	50

**Relationship between the Course Outcomes (COs) and Program Outcomes (POs)**

Mapping between COs and POs		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	<b>comprehend</b> novel interdisciplinary ideas to enrich their scientific interest.	PO1, PO2, PO5, PO6, PO9
CO2	<b>explain</b> their theoretical and practical knowledge for identification, estimating different molecules and their applications.	PO1, PO2, PO3, PO4, PO7
CO3	<b>compare</b> different techniques and analyze the results obtained.	PO1, PO2, PO5, PO6, PO7
CO4	<b>illustrate</b> scientific investigations of interdisciplinary nature.	PO2, PO6, PO7, PO9,
CO5	<b>evaluate</b> evidence-based conclusions.	PO1, PO2, PO3, PO4, PO7

Course Code	Course Title	PO 1	PO 2	PO 3	PO 4	PO 5	PO 6	PO 7	PO 8	PO 9	PO1 0	PO1 1	PO1 2
		PO 1	PO 2	PO 3	PO 4	PO 5	PO 6	PO 7	PO 8	PO 9	PO1 0	PO1 1	PO1 2
		Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking	Problem Identification Ability	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to Environment	Development to Society.	Development to Humanity
<b>IDP140 01</b>	Interdisciplinary Project	3	3	2	2	2	3	3	-	2	-	-	-

1=weakly mapped

2=moderately mapped

3=strongly mapped

BOT 11003	<b>ELECTIVE BOTANY – II (Theory)</b>	<b>L</b>	<b>T</b>	<b>P</b>	<b>C</b>
<b>Version 1.0</b>	<b>Contact Hours - 60</b>	3	1	0	4
<b>Pre-requisites/Exposure</b>	12 <sup>th</sup> with Biology as one subject				
<b>Co-requisites</b>	-				

### Course Objectives:

1. Students will be able to develop basic knowledge in higher groups of plants and also be able to understand various branches of Botany for understanding in details;
2. Students will learn practical implementations of various branches of plant science and their and commercial exploitations;
3. Students will be able to develop fundamental knowledge about surrounding ecosphere and biosphere and their corelations.

### Course Outcomes

On completion of this course, the students will be able to

- CO1. **Illustrate** in detail about *Mosses and Ferns* in various aspects and also their implementation for commercial purposes;
- CO2. **Explain** and Categorize in detail various divisions of *Gymnosperms* and their commercial importance as well harmful effects;
- CO3. **Develop** basic knowledge in Angiospermic plants and also be able to understand various divisions for understanding in details;
- CO4. **Demonstrate** fundamental knowledge about surrounding ecosphere and biosphere and their corelations;
- CO5. **Explain** and summarize the basic concept of pharmacognosy and also can be able implement this knowledge in applied fields;
- CO6. **Develop** fundamental knowledge in plant biotechnology and can be implemented this skill for cultivation practices and commercial uses.

### Catalogue Description

Elective Botany II lecture course covers a vast range of basic plant science. The course takes a broader approach and covers many aspects of higher group of plants as well fundamental knowledge of surrounding environments and their corelations. Moreover, this curriculum covers a practical approach for understanding implementation of gaining knowledge in industrial field. Classroom activities will be designed to encourage students to play an active role in the

construction of their own knowledge and in the design of their own learning strategies. We will combine traditional lectures with other active teaching methodologies using digital platforms, such as analysis of video scenes and debates. Students will be encouraged to actively take part in all group activities and to give an oral group presentation. Students will be expected to interact with media resources, such as, web sites, videos, DVDs, and newspapers etc.

<b>Course Content:</b>	<b>(60 hours)</b>
<b>UNIT I</b>	<b>(10 hours)</b>
<i>Mosses and Ferns:</i>	
<i>Bryophytes:</i> Introduction, Basic classifications, Diagnostic characters of important families with examples, Economic importance.	
<i>Pteridophytes:</i> Introduction, Basic classifications, Diagnostic characters of important families with examples, Economic importance.	
<b>UNIT II</b>	<b>(10 hours)</b>
<i>Gymnosperms:</i> Introduction, Basic classifications, Diagnostic characters of important families with examples, Economic importance.	
<b>UNIT III</b>	<b>(13 hours)</b>
<i>Angiosperm:</i> Brief ultra-structure of plant cell and tissues, Brief description of Plant Architecture (vegetative and reproductive parts); Taxonomic Definitions, principles, Outline of the system of classification with examples, Herbarium, ICBN(ICN), Diagnostic features of important angiospermic families.	
<b>UNIT IV</b>	<b>(12 hours)</b>
<i>Plant Ecology:</i> Population and community ecology, Ecological Succession, Major Ecosystems and ecological adaptations, Environmental Pollution and its effects on plants, Biodiversity and conservation, Bio/Phytoremediation and their environmental significance.	
<b>UNIT V</b>	<b>(7 hours)</b>
<i>Pharmacognosy:</i> A brief idea about pharmacognosy, discuss about- active principles; Pharmacopeia and adulteration; Study of the following drug plants (Diagnostic features, active principles and uses): <i>Rauwolfia serpentina</i> (root), <i>Adhatodavatica</i> (leaf), <i>Strychnos nuxvomica</i> (seed), <i>Cinchona succirubra</i> (bark), Business review of herbal industry.	
<b>UNIT VI</b>	<b>(8 hours)</b>
<i>Plant Biotechnology:</i> Plant Tissue Culture: Introduction, Composition of media; Nutrient and hormone requirements, Types, Applications, In vitro germplasm conservation; Methods of gene transfer: Agrobacterium-mediated, Direct gene transfer methods;	

Applications of Biotechnology: Problems and prospects of transgenic crops and their commercial utilizations

### Text Books

- 1 Bhojwani S S&Dantu P K Plant Tissue Culture: An introductory text
2. Odum, E. P. 1971. Fundamentals of Ecology. W.B Sounders Co., Philadelphia
3. Trigiano R N & Gray D J Plant Tissue Culture, Development and Biotechnology
4. Trease & Evans: Pharmacognosy

### Reference Books

1. College Botany Vol. II By Gangulee and Kar
2. Studies in Botany Vol I & II by J.N. Mitra, D. Mitra, S.K. Chaudhuri

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

### Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and POs		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	<b>illustrate</b> in detail about <i>Mosses and Ferns</i> in various aspects and also their implementation for commercial purposes	PO1, PO2, PO5
CO2	<b>explain</b> and Categorize in detail various divisions of <i>Gymnosperms</i> and their commercial importance as well harmful effects	PO1, PO2, PO4
CO3	<b>illustrate</b> basic knowledge in Angiospermic plants and also be able to understand various divisions for understanding in details	PO1, PO4, PO6
CO4	<b>develop</b> fundamental knowledge about surrounding ecosphere and biosphere and their correlations	PO1, PO3, PO5
CO5	<b>explain</b> and summarize the basic concept of pharmacognosy and also can be able implement this knowledge in applied fields	PO1, PO2, PO3
CO 6	<b>develop</b> fundamental knowledge in plant biotechnology and can be implemented this skill for cultivation practices and commercial uses	PO1, PO3, PO6, PO8

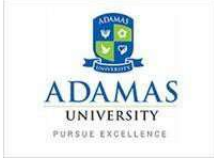
Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
BOT 11003	ELECTIVE BOTANY - I (THEORY)	3	3	1	-	2	1	-	1	-	-	-	-

1=weakly mapped

2= moderately mapped

3=strongly mapped

### Model Question Paper

<b>Name:</b>			
<b>Enrolment No:</b>			
Course: BOT 11003-ELECTIVE BOTANY - II(THEORY) <b>Program: B.Sc. Biochemistry</b> <b>Semester: Odd 2019-20</b>			
		<b>Time: 03 Hrs.</b> <b>Max. Marks: 40</b>	
<b>Instructions:</b> Attempt any <b>four</b> questions from <b>Section A</b> (each carrying 5 marks); any <b>two</b> questions from <b>Section B</b> (each carrying 10 marks).			
<b>SECTION A (Attempt any Four questions)</b>			
1.	Why 'Pteridophytes are claimed as the first true land plants not Bryophytes' explain in details under the light of their morphological and anatomical structure. (An)	5	CO1
2.	Mention the steps of Production of natural drug products. (R)	5	CO5

3.	Write down the subsequent steps of making herbarium sheet. (R)	5	CO3
4.	Briefly discuss about the Population and community ecology. (R)	5	CO4
5	Mention the key differences between plant tissue culture and open field culture. (R)	5	CO6
<b>SECTION B (Attempt any Two questions)</b>			
6.	(a) Bryophytes and other plants have a common ancestor, but neither is ancestral to the other- justify with suitable comparison. (An) (b) Mention the Medicinal use of Bryophytes. (R)	6+4	CO1
7.	(a) Define Plant Tissue Culture. (R) (b) Discuss their achievements, advantages and limitations. (R)	3+7	CO6
8.	(a) Elaborately discuss the characteristic features of Solanaceae family. (R) (b) Draw and label the ultra-structure of a typical plant cell. (U)	5+5	CO3
9	Briefly explain the Basic classifications of Gymnosperm with suitable examples. (U) Mention the key differences between Gymnosperms and Angiosperms. (R)	7+3	CO2

<b>BOT 12004</b>	<b>Elective Botany II LAB(Practical)</b>	<b>L</b>	<b>T</b>	<b>P</b>	<b>C</b>
<b>Version 1.0</b>	<b>Contact Hours - 45</b>	0	0	3	2
<b>Pre-requisites/Exposure</b>	12 <sup>th</sup> with Biology as one subject				
<b>Co-requisites</b>	Elective Botany I				

### **Course Objectives:**

1. Students will be able to demonstrate and design with hands-on activities for applied plant science experiments in laboratory;
2. Students will be able to implement acquired knowledge in commercial field in crop improvement.

### **Course Outcomes**

On completion of this course, the students will be able to

CO1. **Develop** their skill by hands on training in laboratory Experiments on Imbibition in Plants

CO2. **Design** and demonstrate Root Pressure in Plants;

CO3. **Explain** by performing Demonstration of opening & closing of stomata;

CO4. **Develop** fundamental knowledge and can be implemented this skill for demonstrating Transpiration in Plants;

CO5. **Design** and explain Demonstration of respiration;

CO6. **Demonstrate** and design experiments for Osmosis in Plants;

CO7. **Develop** their skill by hands on training in laboratory Experiments on performing Plasmolysis in Plants

CO8. **Illustrate** their skill by hands on training in laboratory for Photosynthesis in Plants;

### **Catalogue Description**

Elective Botany practical is a skill enhancement course covers a vast range of basics as well as applied

implementation of knowledge in inherent metabolism of plants as well technical application for betterment for mankind. The course takes a broader approach and covers many spectra of plant physiological phenomena as well fundamental techniques in plant tissue culture using various technical parameters in laboratory scale which broadly covered by demonstration and hands on trainings. Classroom activities will be designed to encourage students to play an active role in the construction of their own knowledge and in the design of their own learning strategies. We will combine traditional lectures with other active teaching methodologies using digital platforms, such as analysis of video scenes and debates. Students will be encouraged to



actively take part in all group activities and to give an oral group presentation. Students will be expected to interact with media resources, such as, web sites, videos, DVDs, and newspapers etc.

<b>CourseContent:</b>	<b>(60hours)</b>
<b>UNIT I</b> Experiments on Imbibition in Plants	<b>(5hours)</b>
<b>UNIT II</b> Experiment on Root Pressure in Plants	<b>(5hours)</b>
<b>UNIT III</b> Demonstration of opening & closing of stomata	<b>(5hours)</b>
<b>UNIT IV</b> Experiments on Respiration in Plants	<b>(5hours)</b>
<b>UNIT V</b> Experiments on Osmosis in Plants	<b>(5hours)</b>
<b>UNIT VI</b> Experiments on Ascent of Sap in Plants	<b>(5hours)</b>
<b>UNIT VII</b> Experiments on Plasmolysis in Plants	<b>(5hours)</b>
<b>UNIT VIII</b> Experiments on Photosynthesis in Plants	<b>(8hours)</b>
<b>UNIT IX</b> Experiments on Plant growth	<b>(7hours)</b>
<b>UNIT X</b> Basic plant tissue culture techniques: Media composition and Preparation of media Sterilization and contamination, Initiation of aseptic cultures from seed, isolated embryos and other explants	<b>(10hours)</b>
<b>UNIT XI</b> Local Excursions and Field records	

#### **Text Books**

1. Hopkins, W.G. and Huner, P.A. 2008 Introduction to Plant Physiology. John Wiley and Sons.
2. Plant cell culture – A practical approach by Dixon RA. 1995

## Reference Books

1. Practical Botany, Volume II, S CSamanta

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

## Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and POs		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	<b>develop</b> skill by hands on training in laboratory Experiments on Imbibition in Plants	PO1, PO2, PO3, PO8
CO2	<b>design</b> and demonstrate Root Pressure in Plants	PO1, PO2, PO5
CO3	<b>explain</b> by performing Demonstration of opening & closing of stomata	PO1, PO3, PO4, PO5
CO 4	<b>design</b> and explain Demonstration of respiration	PO1, PO2, PO5
CO 5	<b>demonstrate</b> and design experiments for Osmosis in Plants	PO1, PO3, PO4
CO 6	<b>illustrate</b> and design experiments on Ascent of Sap in Plants	PO1, PO2, PO4
CO7	<b>develop</b> their skill by hands on training in laboratory Experiments on performing Plasmolysis in Plants	PO1, PO3, PO6
CO8	<b>explain</b> the importance of plant growth and development by demonstration and hands on training	PO1, PO2, PO8

Course Code	Course Title	
<b>BOT 12004</b>	<b>PLANT PHYSIOLOGY (THEORY)</b>	PO1 Fundamental Knowledge
		PO2 Critical Thinking
		PO3 Skill
		PO4 Technical Knowledge
		PO5 Logical Thinking
		PO6 Problem Identification Ability
		PO7 Analytical Knowledge
		PO8 Career Goals
		PO9 Team Work
		PO10 Sustainable Development to Environment
		PO11 Development to Society.
		PO12 Development to Humanity

1=weakly mapped

2 = moderately mapped

3=strongly mapped

**Model Question Paper**

<b>Name:</b>			
<b>Enrolment No:</b>			
<p>Course: <b>BOT 12004 Elective Botany II (PRACTICAL)</b>  <b>Program: B.Sc. Biochemistry</b>  <b>Semester: Even 2019-20</b>  <b>Instructions:</b>          Attempt any <b>two</b> questions from <b>Section A</b> (each carrying 10 marks); <b>Section B</b> is Compulsory (carrying 10 marks).</p>			
		<b>Time: 03Hrs.</b> <b>Max. Marks: 40</b>	
1.	a) <b>Work out with</b> specimen A. b) <b>Write</b> about the protocol and observations. (R)	6 4	CO1, C9, C7
2.	<b>Design and analysis</b> with specimen B. (An)	10	CO2, CO3 & CO4
3.	<b>Estimate</b> the final outcome with specimen C. (Ap)	10	CO3, CO4, CO5
4.	a) <b>Demonstrate</b> the preparation of provided samples with specimen D. (R) b) <b>Write</b> the procedure in details. (R)	4 6	C8, CO10
<b>SECTION B is compulsory</b>			
5.	<b>Viva-voce</b>	10	CO1, CO2, CO3, CO4, CO5, CO6 CO7, CO8, CO9, CO10
6.	<b>Practical copy</b>	10	CO1, CO2, CO3, CO4, CO5, CO6, CO7, CO8, CO9, CO10

<b>ZOL 11003</b>	<b>ELECTIVE ZOOLOGY II (THEORY)</b>	<b>L</b>	<b>T</b>	<b>P</b>	<b>C</b>
<b>Version 1.0</b>	<b>Contact Hours - 60</b>	3	1	0	4
<b>Pre-requisites/Exposure</b>	PLUS TWO LEVEL BIOLOGY				
<b>Co-requisites</b>	Elective Zoology I				

### Course Objectives

1. To provide those students with some biology background with an introduction to ecology and the study of evolution. This course is designed for students of any major, but will especially benefit biology majors, as well as secondary science education majors.
2. It will also provide an informative elective for 5-8 math/science education majors.
3. Gathering information about ecological community, biodiversity and its structure and function, and how evolutionary concepts and animal behavior enables us to live a more knowledgeable, involved, and environmentally aware life in a science-conscious age.

### Course Outcomes

On completion of this course, the students will be able to

- CO1. **Illustrate** about habitat and niche, ecotone, edge effect, ecological succession.
- CO2. **explain** the significance of energy flow in ecosystem and ecological succession, variation and ecological process which implement changes in learning towards a sustainable development.
- CO3. **Compare** between population growth curves, life strategies.
- CO4. **demonstrate** different animal behaviour, bee dancing which follows future practice of ethical philosophies.
- CO5. **explain** the problems on density, mortality, natality from fecundity tables and life tables and population genetics.

### Catalogue Description

This course covers ecological and evolutionary principles on population, community, ecosystem and biodiversity. The very nature of ecology and evolution requires students to view role of evolutionary process, animal behavior on modern human life. All the lectures will be devoted on discussions of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as power point presentation, audio visual virtual lab session as per requirement. The tutorials will familiarize the students with practical problem-solving techniques led by the course coordinator. Students will strongly grab the basic concepts of the subject via exercise and discussions with the coordinator.

## Course Content

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### **ELECTIVE ZOOLOGY II**

#### **GE 2: ECOLOGICAL PRINCIPLES, EVOLUTION AND BIODIVERSITY (THEORY)**

##### **Unit 1: Introduction to Ecology** **4 Lecture Hours**

History of ecology, Autecology and synecology, Levels of organization.

##### **Unit 2: Population** **9 Lecture Hours**

Unitary and Modular populations. Unique and group attributes of population: Density, natality, mortality, life tables, fecundity tables, survivorship curves, age ratio, sex ratio, dispersal and dispersion. Exponential and logistic growth, equation and patterns, r and K strategies. Population regulation - density-dependent and independent factors.

##### **Unit 3: Community** **9 Lecture Hours**

Community characteristics: species richness, dominance, diversity, Ecotone and edge effect. Ecological succession with one example.

##### **Unit 4: Ecosystem** **8 Lecture Hours**

Types of ecosystems with one example in detail, definition, components, energy flow, food chain, food web, and ecological pyramids.

##### **Unit 5: Historical review of evolutionary concept** **5 Lecture Hours**

Lamarckism, Darwinism, Neo-Darwinism, Geological time scale.

##### **Unit 6: Sources of variations and Population genetics** **10 Lecture Hours**

Heritable variations and their role in evolution, Hardy-Weinberg Law (statement and derivation of equation, application of law to human Population); Evolutionary forces upsetting H-W equilibrium; Natural selection (concept of fitness, selection coefficient, derivation of one unit of selection for a dominant allele, genetic load, mechanism of working, types of selection, density-dependent selection, heterozygous superiority, kin selection, adaptive resemblances, sexual selection. Genetic Drift (mechanism, founder's effect, bottleneck phenomenon; Role of Migration and Mutation in changing allele frequencies), Speciation.

##### **Unit 7: Animal Behaviour** **7 Lecture Hours**

Instinctive and learning behaviour, Fixed action pattern, Communication in honeybees (dance Language), Elements of Sociobiology: Altruism and selfishness.

##### **Unit 8: Biodiversity** **8 Lecture Hours**

Basic concept of Biodiversity, Biodiversity hotspots, Conservation of wild life- purpose & methods, concept of Biosphere Reserve, importance & strategies of wildlife conservation;

conservation act and application. National park & Wildlife Sanctuary, Animal cruelty and prevention act.

### Reference Books

1. Colinvaux, P. A. (1993). Ecology. II Edition. Wiley, John and Sons, Inc
2. Krebs, C. J. (2001). Ecology. VI Edition. Benjamin Cummings.
3. Odum, E.P., (2008). Fundamentals of Ecology. Indian Edition. Brooks/Cole
4. Robert Leo Smith Ecology and field biology Harper and Row publishers
5. Ricklefs, R.E., (2000). Ecology. V Edition. Chiron Press
6. Ridley, M (2004) Evolution III Edition Blackwell publishing
7. Douglas, J. Futuyma (1997). Evolutionary Biology. Sinauer Associates.

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

### Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and Pos		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	<b>illustrate</b> about habitat and niche, ecotone, edge effect, ecological succession.	PO1, PO4, PO5, PO8
CO2	<b>explain</b> the significance of energy flow in ecosystem and ecological succession, variation and evolutionary process which implement changes in learning towards a sustainable development.	PO1, PO2, PO4, PO10
CO3	<b>compare</b> and analyze between population growth curves, life strategies.	PO1, PO2, PO5, PO6, PO7
CO4	<b>demonstrate</b> different animal behaviour, bee dancing which follows future practice of ethical philosophies.	PO1, PO3, PO5, PO8
CO5	<b>explain</b> problems on density, mortality, natality from fecundity tables and life tables and population genetics.	PO1, PO2, PO5, PO6, PO7

Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12		
<b>ZOL 11003</b>	<b>ELECTIVE ZOOLOGY II (THEORY)</b>	Fundamental Knowledge													
		Critical Thinking													
		Skill													
		Technical Knowledge													
		Logical Thinking													
		Problem Identification Ability													
		Analytical Knowledge													
		Career Goals													
		Team Work													
		Sustainable Development to Environment													
		Development to Society.													
		Development to Humanity													

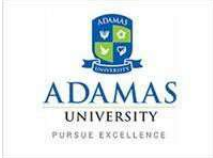
1=weakly mapped

2 = moderately mapped

3=strongly mapped



## Model Question Paper

<b>Name:</b>  <b>Enrolment No:</b>			
<b>Course: ZOL 11003 - ELECTIVE ZOOLOGY II (THEORY)</b> <b>Program: B.Sc. Biochemistry</b> <b>Semester: Even 2020-21</b>			
<b>Time: 03 Hrs.</b> <b>Max. Marks: 40</b>			
<b>Instructions:</b> Attempt any <b>three</b> questions from <b>Section A</b> (each carrying 4 marks); any <b>four</b> questions from <b>Section B</b> (each carrying 7 marks).			
<b>SECTION A ( Attempt any Three questions)</b>			
1.	a) <b>What</b> is ecotone? Write two general features of ecotone? (R) b) <b>What</b> do you understand about the term 'Edge Species'? (U)	2 2	CO1
2.	A community in a wood log depends on the neighbouring community, but a forest community doesn't. <b>Explain</b> this statement by clearly mentioning all the factors involved. (An)	4	CO2
3.	The graph shows the changes in population of bass fish in a lake a) <b>Describe</b> the trend in population growth from 1950 to 1990. Give an example of something that may have happened in 1990 that affected the bass population. (Ap)  b) <b>Explain</b> your reasoning. (An)	2 2	CO3
4.	The mule, which is the offspring of a male donkey and a mare, is a vigorous animal, well suited for hard work. However, it is sterile. <b>Why? Explain</b> your answer with a proper scientific reason. (An)	4	CO4
<b>SECTION B (Attempt any Four questions)</b>			
5.	a) <b>Utilize</b> a proper illustration to explain how a worker bee can inform other fellow worker bees that the food source may be found by flying 30° to the right of the sun when they will leave the hive. (Ap) b) Does the genetic drift play a more important role in large or a small population? <b>Explain</b> with proper reasons. (U)	5 2	CO4 CO5
6.	a) Two populations were sampled for distribution of the MN blood group. <b>Evaluate</b> each of these two populations is in equilibrium or not? (Eva)  (i) 2% M, 96% MN, 2 % N  (ii) 9% M, 42% MN, 49% N  b) If you analysed a population for a trait known to be due to a single pair of genes and your analysis revealed that the population was not in equilibrium for that trait, <b>what possible reasons</b> could you offer for the lack of equilibrium? (An)	3 4	CO5

7.	<p>a) <b>What</b> is mimicry? How does it <b>differ</b> from crypsis? Give examples of both.(U)</p> <p>b) <b>Describe</b> the following equation with respect to the regulation of the population size.(U)</p> $N_{t+1}=N_t+B+I-D-E$ <p>Clearly <b>write</b> the name of all the symbols used in the equation.(R)</p>	4 3	CO4 CO1												
8.	<p>a) On a particular island, a population suddenly becomes split into two parts by a mountain range. After very many generations the two parts of the original population come together again. <b>What</b> do you think the nature of this reformed population might be? <b>Indicate</b> the evolutionary processes involved in your answer. (An)</p> <p>b) <b>What</b> do you mean by Batesian mimicry? Give example.(R)</p>	5 2	CO2 CO4												
9.	<p>a) <b>Which</b> country below has the most rapid population growth? Will the pyramids above tell you total population of a country? (An)</p> <div data-bbox="501 1010 1050 1367" style="border: 1px solid black; height: 170px; width: 338px; margin: 10px auto;"></div>	2	CO1												
	<p>b) Using the table <b>calculate</b> the species diversity and species dominance of the community. Also you have to clearly mention in which species is dominant in the community. (Ap)</p> <table border="1" data-bbox="610 1614 1055 2049" style="margin-left: auto; margin-right: auto;"> <thead> <tr> <th>Species Name</th> <th>Species Number</th> </tr> </thead> <tbody> <tr> <td>A</td> <td>54</td> </tr> <tr> <td>B</td> <td>11</td> </tr> <tr> <td>C</td> <td>28</td> </tr> <tr> <td>D</td> <td>1</td> </tr> <tr> <td>E</td> <td>3</td> </tr> </tbody> </table>	Species Name	Species Number	A	54	B	11	C	28	D	1	E	3	5	
Species Name	Species Number														
A	54														
B	11														
C	28														
D	1														
E	3														

ZOL 12004	ELECTIVE ZOOLOGY LAB II (PRACTICAL)	L	T	P	C
Version 1.0	Contact Hours - 45	0	0	3	2
Pre-requisites/Exposure	PLUS TWO LEVEL BIOLOGY				
Co-requisites	Elective Zoology				

### Course Objectives

1. To provide students with hands-on activities designed to encourage interest in the field ecology and evolution, as well as promote greater understanding of the concepts presented in lecture.
2. Students will need to become proficient with calculations, analysis and applications of different types from the hypothetical / data provided.

### Course Outcomes

On completion of this course, the students will be able to

- CO1. **Identify** and compare between fossil specimens and also learn how to construct phylogenetic tree and draw them.
- CO2. **explain** the importance of field trip and they will get more interest on the subject as they learn how to work in the field tripeffectively as an individual, and as a member or leader in diverse teams, and in multidisciplinary settings.
- CO3. **Illustrate.** calculate and analyse Shannon-Weiner diversity index for the same community.
- CO4. **Solve** the analytical problems on Hardy-Weinberg Law by chi square analysis.
- CO5. **Compare** and analyze between survivorship curves of different types from the hypothetical/real data provided.

### Catalogue Description

This course covers laboratory and on field hands on techniques and study of ecology and evolution. The very nature of ecology and evolution lab requires students to learn, calculate and analyze of real different types from the hypothetical / data provided for the community and also learn to construct phylogenetic tree and draw them. All the lectures will be devoted on discussions of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as power point presentation, audio visual virtual lab session, on field excursion as per requirement. The tutorials will familiarize the students with practical problem-solving techniques led by the course coordinator. Students will strongly grab the basic concepts of the subject via exercise and discussions with the coordinator.

## Course Content

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### ELECTIVE ZOOLOGY LAB II

#### ECOLOGICAL PRINCIPLES, EVOLUTION AND BIODIVERSITY (PRACTICAL)

##### 10 Lecture Hours

1. Study of life tables and plotting of survivorship curves of different types from hypothetical/real data provided.

##### 10 Lecture Hours

2. Determination of population density in a natural/hypothetical community by quadrat method and calculation of Shannon-Weiner diversity index for the same community.

##### 10 Lecture Hours

3. Report on a visit to National Park/Biodiversity Park/Wild life sanctuary.

##### 10 Lecture Hours

4. Study of fossils from models/pictures.

##### 10 Lecture Hours

5. Study and verification of Hardy-Weinberg Law by chi square analysis.

##### 10 Lecture Hours

6. Construction of phylogenetic trees and its interpretation.

#### Reference Books

1. Barnes, R.D. (1992). Invertebrate Zoology. Saunders College Pub. USA.
2. Campbell & Reece (2005). Biology, Pearson Education, (Singapore) Pvt. Ltd.
3. Kardong, K. V. (2002). Vertebrates Comparative Anatomy. Function and Evolution. Tata McGraw Hill Publishing Company. New Delhi.
4. Ruppert, Fox and Barnes (2006) Invertebrate Zoology. A functional Evolutionary Approach 7th Edition, Thomson Books/Cole
5. Raven, P. H. and Johnson, G. B. (2004). Biology, 6th edition, Tata McGraw Hill Publications. New Delhi

#### Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

## Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and Pos		
	Course Outcomes (COs)	Mapped Program Outcomes
<b>CO1</b>	<b>Identify</b> and compare between fossil specimens and also learn how to construct phylogenetic tree and draw them.	<b>PO1, PO2, PO3, PO6</b>
<b>CO2</b>	<b>explain</b> the importance of field trip and they will get more interest on the subject as they learn how to work in the field trip effectively as an individual, and as a member or leader in diverse teams, and in multidisciplinary settings.	<b>PO1, PO3, PO4</b>
<b>CO3</b>	<b>demonstrate</b> , calculate and analyse Shanon-Weiner diversity index for the same community.	<b>PO1, PO3, PO5, PO9</b>
<b>CO4</b>	<b>illustrate</b> problems on Hardy-Weinberg Law by chi square analysis.	<b>PO1, PO5, PO6, PO9</b>
<b>CO5</b>	<b>compare</b> and analyze between survivorship curves of different types from the hypothetical/real data provided.	<b>PO1, PO5, PO6, PO7</b>

		Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking	Problem Identification Ability	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to Environment	Development to Society.	Development to Humanity
Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
<b>ZOL 12004</b>	ELECTIVE ZOOLOGY LAB II (PRACTICAL)	3	1	3	1	3	3	1	-	2	-	-	-

1=weakly mapped

2= moderately mapped

3=strongly mapped

## Model Question Paper

<b>Name:</b>  <b>Enrolment No:</b>	
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**Course: ZOL 12004 – ELECTIVE ZOOLOGY II LAB (PRACTICAL)**

**Program: B.Sc. Biochemistry**

**Time: 03Hrs.**

**Semester: Even 2020-21**

**Max. Marks: 40**

**Instructions:**

Attempt any **two** questions from **Section A** (each carrying 10 marks); **Section B** is Compulsory (carrying 10 marks).

### Section A ( Attemptany Two)

1.	Species	Characters									
		1	2	3	4	5	6	7			
	A	+	-	-	+	-	-	+	4 4 2	CO1	
	B	+	+	+	-	-	-	+			
	C	+	+	+	-	-	-	+			
	D	+	-	-	+	+	-	-			
	E	+	-	-	+	+	+	-			
	F	-	+	-	-	-	-	-			
	a) Draw the phenogram and cladogram using the above data table. (An/Ap)  b) <b>Find out</b> which species is the outgroup for both the cases.(U)										
2.	a) <b>Estimate</b> the pH of the given water sample. You have to calculate 3 times and write the average value in the answer sheet.(Ev) b) <b>Interpret</b> the nature of the sample.(An)									8 2	CO2
3.	a) <b>Identify</b> specimen A.(Ap) b) <b>What</b> is the evolutionary significance of the specimen. (R)									4 4	CO1
4.	a) <b>Construct</b> the survivorship graph using the following life table data for <i>Zootocaviviperain</i> the Netherlands.(Cr) b) <b>Interpret</b> which type of survivorship curve this population is following.(U)									7 3	CO5

	Age interval	Number alive at the start of the year		
	0	1000		
	1	763		
	2	308		
	3	158		
	4	57		
	5	10		
	6	7		
	7	2		
	8	0		
<b>SECTION B is compulsory</b>				
5.	<b>Viva-voce (U/An/Ap/R/Ev)</b>		<b>10</b>	<b>CO1 CO2 CO3 CO4 CO5</b>
6.	<b>Practical copy (U/Ap/Ev)</b>		<b>5</b>	<b>CO1</b>
	<b>Field trip diary (Ap/Ev/An)</b>		<b>5</b>	<b>CO2 CO3</b>

<b>PSG13004</b>	Applied Psychology	L	T	P	C
	Contact Hours=60	5	1	0	6
<b>Pre-requisites/Exposure</b>	10 + 2 knowledge of Biology				
<b>Co-requisites</b>	-				

### **Course Objectives:**

- To inform students about the basics of scientific research in applied psychology.
- To make them learn the application of psychology in day to day life.
- To acquaint students with various applications of Psychological principals.

### **Course Outcomes:**

CO1. **Discuss** the application of psychological principles in different areas of psychological research.

CO2. **Explain** the role of psychology in Organizations.

CO3. **Illustrate** the application of psychological principles in school and education.

CO4. **Determine** the role of psychology in regulating mental health.

CO5. **Define** the role of psychological principles in understanding social interactions.

CO6. **Evaluate** the stress level of students and job satisfaction of industry employees.

### **Catalogue description:**

**UNIT-I:** Introduction: Nature and fields of scientific research in applied psychology. Scientific discipline of applied psychology and challenges.

**UNIT-II:** Psychology in industries and organizations: Personnel selection; job analysis; fatigue and accidents; advertising and consumer behaviour.

**UNIT-III:** Psychology in education: School as an agent of socialization. Factors influencing school achievement. Exceptional children: Problems and remedial measures.

**UNIT-IV:** Psychology and mental health: Symptoms and causes of anxiety disorders and schizophrenia, mental retardation. Anti-social behaviour: Crime and delinquency.



**UNIT-V: Psychology and social behaviour: Prejudice and stereotypes; conflict and its resolution. Applying Social Psychology and Social Problems: Intervention and Evaluation with emphasis on Environmental Problems and Poverty in India.**

**UNIT-VI: Practicum**

1. Academic stress questionnaire. – Determine the Subject’s academic stress by administering Student Academic Stress Scale (SASS).
2. To assess the job satisfaction level of employees.

**Readings:**

1. An Introduction to Applied Cognitive Psychology. Psychology Press: New York. Sternberg, R. J. (Ed.) (2000).
2. Anastasi, A. (1979). Fields of applied psychology. New Delhi: McGraw Hill.
3. Applied Psychology: Perceiving Learning and Remembering. Australia: Cengage Learning.
4. Durso, F. T. (2007). Handbook of Applied Psychology (2nd Ed). New West Sussex : Wiley & Sons. Esgate, A. et al. (2005).
5. Goldstein, A. P., & Krasner, B. (1987). Modern applied psychology. Elmford, New York: Pergamon Press.
6. Handbook of intelligence. New York: Cambridge University Press. Sternberg, R. J. (2009).
7. McCormick, E. J., & Ilgen, D. (1980). Industrial psychology. Englewood Cliffs, N.J.: Prentice Hall.

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

	Course Outcomes (COs)	Mapped Program Outcomes
CO1	<b>discuss</b> the application of psychological principles in different areas of psychological research.	PO1, PO2, PO5
CO2	<b>understand</b> the role of psychology in Organizations.	PO1, PO2, PO4
CO3	<b>summarize</b> the application of psychological principles in school and education.	PO1, PO2, PO9
CO4	<b>determine</b> the role of psychology in regulating mental health.	PO1, PO2, PO4, PO5
CO5	<b>define</b> the role of psychological principles in understanding social interactions.	PO1, PO2, PO4
CO6	<b>evaluate</b> the stress level of students and job satisfaction of industry employees.	PO1, PO2, PO4, PO6, PO9, PO11

Course Code	Course Title	PO 1	PO 2	PO 3	PO 4	PO 5	PO 6	PO 7	PO 8	PO 9	PO1 0	PO1 1	PO1 2
		Fundamental Knowledge											
		Critical Thinking											
		Skill											
		Technical Knowledge											
		Logical Thinking											
		Problem Identification Ability											
		Analytical Knowledge											
		Career Goals											
		Team Work											
		Sustainable Development to Environment											
		Development to Society.											
		Development to Humanity											
<b>PSG13 004</b>	<b>Applied Psychology</b>	3	3	-	3	2	1	-	-	1	-	1	-

1=weakly mapped

2= moderately mapped

3=strongly mapped

**ADAMAS UNIVERSITY**  
**SCHOOL OF Liberal Arts & Culture studies**  
**END-SEMESTER EXAMINATION: DECEMBER 2020**

Name of the Program: B.SC.BIOCHEMISTRY

Semester: II

PAPER TITLE: Applied Psychology

PAPER CODE: PSG

13004

Maximum Marks: 40

Time duration: 3 hours

Total No of questions: 10

Total No of Pages: 01

**Instruction for the Candidate:**

1. At top sheet, clearly mention Name, Univ. Roll No., Enrolment No., Paper Name & Code, Date of Exam.
2. All parts of a Question should be answered consecutively. Each Answer should start from a fresh page.
3. Assumptions made if any, should be stated clearly at the beginning of your answer.

<b>Section A (Answer All the Questions) (5 x 1 = 5)</b>			
1.	<b>What</b> is Applied Psychology?	<b>R</b>	<b>CO1</b>
2.	<b>Define</b> Job Analysis.	<b>R</b>	<b>CO2</b>
3.	<b>What</b> is Mental Retardation?	<b>R</b>	<b>CO3</b>
4.	<b>Relate</b> prejudice with social conflicts.	<b>U</b>	<b>CO5</b>
5.	<b>What</b> is Hallucination?	<b>R</b>	<b>CO4</b>
<b>SECTION B (Attempt any Three Questions) (3 x 5 = 15)</b>			
6.	<b>Describe</b> the role of psychological principles in understanding social interactions.	<b>U</b>	<b>CO5</b>
7.	<b>Summarize</b> the different factors influencing school achievement.	<b>U</b>	<b>CO3</b>
8.	<b>Discuss</b> the role of psychological principles in Personnel Selection.	<b>Creating</b>	<b>CO2</b>
9.	<b>Determine</b> the strategies of conflict resolution in social interactions.	<b>Evaluate</b>	<b>CO5</b>
<b>SECTION C (Answer Any Two Questions) (2 x 10 = 20)</b>			
10.	<b>Determine</b> the Subject's academic stress by administering Student Academic Stress Scale (SASS).	<b>Evaluate</b>	<b>CO3/CO6</b>
11.	<b>Assess</b> the job satisfaction level of employees using a suitable scale.	<b>Evaluate</b>	<b>CO2/CO6</b>
12.	<b>Describe</b> the clinical features and etiology of Schizophrenia.	<b>U</b>	<b>CO4</b>

<b>BIC 11010</b>	<b>METABOLISM OF CARBOHYDRATES AND LIPIDS (THEORY)</b>	<b>L</b>	<b>T</b>	<b>P</b>	<b>C</b>
<b>Version 1.0</b>	<b>Contact Hours - 60</b>	3	1	0	4
<b>Pre-requisites/Exposure</b>	BASIC KNOWLEDGE OF CARBOHYDRATE AND LIPID STRUCTURE				
<b>Co-requisites</b>	Biomolecules				

### Course Objectives

1. To provide students the basic understanding of basic design of metabolism.
2. It will also provide in depth knowledge of different biochemical pathways of carbohydrates and their regulations.
3. To elaborate lipid metabolic pathways and their regulations.
4. To understand general overview of starve feed cycle and glucose homeostasis.

### Course Outcomes

On completion of this course, the students will be able to

- CO1. **Explain** basic design of metabolism.
- CO2. **Summarise** and analyse different metabolic pathways leading to carbohydrate metabolism.
- CO3. **Illustrate** and analyse different metabolic pathways leading to lipid metabolism.
- CO4. **Demonstrate** the role regulatory pathways.
- CO5. **Interpret** the starve feed cycle and glucose homeostasis.

### Catalogue Description

The core-course of 'metabolism of carbohydrate and lipid' will help to understand the basic design of metabolism. This course has been designed to gain a wholesome knowledge on metabolic pathways related to carbohydrates and lipids. Biochemical reactions leading to anabolism and catabolism of carbohydrates and lipids are covered in the syllabus. Regulations of these pathways have also been included for discussion. Classes will be conducted by lectures as well as powerpoint presentation, audio visual virtual lab session as per requirement. The tutorials will enable the students with problem-solving ability led by the course coordinator. Students will perceive the basic concepts of the subject via exercise and discussions with the coordinator.

## **Course Content**

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### **METABOLISM OF CARBOHYDRATES AND LIPIDS**

Unit 1: [6 lecture hours]

Basic design of metabolism: Autotrophs, heterotrophs, metabolic pathways, catabolism, anabolism, ATP as energy currency, reducing power of the cell.

Unit 2: [6 lecture hours]

Glycolysis: Glycolysis - a universal pathway, reactions of glycolysis, fermentation, fates of pyruvate, feeder pathways for glycolysis, galactosemia.

Unit 3: [6 lecture hours]

Gluconeogenesis and pentose phosphate pathway: Synthesis of glucose from non-carbohydrate sources, reciprocal regulation of glycolysis and gluconeogenesis, pentose phosphate pathway and its importance.

Unit 4: [6 lecture hours]

Glycogen metabolism: Glycogenesis and glycogenolysis, regulation of glycogen metabolism, glycogen storage diseases.

Unit 5: [6 lecture hours]

Citric acid cycle: Production of acetyl CoA, reactions of citric acid cycle, anaplerotic reactions, amphibolic role, regulation of citric acid cycle, glyoxalate pathway, coordinated regulation of glyoxalate and citric acid pathways.

Unit 6: [6 lecture hours]

Synthesis of carbohydrates: Calvin cycle, regulation of calvin cycle, regulated synthesis of starch and sucrose, photorespiration, C4 and CAM pathways, synthesis of cell wall polysaccharides, integration of carbohydrate metabolism in plant cell.

Unit 7: [6 lecture hours]

Fatty acid oxidation: Digestion, mobilisation and transport of cholesterol and triacyl glycerols, fatty acid transport to mitochondria,  $\beta$  oxidation of saturated, unsaturated, odd and even numbered and branched chain fatty acids, regulation of fatty acid oxidation, peroxisomal oxidation,  $\omega$  oxidation, ketone bodies metabolism, ketoacidosis.

Unit 8: [6 lecture hours]

Fatty acid synthesis: Fatty acid synthase complex. Synthesis of saturated, unsaturated, odd and even chain fatty acids and regulation.

Unit 9: [6 lecture hours]

Biosynthesis of eicosanoids, cholesterol, steroids and isoprenoids Synthesis of prostagladins, leukotrienes and thromboxanes. Synthesis of cholesterol, regulation of cholesterol synthesis. Synthesis of steroids and isoprenoids.

Unit 10: [4 lecture hours]

Biosynthesis of membrane lipids:Synthesis of membrane phospholipids in prokaryotes and eukaryotes, respiratory distress syndrome, biosynthesis of triacylglycerol, biosynthesis of plasmalogens, sphingolipids and glycolipids, lipid storage diseases.

Unit 11. [2 lecture hours]

Starve-feed cycle:Well-fed state, early fasting state, fasting state, early re-fed state, energy requirements, reserves and caloric homeostasis, five phases of glucose homeostasis.

**Textbook:**

1. Lehninger: Principles of Biochemistry (2013) 6<sup>th</sup>ed., Nelson, D.L. and Cox, M.M., W.H. Freeman and Company (New York), ISBN:13: 978-1-4641-0962-1 / ISBN:10:1-4292-3414-8.

**Reference books:**

1. Harpers Illustrated Biochemistry 30<sup>th</sup>edition,Victor W Rodwell, Kathleen M. Botham, PeterJ. Kennelly, P. Anthony Weil, David A. BenderMcGraw-Hill Education – Europe,ISBN: 9780071825344, 0071825347.

2. Biochemistry (2012) 7<sup>th</sup>ed., Berg, J.M., Tymoczko, J.L. and Stryer L., W.H. Freeman and Company (New York),ISBN:10:1-4292

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

### Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

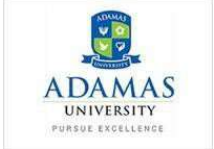
Mapping between COs and POs		
	Course Outcomes (COs)	Mapped Program Outcomes
<b>CO1</b>	<b>explain</b> the basic design of metabolism	<b>PO1, PO2, PO3</b>
<b>CO2</b>	<b>summarise</b> and analyse different metabolic pathways leading to carbohydrate metabolism.	<b>PO1, PO2, PO3, PO5, PO7</b>
<b>CO3</b>	<b>illustrate</b> and analyse different metabolic pathways leading to lipid metabolism.	<b>PO1, PO2, PO5</b>
<b>CO4</b>	<b>demonstrate</b> the role regulatory pathways.	<b>PO2, PO3, PO5, PO6</b>
<b>CO5</b>	<b>interpret</b> the starve feed cycle and glucose homeostasis.	<b>PO1, PO2, PO5, PO6</b>

		Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking	Problem Identification Ability	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to Environment	Development to Society.	Development to Humanity
Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
BIC11010	METABOLISM OF CARBOHYDRATES AND LIPIDS (THEORY)	3	3	2	-	3	2	1	-	-	-	-	-

1=weakly mapped

2= moderately mapped

3=strongly mapped

<b>Name:</b>			
<b>Enrolment No:</b>			
<b>Course: BIC 11010 - METABOLISM OF CARBOHYDRATES AND LIPIDS (THEORY)</b> <b>Program: B.Sc. Biochemistry</b> <b>Semester: Odd2019-20</b>			
		<b>Time: 03Hrs.</b> <b>Max. Marks: 40</b>	
<b>Instructions:</b> Attempt any <b>four</b> questions from <b>Section A</b> (each carrying 5 marks); any <b>two</b> questions from <b>Section B</b> (each carrying 10marks).			
<b>SECTION A ( Attemptany Four questions)</b>			
1.	<b>Explain</b> coupled reaction with a suitable example. (U) What are anabolism and catabolism? (R)	<b>2+3</b>	<b>CO1</b>
2.	<b>Compare</b> between glycolysis and gluconeogenesis? How many ATPs generated by one turn of citric acid cycle. (R)	<b>4+1</b>	<b>CO2</b>
3.	<b>Illustrate</b> the role of HMG CoA reductase in the cholesterol metabolism. What is the role of drug lovastatin? (R)	<b>3+2</b>	<b>CO3</b>
4.	<b>Enlist</b> 3 important enzymes in the citric acid cycle pathway and their mode of regulation. (R)	<b>3+2</b>	<b>CO3</b>
5.	<b>Explain</b> the role of cori cycle in the metabolic pathway? Why pentose PP pathway is also called hexose monophosphate shunt pathway? (U)	<b>3+2</b>	<b>CO1</b>
<b>SECTION B (Attempt any Two questions)</b>			
6.	<b>Illustrate</b> the role of carnitine in the beta oxidation pathway with a schematic diagram. How many turn and what amount of ATPs will be prepared by complete oxidation of a plmytoylcoA molecule. (R)	<b>5+2+3</b>	<b>CO3</b>
7.	What are good and bad cholesterol. <b>Explain</b> the role of cholesterol in atherosclerosis and bile salt formation. <b>Compare</b> energy production between glucose and fatty acid. (R)	<b>2+4+2+2</b>	<b>CO1</b> <b>CO2</b>
8.	<b>Describe</b> the glycogen degradation pathway. <b>Illustrate</b> the regulation of glycogen synthesis and breakdown in details. Briefly describe the formation of triacylglycerol. (U)	<b>2+5+3</b>	<b>CO1</b> <b>CO2</b>
9.	<b>Outline</b> the main features of starve feed cycle by mentioning the role of different hormones. <b>Assess</b> the coordination between glyoxalate cycle and citric acid cycle.	<b>2+3+3+2</b>	<b>CO4</b> <b>CO3</b>



<b>BIC12011</b>	<b>METABOLISM OF CARBOHYDRATES AND LIPIDS LAB (PRACTICAL)</b>	<b>L</b>	<b>T</b>	<b>P</b>	<b>C</b>
<b>Version 1.0</b>	<b>Contact Hours - 45</b>	0	0	3	2
<b>Pre-requisites/Exposure</b>	BASIC KNOWLEDGE OF CARBOHYDRATE AND LIPID BIOCHEMISTRY				
<b>Co-requisites</b>	Theory of Metabolism of Carbohydrate and Lipid.				

### Course Objectives

1. To provide students with hands-on training in the qualitative and quantitative aspects related to carbohydrate and lipid metabolism.
2. Students will need to become proficient with different assays and applications.

### Course Outcomes

On completion of this course, the students will be able to

- CO1. **Explain** the basic theories of carbohydrate and lipid estimation.
- CO2. **Illustrate** basic principles of fermentation and its application in industry.
- CO3. **Comprehend** the assay of salivary amylase.
- CO4. **Demonstrate** modern tools like spectrophotometers, colorimeters, micropipettes etc.
- CO5. **Develop** the knowledge of assay protocols and techniques in biological samples from everyday life.

### Catalogue Description

The core course “metabolism of carbohydrate and lipid lab” is a practical paper which has been designed to provide the knowledge of qualitative and quantitative aspects related to carbohydrate and lipid metabolism. It deals with all the basic principles related to carbohydrate and lipid estimation. Students will be able to understand the basic concepts of spectrophotometers and other techniques and will practice hands-on all of them. They will also learn to compare and use these methods for practical purpose. All the lectures will be devoted on discussions of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as power point presentation, audiovisual virtual lab session as per requirement. The tutorials will familiarize the students with practical problem-solving techniques led by the course coordinator. Students will strongly grab the basic concepts of the subject via exercise and discussions with the coordinator.

## Course Content

### METABOLISM OF CARBOHYDRATES AND LIPIDS LAB

1. Estimation of Blood glucose by glucometer. (5 Hours)
2. Determination of glucose by Benedict's solution.(10 Hours)
3. Determination of glucose by Fehling's method.(5 Hours)
4. Isolation of starch from potato.(5 Hours)
5. Determination of Lipase activity, salivary amylase activity. (5 Hours)
6. Study of Fermentation. (5 Hours)
7. Estimation of Sucrose using Benedict's solution.(10 Hours)
8. Estimation of acid number of oil.(5 Hours)

### SUGGESTED READINGS

1. Lehninger: Principles of Biochemistry (2013) 6<sup>th</sup>ed., Nelson, D.L. and Cox, M.M., W.H. Freeman and Company (New York), ISBN:13:978-1-4641-0962-1 /ISBN:10:1-4641-0962-1.
2. Textbook of Biochemistry with Clinical Correlations (2011) 7<sup>th</sup>ed., Devlin, T.M., John Wiley & Sons, Inc. (New Jersey), ISBN:978-0-470-28173-4.
3. Biochemistry (2012) 7<sup>th</sup>ed., Berg, J.M., Tymoczko, J.L. and Stryer L., W.H. Freeman and Company (New York), ISBN:10:1-4292-2936-5, ISBN:13:978-1-4292-2936-4.

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

### Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and Pos		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	<b>explain</b> the basic theories of carbohydrate and lipid estimation.	PO1, PO2, PO3
CO2	<b>illustrate</b> basic principles of fermentation and its application in industry.	PO1, PO10, PO11
CO3	<b>comprehend</b> the assay of salivary amylase.	PO1, PO2, PO3, PO4, PO9
CO4	<b>demonstrate</b> their concept modern tools like spectrophotometers, colorimeters, micropipettes etc.	PO1, PO3, PO6
CO5	<b>develop</b> the knowledge of assay protocols and techniques in biological samples from everyday life.	PO1, PO2, PO4

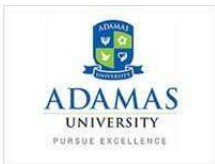
Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking	Problem Identification Ability	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to Environment	Development to Society.	Development to Humanity
<b>BIC12011</b>	<b>METABOLISM OF CARBOHYDRATE S AND LIPIDS LAB (PRACTICAL)</b>	3	2	3	2	-	1	-	1	1	1	1	-												

1=weakly mapped

2= moderately mapped

3=strongly mapped

## Model Question Paper

<b>Name:</b>			
<b>Enrolment No:</b>			
<b>Course: BIC12011 – METABOLISM OF CARBOHYDRATES AND LIPIDS LAB (PRACTICAL)</b>			
<b>Program: B.Sc. Biochemistry</b>		<b>Time: 03Hrs.</b>	
<b>Semester: Odd 2019-20</b>		<b>Max. Marks: 40</b>	
<b>Instructions:</b>			
Attempt any <b>two</b> questions from <b>Section A</b> (each carrying 10 marks); <b>Section B</b> is Compulsory (carrying 10 marks).			
<b>Section A ( Attemptany Two)</b>			
1.	a) Write the principle of fermentation. b) Determine the presence of amylase in human saliva with a simple experiment. (R)	4 6	CO1 CO2
2.	a) Explain the basic theory of GOD-POD method. (R) b) Determine the glucose level of the given serum with the GOD-POD method. (Ap)	4 6	CO3 CO4
3.	a) Write the principle behind ion-exchange chromatography. (R) b) Explain the CO <sub>2</sub> production in fermentation with a simple experiment. (An)	4 6	CO3 CO2
<b>SECTION B is compulsory</b>			
4.	<b>Viva-voce</b> (U/An/Ap/R/Ev)	<b>10</b>	CO1 CO2 CO3 CO4 CO5
5.	<b>Practical copy</b> (U/Ap/Ev)	<b>10</b>	CO1 CO2 CO3

<b>BIC11014</b>	<b>Metabolism of amino acids and Nucleotides (THEORY)</b>	<b>L</b>	<b>T</b>	<b>P</b>	<b>C</b>
<b>Version 1.0</b>	<b>Contact Hours – 60</b>	3	1	0	4
<b>Pre-requisites/Exposure</b>	Basic knowledge of Biomolecules and Enzymology of B.Sc.				
	Level				
<b>Co-requisites</b>	--				

### Course Objectives:

1. The main focus of the course is to cover the metabolic pathways regulating amino acid catabolism and anabolism.
2. It will provide a basic understanding about the biosynthesis and degradation of purines and pyrimidine.
3. The course will finally give an insight about the inter-connections between carbohydrates, proteins, lipids and nucleic acid metabolic pathways.

### Course outcome:

The students will be able to

**CO1. Explain** various biochemical changes that obey the basic thermodynamic principles of anabolism and catabolism.

**CO2. Comprehend** how the living organisms exchange energy and matter with the surroundings for their survival, and store free energy in the form of energy-rich compounds. **CO3. Identify** how the catabolic breakdown of the biomolecules is associated with release of free energy; whereas, free energy is utilized during synthesis of biomolecules i.e., anabolic pathways.

**CO4. Assess** the crucial role of some hormones with regard to the integration of metabolic pathways.

**CO5. Utilize** the knowledge of metabolic pathways to biotechnological and biochemical research.

### Catalogue Description:

The course is going to provide an understanding about the concept of anabolism and catabolism of amino acids and nucleotides and their role as high energy compounds in the cell. They will acquire knowledge related to regulation of various pathways. The course will teach about the importance of biosynthesis and degradation of amino acids and their regulation. The importance of high energy compounds, synthesis of ATP under aerobic and anaerobic conditions will be understood. The role of Urea cycle and its regulation in catabolism of amino acids and related in-born errors of amino acid metabolism as storage molecules will be taught. Further students will acquire the knowledge for biosynthesis and degradation of purines and

pyrimidines. The course will teach about the regulatory pathways of purine and pyrimidine biosynthesis along with biosynthesis of ribonucleotides, deoxyribonucleotides and polynucleotides. Finally, the student will gain insights into metabolic engineering for the production of useful biomolecules.

## Course Content

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### **METABOLISM OF AMINO ACIDS AND NUCLEOTIDES (THEORY)**

#### **Unit I 10**

##### **Lecture Hours**

**Overview of amino acid metabolism:** Nitrogen cycle, incorporation of ammonia into biomolecules. Metabolic fates of amino groups. Digestion and absorption of dietary proteins. Protein calorie malnutrition - Kwashiorkor and Marasmus. Nitrogen balance, transamination, role of pyridoxal phosphate, glucose-alanine cycle, Krebs's bicycle, urea cycle and inherited defects of urea cycle.

#### **Unit II 10**

##### **Lecture Hours**

**Catabolism of amino acids:** Catabolic pathways of individual amino acids. Glucogenic and ketogenic amino acids. Metabolism of one carbon units. Disorders of amino acid metabolism, phenylketonuria, alkaptonuria, maple syrup urine disease, methylmalonic acidemia (MMA), homocystinuria and Hartnup's disease.

#### **Unit III 5**

##### **Lecture Hours**

**Biosynthesis of amino acids:** Overview of amino acid synthesis. Biosynthesis of non-essential amino acids and its regulation.

#### **Unit IV 10 Lecture Hours**

**Precursor functions of amino acids:** Biosynthesis of creatine and creatinine, polyamines (putrescine, spermine, spermidine), catecholamines (dopamine, epinephrine, norepinephrine) and neurotransmitters (serotonin, GABA). Porphyrin biosynthesis, catabolism and disorders of porphyrin metabolism

#### **Unit V 5 Lecture Hours**

**Biosynthesis of purine and pyrimidine nucleotides:** *De novo* synthesis of purine and pyrimidine nucleotides, regulation and salvage pathways.

#### **Unit VI 5 Lecture Hours**

**Deoxyribonucleotides and synthesis of nucleotide triphosphate:** Biosynthesis of deoxyribonucleotides and its regulation, conversion to triphosphates, biosynthesis of coenzyme nucleotides.

#### **Unit VII 10 Lecture Hours**

**Degradation of purine and pyrimidine nucleotides:** Digestion of nucleic acids, degradation of purine and pyrimidine nucleotides. Inhibitors of nucleotide metabolism. Disorders of purine and pyrimidine metabolism—Lesch-Nyhan syndrome, Gout, SCID, adenosine deaminase deficiency.

#### **Unit VIII 5 Lecture Hours**

**Integration of metabolism:** Integration of metabolic pathways (carbohydrate, lipid and amino acid metabolic pathways), tissue specific metabolism (brain, muscle, and liver).

### Suggested Readings

1. Lehninger: Principles of Biochemistry (2013) 6th ed., Nelson, D.L. and Cox, M.M., WHFreeman and Company (New York), ISBN: 13:978-1-4641-0962-1 / ISBN:10:1-4641- 0962-1.
2. Molecular Cell Biology (2013) 7th ed., Lodish, H., Berk, A., Kaiser, C.A., Krieger, M, Bretscher, A., Ploegh, H., Amon, A. and Scott, M.P., W.H. Freeman & Company (New York), ISBN:13:978-1-4641-0981-2.
3. Biochemistry (2010) 4th ed., Garret, R. H. and Grisham, C.M., Cengage Learning (Boston), ISBN-13:978-0-495-11464-2.
4. Principles of Biochemistry (2008) 3rd ed., Voet, D.J., Voet, J.G. and Pratt, C.W., JmWiley & Sons, Inc. (New York), ISBN:13:978-0470-23396-2

### Modes of Examination:

Assignment/Quiz/Project/Presentation/Written Exam Examination

### Scheme:

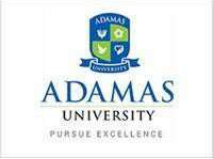
Components	Internal	Attendance	Mid Term	End Term
Weightage (%)	30	10	20	40

### Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and POs		
	Course Outcomes (COs)	Mapped Program Outcomes
CO-1	<b>understand</b> various biochemical changes that obey the basic thermodynamic principles of anabolism and catabolism.	PO1, PO2
CO-2	<b>comprehend</b> how the living organisms exchange energy and matter with the surroundings for their survival, and store free energy in the form of energy-rich compounds.	PO1, PO2, PO5
CO-3	<b>identify</b> how the catabolic breakdown of the biomolecules is associated with release of free energy; whereas, free energy is utilized during synthesis of biomolecules i.e., anabolic pathways.	PO3, PO4, PO2
CO-4	<b>assess</b> the crucial role of some hormones with regard to the integration of metabolic pathways.	PO2, PO4, PO6
CO-5	<b>utilize</b> the knowledge of metabolic pathways to biotechnological and biochemical research.	PO3, PO5, PO7, PO8, PO10

		Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking	Problem Identification Ability	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to	Development to Society.	Development to Humanity
Course Code	Course Title	PO 1	PO 2	PO 3	PO 4	PO 5	PO 6	PO 7	PO 8	PO 9	PO 10	PO 11	PO 12
<b>BIC 11014</b>	<b>Metabolism of amino acids and Nucleotides (THEORY)</b>	2	3	2	2	2	1	1	1	-	1	-	-

**Model  
Question  
Paper**

<b>Name:</b>			
<b>Enrolment No:</b>			
<b>Course: BIC11014 - Metabolism of amino acids and Nucleotides(THEORY)</b> <b>Program: B.Sc.Biochemistry</b> <b>Semester:Odd 2019-20</b>			
		<b>Time: 03Hrs.</b> <b>Max. Marks:40</b>	
<b>Instructions:</b>			
Attempt any <b>four</b> questions from <b>Section A</b> (each carrying 5 marks); any <b>two</b> questions from <b>Section B</b> (each carrying 10 marks).			
<b>SECTION A ( Attemptany Four questions)</b>			
1.	<b>Discuss</b> amine group donors in purine biosynthesis pathway og Inosine Monophosphate (IMP). <b>(Cr)</b>	<b>5</b>	<b>CO1</b>
2.	<b>Explain</b> the importance of hepatocyte cells for amino acid metabolism and Urea cycle.. <b>(Un)</b>	<b>5</b>	<b>CO2</b>
3.	<b>What</b> is the role of isozymes in regulation of biosynthesis pathway? <b>Explain</b> with an example. <b>(Re/Un)</b>	<b>5</b>	<b>CO2</b> <b>CO3</b>
4.	<b>What</b> is the difference between de novo biosynthesis and salvage pathway in metabolism? <b>Explain</b> with an example	<b>3+2</b>	<b>CO4</b>



<b>BIC12015</b>	<b>METABOLISM OF AMINO ACIDS AND NUCLEOTIDES LAB (PRACTICAL)</b>	<b>L</b>	<b>T</b>	<b>P</b>	<b>C</b>
<b>Version 1.0</b>	<b>Contact Hours - 45</b>	0	0	3	2
<b>Pre-requisites/Exposure</b>	BASIC KNOWLEDGE OF BIOCHEMISTRY				
<b>Co-requisites</b>	Theory of Metabolism of Amino acid and Nucleotides.				

### Course Objectives

1. To provide students with hands-on training in the qualitative and quantitative aspects of amino acids and nucleotides metabolism.
2. Students will need to become proficient with different assays and applications.

### Course Outcomes

On completion of this course, the students will be able to

- CO1. **Explain** the basic theories of amino acids and nucleotides metabolism.
- CO2. **Illustrate** basic principles and applications of SGPT-SGOT assay.
- CO3. **Comprehend** the assay of urea and creatinine in blood serum.
- CO4. **Demonstrate** modern tools like spectrophotometers, colorimeters, semi-auto analyser set etc.
- CO5. **Develop** the knowledge of assay protocols and techniques in clinical cases from everyday life.

### Catalogue Description

The core course "metabolism of amino acid and nucleotide lab" is a practical paper which has been designed to provide the knowledge of qualitative and quantitative aspects related to amino acid and nucleotide metabolism. It deals with all the basic principles related to the estimation of different parameters of blood like SGOT-PT, urea, creatinine etc. Students will be able to understand the basic concepts of spectrophotometers and other techniques and will practice hands-on all of them. They will also learn to compare and use these methods for practical purpose. All the lectures will be devoted on discussions of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as power point presentation, audio visual virtual lab session as per requirement. The tutorials will familiarize the students with practical problem-solving techniques led by the course coordinator. Students will strongly grab the basic concepts of the subject via exercise and discussions with the coordinator.

### Course Content

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#### METABOLISM OF AMINO ACID AND NUCLEOTIDE LAB

1. Estimation of SGPT & SGOT. (10 Hours)
2. Estimation of Urea. (10 Hours)
3. Estimation of Bilirubin & Biliverdin. (10 Hours)

4. Estimation of Creatinine. (5 Hours)
5. Estimation of Total protein & serum albumin.(10 Hours)
6. Estimation of ALP.(5 Hours)
7. Estimation of Creatine Kinase (CK).(5 Hours)
8. Estimation of Amylase.(5 Hours)

### SUGGESTED READINGS

1. Lehninger: Principles of Biochemistry (2013) 6<sup>th</sup>ed., Nelson, D.L. and Cox, M.M., W.H. Freeman and Company (New York), ISBN:13:978-1-4641-0962-1 /ISBN:10:1-4641-0962-1.
2. Textbook of Biochemistry with Clinical Correlations (2011) 7<sup>th</sup>ed., Devlin, T.M., John Wiley & Sons, Inc. (New Jersey),ISBN:978-0-470-28173-4.
3. Biochemistry (2012) 7<sup>th</sup>ed., Berg, J.M., Tymoczko, J.L. and Stryer L., W.H. Freeman and Company (New York), ISBN:10:1-4292-2936-5,ISBN:13:978-1-4292-2936-4.

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

### Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and Pos		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	<b>explain</b> the basic theories of amino acids and nucleotides metabolism.	PO1, PO2, PO3
CO2	<b>illustrate</b> basic principles and applications of SGPT-SGOT assay.	PO1, PO3, PO11
CO3	<b>comprehend</b> the assay of urea and creatinine in blood serum.	PO1, PO2, PO3, PO4, PO8
CO4	<b>demonstrate</b> their concept modern tools like spectrophotometers, colorimeters, semi-autoanalysers etc.	PO1, PO3, PO6, PO10
CO5	<b>develop</b> the knowledge of assay protocols and techniques in clinical cases from everyday life	PO1, PO4, PO9, PO11

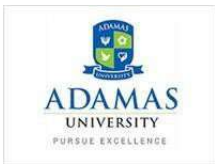
Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12		
		Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking	Problem Identification Ability	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to Environment	Development to Society.	Development to Humanity		
<b>BIC12015</b>	<b>METABOLISM OF AMINO ACID AND NUCLEOTIDE LAB (PRACTICAL)</b>	3	2	3	2	-	1	-	1	1	1	1	-		

1=weakly mapped

2= moderately mapped

3=strongly mapped

## Model Question Paper

<b>Name:</b>  <b>Enrolment No:</b>			
<b>Course: BIC12015 – METABOLISM OF AMINO ACID AND NUCLEOTIDE LAB (PRACTICAL)</b>			
<b>Program: B.Sc. Biochemistry</b>		<b>Time: 03 Hrs.</b>	<b>Max. Marks: 40</b>
<b>Semester: Odd 2019-20</b>			
<b>Instructions:</b> Attempt any <b>two</b> questions from <b>Section A</b> (each carrying 10 marks); <b>Section B</b> is Compulsory (carrying 10 marks).			
<b>Section A (Attempt any Two)</b>			
1.	a) Write the principle of urea estimation. (U) b) Estimate the content of uric acid in the given sample. (Ap)	<b>4</b> <b>6</b>	<b>CO1, CO2</b>
2.	a) Explain the basic theory of GOT method. (U) b) Determine the SGPT level of provided sample. (Ap)	<b>4</b> <b>6</b>	<b>CO3, CO4</b>
3.	a) Write the principle behind creatinine estimation. (U) b) Estimate the content of creatinine in the given sample. (Ap)	<b>4</b> <b>6</b>	<b>CO3, CO2</b>
<b>SECTION B is compulsory</b>			
4.	<b>Viva-voce (U/An/Ap/R/Ev)</b>	<b>10</b>	<b>CO1, CO2, CO3, CO4, CO5</b>
5.	<b>Practical copy (U/Ap/Ev)</b>	<b>10</b>	<b>CO1, CO2, CO3</b>



<b>BIC11016</b>	<b>TOOLS AND TECHNIQUES IN BIOCHEMISTRY</b>	<b>L</b>	<b>T</b>	<b>P</b>	<b>C</b>
<b>Version 1.0</b>	<b>Contact Hours: 30</b>	2	0	0	2
<b>Pre-requisites/Exposure</b>	Basic concepts of Biochemistry at B.Sc Level				
<b>Co-requisites</b>	--				

	of nucleic acid metabolism? <b>.(Re/Un)</b>		
5	<b>How</b> ornithine transcarbamoylase deficiency affects our metabolism? <b>(Re)</b>	<b>5</b>	<b>CO5</b>
	<b>SECTION B (Attempt any Two questions)</b>		
6.	(a) <b>Discuss</b> atleast two important reaction interconnecting carbohydrate and amino acid metabolism. (b) <b>Discuss</b> atleast two important reaction interconnecting carbohydrate and amino acid metabolism. <b>(Re)</b>	<b>5+5</b>	<b>CO1 CO2</b>
7.	<b>What</b> is the role of Carbomoyl phosphate synthetase I (CPS-I) and CPS-II in regulation of amino acid and pyrimidine metabolic pathways? <b>(Re)</b>	<b>10</b>	<b>CO1 CO2</b>
8.	<b>Which</b> important metabolic reaction catalysed by adenosine deaminase (ADA) and how it is responsible for severe combined immunodeficiency disease (SCID)? <b>(Re)</b>	<b>5+5</b>	<b>CO3 CO4</b>
9	<b>How</b> ribonucleotide reductase regulates the formation of pyrimidines? Which two enzymes regulates the activity of Ribonucleotide reductase and <b>how?</b> <b>(Re)</b>	<b>5+5</b>	<b>CO5</b>

### Course Objectives

The objective of this course is to introduce students to understand basics and conduct experiments in biochemistry. The course is designed to teach utility of experimental methods in biochemistry in a problem oriented manner. The basic laboratory safety would help the students to understand possible hazards during experiments.

### Course Outcomes

CO1. **develop** concepts of biochemistry with easy to run experiments.

On completion of this course, the students will be able to

CO1. **develop** concepts of biochemistry with easy to run experiments.

CO2. **experiment** with basic laboratory instruments and understand principle of measurements using those instruments with experiments in biochemistry.

CO3. **assess** various stock solutions and working solutions that will be needed for the course.

CO4. **utilize** spectrophotometer and conduct experiments.

CO5. **develop** knowledge about basic laboratory safety.

### **Catalog Description**

Students will use current biochemical techniques to plan and carry out experiments. They will learn basic laboratory safety before carry out an experiment. They will generate and test hypotheses, analyzed data using biochemical methods and appreciate the limitations of conclusions drawn from experimental data. Trouble-shooting will be stressed in classes and labs.

### **Course Content**

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#### **SEC I: TOOLS AND TECHNIQUES IN BIOCHEMISTRY**

Biochemical reagents and solutions: Safety practices in the laboratory. Preparation and storage of solutions. Concepts of solution concentration and storing solutions. Quantitative transfer of liquids. Concept of a buffer, Henderson-Hasselbach equation, working of a pH meter. [20 hrs]

Exercise: Preparation of a buffer of given pH and molarity. [5 hrs]

Spectrophotometric techniques: Principle and instrumentation of UV-visible and fluorescence spectroscopy. [15 hrs]

Exercises [20hrs]

i. Determination of the absorption maxima and molar extinction coefficient (of a relevant organic molecule).

ii. Measurement of fluorescence spectrum.

iii. Determination of concentration of a protein solution by Lowry/BCA

method. Introduction and importance of virtual labs in biochemistry

### **Reference Books**

1. Physical Biochemistry: Principles and Applications (2010) 2nd ed., Sheehan, D., Wiley Blackwell (West Sussex), ISBN:978-0-470-85602-4 / ISBN:978-0-470-85603-1.
2. Physical Biochemistry: Applications to Biochemistry and Molecular Biology (1982) 2nd ed, Freifelder, D., W.H. Freeman and Company (New York), ISBN:0-7167-1315-2 / ISBN:0-7167-1444-2.
3. An Introduction to Practical Biochemistry (1998) 3rd ed., Plummer D. T., Tata McGraw Hill Education Pvt. Ltd. (New Delhi), ISBN:13: 978-0-07-099487-4 / ISBN:10:0-07-099487-0.

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

**Relationship between the Course Outcomes (COs) and Program Outcomes (POs)**

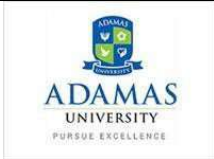
Mapping between COs and POs		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	Develop concepts of biochemistry with easy to run experiments.	PO1, PO4, PO5, PO7
CO2	Experiment with with basic laboratory instruments and understand principle of measurements using those instruments with experiments in biochemistry.	PO1, PO2, PO3, PO4, PO10
CO3	Assess various stock solutions and working solutions that will be needed for the course.	PO3, PO4, PO5, PO6, PO7
CO4	Utilize spectrophotometer and conduct experiments.	PO1, PO3, PO4, PO6, PO7, PO9
CO5	Develop knowledge about basic laboratory safety.	PO3, PO4, PO5, PO6

		Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking	Problem Identification Ability	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to Environment	Development to Society	Development to Humanity
Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
BIC11016	SEC I: TOOLS AND TECHNIQUES IN BIOCHEMISTRY	3	1	3	3	3	3	3	-	1	1	-	-



1=weakly mapped  
2= moderately mapped  
3=strongly mapped

## Model Question Paper

<b>Name:</b>  <b>Enrolment No:</b>	 <b>ADAMAS</b> UNIVERSITY <small>PURSUE EXCELLENCE</small>		
<b>Course: BIC11016 - SEC I: TOOLS AND TECHNIQUES IN BIOCHEMISTRY (THEORY)</b> <b>Program: B.Sc. Biochemistry</b> <b>Semester: Odd2020-21</b>			
<b>Time: 03Hrs.</b> <b>Max. Marks:40</b>			
<b>Instructions:</b> Attempt any <b>three</b> questions from <b>Section A</b> (each carrying 4 marks); any <b>four</b> questions from <b>Section B</b> (each carrying 7 marks).			
<b>SECTION A ( Attemptany Three questions)</b>			
1.	<b>Explain</b> different aspects of basic safety in a laboratory. <b>(Un)</b>	<b>4</b>	<b>CO5</b>
2.	<b>Explain</b> how water can behave both as an acid and base. <b>(Un)</b>	4	<b>CO2</b>
3.	<b>Explain</b> how would you prepare 500 mL of a 1 M solution of NaCl from a 2 M stock of NaCl? <b>(Ev)</b>	4	<b>CO3</b>
4.	<b>Determine</b> hydrogen ion concentration of a solution with a pH = 5.5? <b>(Ev)</b>	4	<b>CO3</b>
<b>SECTION B (Attempt any Four questions)</b>			
5.	<b>How</b> to handle cryogenics in the laboratory? <b>How</b> laboratory waste materials should be discarded? <b>(Re)</b>	<b>3+4</b>	<b>CO2</b> <b>CO5</b>
6.	<b>Illustrate</b> working principle of a spectrophotometer. <b>(Un)</b>	7	<b>CO4</b>
7.	<b>Explain</b> the concept of acid and its conjugate base using suitable examples. <b>(Un)</b>	7	<b>CO1</b>
8.	<b>Classify</b> different types of salts. Give two examples for each category. <b>(Un)</b>	<b>3+4</b>	<b>CO1</b>
9.	<b>Explain</b> how blood buffer system works. <b>Explain</b> how to reduce acidity in human. <b>(Un)</b>	<b>4+3</b>	<b>CO3</b>

<b>BIC11012</b>	<b>Cell Biology</b>	L	T	P	C
<b>Version 1.0</b>	Contact Hours: 60	3	1	0	4
<b>Pre-requisites/Exposure</b>	Class 12 level Biology knowledge				
<b>Co-requisites</b>	Basic Biology Knowledge of B.Sc 1 <sup>st</sup> Year				

### Course Objectives

1. To recall and extend the basic idea about the structure and function of cells.
2. To learn about different physiological characteristics of cells.
3. To think about the applications of microscopy as tools for understanding cell morphology.
4. To learn about cellular reproduction its relation to the key of life.
5. To utilize the knowledge of cell biology to search the causes of human disease.

### Course Outcomes

On completion of this course, the students will be able to

CO1: **develop** the basic idea of structure and function of prokaryotic and eukaryotic cells

CO2: **discuss** detailed perspective of cell including physiological properties, cell composition, growth, metabolic processes, signalling pathways, life cycle.

CO3: **interpret** the applications of different microscopy as tools for understanding cell biology.

CO4: **explain** about cell cycles, cell division and apoptosis.

CO5: **assess** the cell biology in microscopic and molecular level to understand of human health and disease.

### Catalog Description

This course deals with the biology of cells of higher organisms: The structure, function, and biosynthesis of cellular membranes and organelles; cell growth and oncogenic transformation; transport, receptors, and cell signaling; the cytoskeleton, the extracellular matrix, and cell movements; chromatin structure and RNA synthesis.

## Course Content

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### 1. Unit I Introduction to cell biology [5 Lecture Hours]

Prokaryotic (archaea and eubacteria) and eukaryotic cell (animal and plant cells), cells as experimental models.

### 2. Unit II Tools of cell biology [8 Lecture Hours]

Light microscopy, phase contrast microscopy, fluorescence microscopy, confocal microscopy, electron microscopy, FACS. Centrifugation for subcellular fractionation.

### 3. Unit III Structure of different cell organelles [7 Lecture Hours]

Structure of nuclear envelope, nuclear pore complex. ER structure. Organization of Golgi, Lysosome. Structure and functions of mitochondria, chloroplasts and peroxisomes. Zellweger syndrome.

### 4. Unit IV Protein trafficking [20 Lecture Hours]

Selective transport of proteins to and from the nucleus. Regulation of nuclear protein import and export. Targeting proteins to ER, smooth ER and lipid synthesis. Export of proteins and lipids from ER and into ER. Lipid and polysaccharide metabolism in Golgi. Protein sorting and export from Golgi. Mechanism of vesicular transport, cargo selection, coat proteins and vesicle budding, vesicle fusion. Protein import and mitochondrial assembly, protein export from mitochondrial matrix. Import and sorting of chloroplast proteins.

### 5. Unit V Cytoskeletal proteins [10 Lecture Hours]

Structure and organization of actin filaments. Treadmilling and role of ATP in microfilament polymerization, organization of actin filaments. Non-muscle myosin. Intermediate filament proteins, assembly and intracellular organization. Assembly, organization and movement of cilia and flagella.

### 6. Unit VI Cell wall and extracellular matrix [5 Lecture Hours]

Prokaryotic and eukaryotic cell wall, cell matrix proteins. Cell-matrix interactions and cell-cell interactions. Adherence junctions, tight junctions, gap junctions, desmosomes, hemidesmosomes, focal adhesions and plasmodesmata.

### 7. Unit VII Cell cycle, cell death and cell renewal [5 Lecture Hours]

Eukaryotic cell cycle, restriction point, and checkpoints. Cell division. Apoptosis and necrosis - brief outline. Salient features of a transformed cell.

## Reference Books



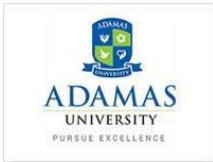
Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
<b>BIC11012</b>	Cell Biology	3	2	3	2	2	1	2	1	-	1	1	-

1=weakly mapped

2= moderately mapped

3=strongly mapped

## Model Question Paper

<b>Name:</b>  <b>Enrolment No:</b>			
<b>Course: BIC11012 – Cell Biology</b>			
<b>Program: B.SCBiochemistry</b> <b>Semester: Odd2019-20</b>	<b>Time: 03Hrs.</b> <b>Max. Marks:40</b>		
<b>Instructions:</b> Attempt any three questions from <b>Section A</b> (each carrying 4 marks); any <b>four Questions</b> from <b>Section B</b> (each carrying 7 marks).			
<b>Section A ( Attemptany Three)</b>			
1.	Pinocytosis refers to the uptake of membrane and solutes by the cell in small vesicles. True/False? <b>Justify. (Ev)</b>	4	CO1
2.	A "signal patch" is the receptor to which a "signal peptide" binds. True/False? <b>Justify. (Ev)</b>	4	CO2
3.	<b>What is apoptosis? (Re)</b>	4	CO5
4.	<b>Show that lipid molecules can move through the membranes? (Ap)</b>	4	CO3
<b>SECTION B (Attempt any fourQuestions)</b>			
5.	a) <b>Compare</b> and contrast the ‘vesicular transport’ and ‘cisternal maturation’ models of golgifunction. b) <b>Why</b> SNARES æ necessary for the processof “cotranslational transport”? <b>(Ap/Re)</b>	4+3	CO2
6.	a) <b>Where</b> Glycosylation is a form of protein covalent modifications that occurs in cell? Describe the process briefly <b>(U)</b>	1+3+3	

	b) Signal peptides are permanent structural components of a protein how can you use it for protein purification?(Ap)		CO4 CO2 CO3
7.	a) <b>Why</b> sulfation required for cell? b) <b>What</b> do you know about membrane	2+3+2	CO2



	liquidity? c) <b>What</b> is the structural role of ER in cell <b>(Re)</b>		
8.	a) <b>Explain</b> the connection between single membrane bound organelles in cell? b) <b>Distinguish</b> between primary and secondary lysosome. <b>(Un/An)</b>	5+2	CO5
9.	a) <b>What</b> are the hall marks of cancer? b) <b>How</b> can you prove velocity gradient centrifugation is inferior to density gradient centrifugation. <b>(Re)</b>	3+4	CO2

<b>BIC12013</b>	Cell Biology Lab	L	T	P	C
<b>Version 1.0</b>	Contact Hours:45	0	0	3	2
<b>Pre-requisites/Exposure</b>	Knowledge of Cell Biology Theory Paper				

## Course Objectives

1. To study the cells from animals and plants.
2. To determine micrographs of different cell components.
3. To discuss the interactions between cellular components.
4. To apply the knowledge of cell biology to select examples of changes or losses in cell function and compare and contrast the events of cell cycle and its regulation.

## Course Outcomes

On completion of this course, the students will be able to

CO1: **Examine** animal and plant cells.

CO2: **Demonstrate** micrographs of different cell components.

CO3: **Illustrate** the interactions of the cells and how cellular components are used to generate and utilize energy in cells.

CO4: **Apply** their knowledge of cell biology to select examples of changes or losses in cell function and compare and contrast the events of cell cycle and its regulation.

CO5: **Appraise** the laws of heredity with practical emphasis on inheritance.

## Catalog Description

Course provides hand-on training in current cell biological methods. A discovery-based component of this course focuses on application of the mentioned methods to study changes in protein expression and cytoskeleton organization in cells exposed to microenvironmental stress stimuli. .

## Course Content

1. Microscope Handling (5 Hours)
2. Plasmolysis and DePlasmolysis of Plant cell (5 Hours)
3. Plasmolysis and DePlasmolysis of animal cell (5 Hours)
4. Observation and Identification of water Eukaryotes in Microscope (10 Hours)
5. Cell counting through Haemocytometer (5 Hours)
6. LeishMann staining to identify and count different WBCs (10 Hours)
7. Giemsa staining to identify and count different WBCs (5 Hours)

8. Identification of electron micograph pictures (5 Hours)
9. Cell fractionation using velocity gradient centrifugation (5 Hours)

### Reference Books

1. The Cell: A Molecular Approach (2009) 5th ed., Cooper, G.M. and Hausman, R.E., ASM Press & Sunderland (Washington DC), Sinauer Associates, MA, ISBN:978-0-87893-300-6.
2. Molecular Cell Biology (2012) 7th ed., Lodish, H., Berk, A., Zipursky, S.L., Matsudaira, P., Baltimore, D. and Darnell, J., W.H. Freeman & Company (New York), ISBN:13:978-1-4641-0981-2 / ISBN:10:1-4641-0981-8.

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

### Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

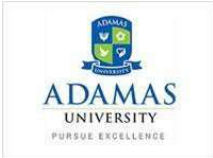
Mapping between COs and Pos		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	Examine animal and plant cells	PO1, PO2, PO4
CO2	Demonstrate micrographs of different cell components	PO1, PO2, PO3, PO4
CO3	Illustrate the interactions of the cells and how cellular components are used to generate and utilize energy in cells	PO3, PO4, PO5, PO6, PO7, PO8
CO4	Utilize their knowledge of cell biology to select examples of changes or losses in cell function and compare and contrast the events of cell cycle and its regulation.	PO7, PO10, PO11, PO12
CO5	Appraise the laws of heredity with practical emphasis on inheritance.	PO1, PO6, PO8, PO9

Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
		Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking Ability	Problem Identification	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to Environment	Development to Society	. Development to Humanity
<b>BIC11012</b>	Cell Biology Lab	2	2	3	3	2	2	2	1	1	1	1	1

1=weakly mapped

2= moderately mapped

3=strongly mapped

<b>Name:</b> <b>Enrolment No:</b>			
<b>Course: BIC11012 – Cell Biology Lab</b>			
<b>Program: B.SC Biochemistry</b> <b>Semester: Odd2019-20</b>		<b>Time: 03Hrs.</b> <b>Max. Marks:40</b>	
<b>Instructions:</b> Attempt all questions			
1	<b>Identify</b> the monocyte from blood smear using Gimseastain. (A p)	10	CO3
2	<b>Evaluate</b> the number of nuclei in the sample with respect to the number of WBC in per ml blood. (Ev)	10	CO3
3	Note book. (Re/Un/Ap/Ev/An)	10	CO1, CO2, CO3, CO4, CO5
4	Viva Voce. (Re/Un/Ap/Ev/An)	10	CO1, CO2, CO3, CO4, CO5

<b>BIC 11017</b>	<b>Clinical Biochemistry (THEORY)</b>	<b>L</b>	<b>T</b>	<b>P</b>	<b>C</b>
<b>Version 1.0</b>	<b>Contact Hours - 60</b>	2	0	0	2
<b>Pre-requisites/Exposure</b>	PLUS TWO (12 <sup>TH</sup> ) LEVEL BIOLOGY AND BASIC KNOWLEDGE ABOUT HUMAN PHYSIOLOGY AT B.SC LEVEL				
<b>Co-requisites</b>	--				

### Course Objectives

1. To provide students basic idea about instrumentation and automation in clinical biochemistry laboratories safety regulations.
2. It will also provide in depth knowledge about different biochemical reactions that are used to determine different disease parameters.
3. Outlining the types of specimen for biochemical analysis.
4. To provide students different parameters like precision, accuracy, quality control, precautions and limitations that are used in clinical biochemistry.

### Course Outcomes

On completion of this course, the students will be able to:

**CO1. Define** the fundamental biochemistry knowledge related to health.

**CO2. Explain** the clinical significance of the laboratory tests.

**CO3. Evaluate** different biochemical changes in different disease condition.

**CO4. Evaluate** the clinical biochemistry tests with their results.

**CO5. Develop** awareness of different lifestyle diseases increasingly found in present day.

### Catalog Description

The core-course of 'Clinical Biochemistry' will help to understand the basic idea about instrumentation and automation in clinical biochemistry. This course includes comprehensive approach through studying different biochemical reactions that are used to determine different disease parameters. Furthermore, the implication of precision, accuracy, quality control, precautions and limitations in different test results will also be illuminated. All the lectures will be devoted on discussions of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as powerpoint

presentation, audio visual virtual lab session as per requirement. The tutorials will enable the students with problem-solving ability led by the course coordinator. Students will perceive the basic concepts of the subject via exercise and discussions with the coordinator.

## **Course Content**

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### **SEC-3: CLINICAL BIOCHEMISTRY**

**Introduction:** Organization of clinical laboratory, Introduction to instrumentation and automation in clinical biochemistry laboratories safety regulations and first aid. General comments on specimen collection, types of specimen for biochemical analysis. Precision, accuracy, quality control, precautions and limitations.

#### **Exercises**

- i. Collection of blood and storage.
- ii. Separation and storage of serum. **[20 Lecture Hours]**

**Evaluation of biochemical changes in diseases:** Basic hepatic, renal and cardiovascular physiology. Biochemical symptoms associated with disease and their evaluation. Diagnostic biochemical profile.

**Assessment of glucose metabolism in blood:** Clinical significance of variations in blood glucose. Diabetes mellitus.

**Exercises:** Estimation of blood glucose by glucose oxidase peroxidase method. **[20 Lecture Hours]**

**Lipid profile:** Composition and functions of lipoproteins. Clinical significance of elevated lipoprotein.

**Exercises:** Estimation of triglycerides.

**Liver function tests Exercises:** Estimation of bilirubin (direct and indirect).

**Renal function tests and urine analysis:** Use of urine strip / dipstick method for urine analysis.

**Exercises:** Quantitative determination of serum creatinine and urea.

**Tests for cardiovascular diseases:** Involvement of enzymes in diagnostics of heart disease including aspartate transaminase, isoenzymes of creatine kinase and lactate dehydrogenase and troponin.

**Exercises:** Estimation of creatine kinase MB. **[20 Lecture Hours]**

## **SUGGESTED READINGS**

1. Medical Laboratory Technology - a Procedure Manual for Routine Diagnostic Tests Vol.I(2010), Mukherjee, K.L., Tata McGraw-Hill Publishing Company Limited (New Delhi). ISBN:9780070076594 / ISBN:9780070076631
2. Medical Laboratory Technology - a Procedure Manual for Routine Diagnostic Tests Vol.II(2010), Mukherjee, K.L., Tata McGraw-Hill Publishing Company Ltd. (New Delhi), ISBN:9780070076648.
3. Medical Biochemistry (2005) 2<sup>nd</sup> ed., Baynes, J.W. and Dominiczak, M.H., Elsevier Mosby Ltd. (Philadelphia), ISBN:0-7234-3341-0.
4. Experimental Biochemistry: A Student Companion (2005) Rao, B.S. and Deshpande, V., IK International Pvt. Ltd. (New Delhi), ISBN:81-88237-41-8.

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination  
Examination Scheme:**

<b>Components</b>	<b>Mid Term</b>	<b>Attendance</b>	<b>Class Assessment</b>	<b>End Term</b>
<b>Weightage (%)</b>	<b>20</b>	<b>10</b>	<b>30</b>	<b>40</b>

**Relationship between the Course Outcomes (COs) and Program Outcomes (POs)**

<b>Mapping between COs and POs</b>		
	<b>Course Outcomes (COs)</b>	<b>Mapped Program Outcomes</b>
<b>CO1</b>	<b>Define</b> the fundamental biochemistry knowledge related to health.	<b>PO1, PO7</b>
<b>CO2</b>	<b>Explain</b> the clinical significance of the laboratory tests	<b>PO1, PO2</b>
<b>CO3</b>	<b>Evaluate</b> different biochemical changes in different disease condition	<b>PO1, PO2, PO3, PO7</b>
<b>CO4</b>	<b>Relate</b> the clinical biochemistry tests with their results	<b>PO1, PO2, PO4, PO7</b>
<b>CO5</b>	<b>Develop</b> awareness of different lifestyle diseases increasingly found in present day	<b>PO3, PO5, PO8, PO11</b>

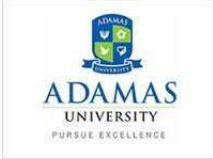


Course Code	Course Title	PO 1	PO 2	PO 3	PO 4	PO 5	PO 6	PO 7	PO 8	PO 9	PO1 0	PO1 1	PO1 2
<b>BIC 11017</b>	Clinical Biochemistry (THEORY)	Fundamental Knowledge	3	3	1	2	2	-	3	1	-	1	-
		Critical Thinking	3	3	1	2	2	-	3	1	-	1	-
		Skill	3	3	1	2	2	-	3	1	-	1	-
		Technical Knowledge	3	3	1	2	2	-	3	1	-	1	-
		Logical Thinking Ability	3	3	1	2	2	-	3	1	-	1	-
		Problem Identification	3	3	1	2	2	-	3	1	-	1	-
		Analytical Knowledge	3	3	1	2	2	-	3	1	-	1	-
		Career Goals	3	3	1	2	2	-	3	1	-	1	-
		Team Work	3	3	1	2	2	-	3	1	-	1	-
		Sustainable Development to Environment	3	3	1	2	2	-	3	1	-	1	-
		Development to Society	3	3	1	2	2	-	3	1	-	1	-
		Development to Humanity	3	3	1	2	2	-	3	1	-	1	-

1=weakly mapped

2= moderately mapped

3=strongly mapped

<b>Name:</b>		 <b>ADAMAS</b> UNIVERSITY <small>PURSUE EXCELLENCE</small>	
<b>Enrolment No:</b>			
<b>Course: BIC 11017–Clinical Biochemistry (THEORY)</b>		<b>Time: 03Hrs.</b>	
<b>Program: B.Sc. Biochemistry</b>		<b>Max. Marks: 40</b>	
<b>Semester: Odd 2019-20</b>			
<b>Instructions:</b>			
Attempt any <b>four</b> questions from <b>Section A</b> (each carrying 5 marks); any <b>two</b> questions from <b>Section B</b> (each carrying 10 marks).			
<b>SECTION A (Attempt any Four questions)</b>			
1.	<b>What</b> is Diabetes mellites? <b>Discuss</b> the cause of this disease.(Re/Un)	<b>2+3</b>	<b>CO1</b>
2.	<b>How</b> can you measure blood glucose level? <b>What</b> is the normal range of blood sugar level? (Ap/Re)	<b>4+1</b>	<b>CO2</b>
3.	<b>What</b> is glycemic index? <b>Explain</b> various mechanisms for regulation of blood glucose.(Re/Un)	<b>2+3</b>	<b>CO3</b>
4.	<b>What</b> is liver function test? <b>Explain</b> your answer..(Re/Un)	<b>2+3</b>	<b>CO4</b>
5	<b>Explain</b> the cause and consequence of fatty liver disease.(Un)	<b>5</b>	<b>CO5</b>
<b>SECTION B (Attempt any Two questions)</b>			
6.	<b>Demonstrate</b> the importance of automation in a clinical laboratory.(Un)	<b>4+1+3+2</b>	<b>CO2</b>
7.	Do you think measuring blood sugar level in fasting condition is a true representation of sugar level for a diabetic patient? <b>Explain</b> . If not, what will be better technique to monitor blood sugar level for a diabetic patient? <b>Why?</b> (Un/An)	<b>2+4+2+2</b>	<b>CO1</b> <b>CO2</b>
8.	<b>Outline</b> the laboratory investigation of kidney disease. <b>Discuss</b> the potential pitfalls and how can you overcome the pitfalls.(Un/An)	<b>4+3+3</b>	<b>CO1</b> <b>CO2</b>
9.	<b>What</b> is Atherosclerosis? <b>What</b> are the risk factors for coronary artery disease? <b>What</b> is the link between smoking and heart disease? <b>Explain</b> your answer.(Un/Re)	<b>2+2+3+3</b>	<b>CO4</b> <b>CO5</b>

<b>SOC 14100</b>	<b>Community Service</b>	L	T	P	C
<b>Version 1.0</b>		-	-	-	1
<b>Pre-requisites/Exposure</b>	Knowledge of Basic English				
<b>Co-requisites</b>	Knowledge of Basic Computer Skills				

### Course Objectives

- To familiarise the students on the concept‘giving back to the society’.
- To familiarize the students on the issues faced by marginalized communities.
- To provide an experiential platform to the students on any one or two issues as an internship.

### Course Outcomes

On completion of this course, the students will be able to

CO1: Explain the concept of social responsibility through an internship.

CO2: Discuss about ‘giving back to the society’ through the concept of social responsibility through an internship.

### Catalog Description

Along with Intelligent Quotient, it is important for students to enhance their Emotional Quotient as well. The Social Internship offers opportunity to the student to be empathetic towards social issues facing our society. To help and support the affected community / cause through a field internship is the essence of the course in ‘giving back to the society’.

### Course Content

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#### Unit I:

Introduction to the course. A brief on social issues facing the society with both global and Indian examples.

#### Unit II:

Minimum 24 hours of field work on a social issue and helping the marginalized / affected community / cause with photographs and testimonies.

#### Unit III:

Submission of individual reflection on the social service rendered.

The benefits that accrue to the students are

#### A.) Subjective

1. Psychosomatic benefits: Volunteering increases overall life satisfaction and also helps to relieve stress and acts as an anti-depressant.
2. Intellectual benefits: Enhances knowledge through new experiences, and develops communication skills.
3. Career benefits : Enhances career prospects by acquisition of work-related skills, builds good references for employers and provides a forum to network with future potential

employers. It also The experience allows gained helps students to take up leadership positions. Letters of recommendation can also be easily sought. Research shows that students who indulge in volunteer work perform better in studies as it invigorates their passion for learning

4. Personal benefits :Real world skills like leadership, problem-solving, collaboration with others, time management and communication skills, learn patience and empathy.
5. Connect learning to real world and enables deeper and lifelong learning.

## B.)Community

1. Collective benefits: Strong interpersonal bonds are created, and leads to increased civic and social awareness and responsibility.

## Further Reading :

1. Tadevosyan, Gohar&Schoenhuth, Michael. Participatory Research Approach :Principles, Challenges and Perspectives.[http://ysu.am/files/01G\\_Tadevosyan\\_M\\_Schoenhuth.pdf](http://ysu.am/files/01G_Tadevosyan_M_Schoenhuth.pdf)
2. Bergold, Jarg& Thomas Stefan. Participatory Research Methods: A Methodological Approach in Motion <http://www.qualitative-research.net/index.php/fqs/article/view/1801/3334>

## Plan of Work

1. Reading on social issues facing the society with both global and Indian examples.
2. Selecting an issue where the student wishes to contribute and wants to make a difference.
3. Areas - The internship may be broadly completed by getting in touch with NGO in your city / town / Police / Municipal Corporation / Local Gram Panchayat / Hospital / State Health Department / Women & Child Development Centre / CSR departments of Corporates / school / Old Age Home / Orphanage / Literacy Drive / Aanganwadi Centres / etc.
4. **Online Discussion** – Through discussion, students elaborate their preferred area of work with reference to the Global Scenario and India. Reason for choosing that area also needs and resources of the people in their area of Social Internship and also submit the testimonials, which include signature of the authority where students initiated their work, or the signature of the authority in whose area students are currently working or photographs of work (photographs must include students working).
5. **Final Report Submission** - Submission of the Testimonials include signatures of the authorities you have worked with, or the signature of the authority in whose area you have worked or photographs of your work (photographs must include you working). Students' accomplishment in their area of operation along with the major successes student experienced and major challenges faced.
6. Students will submit the complete elaborated report along with testimonials and completion certificate in the form of signed Template
  - The registration for all students will open twice, during winter and summer breaks. They may enroll for the internship in either of the two breaks.
  - The student will have to submit a continuous record of their 10 to 15 days internship in the form of photographs and testimonies (wherever required).

## Mode and Scheme of Online Evaluation:

Modes of Evaluation: Online – Quiz / Assignment / Discussions / Case Studies

Examination Scheme:

<b>Components</b>	<b>Attendance</b>	<b>Internal Assessment (Discussion+Initiating Internship Template) MTE</b>	<b>ETE (Detailed Report Submission + Testimonials Photographs / Student Experience Sharing Video)</b>
<b>Weightage (%)</b>	<b>10</b>	<b>30</b>	<b>60</b>

### Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and POs		
	Course Outcomes (COs)	Mapped Program Outcomes
<b>CO1</b>	<b>Explain</b> the concept of social responsibility through an internship.	<b>PO6, PO9, PO12</b>
<b>CO2</b>	<b>Discuss</b> about ‘giving back to the society’ through the concept of social responsibility through an internship.	<b>PO6, PO9, PO12</b>

		Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking	Problem Identification Ability	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to Environment	Development to Society.	Development to Humanity
Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
SOC14100	Community Service						3			3			3

1=weakly mapped

2= moderately mapped

3=strongly mapped

<b>EIC11001</b>	<b>Venture Ideation</b>	L	T	P	C
<b>Version 2.0</b>	Contact Hours: 30	2	0	0	2
<b>Pre-requisites/Exposure</b>	Basic knowledge of English and computer applications such as Internet Explorer and MS Office				
<b>Co-requisites</b>	--				

**Course Objectives**

1. To help the students understand the way to be an Entrepreneur
2. To identify the right business opportunity
3. To empower students to perform a technical feasibility study and thereby developing prototype
4. To help students in identifying their customers using primary and secondary research methods.
5. Expose students to various factors of market and competition with the help of market feasibility study, forecasting techniques, business model canvas and insights about financial statements.
6. To prepare students with finalizing their entrepreneurial Portfolio

**Course Outcomes**

On completion of this course, the students will be able to:

- CO1. Assess personal capacity in the context of the entrepreneurial process
- CO2. Compare characteristics of successful entrepreneurs and entrepreneurial forms and processes
- CO3. Construct resources, research and tools for Entrepreneurial ventures
- CO4. Examine opportunity identification techniques, feasibility terminology, processes and models
- CO5. Develop Ideation and planning documents for entrepreneurial venture

**Catalogue Description**

Over the last decade, the core of our economy has been transitioning from one of industrial might, large monolithic corporations and mass production towards one of networks, flexible enterprises comprising many smaller units and unique value. This new economy is based on innovation originating in creativity and design; it is also disrupting long-standing and established employment patterns and bringing to the fore the importance of entrepreneurship. This core unit will bring together creativity, design and entrepreneurship at the conceptual and more practical level. It aims to explore the nature, determinants and consequences of creativity, design and entrepreneurship as well as the interaction between them.

**Course Content**

**Unit 1. Introduction**

**6 hours**

Preview of the Course, Introduction to the Course, Guest Lecture with U.S. Secretary of Commerce Penny Pritzker – Meaning of Innovation, Entrepreneurial opportunities, Factors influencing the feasibility of an innovation, Innovation strategy: technology-push or market-pull, Product-market fit, How to develop a business model, Walkthrough of the business model canvas, Welcome to Innovation for Entrepreneurs: From Idea to Marketplace.

**Unit 2. Customer Discovery and Validation**

**6 hours**

Customer types, Customer archetypes, Customer segments and business models, Customer segments, value propositions, product features, value mapping, interviewing customer, insights of your customers.

**Unit 3: Product Understanding and Marketing. 9 hours**

Customer value, The DNA of customer-centricity, Crossing the chasm, Qualitative and quantitative marketing research, importance and methods of market segmentation, Focusing on the target market, Beyond the chasm, Strategic implications of beyond the chasm, E-commerce: The internet as a selling platform.

**Unit 4. Prototyping and Testing. 9 hours**

Planning for prototyping, Rapid prototyping and development, Lean startup MVPs, Choosing a wire framing/UX prototyping tool, Anatomy of an experience map, What you'll learn from user testing, Analytics and insight, Troubleshooting your customer discovery, Levels of a product/service.

**Modes of Evaluation: Quiz/Assignment/presentation/extempore/Written Examination**

Components	MTE	Attendance	Presentation/Assignment/ etc	ETE
Weightage (%)	20	10	20	40

**Relationship between the Program Outcomes (POs), Program Specific Outcomes (PSOs) and Course Outcomes (COs)**

Mapping between COs and POs		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	<b>Assess</b> personal capacity in the context of the entrepreneurial process	PO6, PO11
CO2	<b>Compare</b> characteristics of successful entrepreneurs and entrepreneurial forms and processes	PO6, PO11
CO3	<b>Construct</b> resources, research and tools for Entrepreneurial ventures	PO6, PO8, PO11
CO4	<b>Examine</b> opportunity identification techniques, feasibility terminology, processes and models	PO6, PO8, PO11
CO5	<b>Develop</b> Ideation and planning documents for entrepreneurial venture	PO6, PO8, PO11

1=Weakly mapped

2=Moderately mapped

3=Strongly mapped

EIC11001	Course Code	
	Course Title	
	PO1	Engineering Knowledge
	PO2	Problem analysis
	PO3	Design/development of solutions
	PO4	Conduct investigations of complex problems
	PO5	Modern tool usage
3	PO6	The engineer and society
	PO7	Environment and sustainability
3	PO8	Ethics
	PO9	Individual and team work
	PO10	Communication
3	PO11	Project management and finance
	PO12	Life-long Learning
	PSO1	
	PSO2	





## Semester 4

<b>BIC11018</b>	<b>Gene organization, replication and repair</b>	L	T	P	C
<b>Version 1.0</b>	Contact Hours: 60	3	1	0	4
<b>Pre-requisites/Exposure</b>	Knowledge of Theory of Biomolecules at B.Sc Level				
<b>Co-requisites</b>	Class 12 level Biology knowledge -				

### Course Objectives

1. To acquire the concepts of the basic principles of heredity.
2. To understand causal relationships between molecule-cell level phenomena
3. To analyse the molecular consequences about linkage, crossing over and mapping techniques
4. To conceptualize molecular basis of DNA damage
5. To learn about gene repair mechanisms in prokaryotes and eukaryotes.

### Course Outcomes

On completion of this course, the students will be able to

- CO1** build the concepts of the basic principles of inheritance at the molecular, cellular and organismal levels.
- CO2** formulate causal relationships between molecule-cell level phenomena (modern genetics) and organism-level patterns of heredity (classical genetics).
- CO3** discuss linkage, crossing over and mapping techniques.
- CO4** interpret molecular basis of mutations.
- CO5** discuss the inheritance of complex traits & population genetics and evolutionary genetics.

### Catalog Description

DNA is replicated prior to cell division. In this process, DNA polymerase uses an original strand as a template to create a new daughter strand of DNA. Polymerization occurs in the 5' to 3' direction, creating a new strand that is anti-parallel to the original. Polymerization begins at a specific DNA sequence called the Origin of Replication and proceeds in two directions along both template strands. Because of the directionality of synthesis, one strand is synthesized continuously, while the other strand is synthesized in small fragments, which are then joined together by DNA ligase.

### Course Content

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#### 1. Unit I Structure of DNA [15 Lecture Hour]

DNA structure, features of the double helix, various forms of DNA, denaturation and reassociation of DNA.

#### 2. Unit II Genes and genomic organization [10 Lecture Hour]

Genome sequence and chromosome diversity, definition of a gene, organization of genes in

viruses, bacteria, animals and plants. Nucleosome structure and packaging of DNA into higher order structures.

### 3. Unit III Replication of DNA [30 Lecture Hour]

The chemistry of DNA synthesis, DNA polymerase, the replication fork, origin of replication, enzymes and proteins in DNA replication, various modes of replication, stages of replication of E. coli chromosome, relationship between replication and cell division, replication in eukaryotes. Comparison of replication in prokaryotes and eukaryotes. Inhibitors of DNA replication and applications in medicine. Supercoiling of DNA.

### Unit IV Repair of DNA [5 Lecture Hour]

## Reference Books

1. Molecular Biology of the Gene (2008) 6th ed., Watson, J.D., Baker, T.A., Bel S.P., Gann, A., Levine, M. and Losick, R., Cold Spring Harbor Laboratory Press, Cold spring Harbor (New York), ISBN:0-321-50781 / ISBN:978-0-321-50781-5.
2. Lehninger: Principles of Biochemistry (2013) 6th ed., Nelson, D.L. and Cox, M.M., W. H. Freeman & Company (New York), ISBN:13: 978-1-4292-3414-6 / ISBN:10-14641-0962-1.

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

## Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and Pos		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	<b>Build</b> the concepts of the basic principles of inheritance at the molecular, cellular and organismal levels	PO1, PO2
CO2	<b>Formulate</b> causal relationships between molecule-cell level phenomena (modern genetics) and organism-level patterns of heredity (classical genetics).	PO1, PO2, PO3,
CO3	<b>Discuss</b> linkage, crossing over and mapping techniques	PO1, PO3, PO4
CO4	<b>Interpret</b> molecular basis of mutations	PO6, PO7, PO10
CO5	<b>Discuss</b> the inheritance of complex traits & population genetics and evolutionary genetics	PO8, PO9, PO11, PO10, PO12

Course Code	Course Title	PO 1	PO 2	PO 3	PO 4	PO 5	PO 6	PO 7	PO 8	PO 9	PO1 0	PO1 1	PO1 2	
<b>BIC1101</b> <b>8</b>	Gene organization, replication and repair	1	2	3	1	3	2	2	1	1	1	2	1	
		Fundamental Knowledge												
		Critical Thinking												
		Skill												
		Technical Knowledge												
		Logical Thinking Ability												
		Problem Identification												
		Analytical Knowledge												
		Career Goals												
		Team Work												
		Sustainable Development to Environment												
		Development to Society												
		. Development to Humanity												

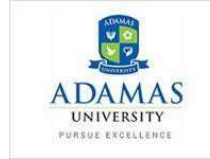
1=weakly mapped

2= moderately mapped

3=strongly mapped

Name:

Enrolment No:



**Course: BIC11018 Gene organization, replication and repair**

**Program: B.SC Biochemistry**

**Time: 03Hrs.**

**Semester: Even 2019-20**

**Max. Marks: 40**

**Instructions:**

Attempt any three questions from Section A (each carrying 4 marks); any four Questions from Section B (each carrying 7 marks).

**Section A ( Attemptany Three)**

1.	<b>Explain</b> semi-conservative replication. (U)	4	CO1
2.	<b>Explain how</b> DNA erosion occurs and <b>how</b> it is prevented. (U), (R)	4	CO2
3.	<b>How</b> can the speed of DNA replication increase while the rate of replication remains constant? (R)	4	CO3
4.	Between the two strands of a DNA segment the nitrogen bases are held together by. <b>Explain.</b> a. covalent bonds b. hydrogen bonds c. ionic bonds d. metallic bonds	4	CO3
<b>SECTION B (Attempt any FOUR Questions)</b>			
5.	<b>What</b> about stem cells? They aren't limited to 50 divisions. Stem cells can divide indefinitely. What makes a stem cell impervious to erosion? (R)	7	CO4
6.	Create the complimentary strand for the DNA strand below. Make sure to label the parts and direction of the strand. 5'-AACGGTCCAGTCCAAGTTACG-3'(Ap)	7	CO3,CO5
7.	Below is a segment of DNA that is ready to be replicated. Show the processes that the segment will go through during replication. Make sure to include the names of the enzymes that are involved. (Ap) <b>AATTGCCTGCTAGTCTCAG</b> <b>TTAACGGACGATCAGAGTC</b>	7	CO4

8.	<b>Describe</b> some common chromosomal mutations: inversions, deletions, duplications, fusions, fissions, and translocations. (U),	7	CO5
9.	<b>Name</b> the two basic kinds of point mutations. (U),	7	CO1

<b>BIC12019</b>	<b>GENE ORGANIZATION, REPLICATION AND REPAIR LAB (PRACTICAL)</b>	<b>L</b>	<b>T</b>	<b>P</b>	<b>C</b>
<b>Version 1.0</b>	<b>Contact Hours - 45</b>	0	0	3	2
<b>Pre-requisites/Exposure</b>	Basic knowledge of Organic Chemistry and Protein Biochemistry, Cell Biology at B.Sc level				
<b>Co-requisites</b>	--				

### Course objective

1. To acquire the concepts of the basic principles of heredity.
2. To understand causal relationships between molecule-cell level phenomena
3. To analyse the molecular consequences about linkage, crossing over and mapping techniques
4. To conceptualize molecular basis of DNA damage
5. To learn about gene repair mechanisms in prokaryotes and eukaryotes.

### Course Outcomes

On completion of this course, the students will be able to

- CO1** build the concept of the basic principles of inheritance at the molecular, cellular and organismal levels.
- CO2** formulate causal relationships between molecule-cell level phenomena (modern genetics) and organism-level patterns of heredity (classical genetics).
- CO3** discuss linkage, crossing over and mapping techniques.
- CO4** interpret molecular basis of mutations.
- CO5** discuss the inheritance of complex traits & population genetics and evolutionary genetics.

### Content

1. Verification of Chargaff's rule by paper chromatography.
2. Ultraviolet absorption spectrum of DNA and RNA.
3. Determination of DNA and RNA concentration by A<sub>260</sub>nm.
4. Determination of the melting temperature and GC content of DNA.
5. To study the viscosity of DNA solutions.
6. Isolation of chromosomal DNA from *E. coli* cells.

### SUGGESTED READINGS

1. Molecular Biology of the Gene (2008) 6th ed., Watson, J.D., Baker, T.A., Bell, S.P., Gann, A., Levine, M. and Losick, R., Cold Spring Harbor Laboratory Press, Cold Spring Harbor (New York), ISBN:0-321-50781 / ISBN:978-0-321-50781-5.
2. Lehninger: Principles of Biochemistry (2013) 6th ed., Nelson, D.L. and Cox, M.M., W. H. Freeman & Company (New York), ISBN:13: 978-1-4292-3414-6 / ISBN:10-14641-0962-1.
3. Principles of Genetics (2010) 5th ed., Snustad, D.P. and Simmons, M.J., John Wiley & Sons Asia, ISBN:978-0-470-39842-5.

<b>BIC11022</b>	<b>Membrane Biology and Bioenergetics (THEORY)</b>	<b>L</b>	<b>T</b>	<b>P</b>	<b>C</b>
<b>Version 1.0</b>	<b>Contact Hours - 60</b>	3	1	0	4
<b>Pre-requisites/Exposure</b>	Basic knowledge of Organic Chemistry and Protein Biochemistry, Cell Biology at B.Sc level				
<b>Co-requisites</b>	--				

### Course Objectives:

1. The main focus of the course is to cover the fundamental aspects of the structure and function of biological membranes and energy transformation in living organisms at a molecular level with the emphasis on modern methods of investigation.
2. It will provide a basic understanding about the signal transduction mechanism across the cell membrane important to run the cellular processes in homeostasis.
3. The course will further provide a details about membrane proteins in terms of important drug targets.

### Course outcome:

The students will be able to

1. **develop** the knowledge about fundamental aspects of composition, structure and functioning of biological membranes and energy transformation in living organisms.
2. **Explain** the laws of chemical thermodynamics, in terms of energetic processes in living cells, biological role of membrane structures, and the associated energy transformation mechanisms.
3. **Identify** the ways of energy transformation in animal and plant cells, archaea and bacteria, to describe the membrane transport mechanisms; to describe the process of synthesis of ATP by chemiosmosis.
4. **Examine** the preparation of artificial membranes, with their applications and limitations in membrane research methods;
5. **Utilize** methods of studies of biological membranes and bioenergetics to solve fundamental problems in biomedicine and to use for the applied research in biotechnology.

### Catalogue Description:

Membrane Biology and bioenergetics is theoretical course aimed at enlightening students on the structure-function relationship of biological membranes. The course outline includes membrane structure and function, role of membrane in cellular processes (signal transduction, solute transport, signalling, etc.), plasma membrane - chemical composition, fluidity, dynamic nature (cell fusion and protein diffusion), membrane transport (diffusion, facilitated and active transport, energetics), membrane potential and nerve impulses, structure and function of endomembrane system (endoplasmic reticulum, Golgi complex, endosomes, lysosomes, cellular uptake (phagocytosis and endocytosis), chemical thermodynamics, high-energy compounds, electron transport system and oxidative phosphorylation, regulation of ATP production, oxidations-reductions, chemical potentials and electrochemical potentials.

### Course Content

#### Membrane Biology and Bio-energetics (THEORY)



## Unit I

15 Lecture Hours

**Introduction to biomembranes:** Composition of biomembranes - prokaryotic, eukaryotic, neuronal and subcellular membranes. Study of membrane proteins. Fluid mosaic model with experimental proof. Monolayer, planer bilayer and liposomes as model membrane systems. Membrane structures and dynamics: Polymorphic structures of amphiphilic molecules in aqueous solutions - micelles and bilayers. CMC, critical packing parameter. Membrane asymmetry. Macro and micro domains in membranes. Membrane skeleton, lipid rafts, caveolae and tight junctions. RBC membrane architecture. Lateral, transverse and rotational motion of lipids and proteins. Techniques used to study membrane dynamics - FRAP, TNBS labelling etc. Transition studies of lipid bilayer, transition temperature. Membrane fluidity, factors affecting membrane fluidity.

## Unit II

15 Lecture Hours

**Membrane transport:** Thermodynamics of transport. Simple diffusion and facilitated diffusion. Passive transport - glucose transporter, anion transporter and porins. Primary active transporters - P type ATPases, V type ATPases, F type ATPases. Secondary active transporters - lactose permease, Na<sup>+</sup>/glucose symporter. ABC family of transporters - MDR, CFTR. Group translocation. Ion channels - voltage-gated ion channels (Na<sup>+</sup>/K<sup>+</sup> voltage-gated channel), ligand-gated ion channels (acetyl choline receptor), aquaporins, and bacteriorhodopsin. Ionophores - valinomycin, gramicidin.

## Unit III

5 Lecture Hours

**Vesicular transport and membrane fusion:** Types of vesicle transport and their function - clathrin, COP I and COP II coated vesicles. Molecular mechanism of vesicular transport. Membrane fusion. Receptor mediated endocytosis of transferrin.

## Unit IV

10 Lecture Hours

**Laws of thermodynamics,** state functions, equilibrium constant, coupled reactions, energy charge, ATP cycle, phosphorylation potential, phosphoryl group transfers. Chemical basis of high standard energy of hydrolysis of ATP, other phosphorylated compounds and thioesters. Redox reactions, standard redox potentials and Nernst equation. Universal electron carriers.

## Unit V

15 Lecture Hours

**Oxidative phosphorylation and photophosphorylation:** Mitochondria Electron transport chain - its organization and function. Inhibitors of ETC and uncouplers. Peter Mitchell's chemiosmotic hypothesis. Proton motive force. Fo-F<sub>1</sub>ATP synthase, structure and mechanism of ATP synthesis. Metabolite transporters in mitochondria. Regulation of oxidative phosphorylation. ROS production and antioxidant mechanisms. Thermogenesis. Alternative respiratory pathways in plants. General features of photophosphorylation, historical background, Hill's reaction, and photosynthetic pigments, light harvesting systems of plants and microbes and resonance energy transfer. Bacterial photophosphorylation in purple bacteria, Green sulphur bacteria and *Halobacterium salinarum*. Photophosphorylation in plants - structure of chloroplast, molecular architecture of Photosystem I and Photosystem II, Z-scheme of photosynthetic electron flow, oxygen evolving complex and action of herbicides. Cyclic photophosphorylation and its significance. Photoinhibition. Evolution of oxygenic photosynthesis.

Suggested Books:

1. Lehninger: Principles of Biochemistry (2013) 6th ed., Nelson, D.L. and Cox, M.M., WH Freeman and Company (New York), ISBN: 13:978-1-4641-0962-1 / ISBN:10:1-4641-0962-1.

2. Molecular Cell Biology (2013) 7th ed., Lodish, H., Berk, A., Kaiser, C.A., Krieger, M., Bretscher, A., Ploegh, H., Amon, A. and Scott, M.P., W.H. Freeman & Company (New York), ISBN:13:978-1-4641-0981-2.
3. Biochemistry (2010) 4th ed., Garret, R. H. and Grisham, C.M., Cengage Learning (Boston), ISBN-13:978-0-495-11464-2.
4. Principles of Biochemistry (2008) 3rd ed., Voet, D.J., Voet, J.G. and Pratt, C.W., John Wiley & Sons, Inc. (New York), ISBN:13:978-0470-23396-2

**Modes of Examination: Assignment/Quiz/Project/Presentation/Written Exam**

**Examination Scheme:**

Components	Internal	Attendance	Mid Term	End Term
Weightage (%)	30	10	20	40

**Relationship between the Course Outcomes (COs) and Program Outcomes (POs)**

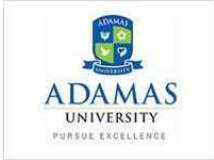
Mapping between COs and POs		
	Course Outcomes (COs)	Mapped Program Outcomes
CO-1	<b>Explain</b> the fundamental aspects of composition, structure and functioning of biological membranes and energy transformation in living organisms	PO1, PO2
CO-2	<b>Develop</b> the laws of chemical thermodynamics, in terms of energetically process in living cells, biological role of membrane structures, and the associated energy transformation mechanisms.	PO1, PO2, PO5
CO-3	<b>Identify</b> the ways of energy transformation in animal and plant cells, archaea and bacteria, to describe the membrane transport mechanisms; to describe the process of synthesis of ATP by chemiosmosis.	PO3, PO4
CO-4	<b>Examine</b> the preparation of artificial membranes, with their applications and limitations in membrane research methods	PO2, PO4, PO6
CO-5	<b>Utilize</b> the methods of studies of biological membranes and bioenergetics to solve fundamental problems in biomedicine and to use for the applied research in biotechnology.	PO3, PO5, PO7, PO8, PO10

Course Code	Course Title	PO 1	PO 2	PO 3	PO 4	PO 5	PO 6	PO 7	PO 8	PO 9	PO 10	PO 11	PO12
		Fundamental Knowledge	Critical Thinking	Skills	Technical Knowledge	Logical Thinking	Problem identification ability	Analytical Knowledge	Career goals	Team Work	Sustainable Development to environment	Development to society	Development to humanity
<b>BIC11022</b>	<b>Membrane Biology and Bioenergetics (THEORY)</b>	2	3	2	2	2	1	1	1	-	1	-	-

1=weakly mapped

2= moderately mapped

3=strongly mapped

<b>Name:</b>		 <b>ADAMAS</b> UNIVERSITY <small>PURSUE EXCELLENCE</small>	
<b>Enrolment No:</b>			
<b>Course: BIC11022 - Membrane Biology and Bioenergetics(THEORY)</b> <b>Program: B.Sc.Biochemistry</b> <b>Semester:Even2020-21</b>			
		<b>Time: 03Hrs.</b> <b>Max. Marks:40</b>	
<b>Instructions:</b> Attempt any <b>four</b> questions from <b>Section A</b> (each carrying 5 marks); any <b>two</b> questions from <b>Section B</b> (each carrying 10 marks).			
<b>SECTION A ( Attemptany Four questions)</b>			
1.	<b>Outline</b> antiporters with suitable example.(Un)	<b>5</b>	<b>CO1</b>
2.	<b>Explain</b> the role of physicochemical properties in relation to biological activity and drug design. (Un)	<b>5</b>	<b>CO2</b>
3.	<b>Discuss</b> active transport mechanism across plasma membrane using an example of Na <sup>+</sup> -K <sup>+</sup> pump. (Cr)	<b>5</b>	<b>CO2</b> <b>CO3</b>
4.	<b>How</b> plasma membrane components influence the folding of membrane proteins? (Re)	<b>5</b>	<b>CO4</b>
5	<b>How</b> hydropathy plots contribute for analysis of membrane protein? (Re)	<b>5</b>	<b>CO5</b>
<b>SECTION B (Attempt any Two questions)</b>			
6.	<b>Which</b> receptor regulate the activation mechanism of GLUT transporters and how? <b>What</b> are the differences between receptors and transporters?(Re)	<b>5+5</b>	<b>CO1</b> <b>CO2</b>
7.	<b>Classify</b> three important components of plasma membrane required for its fluid mosaic properties and functionalities.(Un)	<b>10</b>	<b>CO1</b> <b>CO2</b>
8.	<b>How</b> electron transport chain is coupled to ATP synthase complex in oxidative phosphorylation? Does ETC have a role in anaerobic bacteria? (Re)	<b>6+4</b>	<b>CO3</b> <b>CO4</b>
9	<b>What</b> are the forces involved in drug-receptor interaction important to modify the downstream signaling pathway? <b>What</b> is biased agonism in terms of G-protein coupled receptor signaling pathway? (Re)	<b>5+5</b>	<b>CO5</b>

<b>BIC11020</b>	<b>Immunology</b>	<b>L</b>	<b>T</b>	<b>P</b>	<b>C</b>
<b>Version 1.0</b>	<b>Contact Hours - 60</b>	4	0	0	4
<b>Pre-requisites/Exposure</b>	Basic Human Physiology at B.Sc level				
<b>Co-requisites</b>	--				

### Course Objectives:

1. To provide basic understanding of our immune system and its medical implication.
2. To provide basic understanding B-cell, T-cell, antibody structure and their interaction with antigen.
3. To provide basic understanding of the activation, mechanism and regulation of the immune system and Host pathogen interaction.

### Course Outcomes

On completion of this course, the students will be able to

- CO1. demonstrate a coherent understanding of the fundamental concepts in Immunology and understanding the complex relations among components of the immune system.
- CO2. develop concept about immunoglobulin structures and diversity of antibodies, B-cell, T-cell, and their interaction with antigen.
- CO3. explain the knowledge about basic mechanisms of hypersensitivity responses and their associations with different diseases.
- CO4. discuss basic understanding of immunology and immune responses in response to various infectious and non-infectious diseases.
- CO5. develop awareness of the current research activities in the field and possible applications of this knowledge for the betterment of humanity.

### Catalog Description:

Immunology and Medical Biotechnology course will provide an through understanding of the principles and mechanisms of the immune system and immune responses in the context of infection, malignancy and immunological disorders. All the lectures will be devoted on discussions of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as power point presentation, audio visual virtual lab session as per requirement. The tutorials will familiarize the students with

practical problem-solving techniques led by the course coordinator. Students will strongly grasp the basic concepts of the subject via exercise and discussions with the coordinator.

### **Course Content:**

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## **IMMUNOLOGY**

### **UNIT I                      10 LectureHours**

Introduction:

Concept of Innate and Adaptive immunity; Contributions of following scientists to the development of field of immunology - Edward Jenner, Karl Landsteiner, Robert Koch, Paul Ehrlich, Elie Metchnikoff, Peter Medawar, MacFarlane Burnet, NeilsK Jerne, Rodney Porter and Susumu Tonegawa.

### **UNIT II                      5 LectureHours**

Immune Cells and Organs:

Structure, Functions and Properties of: Immune Cells – Stem cell, T cell, B cell, NK cell, Macrophage, Neutrophil, Eosinophil, Basophil, Mast cell, Dendritic cell; and Immune Organs – Bone Marrow, Thymus, Lymph Node, Spleen, GALT, MALT, CALT

### **UNIT III                      5 LectureHours**

Antigens:

Characteristics of an antigen (Foreignness, Molecular size and Heterogeneity); Haptens; Epitopes (T & B cell epitopes); T-dependent and T-independent antigens; Adjuvants. Cytokines: Different Types and Functions.

### **UNIT IV                      7 LectureHours**

Antibodies:

Structure, Types, Functions and Properties of antibodies; Antigenic determinants on antibodies (Isotypic, allotypic, idiotypic); VDJ rearrangements; Monoclonal and Chimeric antibodies.

### **UNIT V                      7 LectureHours**

Major Histocompatibility Complex:

Organization of MHC locus (Mice & Human); Structure and Functions of MHC I & II molecules; Antigen processing and presentation (Cytosolic and Endocytic pathways)

### **UNIT VI                      7 LectureHours**

Complement System:

Components of the Complement system; Activation pathways (Classical, Alternative and Lectin pathways); Biological consequences of complement Activation

### **UNIT VII                      9 LectureHours**

Generation of Immune Response:

Primary and Secondary Immune Response; Generation of Humoral Immune Response (Plasma and Memory cells); Generation of Cell Mediated Immune Response (Self MHC restriction, Tcell activation, Co- stimulatory signals); Killing Mechanisms by CTL and NK cells, Introduction to tolerance. : Immunological basis of graft rejection, clinical manifestations, immunosuppressive therapy and privileged sites. Vaccines - active and passive immunization, types of vaccines.

**UNITVIII 5 LectureHours**

Immunological Disorders and Tumor Immunity:

Types of Autoimmunity and Hypersensitivity with examples; Immunodeficiencies - Animal models (Nude and SCID mice), SCID, DiGeorge syndrome, Chediak- Higashi syndrome, Leukocyteadhesiondeficiency,CGD;Typesoftumours,tumourAntigens,causesandtherapyfor cancers.

**UNITIX 5 LectureHours**

Immunological Techniques:

Principles of Precipitation, Agglutination, Immunodiffusion, Immunoelectrophoresis, ELISA, ELISPOT, Western blotting, Immunofluoresence, Flow cytometry, Immunoelectron microscopy.

**Reference Books****Suggested Books:**

- 1.Kuby Immunology by Judy Owen, Jenni Punt, Sharon Stranford, 2013
2. Roitt's Essential Immunology (Essentials) by Ivan M. Roitt,2016
3. Medical Microbiology & Immunology by Warren Levinson, 2004
4. Basic and Clinical Immunology by Mark Peakman

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

**Relationship between the Course Outcomes (COs) and Program Outcomes (POs)**

Mapping between COs and POs		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	<b>Demonstrate</b> a coherent understanding of the fundamental concepts in Immunology and understanding the complex relations among components of the immune system.	PO1, PO2, PO5, PO6,
CO2	<b>Develop</b> knowledge about immunoglobulin structures and diversity of antibodies, B-cell, T-cell, and their interaction with antigen.	PO1, PO2, PO5, PO6, PO9, PO3
CO3	<b>Explain</b> the basic mechanisms of hypersensitivity responses and their associations with different diseases.	PO1, PO2, PO3, PO4, PO5, PO6, PO7
CO4	<b>Discuss</b> basic understanding of immunology and immune responses in response to various infectious and non-infectious diseases.	PO1, PO2, PO4, PO5, PO6, PO7
CO5	<b>Develop</b> awareness of the current research activities in the field and possible applications of this knowledge for the betterment of humanity.	PO1, PO8, PO9, PO10, PO11, PO12

Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
<b>BIC11020</b>	Immunology (THEORY)	3	3	1	1	3	3	2	2	1	1	1	1
		Fundamental Knowledge	Critical Thinking	Skills	Technical Knowledge	Logical Thinking	Problem identification ability	Analytical Knowledge	Career goals	Team Work	Sustainable Development to environment	Development to society	Development to humanity

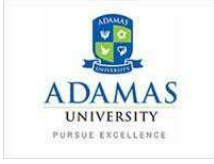
1=weakly mapped

2= moderately mapped

3=strongly mapped



## Model Question Paper

<b>Name:</b>  <b>Enrolment No:</b>	 <p style="font-size: small;">ADAMAS UNIVERSITY PURSUE EXCELLENCE</p>		
<b>Course: BIC11020–IMMUNOLOGY (THEORY)</b> <b>Program: B.Sc. Biochemistry</b> <b>Time: 03 Hrs.</b> <b>Semester: Odd20xx-20xx</b>			
<b>Max. Marks:40</b>			
<b>Instructions:</b> Attempt <b>all</b> the questions from <b>Section A</b> (each carrying 5 marks); <b>all</b> the questions from <b>Section B</b> (each carrying 10 marks).			
<b>SECTION A (Attempt all the questions)</b>			
1.	<b>Explain</b> the working principle of an adjuvant. <b>List</b> two Examples of adjuvant. (R) (U)	5	<b>CO1, CO2, CO3, CO5</b>
2.	<p>(i) A 17-year-old boy suffered an injury to his left eye when, during a car crash near Barasat railway station, a sharp sliver of glass penetrated his eye, damaging his lens and uveal tract. The glass was removed and the injury repaired with complete recovery. However, 3 weeks later he noticed some redness in the left eye and photophobia, followed by pain and severe visual impairment. The left eye was removed and histologic examination showed an extensively infiltrated uveal tract with abundant lymphocytes and mononuclear cells. Two weeks later the other eye began to show the same symptoms. Since the lens was damaged the most likely scenario is that: <b>(Explain your answer)</b>(U)(Ev)</p> <p>A. this is an example of immediate hypersensitivity.          B. sequestered antigen was released and initiated an immune response.          C. some of the glass must have penetrated the right eye as well.          D. the bacteria entered the damaged eye and is causing an infection in both eyes.</p>	5	<b>CO2, CO3</b>
3.	<b>Explain</b> the cytokine signalling through JAK-STAT pathway.(U)	5	<b>CO1</b>
4.	<b>Compare and contrast</b> innate and adaptive immunity by <b>identifying</b> (matching) the following characteristics with the correct arm of immunity, using I for innate and A for adaptive: (U)(Ap)	5	<b>CO2, CO4</b>
	<p>(A) Is the first to engage upon initial encounter with antigen          (B) Is the most pathogen specific          (C) Employs T and B lymphocytes          (D) Adapts during the response</p>		

	<b>SECTION B (Attempt all the questions)</b>		
5.	(a) <b>Explain</b> the immune mechanism involved in Rheumatoid arthritis (U) (b) <b>Compare</b> and <b>contrast</b> immunological basis of Grave's disease and Myasthenia gravis. (U)(A)	<b>10</b>	<b>CO4</b>
6.	<b>Explain</b> the classical and alternative pathway of compliment activation. (Ev)	<b>10</b>	<b>CO3, CO4</b>

<b>BIC12021</b>	<b>IMMUNOLOGY LAB (PRACTICAL)</b>	<b>L</b>	<b>T</b>	<b>P</b>	<b>C</b>
<b>Version 1.0</b>	<b>Contact Hours - 45</b>	0	0	3	2
<b>Pre-requisites/Exposure</b>	Basic knowledge about Immunology Theory B.Sc level				
<b>Co-requisites</b>	--				

### Course Objectives

1. to demonstrate and interpret different antigen-antibody interactions.
2. to acquaint with various components of the immune system and apply this knowledge in immunodiagnostics.
3. to apply various immunological techniques for clinical and research purpose.
4. to do quantitative/qualitative measurement of antigen/ antibody in different samples.
5. to analyze and assess antigen-antibody interaction

### Course Outcomes

On completion of this course, the students will be able to

- CO1. **evaluate** different antigen-antibody interactions.
- CO2. **identify** different components of immune system in human system
- CO3. **utilize** different immunological techniques for research and clinical purposes.
- CO4. **estimate and compare amount of antigen/ antibody present in different samples**
- CO5. **assess** antigen-antibody interaction

### Catalog Description

The student will be able to use the knowledge obtained to perform and analyze different types of antigen-antibody interaction. Identification of different components of the immune system is possible with the concept obtained. Students will gain the ability to apply different immunological techniques for research and clinical purposes. All the experiments will be based on hands-on training in laboratory setup along with discussions of basic theories and advanced topics for practical implementation of knowledge. Classes will be conducted by hands-on lab training and/or audio-visual virtual lab session as per requirement. Students will perceive the basic concepts of the subject via exercise and discussions with the coordinator.

### Course Content

#### IMMUNOLOGY LAB

1. Identification of human blood groups. (5 Hours)
2. To perform Total Leukocyte Count of the given blood sample. (5 Hours)

3. To perform Differential Leukocyte Count of the given blood sample. (5 Hous)
4. To separate serum from the blood sample (demonstration). (5 Hours)
5. To perform immunodiffusion by Ouchterlony method. (10 Hours)
6. To perform DOT ELISA. (10 Hours)
7. To perform immunodiffusion by radial method. (10 Hours)
8. Demonstration of Phagocytosis. (10 Hours)

**Text Book(s)**

1. Immunology Lab Manual by Wilmore Weberly,2015

2. Immunology methods manual - The comprehensive source book by Lefkovits. ,1996
3. Manual of clinical laboratory immunology by Rose NR,2002
5. Laboratory Immunology by BradshawLJ.1997

### Reference books

Owen, J.A.; Punt, J.; Kuby, J.; Stranford, S.A. Kuby immunology. W.H. Freeman: 2013.

### Modes of Examination: Assignment/Quiz/Project/Presentation/Written Exam

#### Examination Scheme:

Components	Internal	Attendance	Mid Term	End Term
Weightage (%)	30	10	20	40

### Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

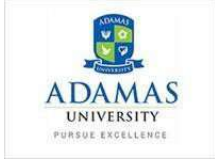
Mapping between COs and POs		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	Evaluate different antigen-antibody interactions.	PO1, PO2, PO3, PO5
CO2	Identify different components of immune system in human system	PO2, PO3, PO5
CO3	Utilize different immunological techniques for research and clinical purposes.	PO2, PO4, PO5, PO7, PO8
CO4	Estimate and compare amount of antigen/antibody present in different samples	PO2, PO5, PO6, PO7, PO8
CO5	Assess different host pathogen interactions.	PO2, PO4, PO5, PO7, PO8

Course Code	Course Title	
<b>BIC12021</b>	Immunology lab(practical)	PO1 Fundamental Knowledge
		PO2 Critical thinking
		PO3 Skills
		PO4 Technical Knowledge
		PO5 Logical Thinking
		PO6 Problem identification ability
		PO7 Analytical Knowledge
		PO8 Career goals
		PO9 Team Work
		PO10 Sustainable Development to environment
		PO11 Development to society
		PO12 Development to humanity

1=weakly mapped

2= moderately mapped 3=strongly mapped

## Model Question Paper

<b>Name:</b>			
<b>Enrolment No:</b>			
<b>Course: BIC12021–IMMUNOLOGY LAB (PRACTICAL)</b>		<b>Time: 03Hrs.</b>	
<b>Program: B.Sc. Biochemistry</b>		<b>Max. Marks: 40</b>	
<b>Semester: Even 2020-21</b>			
<b>Instructions:</b> Answer the following questions”			
<b>SECTION A</b>			
1.	<b>Summarize</b> the principle ABO blood grouping. (Un)	<b>10</b>	<b>CO1</b>
2.	<b>Illustrate</b> the procedure for Giemsa staining for the given blood sample/s and evaluate the results. (Cr) <i>(Practical skill- 5, Method accuracy -5)</i>	<b>10</b>	<b>CO1, CO2</b>
<b>SECTION B</b>			
6.	Lab note book (U/An/Ap/R/Ev)	<b>10</b>	<b>CO1, CO2, CO3, CO4, CO5</b>
7.	Viva-Voce (U/An/Ap/R/Ev)	<b>10</b>	<b>CO1, CO2, CO3, CO4, CO5</b>





<b>BIC12023</b>	<b>MEMBRANE BIOLOGY AND BIOENERGETICS LAB (PRACTICAL)</b>	<b>L</b>	<b>T</b>	<b>P</b>	<b>C</b>
<b>Version 1.0</b>	<b>Contact Hours - 45</b>	0	0	3	2
<b>Pre-requisites/Exposure</b>	BASIC KNOWLEDGE OF MEMBRANE BIOLOGY AND BIOENERGETICS THEORY PAPER OF B.SC. LEVEL				
<b>Co-requisites</b>	-				

### Course Objectives

1. To provide students with hands-on training in the field of membrane biology.
2. Students will need to become proficient with different practical applications of membrane biology.

### Course Outcomes

On completion of this course, the students will be able to

- CO1. **Explain** the basic principles of membrane biology and its structural aspects.
- CO2. **Develop** fundamental knowledge about micelle formation and their properties.
- CO3. **Comprehend** different assay systems of marker enzymes.
- CO4. **Build** concept about modern tools like TLC, conductometry etc.
- CO5. **Utilize** the knowledge of membrane biology to understand important physiological processes and organelles.

### Catalog Description

The core course “membrane biology and bioenergetics lab” is a practical paper which has been designed to provide the knowledge of different aspects membrane biology. It will provide understanding of important physiological processes and organelles. Students will be able to understand basic concepts TLC and other techniques and will practice hands-on all of them. They will also learn to compare and use these methods for practical purpose like CMC determination, marker enzyme assay etc. All the lectures will be devoted on discussions of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as power point presentation, audio visual virtual lab session as per requirement. The tutorials will familiarize the students with practical problem-solving techniques by the course coordinator. Students will strongly grasp the basic concepts of the

subject via exercise and discussions with the coordinator.

### Course Content

#### MEMBRANE BIOLOGY AND BIOENERGETICS LAB

1. Determination of CMC of a detergent. (10 Hours)
2. Effect of temperature on the permeability of a cell. (5 Hous)
3. Effect of organic solvent concentration on the permeability of a cell. (5 Hous)
4. Study of Osmosis. (5 Hous)
5. Isolation of RBC Membrane. (5 Hous)
6. Egg Lab: Study of Tonicity. (5 Hous)
7. Dialysis tube Lab. (5 Hous)
8. Liposome Preparation using Phospholipid.(10 Hours)

#### SUGGESTED READINGS

1. Introduction to Experimental Biophysics: Biological Methods for Physical Scientists: Volume 1& 2 (Foundations of Biochemistry and Biophysics). Jay L. Nadeau; CBC Press.
2. Textbook of Biochemistry with Clinical Correlations (2011) 7<sup>th</sup>ed., Devlin, T.M., John Wiley & Sons, Inc. (New Jersey), ISBN:978-0-470-28173-4.
3. Biochemistry (2012) 7<sup>th</sup>ed., Berg, J.M., Tymoczko, J.L. and Stryer L., W.H. Freeman and Company (New York), ISBN:10:1-4292-2936-5, ISBN:13:978-1-4292-2936-4.

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

#### Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and Pos	
	<b>Course Outcomes (COs)</b>
	<b>Mapped Program</b>

		Outcomes
CO1	<b>Explain</b> the basic principles of membrane biology and its structural aspects.	PO1, PO2, PO3
CO2	<b>Develop</b> fundamental knowledge about micelle formation and their properties.	PO1, PO10, PO11
CO3	<b>Comprehend</b> different assay systems of marker enzymes.	PO1, PO2, PO3, PO4, PO8
CO4	<b>Build</b> their concept on modern tools like TLC, conductometry etc.	PO1, PO3, PO6, PO8
CO5	<b>Utilize</b> the knowledge of membrane biology to understand important physiological processes and organelles.	PO1, PO4, PO8,

		Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking Ability	Problem Identification	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to Environment	Development to Society	. Development to Humanity
Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
<b>BIC12023</b>	MEMBRANE BIOLOGY AND BIOENERGETICS LAB (PRACTICAL)	3	2	3	2	-	1	-	3	1	1	1	-

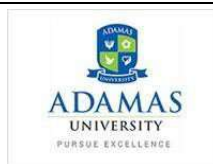
1=weakly mapped

2= moderately mapped

3=strongly mapped

Name:

Enrolment No:



**Course: BIC12023– MEMBRANE BIOLOGY AND BIOENERGETICS LAB  
(PRACTICAL)**

**Program: B.Sc. Biochemistry**

**Semester: Odd 2019-20**

**Time: 03Hrs.**

**Max. Marks: 40**

**Instructions:**

Attempt any **two** questions from **Section A** (each carrying 10 marks); **Section B** is Compulsory (carrying 10 marks).

**Section A ( Attemptany Two)**

1.	a) <b>Outline</b> the principle of micelle formation.(U) b) <b>Determine</b> the effect of temperature on a membrane.(Ap)	4 6	CO1 CO2
2.	a) <b>Explain</b> the basic theory of TLC.(U) b) <b>Determine</b> the CMC of a detergent.(Ap)	4 6	CO3 CO4
3.	a) <b>Define</b> the principle behind ion-exchange chromatography.(Re) b) <b>Demonstrate</b> the O <sub>2</sub> production in hydrilla with a simple experiment.(Ap)	4 6	CO3 CO2
	<b>SECTION B is compulsory</b>		
4.	<b>Viva-voce</b> (U/An/Ap/R/Ev)	10	CO1 CO2 CO3 CO4 CO5
5.	<b>Practical copy</b> (U/Ap/Ev)	10	CO1 CO2 CO3



(a) these issues? (Ap)

<b>BIC11024</b>	<b>Sec II Protein Purification Techniques (THEORY)</b>	<b>L</b>	<b>T</b>	<b>P</b>	<b>C</b>
<b>Version 1.0</b>	<b>Contact Hours - 30</b>	2	0	0	2
<b>Pre-requisites/Exposure</b>	Knowledge About Protein at B.Sc. Level				
<b>Co-requisites</b>	--				

### Course Objectives

1. To provide students basic idea about different column purification techniques used to purify proteins.
2. It will also provide in depth knowledge about which techniques should be used for purification based on the nature of proteins.
3. To differentiate between different methodologies used for protein purification.
4. To provide students different ways of identifying proteins in a mixture of protein lysate.

### Course Outcomes

On completion of this course, the students will be able to:

1. **Define** the fundamental biochemical reaction used in different protein purification techniques.
2. **Identify** the appropriate purification method based on nature of the protein.
3. **Develop** different methods for sample preparation that will be used for purification.
4. **Utilize** the electrophoresis techniques to identify protein.
5. **Identify** a particular protein from a heterogeneous mixture of proteins.

### Catalog Description

The core-course of 'Protein purification Techniques' will help to understand fundamental biochemical reaction used in different protein purification techniques. This course includes comprehensive approach through studying different biochemical reactions that are used to determine different methods for sample preparation that will be used for purification. Furthermore, the implication of the electrophoresis techniques to identify protein will also be illuminated. All the lectures will be devoted to discussion of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as power point presentation, audio visual virtual lab session as per requirement. The tutorials will enable the students with problem-solving ability led by the course coordinator. Students will perceive the basic concepts of the subject via exercise and discussions with the coordinator.

### Course Content

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#### SEC-2 : PROTEIN PURIFICATION TECHNIQUES

1. **Purification and characterization of a protein from a complex mixture (native or heterologously expressed) involving the following methods/techniques**

**Exercises:**

- i. Preparation of the sample. [10 LectureHours]
- ii. Ion-exchange chromatography. [10 LectureHours]
- iii. Gel filtration chromatography. [10 Lecture Hours]
- iv. Affinity chromatography. [10 LectureHours]
- v. Electrophoresis. [10 Lecture Hours]

**2. Demonstration of High-Performance Liquid Chromatography (HPLC) [10 Lecture Hours]****SUGGESTED READINGS**

2. Physical Biochemistry: Principles and Applications (2010) 2<sup>nd</sup>ed., Sheehan, D., Wiley Blackwell (West Sussex), ISBN:978-0-470-85602-4 / ISBN:978-0-470-85603-1.
3. Physical Biochemistry: Applications to Biochemistry and Molecular Biology (1982) 2<sup>nd</sup> ed., Freifelder, D., W.H. Freeman and Company (New York), ISBN:0-7167-1315-2 / ISBN:0-7167-1444-2.
4. An Introduction to Practical Biochemistry (1998) 3<sup>rd</sup>ed., Plummer D. T., Tata McGrawHill Education Pvt. Ltd. (New Delhi), ISBN:13: 978-0-07-099487-4 / ISBN:10: 0-07-099487-0.

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

**Relationship between the Course Outcomes (COs) and Program Outcomes (POs)**

Mapping between COs and POs		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	Define the fundamental biochemical reaction used in different protein purification techniques.	PO1,PO5
CO2	Identify the appropriate purification method based on nature of the protein.	PO1, PO2,PO7
CO3	Develop different methods for sample preparation that will be used for purification.	PO1, PO2, PO3,PO5
CO4	Utilize the electrophoresis techniques to identify protein.	PO1, PO2, PO4, PO7
CO5	Identify a particular protein from a heterogenous mixture of proteins.	PO3, PO5, PO7,PO8, PO10

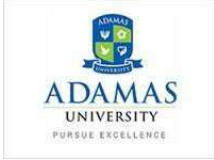


Course Code	Course Title	
<b>BIC110</b> <b>24</b>	Protein Purification Techniques(THEORY)	PO 1 Fundamental Knowledge
		PO 2 Critical Thinking
		PO 3 Skill
		PO 4 Technical Knowledge
		PO 5 Logical Thinking Ability
		PO 6 Problem Identification
		PO 7 Analytical Knowledge
		PO 8 Career Goals
		PO 9 Team Work
		PO 10 Sustainable Development to Environment
		PO 11 Development to Society
		PO 12 . Development to Humanity

1=weakly mapped

2= moderately mapped

3=strongly mapped

<b>Name:</b>			
<b>Enrolment No:</b>			
<b>Course: BIC11024–Protein Purification Technique (THEORY)</b> <b>Program: B.Sc. Biochemistry</b> <b>Semester: Even 2020-21</b>			
		<b>Time: 03Hrs.</b> <b>Max. Marks: 40</b>	
<b>Instructions:</b> Attempt any <b>four</b> questions from <b>Section A</b> (each carrying 5 marks); any <b>two</b> questions from <b>Section B</b> (each carrying 10 marks).			
<b>SECTION A (Attempt any Four questions)</b>			
1.	<b>Define</b> partition coefficient value of paper chromatography. How will you separate arginine and glutamic acid from their mixture by paper electrophoresis?(Re)	<b>2+3</b>	<b>CO1</b>
2.	<b>What</b> will be the order of elution of following proteins from Sephadex-G50 column: catalase (222 KD), $\alpha$ -chymotrypsin (21.6 KD), concanavalin B (42.5 KD), lipase (6.7 KD) and RNase A (12.6 KD)? <b>Explain</b> your answer.(Un)	<b>2+3</b>	<b>CO2</b>
3.	<b>Distinguish</b> between Anion exchange chromatography and Affinity Chromatography.(An)	<b>5</b>	<b>CO3</b>
4.	<b>Why</b> two different Tris buffers of pH 6.8 and 8.8 are used in native and SDS-PAGE? <b>Identify</b> the extra advantage of the usage of SDS in the polyacrylamide gel electrophoresis.(Re/Un)	<b>2+3</b>	<b>CO4</b>
5.	<b>How</b> ammonium sulphate helps in salting out of a specific protein? <b>What</b> is the principle of isoelectric precipitation during protein purification? (Re/Un)	<b>2+3</b>	<b>CO5</b>
<b>SECTION B (Attempt any Two questions)</b>			
6.	<b>What</b> is the major difference between polyacrylamide gel electrophoresis and gel filtration chromatography? Write down the significance of molecular exclusion limit of gel filtration chromatography. <b>How</b> will you confirm about the presence of amino acids in the thin layer chromatography plates?(An/Re)	<b>4+4+2</b>	<b>CO2</b> <b>CO4</b>
7.	Pepsin is the name given to a mix of several digestive enzymes secreted (as larger precursor proteins) by glands that line the stomach. These glands also secrete hydrochloric acid, which dissolves the particulate matter in food, allowing pepsin to enzymatically cleave individual protein molecules. The resulting mixture of food, HCl, and digestive enzymes is known as chyme and has a pH near 1.5. <b>What</b> pI would you predict for the pepsin proteins? <b>What</b> functional groups must be present to confer this pI on pepsin? <b>Which</b> amino acids in the proteins would contribute such groups?(Re/Un/An)	<b>2+4+4</b>	<b>CO1</b> <b>CO2</b>

8.	<p>Histones are proteins found in eukaryotic cell nuclei, tightly bound to DNA, which has many phosphate groups. The pI of histones is very high, about 10.8. <b>Which</b> amino acid residues must be present in relatively large numbers in histones? <b>Identify</b> the way through which these residues contribute to the strong binding of histones to DNA? (Re/Un)</p>	<b>5+5</b>	<b>CO1</b> <b>CO2</b>
9	<p>A protein has a molecular mass of 400 kDa when measured by gel filtration. When subjected to gel electrophoresis in the presence of sodium dodecyl sulfate (SDS), the protein gives three bands with molecular masses of 180, 160, and 60 kDa. When electrophoresis is carried out in the presence of SDS and dithiothreitol, three bands are again formed, this time with molecular masses of 160, 90, and 60 kDa. <b>Determine</b> the subunit composition of the protein. (An)</p>	<b>3+3+4</b>	<b>CO4</b> <b>CO5</b>

<b>BIC 11025</b>	<b>Recombinant DNA technology</b>	L	T	P	C
<b>Version 1.0</b>	Contact Hours: 60	2	0	0	2
<b>Pre-requisites/Exposure</b>	Basic knowledge in Molecular Biology				
<b>Co-requisites</b>	-				

### Course Objectives

1. To understand the Restriction and modification systems, ligation systems in bacteria and viruses and apply their appropriate use in recombinant DNA technology
2. To conceptualize about isolation of cloning vectors for prokaryotes and eukaryotes and summarize their appropriate use in recombinant DNA technology
3. To provide knowledge of PCR, DNA sequencing and choose their appropriate use in recombinant DNA technology
4. Able to understand recombinant DNA technology in gene expression through blue-white screening
5. To **demonstrate** and apply RDT in Recombinant vaccines Gene therapy, agriculture- plant genetic engineering, herbicide resistant crops etc.

### Course Outcomes

On completion of this course, the students will be able to

- CO1** **demonstrate** the Restriction and modification systems, ligation systems in bacteria and viruses and apply their appropriate use in recombinant DNA technology.
- CO2** **develop** the concept about isolation of cloning vectors for prokaryotes and eukaryotes and **apply** their appropriate use in recombinant DNA technology.
- CO3** **demonstrate** PCR, DNA sequencing and choose their appropriate use in recombinant DNA technology.
- CO4** **apply** recombinant DNA technology in gene expression through blue-white screening.
- CO5** **demonstrate** and apply RDT in Recombinant vaccines Gene therapy, agriculture - plant genetic engineering, herbicide resistant crops etc.

### Catalog Description

Construction Management is the overall planning, coordination and control of a project from inception to completion aimed at meeting a client's requirements in order to produce a functionally and financially viable project. This course includes specific activities like defining the responsibilities and management structure of the project management team, planning methods and implementing it in project controls (time and cost), defining roles and

responsibilities of personnel in the organization, equipments and safety measures in construction. All the lectures will be devoted on discussions of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as power point presentation, audio visual virtual lab session as per requirement. The tutorials will familiarize the students with practical problem-solving techniques led by the course coordinator. Students will strongly grab the basic concepts of the subject via exercise and discussions with the coordinator.

## Course Content

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### UNIT 1 Introduction to Genetic Engineering

2hrs

Milestones in genetic engineering and biotechnology

### UNIT 2 Molecular Cloning- Tools and Strategies

12 hrs

Cloning Tools; Restriction modification systems: Types I, II and III. Mode of action, nomenclature, applications of Type II restriction enzymes in genetic engineering DNA modifying enzymes and their applications: DNA polymerases. Terminal deoxynucleotidyl transferase, kinases and phosphatases, and DNA ligases ; Cloning Vectors: Definition and Properties; Plasmid vectors: pBR and pUC series; Bacteriophage lambda and M13 based vectors; Cosmids, Phagemids, BACs, YACs; Use of linkers and adaptors; Expression vectors: *E.coli* lac and T7 promoter-based vectors, yeast YIp, YEp and YCp vectors, Baculovirus based vectors, mammalian SV40-based expression vectors

### UNIT 3 Methods in Molecular Cloning

12 hrs

Transformation of DNA: Chemical method, Electroporation, Gene delivery: Microinjection, electroporation, biolistic method (gene gun), liposome and viral mediated delivery, *Agrobacterium* (Ti plasmid)- mediated delivery; DNA, RNA and Protein analysis: Agarose gel electrophoresis, PAGE, Southern and Northern blotting techniques, dot blot, DNA microarray analysis, SDS-PAGE and Western blotting.

### UNIT 4 DNA Amplification and DNA sequencing

12 hrs

PCR: Basics of PCR, RT-PCR, Real-Time PCR; Sanger's method and Dideoxy method of DNA Sequencing: traditional and automated sequencing; Primer walking and shotgun sequencing

### UNIT 5 Construction and Screening of Genomic and cDNA libraries

10 hrs

Genomic and cDNA libraries: Preparation and uses, Screening of libraries: Colony hybridization and colony PCR, Chromosome walking and chromosome jumping

### UNIT 6 Applications of Recombinant DNA Technology

12hrs

Products of recombinant DNA technology: Products of human therapeutic interest - insulin, hGH, antisense molecules. Bt transgenic - cotton, brinjal, Gene therapy, recombinant vaccines, protein engineering and site directed mutagenesis.

## TEXT BOOKS

1. Brown TA. (2010). Gene Cloning and DNA Analysis. 6th edition. Blackwell Publishing, Oxford, U.K

## REFERENCE BOOKS

1. Primrose SB and Twyman RM. (2006). Principles of Gene Manipulation and Genomics, 7th edition. Blackwell Publishing, Oxford, U.K.
2. Clark DP and Pazdernik NJ. (2009). Biotechnology: Applying the Genetic Revolution. Elsevier Academic Press, USA
3. Glick, B.R.; Pasternak, J.J.; Patten, C.L. (2010). Molecular biotechnology: Principles and applications of recombinant DNA. ASM Press.

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination  
Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

**Relationship between the Course Outcomes (COs) and Program Outcomes (POs)**

Mapping between COs and Pos		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	<b>Demonstrate</b> the Restriction and modification systems, ligation systems in bacteria and viruses and apply their appropriate use in recombinant DNA technology	PO1, PO2

CO2	<b>Develop</b> the concept about isolation of cloning vectors for prokaryotes and eukaryotes and <b>apply</b> their appropriate use in recombinant DNA technology	PO1, PO2, PO3,
CO3	<b>Demonstrate</b> PCR, DNA sequencing and choose their appropriate use in recombinant DNA technology	PO1, PO2, PO3, PO4
CO4	<b>Apply</b> recombinant DNA technology in gene expression through blue-white screening	PO1, PO5, PO10
CO5	<b>Demonstrate</b> and apply RDT in Recombinant vaccines Gene therapy, agriculture - plant genetic engineering, herbicide resistant crops etc	PO1, PO6, PO8

<b>SDS11507</b>	<b>Elective Statistics II</b>	<b>L</b>	<b>T</b>	<b>P</b>	<b>C</b>
<b>Version 1.0</b>	<b>Contact Hours -90</b>	5	1	0	6
<b>Pre-requisites/Exposure</b>	Statistics I				
<b>Co-requisites</b>	--				

### Course Objectives:

This course aims to build up the advanced knowledge on the basic statistics. Here students will learn more tools and techniques which are useful for analyzing economic issues in real life. The difference between population and sample, why sampling is required for any study, has to be understood clearly before one delves into statistical analysis. In this paper students will get an idea of sampling theory and techniques, sampling distribution and its different forms, test of hypothesis and also learn business index numbers.

### Course Outcomes

On completion of this course, the students will be able to

- CO1. **Develop** the knowledge basic terminologies of sampling, hypothesis testing.
- CO2. **Illustrate** sampling distribution of statistics and test of significance for large sample and small sample.
- CO3. **Explain** several methods of estimation to estimate population parameters.
- CO4. **Classify** several types of index numbers to measure relative changes.

### Catalogue Description:

This course introduces basic concepts and techniques of statistical theory. It emphasizes the intuitive logic that underlies the theory and techniques, and valid interpretation of the results obtained using the techniques.

This course contains sampling techniques, estimation, test of hypothesis and index numbers. All the lectures will be devoted to discussions of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as power point presentation, audio visual virtual lab session as per requirement. The tutorials will familiarize the students with practical problem-solving techniques led by the course coordinator. Students will strongly grasp the basic concepts of the subject via exercise and discussions with the coordinator.

### Course Content:

#### Unit-I

(23 Lecture Hours)

Definitions of random sample, parameter and statistic, null and alternative hypotheses, simple and composite hypotheses, level of significance and probabilities of Type I and Type II errors, power of a test and critical region. Sampling distribution of a statistic, sampling distribution of sample mean, standard error of sample mean.

**Unit-II**

(28 LectureHours)

Large sample tests for single mean, difference of means, standard deviation and difference of standard deviations. Sampling distributions of chi-sq, t and F: definitions, properties and relationships between them. Tests of Significance based on Chi-square (goodness of fit and independence of attributes), t-distribution and F-distribution using classical and p-value approach.

**Unit-III**

(28 Lecturehours)

Methods of estimation: maximum likelihood, least squares and minimum variance, statement of Rao-Blackwell theorem and Lehmann-Schaffer theorem. Properties of maximum likelihood estimators (illustration). Interval Estimation: confidence intervals for the parameters of normal distribution, confidence intervals for difference of mean and for ratio of variances.

**Unit-IV**

(11 LectureHours)

Basic concept of index numbers – simple and weighted index numbers – concept of weights – types of index numbers – Business index number – CPT, WPI, Sensex, Nifty, Production Index.

**Text book**

**T1.** A.M. Goon, M.K. Gupta and B. Dasgupta (2003): *An outline of Statistical Theory* (Vol.I), 4th Ed., World Press, Kolkata.

**T2.** S.C. Gupta and V.K. Kapoor (2007): *Fundamentals of Mathematical Statistics*, 11th Ed., Sultan Chand and Sons.

**Reference book**

**R1.** V.K. Rohtagi and A.K. Md. E. Saleh (2009): *An Introduction to Probability and Statistics*, 2nd Edition, John Wiley and Sons.

**Modes of Examination: Assignment/Quiz/Project/Presentation/Written Exam****Examination Scheme:**

Components	Mid term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40





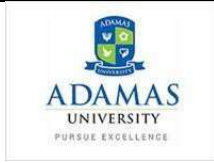
Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
SDS11507	Elective Statistics II		3			3							

1=weakly mapped

2= moderately mapped

3=strongly mapped

### Model Question Paper

<b>Name:</b> <b>Enrolment No:</b>			
<b>Course: Elective Statistics II</b> <b>Program: SDS11507 B.Sc.inBiochemistry(H)</b> <b>Semester: Even 2020-21</b>			
<b>Time: 03Hrs.</b> <b>Max. Marks: 40</b>			
<b>Instructions:</b> Attempt any three questions from <b>Section A</b> (each carrying 5 marks); any <b>Two Questions</b> from <b>Section B</b> (each carrying 10 marks).			
<b>Section A ( Answerall)</b>			
1.	<b>Find</b> the distribution of sample mean taken from a population which follows normal distribution. (R)	<b>5</b>	<b>CO1</b>
2.	A drug given to each of the 12 persons resulted in the following changes in the blood pressure from normal $-3, 2, 8, -1, 3, 0, 7, -2, 1, 5, 0, 4$ . <b>Explain</b> whether the differences insignificant or not. Given, $t_{11, 0.05} = 2.228$ . (U)	<b>5</b>	<b>CO2</b>
3.	<b>Show</b> that if $\hat{\theta}$ is an unbiased estimator of $\theta$ , then it does not imply that $\sqrt{\hat{\theta}}$ is also an unbiased estimator of $\sqrt{\theta}$ . (U)	<b>5</b>	<b>CO3</b>

4.	State and <b>explain</b> Rao-Blackwell theorem and Lehmann-Scheffe theorem. (U)	<b>5</b>	<b>CO3</b>
<b>SECTION B (Attempt any Two Questions)</b>			
5.	<b>Find</b> the MLE of the mean of an exponential distribution? (U)	<b>10</b>	<b>CO3</b>
6.	a) State and <b>explain</b> Type-I error and Type-II error. (R) b) A random sample of size 3 is drawn without replacement from a population of size 5 having units 3, 6, 7, 8, 10. Verify that the sample mean is an unbiased estimator of population mean. And <b>show</b> that the sample variance of sample mean is $\sigma^2 \times \frac{N-n}{N-1}$ , where $\sigma^2$ is the population variance. (U)	<b>4</b> <b>6</b>	<b>CO1</b> <b>CO3</b>
7.	a) A sample of 1600 members is found to have a mean of 5.7. <b>Explain</b> it could be reasonably regarded as a simple sample from a large population whose mean is 4.5 and standard deviation is 2.8? (U) b) <b>Explain</b> business index numbers WPI and Sensex. (U)	<b>6</b> <b>4</b>	<b>CO2</b> <b>CO4</b>

<b>CHM 11153</b>	<b>Elective Chemistry II</b>	L	T	P	C
<b>Version 1.0</b>	Contact Hour: 60	3	1	0	4
<b>Pre-requisites/Exposure</b>	Physics and Chemistry of class 12 or 10+2 level				
<b>Co-requisites</b>	Partial differentiation, model making, graph plotting				

### Course Objectives

1. To introduce important concepts required in the field of the course advanced elective chemistry. This course gives students a thorough understanding regarding fundamental knowledge of various branches of chemistry.
2. To introduce clear understanding of regarding the stabilization of colloidal systems and how solution properties are affected with different dissolutions.
3. To impart the basic notions of chemical equilibrium.
4. To impart detailed descriptions of basic properties of organic molecules and their related reaction mechanism which play major roles in everyday lifecycle.
5. To learn the elementary concepts of acid-base chemistry required for daily life chemistry.
6. To understand the major role of inorganic complexes in living organisms which are very essential concepts in the course curriculum of some disciplines.
7. To identify important tools of different spectroscopic methods required in structure analysis of molecules.

## Course Outcomes for CHM 11153

On completion of this course, the students will be able to

CO-1: Explain fundamental principles and theories for stabilisation/destabilisation of colloidal systems.

CO- 2: Utilize LeChâtelier's Principle, how the equilibrium quantities of reactants and products are shifted by changes in the parameters of the chemical reactions.

CO-3: Develop the knowledge about the properties of solutions that depends on the number of dissolved particles in solution, but not on the identities of the solutes.

CO-4: Explain reasonable mechanism for a chemical reaction.

CO-5: Develop the acid-base concept in aqueous and non-aqueous media and reactions in non-aqueous media.

CO-6: Infer about the fundamental tasks performed by inorganic elements in living organisms as well as the related methods.

CO-7 :Identify the structure of unknown/new compounds with the help of different spectroscopic methods like UV-Visible, IR and NMR spectroscopic Technique.

### Catalogue Description:

This course gives a detailed understanding of the basics of physical, organic, bioinorganic and spectroscopic knowledge required in other disciplines. This course will include expert instructors who will introduce the importance of chemical equilibrium, property of colloidal states, preliminary concepts of organic chemistry, stereochemistry and some and their various mechanisms, basic bioinorganic chemistry and spectroscopic methods required in analysing chemical structures. All the lectures will be devoted on discussions of elementary concepts and cutting-edge topics, focusing on practical implementation of knowledge. Instructors will conduct theory classes by taking lecture as well as power point presentations, audio visual virtual lab sessions as per requirement of the course. After finishing this course, students from different disciplines will strongly grasp the basic concepts of the subject via exercise and discussions with the coordinator.

### Course Content:

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#### Physical Chemistry-II

##### Unit-I: Colloids

(5L)

Colloids and crystalloids; classification of colloids; Preparation and purification of colloids; Properties of colloids: Brownian motion, peptization, dialysis, Tyndal effect and its applications. Protecting colloids, Gold number, Isoelectric points, Coagulation of colloids by electrolytes, Schulze-Hardy rule

##### Unit II: Chemical and Ionic equilibrium

(10 L)

Concept of Gibbs Free Energy; Criteria for thermodynamic equilibrium and spontaneity of a process; Chemical equilibria of homogeneous and heterogeneous systems, Derivation of expression of equilibrium constants; Temperature, pressure and concentration dependence of equilibrium constants ( $K_p$ ,  $K_c$ ,  $K_x$ ); Le Chatelier's principle of dynamic equilibrium. Strong, moderate and weak electrolytes, degree of ionization, factors affecting degree of ionization, ionization constant and ionic product of water. Ionization of weak acids and bases, pH scale, common ion effect; dissociation constants of mono-, di- and triprotic acids (exact treatment). Concept of salt hydrolysis; Buffer solution and buffer capacity.

## Organic Chemistry I

### Unit-I:

#### Fundamentals of Organic Chemistry and Stereochemistry: [5 Lecture Hours]

Functional group-based classification and nomenclature; Sources of origin of different compounds; Concept of hybridization; resonance (including hyperconjugation); inductive effect; steric effect; steric inhibition of resonance. Orbital pictures of bonding ( $sp^3$ ,  $sp^2$ ,  $sp$ : C-C, C-N & C-O system).

Stereochemistry: Symmetry elements, Molecular chirality, Concept of Stereo Centre, Representation of molecules in Fischer projection, Concept of E/Z and Cis-Trans stereo-isomers.

### Unit-II: [3 Lecture Hours]

Mechanistic classification: ionic, radical and pericyclic; heterolytic and homolytic bond cleavage and bond formation; representation of mechanistic steps using formalism. Reactive intermediates: carbocations (benzenium and carbonium ions), Carbanions, Carbon radicals, Carbenes-structure using orbital picture, Electrophilic/nucleophilic behaviour, Stability, generation and fate (elementary idea); Nucleophilic and electrophilic substitution reaction (only  $sp^2$  centre); Introduction to Elimination reaction and its types.

### Unit-III: [7 Lecture Hours]

Basic Organic Reactions: (Addition, Substitution, Elimination, Rearrangement Reactions) Addition Reactions: Halogenations, Hydration, Hydrogenation, Epoxidation, Hydroxylation, Ozonolysis, electrophilic addition to diene; Hydroboration-oxidation reaction; Radical addition: HBr addition, Birch Reduction. Nucleophilic addition to carbonyl group.

Substitution Reactions:  $S_N1$ ,  $S_N2$ , NGP, Elimination Reactions:  $E1$ ,  $E2$ , Elimination vs. Substitution, Rearrangement Reactions: Rearrangement to electron-deficient carbon: Wagner-Meerwein rearrangement, pinacol-pinacolone and related rearrangements, dienone-phenol.

## Inorganic Chemistry-II:

### Unit-I: [7 Lecture Hours]

Acids-Bases and Redox: Bronsted- Lowry concept of acid-base reaction, solvated proton, types of acid-base reactions, levelling solvents, Lewis acid-base concept, Classification of Lewis acids, Hard and Soft Acids and Bases (HSAB) Application of HSAB principle. Theory of acid-base titration and significance of Acid-base indicators. Common ion effect ; Ion-electron method of balancing equation of redox reaction. Elementary idea on standard redox potentials with sign conventions; Nernst equation (without derivation); redox indicators

### **Unit-II:**

#### **Bioinorganic Chemistry:**

**[8 Lecture Hours]**

Elements of life: essential major, trace and ultra-trace elements; Basic chemical reactions in the biological systems and the role of metal ions (specially  $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{Mg}^{2+}$ ,  $\text{Ca}^{2+}$ ,  $\text{Fe}^{3+}$ ,  $\text{Fe}^{2+}$ ,  $\text{Cu}^{2+}$ ,  $\text{Zn}^{2+}$ ); Biological functions of haemoglobin and myoglobin.

#### **Spectroscopy: [15 Lecture Hours]**

##### **Unit-I:**

UV-Vis Spectra: Electronic transition, relative positions of  $\lambda_{\text{max}}$ , Woodward's empirical rule; Lambert-Beers Law.

##### **Unit-II:**

IR Spectra: Modes of molecular vibrations, application of Hooke's law, characteristic stretching frequencies and factors effecting stretching frequencies.

##### **Unit-III:**

NMR Spectra: Preliminary idea of NMR, Nuclear spin, NMR active nuclei, Equivalent and non-equivalent carbons and protons; Chemical shift  $\delta$ ; Shielding deshielding, Upfield and Downfield shifts.

##### **Unit-IV:**

Photochemistry: Fluorescence and phosphorescence; Quantum Yield; Jablonsky diagram

#### **Reference Books:**

Physical Chemistry:

1. D. A. Mcquarrie and J. D. Simon: Physical Chemistry — A Molecular Approach
2. G. W. Castellan: Physical Chemistry
3. P. W. Atkins: Physical Chemistry

Organic Chemistry:

1. D. Nasipuri: Stereochemistry of organic compounds: Principles and Applications
2. P. Sykes: A Guide to Mechanism in Organic Chemistry

3. R. T. Morrison and R. N. Boyd: Organic Chemistry

Inorganic Chemistry

1. Bioinorganic Chemistry. Asim K. Das.

Spectroscopy.

2. Organic Spectroscopy. William Kemp.

**Modes of Examination: Assignment/Quiz/Project/Presentation/Written Exam**

**Examination Scheme:**

Components	Mid term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

**Relationship between the Course Outcomes (COs) with Program Outcomes (POs) and Program Educational Outcomes (PEOs)**

**Mapping between COs, POs and PEOs**

CO-1	<b>Explain</b> fundamental principles and theories for stabilisation/destabilisation of colloidal systems.	<b>PO1, PO2, PO4</b>
CO-2	<b>Utilize</b> LeChâtelier's Principle, how the equilibrium quantities of reactants and products are shifted by changes in the parameters of the chemical reactions.	<b>PO2, PO6, PO9</b>
CO-3	<b>Develop</b> the knowledge about the properties of solutions that depends on the number of dissolved particles in solution, but not on the identities of the solutes.	<b>PO1, PO2, PO3</b>
CO-4	<b>Explain</b> reasonable mechanism for a chemical reaction.	<b>PO1, PO5</b>
CO-5	<b>Develop</b> the acid-base concept in aqueous and non-aqueous media and reactions in non-aqueous media.	<b>PO2, PO6</b>
CO-6	<b>Infer</b> the fundamental tasks performed by inorganic elements in living organisms as well as the related methods.	<b>PO1, PO2, PO5</b>
CO-7	<b>Identify</b> the structure of unknown/new compounds with the help of different spectroscopic methods like UV-Visible, IR and NMR spectroscopic Technique.	<b>PO2, PO9</b>



Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
<b>CHM 11153</b>	Elective Chemistry II	Fundamental Knowledge											
		Critical Thinking	3										
		Skill		3									
		Technical Knowledge			1								
		Logical Thinking Ability				1							
		Problem Identification					2						
		Analytical Knowledge						2					
		Career Goals							-				
		Team Work								-			
		Sustainable Development to Environment									2		
		Development to Society										-	
		Development to Humanity											-


1=weakly mapped

2= moderately mapped

3=strongly mapped



## Model Question Paper

	<b>Name:</b> <b>Enrolment No:</b>		 <small>ADAMA UNIVERSITY PAROLE EXCELLENCE</small>	CO
	<b>Course: Elective Chemistry II</b> <b>Program: B.Sc.Biochemistry</b> <b>Semester: Even'2020-21</b>	<b>Time: 03Hrs.</b> <b>Max. Marks: 40</b>		
1	a) <b>Change</b> 1-propene to 1-propanol and 2-propanol. b) <b>Outline</b> the mechanism and the rate law of E2reaction.	3+1	(Un/Cor)	CO3
2	<b>Distinguish</b> between SN1 and SN2 reactions showing the mechanism of both. b) What are enantiomers and diastereoisomers? <b>Explain</b> with examples.	3+4	(An)	CO3, CO4
3	a) <b>What</b> is configuration and conformation? Explain with examples. b) <b>Explain</b> the mechanism of formation of nitrobenzene from benzene.	1+2+2	(Re/Ap)	CO3
4	<b>Outline</b> the properties of a stable colloidal solution. Is soap a colloid?	3	(Un)	CO1
5	<b>Summarise</b> the effects of temperature and pressure on a chemical equilibrium.	2+2	(Un)	CO2
6	<b>Why</b> is aniline a weaker base compared to methyl amine? What is optical rotation?	2+2	(Re)	CO3, CO4
7	<b>Compare</b> the structural differences between haemoglobin and myoglobin.	2	(An)	CO5
8	<b>What</b> is Bohr effect? <b>Relate</b> the functional differences of haemoglobin and cytochrome C.	2+3	(Re/Un)	CO6
9	<b>Outline</b> Lambert-Beers law. <b>Explain</b> the required characteristics of a molecule to execute UV-Vis spectra.	2+3	(Un)	CO7
10	<b>Why</b> pyridine is more basic than aniline? <b>Tell</b> the required characteristics of a molecule to show chirality.	3+2	(Re)	CO4

Course Code CHM12154	Course Name Elective Chemistry II lab	L	T	P	Credit
		0	0	3	2
Course Outcome	Course Outcome Statement				
<b>Course Content:</b>					
<ol style="list-style-type: none"> <li>1. To study the kinetics of inversion of sucrose using polarimeter.</li> <li>2. Experiment A: Detection of special elements (N, Cl, and S) in organic compounds. Experiment B: Solubility and Classification (solvents: H<sub>2</sub>O, dil. HCl, dil. NaOH) <ol style="list-style-type: none"> <li>a. Experiment C: Detection of functional groups -NO<sub>2</sub>, -NH<sub>2</sub>, -COOH, carbonyl</li> <li>b. (-CHO, &gt;C=O), -OH (phenolic) in solid organic compounds.</li> </ol> </li> </ol>					
<b>Text Book:</b>					
<b>Reference Books:</b>					
<b>Organic Chemistry:</b>					
<ol style="list-style-type: none"> <li>1. D. Nasipuri: Stereochemistry of organic compounds: Principles and Applications</li> <li>2. P. Sykes: A Guide to Mechanism in Organic Chemistry</li> <li>3. J. March: Advanced Organic Chemistry</li> <li>4. I. L. Finar: Organic Chemistry (Vol. I),37</li> <li>5. R. T. Morrison and R. N. Boyd: Organic Chemistry</li> </ol>					

<b>CSE11643</b>	<b>Elective Computer Science II</b>	L	T	P	C
<b>Version 1.0</b>	Contact Hours:60	4	0	0	4
<b>Pre-requisites/Exposure</b>	Basics of computer knowledge				
<b>Co-requisites</b>					

### Course Objectives

1. To understand the basic procedural programming skills.
2. To apply the conditional constructs to solution of different problems.
3. To understand the iterative way of programming.
4. To design solution to different problem scenarios using the programming constructs.

### Course Outcomes

On completion of this course, the students will be able to

CO1: develop the concept about a functional hierarchical code organization.

CO2: define and manage data structures based on problem subject domain.

CO3: experiment with textual information, characters and strings.

CO4: design algorithms to solve simple problems.

CO5: demonstrate the ability to correct, test and debug Processing programs.

### Catalog Description

This course introduces the students to the basics of programming skills. It familiarizes them with the procedural programming approach and design solutions to problems using conditional constructs, iterative programming and functions. They learn to create user-defined functions to solve different problem scenarios and grasp knowledge about string handling functions.

### Course Content

#### Unit I:

**8 Lecture Hours**

**Basics of C Programming:** Characters used in C, Identifiers, Keywords, Data type & sizes, Constants & Variables, Different types of Operators, Standard Input/output functions.

#### Unit II:

**8 Lecture Hours**

**Control Flow:** Control Flow, if-else, switch-case, Loop Control Statements, for loop, while loop, do-while loop, nested loop, break, continue, goto label and exit( ) function.

**Unit III:**

**15 Lecture Hours**

**Arrays:** Definition, Single and Multidimensional Arrays, Representation of Arrays - Row Major Order, and Column Major Order, Application of arrays – searching and sorting, Sparse Matrices and their representations.

**Unit IV:**

**9 Lecture Hours**

**String:** Definition of a String, Declaration of a String, Initialization of a String, Various String Handling Functions with example.

**Unit V:**

**10Lecture Hours**

**Pointers:** Definition of Pointer, Declaration of Pointer, Operators used in Pointer, Pointer Arithmetic, Functions with Pointer.

**UnitVI:**

**10 lecturehours**

**Functions:**BasicConceptofFunction,DeclarationorPrototypeofFunction,TypesofFunctions, CallbyValue,CallbyReference,Recursion,TailRecursion.

**Reference Books**

**List of Books:**

1. “The C Programming Language”, 2nd Edition, Brian W. Kernighan, Dennis M. Ritchie,PHI
2. “Schaum'sOutline of Programming with C”, 2nd Edition, Byron S. Gottfried, McgrawHill Education
3. “TheCompleteReference”, 4thEditionbyHerbertSchildt, TataMcgrawHillEducation

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

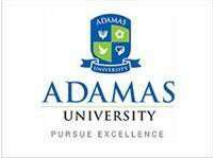
## Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and POs		
	Course Outcomes (COs)	Mapped Program Outcomes
<b>CO1</b>	<b>Develop</b> concept a functional hierarchical code organization.	<b>PO5</b>
<b>CO2</b>	<b>Define</b> and manage data structures based on problem subject domain.	<b>PO8,PO12</b>
<b>CO3</b>	<b>Experiment with</b> textual information, characters and strings.	<b>PO8,PO5</b>
<b>CO4</b>	<b>Design</b> algorithms to solve simple problems.	<b>PO5,PO8</b>
<b>CO5</b>	<b>Demonstrate</b> the ability to correct, test and debug Processing programs.	<b>PO5</b>

Course Code	Course Title	Fundamental Knowledge	Critical Thinking	Skills	Technical Knowledge	Logical Thinking	Problem identification ability	Analytical Knowledge	Career goals	Team Work	Sustainable Development to environment	Development to society	Development to humanity
		PO 1	PO 2	PO 3	PO 4	PO 5	PO 6	PO 7	PO 8	PO 9	PO10	PO1 1	PO1 2
<b>CSE1164 3</b>	Elective Computer Science II	-	-	-	-	2	-	-	3	-	-	-	1

1 = weakly mapped

2 = moderately mapped 3 = strongly mapped

<b>Name:</b>			
<b>Enrolment No:</b>			
<b>END-SEMESTER EXAMINATION (DECEMBER 2019)</b> (Academic Session: 2019 – 20, Semester Term: Aug. 2019 – Dec. 2019)			
<b>Name of the Program: B.Sc. Biochemistry</b>		<b>Semester: Even'2020-21</b>	
<b>PAPER TITLE: Elective Computer Science II</b>		<b>PAPER CODE: CSE11643</b>	
<b>Maximum Marks: 40</b>		<b>Time duration: 3 hours</b>	
<b>Instructions:</b>			
<b>Attempt any three questions from Section A (each carrying 4 marks); any four from Section B (each carrying 7 marks).</b>			
<b>SECTION A ( Attempt any Three questions)</b>			
1	<b>What</b> is a data type? <b>Explain</b> about any two data types.(Re/Un)	2 2	CO1
2	<b>Explain</b> nested if else statement with example.(Un)	4	CO2
3	<b>Explain</b> a C program to display the reverse of a number. <b>What</b> is ternary operator?(Re/Un))	3 1	CO4
4	<b>What</b> is the use of break statement? <b>Explain</b> with a suitable example(Re/Un).	2 2	CO2
<b>SECTION B (Attempt any Four questions)</b>			
5	a) <b>Illustrate</b> a C program to find the largest among three numbers. b) <b>What</b> is string? What is the use of strlen() function?(Re/Un)	3 2 2	CO3
6	a) <b>What</b> is a pointer? <b>Explain</b> with suitable example. b) <b>Elaborate</b> Pointer arithmetic.(Re/Un/Cr)	3 2 2	CO3
7	a) <b>Distinguish</b> between row major and column major array representation. b) <b>Outline</b> a C program to swap two variables without using third variable. c) <b>Outline</b> a C program to print the elements of a one-dimensional array. (An/Un)	2 3 2	CO3, CO5
8	a) <b>Outline</b> a C program to search for an element in a given array. b) <b>What</b> is a two-dimensional array? Give an example. c) <b>What</b> is a function?(Re/Un)	3 2 2	CO4, CO5
9	a) <b>Distinguish</b> between call by value and call by reference method of passing the parameters. b) <b>Construct</b> a C program to check whether a number is palindrome by using user-defined function.(An/Ap)	3 4	CO1, CO2, CO5





<b>CSE12644</b>	<b>Elective computer science II Lab</b>	L	T	P	C
<b>Version 1.0</b>	Contact Hours: 45	0	0	3	2
<b>Pre-requisites/Exposure</b>	Basics knowledge about Elective Computer Science II Theory Paper				
<b>Co-requisites</b>					

### Course Objectives

1. To understand the basic procedural programmingskills.
2. To apply the conditional constructs to solution of different problems.3.
- To understand the iterative way ofprogramming.
4. To design solution to different problem scenarios using the programmingconstructs.

### Course Outcomes

On completion of this course, the students will be able to

CO1: identify situations where computational methods and computers would be useful.

CO2: solve a computational problem, identify and abstractthe programming task involved.

CO3: formulate the programming tasks using techniques learned and write pseudo-code.

CO4:choosetherightdatarepresentationformatsbasedontherequirementsoftheproblem.

CO5:outlinetheprogramonacomputer,edit,compile,debug, correct,recompileandrunit.

CO6: identify tasks in which the numerical techniques learned are applicable and apply them to write programs, and hence use computers effectively to solve the task.

### Catalog Description

This course introduces the students to the basics of programming skills. It familiarizes them with the procedural programming approach and design solutions to problems using conditional constructs, iterative programming and functions. They learn to create user-defined functions to solve different problem scenarios and grasp knowledge about string handling functions.

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## Course Content

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### List of experiments:

1. Introduction to C Programming [10 lecturehours]
2. C program to implement different aspects of Control Flow[10 lecture hours]3.  
C program to implement different aspects of Arrays[10 lecturehours]
4. C program to implement different aspects of String[10 lecture hours]
5. C program to implement different aspects of Pointers[10 lecturehours]
6. C program to implement different aspects of Functions[10 lecture hours]

### Reference Books

#### List of Books:

1. “TheCProgrammingLanguage”,2ndEdition, BrianW.Kernighan,DennisM. Ritchie,PHI
2. “Schaum'sOutline of Programming with C”, 2nd Edition, Byron S. Gottfried, McGrawHill Education
3. “TheCompleteReference”,4thEdition byHerbertSchildt, TataMcgrawHillEducation

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written ExaminationExamination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

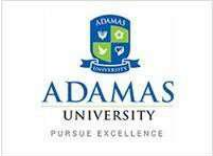
### Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and POs		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	<b>Identify</b> situations where computational methods and computers would be useful.	PO1,PO3,PO5
CO2	<b>Solve</b> a computational problem, identify and abstract the programming task involved.	PO3,PO5,PO7
CO3	<b>Formulate</b> the programming tasks using techniques learned and write pseudo-code.	PO1,PO3,PO5
CO4	<b>Choose</b> the right data representation formats based on the requirements of the problem.	PO1,PO3,PO5
CO5	<b>Outline</b> the program on a computer, edit, compile,	PO1,PO3,PO5

	debug, correct, recompile and run it.	
<b>CO6</b>	<b>Identify</b> tasks in which the numerical techniques learned are applicable and apply them to write programs, and hence use computers effectively to solve the task.	<b>PO3,PO5,PO7,PO8</b>

		Fundamental Knowledge	Critical Thinking	Skills	Technical Knowledge	Logical Thinking	Problem identification ability	Analytical Knowledge	Career goals	Team Work	Sustainable Development to environment	Development to society	Development to humanity
Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
<b>CSE12644</b>	Elective Computer Science II Lab	3	-	3	-	3	-	2	2	-	-	-	-

- 1 = weakly mapped  
2 = moderately mapped  
3 = strongly mapped

<b>Name:</b>  <b>Enrolment No:</b>			
<b>END-SEMESTER EXAMINATION (DECEMBER 2019)</b> (Academic Session: 2019 – 20, Semester Term: Aug. 2019 – Dec. 2019)			
<b>Name of the Program: B.Sc. Biotechnology</b> <b>PAPER TITLE: Elective computer science IILab</b> <b>CSE12644 Maximum Marks: 40</b> <b>Instructions:</b> <b>Attempt any two questions from Section A (each carrying 10 marks); Section B is Compulsory (carrying 10 marks).</b>		<b>Semester: Even 2020-21</b> <b>PAPER CODE:</b> <b>Time duration: 3 hours</b>	
<b>Section A ( Attempt any Two)</b>			
1	a. To <b>change</b> the temperature from Celsius to Fahrenheit. b. To <b>predict</b> whether a number is odd or even. c. To <b>show</b> the elements of an 1D array. (Un/Cr)	3 3 4	CO3,C 05,CO 6,CO1, CO2
2	a. To print the reverse of a number. b. To <b>evaluate</b> whether a string is palindrome or not. c. To swap two variables using third variable (Ev)	3 3 4	CO3,C 05,CO 6,CO1, CO2
3	a. To print the largest among three numbers. b. To <b>find</b> for a given element in an array. c. To print the sum of first n natural numbers (Re)	3 3 4	CO3,C 05,CO 6,CO1, CO2
4	a. To <b>utilize</b> user defined function to find the Fibonacci series upto n terms. b. To <b>show</b> the factorial of a given number. c. To <b>find</b> the average of an array of integers. (Un/ Ap)	4 3 3	CO3,C 05,CO 6,CO1, CO2,C 04
<b>SECTION B is compulsory</b>			
5	Viva-voce (U/An/ Ap/R/Ev)	10	CO1 CO2 CO3 CO4 CO5

			<b>CO6</b>
6	Practical copy (U/Ap/Ev)	10	<b>CO1</b> <b>CO2</b> <b>CO3</b> <b>CO4</b> <b>CO5</b> <b>CO6</b>

<b>PSG11021</b>	<b>Human Values and Professional Ethics</b>	<b>L</b>	<b>T</b>	<b>P</b>	<b>C</b>
<b>Version 1.0</b>	<b>Contact hours: 30</b>	<b>2</b>	<b>0</b>	<b>0</b>	<b>2</b>
<b>Pre-requisites/Exposure</b>	--				
<b>Co-requisites</b>	--				

## Course

### Objectives

- To inculcate human values and professional ethics in students.
- To enhance the understanding of students towards personal, professional & societal relationships and achieve harmony in life.
- To develop moral responsibilities and ethical vision.

## Course

### Outcomes

At the completion of the course, the students will be able to:

CO1. **Show** the importance of values, ethics, harmony and lifelong learning in personal and professional life.

CO2. **Utilize** the knowledge to perform self-exploration and transformation augmenting harmony, peace and positivity in the surroundings.

CO3. **Appraise** the core values that shape the ethical behaviour of a professional

### Catalog Description

This course aims to develop an understanding for a movement from rule based society to a relationship based society. Apart from teaching values, this course encourages students to discover what values are for them and for society. Self-explorational so enables the mtocritically evaluate the irpre-conditionings and present beliefs. It is designed in a way where students get familiar with the Ethical Code of Conduct, Ethical Dilemma, Conflict of Interest and all this will help them eventually in their professional life.

### Course Content

#### Unit I: Introduction to Human [7 hours Lecture]

#### Values:

Character, Integrity, Credibility, Mutual Respect, Dedication, Perseverance, Humility and Perception  
 .Self-Assessment & Analysis, Setting Life Goals, Consciousness and Self-

Transformation. Team Work, Conflict Resolution, Influencing and Winning People, Anger Management, Forgiveness and Peace, Morality, Conscience. Yoga and Spirituality

**Unit II: Harmony and Life Long Learning: [7 hours Lecture]**

Harmony in human being, Nature and Existence. Harmony in family and society – Responsibilities towards society, Respecting teachers. Transition from School to College – Freedom & Responsibilities, Respecting Cultural Diversity, Learning beyond the Classrooms, Independent study and research

**Unit III: Introduction to Professional Ethics: [7 hours Lecture]**

Work Ethics, Engineering Ethics, Moral Dilemma, Moral Development Theories, Ethical Theories – Kantianism, Utilitarianism, etc, Case Studies for Choice of the theory, Code of Ethics

**Unit IV: Individual to Global Issues: [7 hours Lecture]**

Industrial Standards, A Balanced Outlook on Law, Safety, Responsibility, Rights, Confidentiality, Conflict of Interest, Occupational Crime, Whistle Blowing, Environmental Ethics, Business Conduct in MNC, E-Professionalism (IPR, Internet Ethics & Privacy issues)

**Text Books**

2. Shetty, Foundation Course in Human Values and Professional Ethics [R.R. Gaur, R. Sangal, G.P. Bagaria]



**Modes of Evaluation: Quiz/Assignment/ Seminar/Written Examination**

**Scheme:**

<b>Components</b>	<b>MTE</b>	<b>Attendance</b>	<b>Presentation/Assignment/ etc</b>	<b>ETE</b>
<b>Weightage (%)</b>	<b>20</b>	<b>10</b>	<b>20</b>	<b>40</b>

**Relationship between the Course Outcomes (COs) and Program Outcomes (POs)**

<b>Mapping between COs and POs</b>		
	<b>Course Outcomes (COs)</b>	<b>Mapped Programme Outcomes</b>
<b>CO1</b>	<b>Show</b> the importance of values, ethics, harmony and lifelong learning in personal and professional life	<b>PO6, PO8</b>
<b>CO2</b>	<b>Utilize</b> the knowledge to perform self-exploration and transformation augmenting harmony, peace and positivity in the surroundings	<b>PO6, PO8</b>
<b>CO3</b>	<b>Appraise</b> the core values that shape the ethical behaviour of a professional	<b>PO8</b>

Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	
<b>PSG11021</b>	Human Values and Ethics	Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking	Problem Identification Ability	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to Environment	Development to Society.	Development to Humanity	
		2												
										3				

1=weakly mapped

2= moderately mapped

3=strongly mapped

**MODEL QUESTION**

**Course: PSG11021 - Human Values and Professional Ethics**

**Programme: UGAllprogram**

**Semester: I Time: 03 hrs.**

**Max.Marks:60**

**Instructions:**

Attempt any **Four Questions** from **Section A** (each carrying 6 marks); any **Two Questions** from **Section B** (each carrying 10 marks). **Section C** is Compulsory (carrying 16 marks).

**SECTION A (Attempt any Four Questions)**

1.	What do you mean by happiness and Prosperity? Critically examine the prevailing notions of happiness in the society and their consequences. (Ap)	<b>[06]</b>
2.	How do the current world views lead to contradictions and dilemmas in professional life? – Explain. (An)	<b>[06]</b>
3.	What do you mean by ‘Universal Human Order’? (U)	<b>[06]</b>
4.	“Physical facilities are necessary and complete for animals, while they are necessary but not complete for humans.” Comment. (An)	<b>[06]</b>
5.	Why do you think that there should be emphasis on Life Long Learning in the current academic setting? (Ap)	<b>[06]</b>

**SECTION B (Attempt any Two Questions)**

6.	Critically examine the issues in professional ethics in the current scenario. List any five unethical practices in profession today and the methods being tried to curb them. (Ap)	<b>[10]</b>
7.	What are the implications of value based living at all four levels of living? Explain. (Ap)	<b>[10]</b>
8.	Discuss the Basic Aspects and Characteristic Features of Kohlberg’s Theory and Gilligan’s Theory. (U)	<b>[10]</b>

**SECTION C is Compulsory**

9.	<p><b>Case Study VI HI FI Hose Company</b></p> <p>Anhydrous ammonia is used to fertilize the crops. The anhydrous ammonia reacts violently with water. Pressurized tanks provided with wheels carry this fertilizer, and tanks are pulled by tractors. Farmers take these tanks on rent. They take on rent or purchase the hose to carry this ammonia from the tank to perforated blades that dig into the soil and spread ammonia. Leaks from the hose are very dangerous.</p> <p>In the past, the hoses were made of steel-mesh reinforced rubber, which were similar to automobile tyres. Later, the reinforced-plastic hoses were introduced and they satisfied the standards. The VI HI FI has been marketing these hoses to the farmers. The officials of the company arranged for testing the hoses as a consultancy work in the Agricultural College. The tests indicated that the plastic did not react initially to the anhydrous</p>	<b>[8+8]</b>
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	<p>ammonia. But over the years, the plastic was found to degrade and lose some mechanical properties. Hence, the company attached warnings on all the hoses, indicating that they should be replaced periodically.</p> <p>After a few years of use of the product in the market, several accidents occurred where the hoses ruptured during use and severely injured and blinded the farmers. Legal action followed and the company argued in defense that the farmers had misused the hoses and not heeded the replacement warnings. But they have to make substantial out-of-court settlements. The company then dropped the product line and advertised in the press asking the farmer to turn in their hoses for full refunds. The advertisement stated that the hoses are 'obsolete', and not that a unsafe.</p>	
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(a) What are the factual, conceptual and normative issues? (An)

(b) What are the methods suggested for resolving

**Course Objectives**

<b>CHM 11151</b>	<b>Elective Chemistry I</b>	L	T	P	C
<b>Version 1.0</b>	Contact Hours: 60	3	1	0	4
<b>Pre-requisites/Exposure</b>	Physics and Chemistry of class 12 or 10+2 level				
<b>Co-requisites</b>	Partial differentiation, model making, graph plotting				

1. To introduce important concepts required in the field of the course elective chemistry. This course gives students a thorough understanding regarding the prerequisites of basic chemistry knowledges in their course curriculums.
2. To introduce clear understanding of energy conditions necessary to execute a feasible chemical reaction.
3. To impart the basic notions of different properties of liquid states of chemical compounds and their effects with atmosphere.
4. To impart the concepts required for kinetics of a reaction mechanism and deeper understanding of the molecular interactions which can influence chemical reactivity. Students can understand the various kinds of reaction mechanisms occurring in their daily lifecycle.
5. To learn the basic understanding of atomic structure of molecules important in our daily life and how nuclear reactions are pertinent to their structure.
6. To conceptualize the essence of molecular bonding of necessary molecules.

### Course Outcomes

On completion of this course, the students will be able to

CO-1. Identify the exclusive terminologies associated with thermodynamics and explain the basic concepts of thermodynamics i.e. heat transfer and its consequences with the thermodynamic system.

CO- 2 . Explain the difference between what the molecules are doing in a solid, liquid, and gas, including movement, spacing, and organization, and how this explains the physical characteristics of these states.

CO-3. Summarise the properties of solutions that depends on the number of dissolved particles in solution, but not on the identities of the solutes.

CO-4..Develop the concept of reaction rates and be able to use to predict products, yields etc.

CO-5.Distinguish between the symbols for protons, neutrons, electrons, positrons, alpha particles, beta particles, and gamma rays.

CO-6 .Identify the periodic properties of elements, principles in molecular theory and bonding models to the study of inorganic compound.

### Catalogue Description:

This course gives a detailed understanding of the basics of physical and inorganic chemistry required in other disciplines. This course will include expert instructors who will introduce thermodynamics of chemical reaction, colligative properties of liquid states, the structures of nucleus and subatomic particles and their relations with the chemical properties and especially molecular bonding of important molecules of our daily life. All the lectures will be devoted on discussion of elementary concepts and cutting-edge topics, focusing on practical implementation of knowledge. Instructors will conduct theory classes by taking lecture as well as power point

presentations, audio visual virtual lab sessions as prerequisite of the course. The tutorials and required assignments will acquaint the students with practical problem-solving techniques led by the course coordinator. After finishing this course, students from different disciplines will strongly grasp the basic concepts of the subject via exercise and discussions with the coordinator.

## **Course Content:**

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### **Physical Chemistry-I**

#### **Unit- 1-Thermodynamics [10 Lecture Hours]**

Thermodynamics: Definition of thermodynamic terms; Concept of heat and work; First law of thermodynamics; Concept of enthalpy (H); Expansion of ideal gas under isothermal and adiabatic conditions for reversible and irreversible processes; Concept of standard state, Standard enthalpy changes of physical and chemical transformations: fusion, sublimation, vaporization, ... - solution, dilution, neutralization, ionization.; Hess's law of constant heat summation, Second law of thermodynamics; Heat engine; Carnot cycle and its efficiency; Entropy (S) as a state function. Spontaneous processes; Concept of Free Energy (G and A).

#### **Unit-2-Liquid state: [4 Lecture Hours]**

Liquid States and Viscosity of Fluids: Nature of the liquid state (short range order and long range disorder); Physical properties of liquids; Vapor pressure, Surface tension, Surface energy, General features of fluid flow (streamline flow and turbulent flow); Coefficient of viscosity and their determination.

#### **Unit-3-Colligative properties [6 Lecture Hours]**

Colligative Properties: What are colligative properties? Dependence of colligative properties; Freezing point depression; boiling point elevation, Raoult's Law and Vapor Pressure Lowering ; osmotic pressure.

#### **Unit-4:Chemical kinetics [10 Lecture Hours]**

Chemical kinetics and catalysis: Order and molecularity of reactions; Rate laws and rate equations for first order and second order reactions (differential and integrated forms); Zero order reactions; Determination of order of reactions; Temperature dependence of reaction rate, energy of activation; Catalytic reactions: homogeneous and heterogeneous catalytic reactions. Enzyme catalysis

### **Inorganic Chemistry-I**

## **Unit-I: Atomic structure [10 Lecture Hours]**

Extra-nuclear Structure of atoms, Bohr's model. quantum numbers and their significance, Pauli's exclusion principle, Hund's rule, electronic configuration of many-, electron atoms, Aufbau principle.

## **Unit-II: Chemical Periodicity [5 Lecture Hours]**

Classification of elements on the basis of electronic configuration; Position of hydrogen and noble gases; Atomic and ionic radii; ionization potential; electron affinity; and electronegativity; periodic and group-wise variation of above properties in respect of s- and p- block elements.

## **Unit-III: Radioactivity and Nuclear Structure of Atoms [5 Lecture Hours]**

Natural radioactivity, group displacement law, law of radioactive decay, half-life of radio elements. Atomic Nucleus: Stability of atomic nucleus, nuclear binding energy, Nuclear reactions: fission, fusion, transmutation of elements.

## **Unit-IV: Chemical Bonding [10 Lecture Hours]**

Ionic Bonding: General characteristics of ionic compounds; Lattice energy; Born Haber cycle. Covalent bonding: General characteristics of covalent compounds; valence-bond approach, directional character of covalent bond; hybridization involving s-, p-, d orbitals; multiple bonding; Valence Shell Electron Pair Repulsion (VSEPR) concept; Partial ionic character of covalent bonds; Fajan's rules. Hydrogen bonding and its effect on physical and chemical properties.

### **Reference Books:**

1. D. A. McQuarrie and J. D. Simon: Physical Chemistry — A Molecular Approach
2. G. W. Castellan: Physical Chemistry
3. P. W. Atkins: Physical Chemistry
4. J. E. Huheey, E. A. Keiter, R. L. Keiter: Inorganic Chemistry (Principle and structure and reactivity).
5. N. N. Greenwood, A. Earnshaw: Chemistry of the Elements
6. D. F. Shriver, P. W. Atkins, C. H. Langford: Inorganic Chemistry

**Modes of Examination: Assignment/Quiz/Project/Presentation/Written Exam**

**Examination Scheme:**

Components	Mid term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

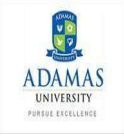
**Relationship between the Course Outcomes (COs) with Program Outcomes (POs) and Program Educational Outcomes (PEOs)****Mapping between Cos and POs**

CO-1	<b>Identify</b> to recognize the exclusive terminologies associated with thermodynamics and explain the basic concepts of thermodynamics i.e. heat transfer and its consequences with the thermodynamic system.	<b>PO1, PO2, PO4</b>
CO-2	<b>Explain</b> the difference between what the molecules are doing in a solid, liquid, and gas, including movement, spacing, and organization, and how this explains the physical characteristics of these states.	<b>PO9, PO2,</b>
CO-3	<b>Summarise</b> the properties of solutions that depends on the number of dissolved particles in solution, but not on the identities of the solutes.	<b>PO1, PO3, PO2</b>
CO-4	<b>Develop</b> the concept of reaction rates and be able to use to predict products, yields etc.	<b>PO5, PO1, PO9</b>
CO-5	<b>Distinguish</b> between of using the symbols for protons, neutrons, electrons, positrons, alpha particles, beta particles, and gamma rays.	<b>PO6, PO2</b>
CO-6	<b>Identify</b> periodic properties of elements, principles in molecular theory and bonding models to the study of inorganic compound.	<b>PO1, PO2, PO5, PO9</b>





## Model Question Paper

	<b>Name:</b> <b>Enrolment No:</b>	 <small>ADAMAS UNIVERSITY PURSUE EXCELLENCE</small>	
	<b>Course: Elective Chemistry I</b> <b>Program: B.Sc.Biochemistry</b> <b>Semester: Odd 2020-21</b>	<b>Time: 03Hrs.</b> <b>Max. Marks:40</b>	
	<b>Instructions:</b> Attempt any five questions from Section A (each carrying 5 marks); all Questions from Section B (each carrying 5 marks)		
	<b>Section A</b>		
1.	<b>Show</b> the radial probability distribution function diagram of the orbitals of 3s, 3p and 3d in a Hydrogen atom. Explain number of radial nodes of 3s and 4p orbitals.	(Ap/Un)	CO5
2.	<b>Define</b> Hund's rule of maximum spin multiplicity and using this write down the electronic configurations of the elements having atomic no 14 and 22? 4+1 What are the possible values of ml (magnetic quantum number) for l=1?	(Un)	CO5
3.	<b>Measure</b> the difference in electronegativity of carbon atom in C <sub>2</sub> H <sub>6</sub> , C <sub>2</sub> H <sub>4</sub> and C <sub>2</sub> H <sub>2</sub> . Explain the hybridisation of CH <sub>4</sub> .	(Ev)	CO6
4.	<b>Explain</b> Pauli's Exclusion Principle and apply this to predict the maximum capacity of s-subshell for accommodating e <sup>-</sup> . <b>Relate</b> between momentum of a particle & the wavelength associated with its wave.	(Un/An)	CO6
5.	<b>Explain</b> the shapes of SOF <sub>4</sub> , IF <sub>2</sub> <sup>-</sup> and XeF <sub>2</sub> according to VSEPR principle. b) <b>Why</b> F <sub>2</sub> O has smaller <FOF angle than H <sub>2</sub> O?	(Re/Un)	CO5, CO6
6.	<b>Show</b> the d orbital wave the d orbital wave functions of dxz, dyz and dz <sup>2</sup> orbitals. <b>Explain</b> the hybridizations of H <sub>2</sub> O and NH <sub>3</sub> molecules.	(Un)	CO6
7.	Electron affinity of carbon is greater than boron- <b>Explain? Why</b> does atomic radii increase down the group and decrease across a period? (AP)	(Re/Un)	CO5, Co6
	<b>Section B</b>		
8.	<b>Explain</b> Carnot cycle. <b>What</b> is the prerequisite of a spontaneous reaction?	(Un/Re)	CO1
9.	<b>Define</b> the relation between half-life and rate constant between a second order reaction. <b>Explain</b> the reason for the depression of freezing point.	(Un)	CO3, CO4
10.	<b>Explain</b> the entropy and enthalpy of a chemical reaction. <b>What</b> is the difference between molecularity and order of a chemical reaction?	(Re/Un)	CO4



<b>CHM 12152</b>	<b>Elective Chemistry Lab II</b>	L	T	P	C
<b>Version 1.0</b>	Contact Hours:45	0	0	3	2
<b>Pre-requisites/Exposure</b>	Physics and Chemistry of class 12 or 10+2 level				
<b>Co-requisites</b>	Knowledge of Elective Chemistry III Theory				

### Course Objectives

- 1.To introduce important concepts required in the field of the practical field of elective chemistry. This course gives students a detailed understanding of lab-based chemistry knowledges in their course curriculums.
- 2.To introduce hands on training of standard solutions essential in every practical courses. 3. To impart the elementary ideas of physical methods of determination of surface tension, viscosity of organic solvents and acid catalysed hydrolysis of ester.
- 4.To learn the basic quantitative methods of titration of alkaline mixtures using various indicators.
- 5.To learn the determination methods of ionization constant of a weak acid by conductometric method.
- 6.To introduce the pH metric determination procedure of neutralization of acid-base titration.
- 7.To impart the determination method for rate constant of decomposition of H<sub>2</sub>O<sub>2</sub> by acidified KI solution using clock reactions.

### Course Outcomes for SCY32207

On completion of this course, the students will be able to

CO-1. Develop skills in the proper handling of apparatus and chemicals.

CO-2. Experiment with volumetric analysis and determination of physical properties of substances.

CO-3. Determine the "overall reaction order" for a chemical reaction using the (differential) rate law

CO-4. Illustrate the monitoring a titration with a pH electrode and determining the equivalence point.

### Catalogue Description:

This course gives a detailed understanding of the basics of chemistry lab techniques required in other disciplines. This course will include expert instructors who will introduce a detailed description of lab-based chemistry knowledges in their course curriculums, methods of determination of surface tension and viscosity of common liquids, correlation of theories of kinetics in the light of acid catalysed hydrolysis of ester, different quantitative methods of acid-base titrations using direct and pH mediated methods, determination of ionization constants of weak acids by conductometric titration and clock reaction mediated rate constant determinations. All the lectures will be devoted on discussions of elementary concepts and cutting-edge topics, focusing on practical implementation of knowledge. Instructors will conduct demonstration

classes by taking lecture followed by practical hands on training per requirement of the course. The tutorials and required assignments will acquaint the students with practical problem-solving techniques led by the course coordinator. After finishing this course, students from different disciplines will strongly acquire the hands-on training via experiencing practical lab sessions with the coordinator.

### Course Content:

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#### General Chemistry Lab:

[30 Lecture Hours]

1. Preparation of Solution: Normal Solution; Molar Solution
2. Determination of surface tension of a given solution by drop weight method using stalagmometer, considering aqueous solutions of NaCl, acetic acid, ethanol etc, as systems.
3. Determination of viscosity of organic solvents with Ostwald Viscometer at room temperature.
4. To determine the rate constant for the acid catalysed hydrolysis of an ester.

#### Inorganic Chemistry Lab:

[30 Lecture Hours]

1. Titration of  $\text{Na}_2\text{CO}_3 + \text{NaHCO}_3$  mixture vs HCl using phenolphthalein and methyl orange indicators.
2. Determination of ionization constant of a weak acid by conductometric method
3. Determination of neutralization point of the reaction between HCl and NaOH with the help of pH meter.
4. Determination of rate constant of decomposition of  $\text{H}_2\text{O}_2$  by acidified KI solution using clock reactions.

#### References:

1. Palit, S.R., De, S. K. Practical Physical Chemistry Science Book Agency
2. Handbook of Inorganic Analysis (First Edition): U.N Dhur & Sons Private Ltd.
3. Das, S.C. Advanced Practical Chemistry:

#### Modes of Examination: Assignment/Quiz/Project/Presentation/Written Exam

#### Examination Scheme:

Components	Mid term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

**Relationship between the Course Outcomes (COs) with Program Outcomes (POs) and Program Educational Outcomes (PEOs)**

### Mapping between COs, POs and PEOs

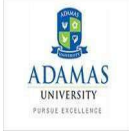
CO- 1	<b>Develop</b> skills in the proper handling of apparatus and chemicals.	PO1, PO3, PO7
CO- 2	<b>Experiment with</b> quantitative volumetric analysis and determination of physical properties of substances.	PO2, PO4, PO3, PO9
CO-3	<b>Determine</b> the "overall reaction order" for a chemical reaction using the (differential) rate law	PO3, PO4
CO- 4	<b>Illustrate</b> the monitoring a titration with a pH electrode and determining the equivalence point.	PO7, PO4, PO9

		Fundamental Knowledge	Critical Thinking	Skills	Technical Knowledge	Logical Thinking	Problem identification ability	Analytical Knowledge	Career goals	Team Work	Sustainable Development to environment	Development to society	Development to humanity
Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
<b>CHM 12152</b>	Elective Chemistry Lab II	1	1	2	2	-	2	3	-	2	-	-	-

1=weakly mapped

2= moderately mapped

3=strongly mapped

	<b>Name:</b>		
	<b>Enrolment No:</b>		
	<b>Course: Elective Chemistry Lab I</b> <b>Program: B.Sc.Biochemistry</b> <b>Semester: Odd2020-21</b> <b>Marks: 40</b>	<b>Time: 03 Hrs.</b> <b>Max.</b>	
	<b>Section A</b>		
1.	<b>Determine</b> the surface tension of toluene and acetone at room temperature.	(An)	CO1, CO2
2.	<b>Determine</b> the amount of Na <sub>2</sub> CO <sub>3</sub> and NaHCO <sub>3</sub> in an unknown mixture by acid-base titration.	(An)	CO1, CO2
3.	<b>Find</b> out the strength of unknown HCl using 0.01 N NaOH solution via conductometric titration.	(An)	CO3
4.	HNO <sub>3</sub> and H <sub>2</sub> SO <sub>4</sub> have the same molarity. <b>Why</b> did H <sub>2</sub> SO <sub>4</sub> require about twice the volume of NaOH solution for the titration?	(Un)	CO4
5.	<b>Explain</b> the effect of each of the following sources of error on the molarity of NaOH as determined in this experiment stating whether the concentration of NaOH obtained would be too high or too low. a. The burette is contaminated with an acid solution. b. The burette contains a large air bubble in the tip, which disappears in the course of the titration. c. A small volume of the acid solution is spilled when you transfer it into the Erlenmeyer flask.	(Un/An)	CO1, CO2, CO4





<b>CSE 11641</b>	<b>Elective Computer Science I</b>	L	T	P	C
<b>Version 1.0</b>	Contact Hours: 60	4	0	0	4
<b>Pre-requisites/Exposure</b>	Basics of Computer knowledge				
<b>Co-requisites</b>					

### Course Objectives

1. To understand the usage of computers in daily life applications.
2. To apply the Office productivity software applications in performing different tasks.
3. To understand about operating system about how it works.
4. To study about the different types of networks and their applications.

### Course Outcomes

On completion of this course, the students will be able to

CO1: Outline the usage of computers and why computers are essential components in business and society.

CO2: Experiment with a range of current, standard, Office Productivity software applications.

CO3: Evaluate office productivity software appropriate to a given situation

CO4: Utilize the Internet Web resources and evaluate on-line e-business system.

CO5: Solve common business problems using appropriate Information Technology applications and systems.

### Catalog Description

This course introduces the student to the world of computers and their basics. It helps them to grasp knowledge about the different Office productivity software applications and their practical applications scenarios. They become familiar with the basics of Operating System and its different types and also get knowledge about computer networks and their principal network components.

### Course Content

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#### Unit I: Knowing computer:

## **5 LectureHours**

What is Computer, Basic Applications of Computer; Components of Computer System, Central Processing Unit (CPU), VDU, Keyboard and Mouse, Other input/output Devices, Computer Memory, Concepts of Hardware and Software; Concept of Computing, Data and Information; Applications of IECT; Connecting keyboard, mouse, monitor and printer to CPU and checking power supply.

### **Unit II: Operating Computer using GUI Based Operating System:**

**8 Lecture Hours**

What is an Operating System; Basics of Popular Operating Systems; The User Interface, Status Bar, Using Menu and Menu-selection, Running an Application, Viewing of File, Folders and Directories, Creating and Renaming of files and folders, Opening and closing of different Windows; Using help; Creating Short cuts, Basics of O.S Setup; Common utilities.

### **Unit III: Understanding Word Processing:**

**10 Lecture Hours**

Word Processing Basics; Opening and Closing of documents; Text creation and Manipulation; Formatting of text; Table handling; Spell check, language setting and thesaurus; Mail merge, Printing of word document.

### **Unit IV: Using Spread Sheet:**

**9 Lecture Hours**

Basics of Spreadsheet; Manipulation of cells; Formulas and Functions; Editing of Spread Sheet, printing of Spread Sheet, Macro.

**Making Small Presentation:** Basics of presentation software; Creating Presentation; Preparation and Presentation of Slides; Slide Show; Taking printouts of presentation/handouts.

### **Unit V: Communications and collaboration:**

**15 Lecture Hours**

Basics of electronic mail; Getting an email account; Sending and receiving emails; Accessing sent emails; Using Emails; Document collaboration; InstantMessaging; Netiquettes.

### **Unit VI: Introduction to Internet, WWW and Web Browsers:**

**20 lecture hours**

Basic of Computer networks; LAN, WAN; Concept of Internet; Applications of Internet; connecting to internet; What is ISP; Knowing the Internet; Basics of internet connectivity related troubleshooting, World Wide Web; Web Browsing software, Search Engines; Understanding URL; Domain name; IP Address; Using e-governance website

## **Reference Books**

**List of Books:**

1. Introduction to Computers with MS-Office, Leon, TMH
2. Personal Computer Software, EXCELBOOKS
3. A First Course in Computers 2003, Saxena, VIKAS
4. Windows & MS-Office 2000, Krishnan, SCITECH

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

<b>Components</b>	<b>Mid Term</b>	<b>Attendance</b>	<b>Class Assessment</b>	<b>End Term</b>
<b>Weightage (%)</b>	<b>20</b>	<b>10</b>	<b>30</b>	<b>40</b>

**Relationship between the Course Outcomes (COs) and Program Outcomes (POs)**

<b>Mapping between COs and POs</b>		
	<b>Course Outcomes (COs)</b>	<b>Mapped Program Outcomes</b>
<b>CO1</b>	<b>Outline</b> the usage of computers and why computers are essential components in business and society.	<b>PO1, PO3, PO5, PO8</b>
<b>CO2</b>	<b>Experiment with</b> a range of current, standard, Office Productivity software applications.	<b>PO1, PO3, PO5</b>
<b>CO3</b>	<b>Evaluate</b> office productivity software appropriate to a given situation	<b>PO1, PO3, PO5</b>
<b>CO4</b>	<b>Utilize</b> the Internet Web resources and evaluate on-line e-business system.	<b>PO1, PO3, PO8</b>
<b>CO5</b>	<b>Solve</b> common business problems using appropriate Information Technology applications and systems.	<b>PO1, PO3, PO5, PO8, PO12</b>

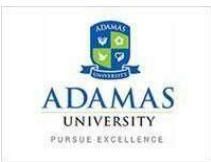
Course Code	Course Title	PO 1	PO2	PO 3	PO 4	PO5	PO 6	PO 7	PO 8	PO9	PO 10	PO 11	PO 12
		Fundamental Knowledge	Critical Thinking	Skills	Technical Knowledge	Logical Thinking	Problem identification ability	Analytical Knowledge	Career goals	Team Work	Sustainable Development to environment	Development to society	Development to humanity
<b>CSE 11641</b>	Elective Computer Science I	3	-	3	-	3	-	-	3	-	-	-	1

1 = weakly mapped

2 = moderately mapped

3 = strongly mapped

## Model Question Paper

<b>Name:</b>			
<b>Enrolment No:</b>			
<b>END-SEMESTER EXAMINATION (DECEMBER 2019)</b> (Academic Session: 2019 – 20, Semester Term: Aug. 2019 – Dec. 2019)			
<b>Name of the Program: B.Sc. Biochemistry</b>		<b>Semester: Odd 2020-21</b>	
<b>PAPER TITLE: ELECTIVE COMPUTERSCIENCE I PAPER CODE: CSE 11641</b>			
<b>Maximum Marks: 40</b>		<b>Time duration: 3 hours</b>	
<b>Instructions:</b>			
<b>Attempt any three questions from Section A (each carrying 4 marks); any four from Section B (each carrying 7 marks).</b>			
<b>SECTION A ( Attempt any Three questions)</b>			
1.	<b>Distinguish</b> between Random Access Memory and Read Only Memory.(An)	4	CO1
2.	<b>Outline</b> on: Keyboard.(Un)	4	CO1
3.	<b>Define</b> utility software and its significance.(Re)	4	CO2
4.	<b>What</b> is a function in Ms Excel? <b>What</b> function is used for counting the number?(Re)	3 1	CO3
<b>SECTION B (Attempt any Four questions)</b>			
5.	(a) <b>What</b> is the slide show View in Ms powerpoint?(Re) (b) <b>What</b> are Web sites & URL(s)? Define their usage in practical scenarios.(Re) (c) <b>What</b> is the difference between “operating system” & “Application software”?(Re)	1 2 4	CO2 CO3
6.	(a) <b>Outline</b> the significance of E-mail.(Un) (b) <b>What</b> are the productivity tools in an Email account? <b>Define</b> them.(Re)	2 5	CO4
7.	<b>Explain</b> the keyboard shortcuts for the following functions:- a) Making fonts bold: b) Making fonts underlines: c) Making fonts italic: d) Increasing font size: e) Making a paragraph right-aligned: What is the procedure of inserting Symbols in an MS Word document? (Un)	5 2	CO3, CO2
8.	a) <b>What</b> are the uses of Header and Footer in MS Word? b) <b>How</b> can you correct the spelling and grammatical mistakes in MS Word?(Re)	5	CO2, CO3

		<b>2</b>	
9.	(a) <b>Discuss</b> about Mail Merge in MS Word?	<b>3</b>	<b>CO2, CO3</b>
	(b) <b>What</b> is chart in MS Excel? Name three types ofcharts?	<b>3</b>	
	(c) <b>What</b> is slide transition? (Re /Cr)	<b>1</b>	

<b>CSE12642</b>	<b>Elective Computer Science I Lab</b>	L	T	P	C
<b>Version 1.0</b>	Contact Hours:45	0	0	3	2
<b>Pre-requisites/Exposure</b>	Theory of Elective Computer Science I at B.Sc Level				
<b>Co-requisites</b>					

### Course Objectives

1. To understand the usage of computers in daily life applications.
2. To apply formatting features in Word application to create documents.
3. To apply different formula in Excel to calculate and process information.
4. To apply slide transitions, themes and formatting features to create a power point presentation.

### Course Outcomes

On completion of this course, the students will be able to

- CO1: Develop the concept about word and its different features.
- CO2: Create different documents using Word.
- CO3: Utilize mail merge to create template using word.
- CO4: Develop the concept excel and its different formulas.
- CO5: Create presentation with animation and effects.

### Catalog Description

This course introduces the student to the world of computers and their basics. It helps them to grasp knowledge about the different Office productivity software applications and their practical applications scenarios. They become familiar with the basics of Operating System and its different types and also get knowledge about computer networks and their principal network components.

### Course Content

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#### List of experiments:

1. Introduction and familiarization with word with different type of stylings. [4 Lecture Hours]
2. Table creation in and basic formatting. [4 Lecture Hours]
3. Inclusion of image and editing image using Word and some basic designing features. [4 Lecture Hours]

4. Example of Mail-merge. [4 LectureHours]
5. Create spreadsheet with some basic calculation. [4 LectureHours]
6. Creating Spreadsheet with some advance level formula and conditions. [4 Lecture Hours]
7. Creating Macro in spreadsheet. [4 LectureHours]
8. Creating colour conditioning in spreadsheet. [4 LectureHours]
9. Creating different kinds of charts in spreadsheet. [4 Lecture Hours]10. Creating basic presentation. [4 LectureHours]
11. Inclusion of different levels of animations in the presentation. [5 Lecture Hours]
12. Project on Word. [5 Lecture Hours]
13. Project on Excel. [5 LectureHours]
14. Project on power-point. [5 LectureHours]

## Reference Books

### List of Books:

1. Introduction to Computers with MS-Office, Leon, TMH
2. Personal Computer Software, EXCELBOOKS
3. A First Course in Computers 2003, Saxena, VIKAS
4. Windows & MS-Office 2000, Krishnan, SCITECH

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination**  
**Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40


**Relationship between the Course Outcomes (COs) and Program Outcomes (POs)**



Mapping between COs and POs		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	Develop the concept about word and its different features.	PO1,PO3
CO2	Create different documents using Word.	PO3,PO5,PO7,PO8
CO3	Utilize mail merge to create template using word.	PO3,PO5,PO7
CO4	Develop the concept excel and its different formulas.	PO3,PO5,PO7
CO5	Create presentation with animation and effects.	PO3,PO5,PO7

		Fundamental Knowledge	Critical Thinking	Skills	Technical Knowledge	Logical Thinking	Problem identification ability	Analytical Knowledge	Career goals	Team Work	Sustainable Development to environment	Development to society	Development to humanity
Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
CSE12642	Elective Computer Science I Lab	1	-	3	-	3	-	3	-	-	-	-	-

- 1 = weakly mapped  
2 = moderately mapped  
3 = strongly mapped

<b>Name:</b>			
<b>Enrolment No:</b>			
<b>END-SEMESTER EXAMINATION (DECEMBER2019)</b> (Academic Session: 2019 – 20, Semester Term: Aug. 2019 – Dec.2019)			
<b>Name of theProgram:B.Sc.Biochemistry</b>		<b>Semester: Odd 2020-21</b>	
<b>PAPER TITLE: Elective Computer ScienceII</b>		<b>PAPER CODE:</b>	
<b>CSE12642</b>		<b>MaximumMarks:40</b>	
<b>Instructions:</b>		<b>Time duration: 3 hours</b>	
<b>Attempt any two questions from Section A (each carrying 10 marks); Section B is Compulsory (carrying 10 marks).</b>			
<b>Section A ( Attempt any Two)</b>			
1	a) <b>Create</b> your bio-data in Worddocument.	5	<b>CO1</b>
	b) <b>Create</b> a table with basicformatting.	3	<b>CO2</b>
	c) Insert image using Word and apply bordereffects.(Cr)	2	<b>CO3</b>
2	a) <b>Illustrate</b> Mail-merge inWord.	5	<b>CO1</b>
	b) <b>Create</b> spreadsheet with some basiccalculation.	3	<b>CO4</b>
	c) <b>Create</b> a basic presentation and apply a suitabletheme.(Un/Cr)	2	<b>CO5</b>
3	a) <b>Create</b> Spreadsheet with some advance level formula andconditions.	4	<b>CO4</b>
	b) <b>Create</b> Macro inspreadsheet.	3	
	c) <b>Create</b> colour conditioning in spreadsheet.(Cr)	3	
4	a) <b>Create</b> different kinds of charts inspreadsheet.	4	<b>CO1</b>
	b) <b>Discuss</b> different levels of animations in thepresentation.	3	<b>CO4</b>
	c) <b>Create</b> Word document and insert hierarchical smart art.(Cr)	3	<b>CO5</b>
<b>SECTION B is compulsory</b>			
5	Viva-voce (U/An/Ap/R/Ev)	10	<b>CO1</b> <b>CO2</b> <b>CO3</b> <b>CO4</b> <b>CO5</b>

<b>SDS11506</b>	<b>Elective Statistics-I</b>	<b>L</b>	<b>T</b>	<b>P</b>	<b>C</b>
<b>Version 1.0</b>	<b>Contact Hours :90</b>	5	1	0	6
<b>Pre-requisites/Exposure</b>	12 <sup>th</sup> level Mathematics				
<b>Co-requisites</b>	--				

### Course Objectives:

The objective of this course for the graduate student is:

1. To provide a basic understanding of statistical data with preparation and presentation of data.

To develop the statistical concepts of the discrete and continuous variables or data and its

various central and dispersion measures, regression, and correlation analysis with applications in simple real life examples.

### Course Outcomes:

On completion of this course, the students will be able to:

**CO1: Define** different types of statistical data, attributes, and variables (discrete and continuous) with frequency distribution.

**CO2: Find** various measures of central tendency and dispersion for grouped and ungrouped data, regression lines and correlation coefficients.

**CO3: Summarize**, collect, and present the different types of data graphically and numerically.

**CO4: Compare** the results obtained from various central and dispersion measures, regression, and correlation analysis.

**CO5: Utilize** the concept of correlation and regression and its properties to obtain the solution of simple statistical/business/economics problems.

### Catalogue Description:

Knowledge of basic statistics and methods is necessary to work on statistical data for the beginners of graduate students. This course gives an idea and understanding about the several statistical methods and measures used to extract the information from various types of data comes from statistical problems. This course deals with data collection, preparation and presentation with frequency distribution, various measures of central tendency and dispersion, correlation, regression analysis, and its application in statistical problems. Classes will be conducted by lecture as well as power point presentation, audio visual session as per requirement. The tutorials will familiarize the students with practical problem-solving techniques.

guided by the course coordinator. Students will strongly grab the basic concepts of the subject via exercise and discussions with the coordinator.

## **Elective Statistics -I**

**Paper Code:**

L	T	P	C
5	1	0	6

### **Course Content**

**Unit-I** (12L)

#### **Collection and Scrutiny of Data**

Statistical data: Primary Data and Secondary Data, Collection of Data, Presentation of data, tabular representation of data, Scrutiny of Data.

**Unit-II** (14L)

#### **Frequency Distribution**

Attribute and variable, Discrete variable and continuous variable, Frequency Distribution of an Attribute, Frequency Distribution of a variable, Case of a discrete variable, Case of a continuous variable, Graphical Representation of a frequency Distribution, Frequency curve.

**Unit-III** (8L)

#### **Presentation of Data**

Frequency data and non-frequency data, Textual presentation of Data, Tabulation of Data, Diagrammatic presentation of Data (Bar chart, pie diagram, Histogram, Ogives).

**Unit-IV** (17L)

#### **Measures of Central Tendency:**

Meaning of Central Tendency, Common measure of Central Tendency, Requirements of an ideal Average, Comparison of Mean, Median and Mode, Geometric Mean and Harmonic Mean, weighted Means.

**Unit-V** (15L)

#### **Measures of Dispersion:**

Range, Mean Deviation, Standard Deviation, Quantiles and Percentiles, Quantile Deviation, Comparison of the Measures of Dispersion, Some important relations, Measures of relative Dispersion.

**Unit-VI** (13L)

#### **Moments and Measures of Skewness and Kurtosis:**

Moments, Relationship between central and ordinary moments, Skewness, Kurtosis, Some important relations.

**Unit-VII** (13L)

#### **Correlation and Regression:**

Correlation: Scatter diagram, Karl-Pearson's correlation, concurrent deviation method, rank correlation, uses of correlation in business regression, regression lines, regression coefficients, properties of regression coefficients, Use of regression in business problems.

### Books Recommended

**T1.** A.M. Goon, M.K. Gupta and B. Dasgupta (2005): *Fundamentals of Statistics*, Vol. I, 8th Ed., World Press, Kolkata

**T2.** S.C. Gupta and V.K. Kapoor (2007): *Fundamentals of Mathematical Statistics*, 11th Ed., Sultan Chand and Sons.

### Reference book:

**R1.** N. G. Das (2009): *Statistical Methods*, combined edition (vol I & II), McGraw Hill Education (India).

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

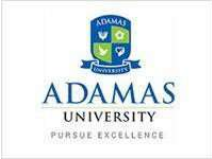
Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

### Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and POs		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	<b>Define</b> different types of statistical data, attributes, and variables (discrete and continuous) with frequency distribution.	PO1, PO3, PO7
CO2	<b>Find</b> various measures of central tendency and dispersion for grouped and ungrouped data, regression lines and correlation coefficients.	PO1, PO3, PO7
CO3	<b>Summarize</b> , collect, and present the different types of data graphically and numerically.	PO1, PO3, PO5, PO7
CO4	<b>Compare</b> the results obtain from various central and dispersion measures, regression, and correlation Analysis.	PO1, PO3, PO5, PO7
CO5	<b>Utilize</b> the concept of correlation and regression and its properties to obtain the solution of simple statistical/business/economics problems.	PO1, PO3, PO5, PO7



## Model Question Paper

<b>Name:</b>  <b>Enrolment No:</b>																																															
<b>Course: Elective Statistics-I</b> <b>Program: B.Sc.inBiochemistry</b> <b>Semester:Odd 2020-21</b>		<b>Time: 03Hrs.</b> <b>Max. Marks:40</b>																																													
<b>Instructions:</b> Attempt any three questions from <b>Section A</b> (each carrying 4 marks); any <b>Two Questions</b> from <b>Section B</b> (each carrying 10 marks). <b>Section C</b> is Compulsory (carrying 8 marks).																																															
<b>Section A (Attempt any Three)</b>																																															
1.	a) <b>Define</b> primary and secondary data.(R)  b) <b>Explain</b> with illustration the distinction between an attribute and a variable.(U)	2  2	CO1  CO3																																												
2.	Draw the histogram and frequency polygon to <b>illustrate</b> the following frequency distribution: <table border="1" style="margin: 10px auto; border-collapse: collapse; text-align: center;"> <tr> <td style="padding: 5px;">Wages(Rs.)</td> <td style="padding: 5px;">50-59</td> <td style="padding: 5px;">60-69</td> <td style="padding: 5px;">70-79</td> <td style="padding: 5px;">80-89</td> <td style="padding: 5px;">90-99</td> <td style="padding: 5px;">100-109</td> <td style="padding: 5px;">110-119</td> </tr> <tr> <td style="padding: 5px;">No. of employees</td> <td style="padding: 5px;">8</td> <td style="padding: 5px;">10</td> <td style="padding: 5px;">16</td> <td style="padding: 5px;">14</td> <td style="padding: 5px;">10</td> <td style="padding: 5px;">5</td> <td style="padding: 5px;">2</td> </tr> </table> . (U)	Wages(Rs.)	50-59	60-69	70-79	80-89	90-99	100-109	110-119	No. of employees	8	10	16	14	10	5	2	4	CO3																												
Wages(Rs.)	50-59	60-69	70-79	80-89	90-99	100-109	110-119																																								
No. of employees	8	10	16	14	10	5	2																																								
3.	a) <b>What</b> is moment of distribution? <b>What</b> are the raw and the central moment? ( R)  b) From the data given below, <b>find</b> the coefficient of variance: Pearson's measure of skewness = 0.42, AM = 86; median = 80. ( R)	2  2	CO2  CO2																																												
4.	Ten hand-writing were ranked by two judges in a competition. The ranking are given below. <b>Make use of</b> rank correlation coefficient method, what conclusion do you draw from the result? <table border="1" style="margin: 10px auto; border-collapse: collapse; text-align: center;"> <tr> <td></td> <td colspan="10" style="padding: 5px;">Hand- writing</td> </tr> <tr> <td></td> <td style="padding: 5px;">A</td> <td style="padding: 5px;">B</td> <td style="padding: 5px;">C</td> <td style="padding: 5px;">D</td> <td style="padding: 5px;">E</td> <td style="padding: 5px;">F</td> <td style="padding: 5px;">G</td> <td style="padding: 5px;">H</td> <td style="padding: 5px;">I</td> <td style="padding: 5px;">J</td> </tr> <tr> <td style="padding: 5px;">Judge 1</td> <td style="padding: 5px;">3</td> <td style="padding: 5px;">8</td> <td style="padding: 5px;">5</td> <td style="padding: 5px;">4</td> <td style="padding: 5px;">7</td> <td style="padding: 5px;">10</td> <td style="padding: 5px;">1</td> <td style="padding: 5px;">2</td> <td style="padding: 5px;">6</td> <td style="padding: 5px;">9</td> </tr> <tr> <td style="padding: 5px;">Judge 2</td> <td style="padding: 5px;">6</td> <td style="padding: 5px;">4</td> <td style="padding: 5px;">7</td> <td style="padding: 5px;">5</td> <td style="padding: 5px;">10</td> <td style="padding: 5px;">3</td> <td style="padding: 5px;">2</td> <td style="padding: 5px;">1</td> <td style="padding: 5px;">9</td> <td style="padding: 5px;">8</td> </tr> </table> . (Ap)		Hand- writing											A	B	C	D	E	F	G	H	I	J	Judge 1	3	8	5	4	7	10	1	2	6	9	Judge 2	6	4	7	5	10	3	2	1	9	8	4	CO5
	Hand- writing																																														
	A	B	C	D	E	F	G	H	I	J																																					
Judge 1	3	8	5	4	7	10	1	2	6	9																																					
Judge 2	6	4	7	5	10	3	2	1	9	8																																					
<b>SECTION B (Attempt any Two Questions)</b>																																															

5.	<p>The score of two batsmen, A and B, in ten innings during a certain season, are as under:</p> <table border="1" data-bbox="293 289 1149 365"> <tr> <td>A</td><td>32</td><td>28</td><td>47</td><td>63</td><td>71</td><td>39</td><td>10</td><td>60</td><td>96</td><td>14</td> </tr> <tr> <td>B</td><td>19</td><td>31</td><td>48</td><td>53</td><td>67</td><td>90</td><td>10</td><td>62</td><td>40</td><td>80</td> </tr> </table> <p><b>Compare</b> the batsmen to identify who is better score getter and who is more consistent. (U)</p>	A	32	28	47	63	71	39	10	60	96	14	B	19	31	48	53	67	90	10	62	40	80	10	CO4
A	32	28	47	63	71	39	10	60	96	14															
B	19	31	48	53	67	90	10	62	40	80															
6.	<p>a) The arithmetic mean of two observations is 25 and their geometric mean is 15. <b>Find</b> (i) their harmonic mean and (ii) the two observations. (R)</p> <p>b) The number of telephone calls received in 245 successive one minute intervals at an exchange are shown in the following frequency distribution:</p> <table border="1" data-bbox="293 697 1166 823"> <tr> <td>No. of calls</td><td>0</td><td>1</td><td>2</td><td>3</td><td>4</td><td>5</td><td>6</td><td>7</td><td>Total</td> </tr> <tr> <td>Frequency</td><td>14</td><td>21</td><td>25</td><td>43</td><td>51</td><td>40</td><td>39</td><td>12</td><td>245</td> </tr> </table> <p><b>Find</b> the mean, median, and mode. (R)</p>	No. of calls	0	1	2	3	4	5	6	7	Total	Frequency	14	21	25	43	51	40	39	12	245	4 6	CO2  CO2		
No. of calls	0	1	2	3	4	5	6	7	Total																
Frequency	14	21	25	43	51	40	39	12	245																
7.	<p>While calculating the coefficient of correlation between two variables <math>x</math> and <math>y</math>, the following results were obtained: <math>n=25</math>, <math>\sum x=125</math>, <math>\sum y=100</math>, <math>\sum x^2=650</math>, <math>\sum y^2=460</math>, <math>\sum xy=508</math>. It was however later discovered at the time of checking that two pair of observations <math>(x, y)</math> were copied <math>(6,14)</math> and <math>(8,6)</math>, while the correct values were <math>(8,12)</math> and <math>(6,8)</math> respectively. <b>Find</b> the correct value of the coefficient of correlation. (R)</p>	10	CO2																						
<b>SECTION C is Compulsory</b>																									
8.	<p>From the following results, <b>find</b> the two regression equations and <b>utilize</b> these two equations to estimate the yield of crops when the rainfall is 22 cms, and the rainfall when the yield is 600 kg:</p> <table border="1" data-bbox="293 1402 1143 1558"> <tr> <td></td><td>y (Yield in kg.)</td><td>x (Rainfall in cm.)</td> </tr> <tr> <td>Mean</td><td>508.4</td><td>26.7</td> </tr> <tr> <td>S.D.</td><td>36.8</td><td>4.6</td> </tr> </table> <p>Coefficient of correlation between yield and rainfall = 0.52. (R, Ap)</p>		y (Yield in kg.)	x (Rainfall in cm.)	Mean	508.4	26.7	S.D.	36.8	4.6	5 3	CO2  CO5													
	y (Yield in kg.)	x (Rainfall in cm.)																							
Mean	508.4	26.7																							
S.D.	36.8	4.6																							



## SEMESTER 5

Course Title	CONCEPT OF GENETICS	L	T	P	C
BIC11026		3	1	0	4
Contact Hours	60				
Pre-requisites/Exposure	Basic Understanding of Cell Biology and Biochemistry				

### Course Objectives

1. To gain a deeper understanding in the history and developments in the field of Genetics.
2. To be able to perform and explain genetic experiments.

### Course Outcomes

On completion of this course, the students will be able to

- CO1 gather **knowledge** about the working principle of all instruments needed for genetics practical. CO2 **illustrate** the process of mitosis and meiosis.
- CO3 **perform** karyotyping with the help of photographs.
- CO4 **develop** pedigree charts of some common characters like blood group, colour blindness and PTC tasting.
- CO5 **demonstrate** Barr Body.

### **Catalog Description:**

The core-course of 'Concept of Genetics' will help to understand the concept of genetics with experimental point of views. This course includes comprehensive approach through studying karyotyping, different stages of cell division, role of Barr body and to check the presence of Barr body following conclusion. All the lectures will be devoted on discussions of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as power point presentation, audio visual virtual lab session as per requirement. The tutorials will enable the students with problem-solving ability led by the course coordinator. Students will perceive the basic concepts of the subject via exercise and discussions with the coordinator.

### **CONCEPT IN GENETICS**

1. Introduction to model organisms and Mendelism: Model organisms: Escherichia coli, Saccharomyces cerevisiae, Drosophila melanogaster, Caenorhabditis elegans, Danio rerio and Arabidopsis thaliana, Basic principles of heredity.
2. Applications of Mendel's principles & chromosomal basis of heredity: Laws of probability & binomial expansion, formulating and testing genetic hypothesis, chromosomal basis of Mendelism - Sutton and Boveri hypothesis with experimental evidences.
3. Extensions of Mendelism: Allelic variation and gene function - dominance relationships, multiple alleles, lethal alleles and null alleles. Pleiotropy gene interaction - epistatic and non epistatic, interaction between gene(s) and environment.

Penetrance and expressivity, norm of reaction and phenocopy.

4. Genetic definition of a gene: Complementation test, limitations of cis-trans test, intragenic complementation, rII locus of phage T4 and concept of cistron
5. Genetics of bacteria and viruses: Mechanism of genetic exchange - conjugation, transformation and transduction. Gene mapping in bacteria.
6. Linkage, crossing over and mapping techniques
7. Linkage and crossing over, genetic mapping in eukaryotes, centromere mapping with ordered tetrads, cytogenetic mapping with deletions and duplications in *Drosophila*, detection of linked loci by pedigree analysis in humans and somatic cell hybridization for positioning genes on chromosomes.
8. Human pedigree analysis: Pedigree conventions, characteristics of dominant and recessive inheritance. Applications of pedigree analysis.
9. The genetic control of development and sex determination: Model organism for genetic analysis, *Drosophila* development, maternal effect genes, morphogens and zygotic gene activity in development, sex chromosomes and sex determination, dosage compensation of X-linked genes.
10. Organelle heredity and epigenetics: Extra nuclear inheritance, tests for organelle heredity and maternal effect, epigenetic mechanisms of transcriptional regulation & genomic imprinting.
11. Chromosomal aberrations: Variations in chromosome number- monosomy and trisomy of sex and autosomes. Variations in chromosome structure - inversions, deletions, duplications and translocations.
12. Inheritance of complex traits & population genetics: Inheritance of complex trait, analysis of quantitative traits, narrow and broad sense heritability, quantitative trait loci (QTL) and their identification. Hardy- Weinberg law, predicting allele and genotype frequencies and exceptions to Hardy-Weinberg principle.
13. Evolutionary genetics: Molecular evolution - analysis of nucleotide and amino acid sequences, molecular phylogenies, homologous sequences, phenotypic evolution and speciation.

## **SUGGESTED READINGS**

1. Genetics (2012) 6<sup>th</sup> ed., Snustad, D.P. and Simmons, M.J., John Wiley & Sons. (Singapore), ISBN: 978-1-118-09242-2.
2. Genetics - A Conceptual Approach (2012), 4<sup>th</sup> ed., Pierce, B.A., W.H. Freeman & Co. (New York), ISBN:13:978-1-4292-7606-1 / ISBN:10:1-4292-7606-1.
3. An Introduction to Genetic Analysis (2010), 10<sup>th</sup> ed., Griffiths, A.J.F, Wessler, S. R, Carroll, S. B. and Doebley, J., W.H. Freeman & Company (New York), ISBN:10: 1-4292-2943-8.

Modes of Examination: Assignment/Quiz/Project/Presentation/Written Exam

Examination Scheme:

Components	Internal	Attendance	Mid Term	End Term
Weightage (%)	30	10	20	40

### Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

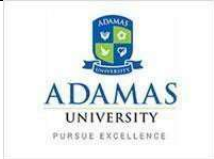
Mapping between COs and POs		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	gather <b>knowledge</b> about the working principle of all instruments needed for genetics practical.	PO1, PO3, PO4
CO2	<b>illustrate</b> the process of mitosis and meiosis.	PO1, PO2, PO4
CO3	<b>perform</b> Karyotyping with the help of photographs.	PO1, PO2, PO3, PO4, PO5
CO4	<b>develop</b> pedigree charts of some common characters like blood group, colour blindness and PTC tasting.	PO1, PO2, PO3, PO4, PO5, PO6, PO10, PO11
CO5	<b>demonstrate</b> Barr Body.	PO1, PO2, PO3, PO4, PO5, PO7, PO8

		Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking Ability	Problem Identification	Analytical Knowledge	Career Goals	Team Work	Sustainable Environment	Development to Society	Development to Humanity
Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
<b>BIC12026</b>	Concept of Genetics	3	2	2	3	3	1	1	1	-	1	1	-

1=weakly mapped

2= moderately mapped

3=strongly mapped

<b>Name:</b>			
<b>Enrolment No:</b>			
<b>Course: BIC12027 Concept of GENETICS</b>		<b>Time: 03Hrs.</b> <b>Max. Marks:40</b>	
<b>Program: B.Sc. Biochemistry</b>			
<b>Semester: Odd2020-21</b>			
<b>Instructions:</b> Attempt any <b>Two</b> questions from <b>Section A</b> (each carrying 10 marks)			
<b>SECTION A ( Attemptany Two questions)</b>			
1.	Perform the given experiment A and write down its Principle and <b>interpret</b> the result. (An)	<b>10</b>	<b>CO1,</b> <b>CO2</b>
2.	Perform the given experiment B and write down its principle and <b>illustrate</b> the result.(Ev)	<b>10</b>	<b>CO1,</b> <b>CO2</b>
3.	Perform the given experiment C and write down its principle and <b>interpret</b> the result.(An)	<b>10</b>	<b>CO1,</b> <b>CO2</b>  <b>CO3</b>
4.	Perform the given experiment D and write down its Principle and <b>interpret</b> the result.(An)	<b>10</b>	<b>CO4</b> <b>CO5</b>
<b>SECTION B is compulsory</b>			
5.	<b>Viva-voce</b> (U/An/Ap/R/Ev)	<b>10</b>	<b>CO1</b> <b>CO2</b> <b>CO3</b> <b>CO4</b> <b>CO5</b>
6.	<b>Practical copy</b> (U/Ap/Ev)	<b>10</b>	<b>CO1</b> <b>CO2</b> <b>CO3</b> <b>CO4</b> <b>CO5</b>

Course Title	CONCEPT OF GENETICS LAB	L	T	P	C
BIC12027		0	0	3	2
Contact Hours	45				
Pre-requisites/Exposure	Basic Understanding of Cell Biology and Biochemistry				

### Course Objectives

2. To gain a deeper understanding in the history and developments in the field of Genetics. 2. To be able to perform and explain genetic experiments.

### Course Outcomes

On completion of this course, the students will be able to

- CO1 gather **knowledge** about the working principle of all instruments needed for genetics practical. CO2 **illustrate** the process of mitosis and meiosis.
- CO3 **perform** karyotyping with the help of photographs.
- CO4 **develop** pedigree charts of some common characters like blood group, colour blindness and PTC tasting.
- CO5 **demonstrate** Barr Body.

### **Catalog Description:**

The core-course of 'Concept of Genetics lab' will help to understand the concept of genetics with experimental point of views. This course includes comprehensive approach through studying karyotyping, different stages of cell division, role of Barr body and to check the presence of Barr body following conclusion. All the lectures will be devoted on discussions of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as power point presentation, audio visual virtual lab session as per requirement. The tutorials will enable the students with problem-solving ability led by the course coordinator. Students will perceive the basic concepts of the subject via exercise and discussions with the coordinator.

## Course Content

### Concept of GENETICS LAB

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#### Topic

1. Observation of Mitosis in onion root tip. (5 Hours)
2. Observation of Mitosis in garlic root tip. (5 Hours)
3. Identification of different stages of meiosis in onion flower buds.(5 Hours)
4. Identification of different stages of meiosis in garlic.(5 Hours)
5. Demonstration of - Barr Body from epithelial cells of human female. (10 Hours)
6. Study of polyploidy in onion root tip by colchicine treatment. (10 Hours)
7. Karyotyping with the help of photographs.(10 Hours)
8. Mendelian deviations in dihybrid crosses.(10 Hours)

### Books & Other Resources

1. Gardner, E.J., Simmons, M.J., Snustad, D.P. (2006). Principles of Genetics.VIII Edition John Wiley & Sons.
2. Snustad, D.P., Simmons, M.J. (2009). Principles of Genetics.V Edition. John Wiley and Sons Inc.
3. Klug, W.S., Cummings, M.R., Spencer, C.A. (2009). Concepts of Genetics. IX Edition. Benjamin Cummings.
4. Russell, P.J. (2009). Genetics- A Molecular Approach. III Edition. Benjamin Cummings.
5. Griffiths, A.J.F., Wessler, S.R., Lewontin, R.C. and Carroll, S.B. IX Edition. Introduction to Genetic Analysis, W. H. Freeman & Co.

Modes of Examination: Assignment/Quiz/Project/Presentation/Written Exam

Examination Scheme:

Components	Internal	Attendance	Mid Term	End Term
Weightage (%)	30	10	20	40

### Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and POs		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	gather <b>knowledge</b> about the working principle of all instruments needed for genetics practical.	PO1, PO3, PO4
CO2	<b>illustrate</b> the process of mitosis and meiosis.	PO1, PO2, PO4
CO3	<b>perform</b> Karyotyping with the help of photographs.	PO1, PO2, PO3, PO4, PO5
CO4	<b>develop</b> pedigree charts of some common characters like blood group, colour blindness and PTC tasting.	PO1, PO2, PO3, PO4, PO5, PO6, PO10, PO11
CO5	<b>demonstrate</b> Barr Body.	PO1, PO2, PO3, PO4, PO5, PO7, PO8

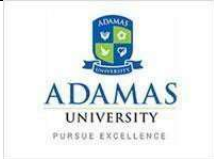
		Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking Ability	Problem Identification	Analytical Knowledge	Career Goals	Team Work	Sustainable Environment	Development to Society	Development to Humanity
Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
<b>BIC12027</b>	Concept of Genetics Lab	3	2	2	3	3	1	1	1	-	1	1	-



1=weakly mapped

2= moderately mapped

3=strongly mapped

<b>Name:</b>			
<b>Enrolment No:</b>			
<b>Course: BIC12027 Concept of GENETICS LAB(Practical)</b>			
<b>Program: B.Sc. Biochemistry</b>		<b>Time: 03Hrs.</b>	
<b>Semester: Odd2020-21</b>		<b>Max. Marks: 40</b>	
<b>Instructions:</b>			
Attempt any <b>Two</b> questions from <b>Section A</b> (each carrying 10 marks)			
<b>SECTION A ( Attemptany Two questions)</b>			
1.	Perform the given experiment A and write down its Principle and <b>interpret</b> the result. (An)	<b>10</b>	<b>CO1, CO2</b>
2.	Perform the given experiment B and write down its principle and <b>illustrate</b> the result.(Ev)	<b>10</b>	<b>CO1, CO2</b>
3.	Perform the given experiment C and write down its principle and <b>interpret</b> the result.(An)	<b>10</b>	<b>CO1, CO2 CO3</b>
4.	Perform the given experiment D and write down its Principle and <b>interpret</b> the result.(An)	<b>10</b>	<b>CO4 CO5</b>
<b>SECTION B is compulsory</b>			
5.	<b>Viva-voce</b> (U/An/Ap/R/Ev)	<b>10</b>	<b>CO1 CO2 CO3 CO4 CO5</b>
6.	<b>Practical copy</b> (U/Ap/Ev)	<b>10</b>	<b>CO1 CO2 CO3 CO4 CO5</b>

<b>BIC11028</b>	<b>Gene expression and regulation</b>	L	T	P	C
<b>Version 1.0</b>	Contact Hours:60	3	1	0	4
<b>Pre-requisites/Exposure</b>	Basic Understanding of Biochemistry				
<b>Co-requisites</b>	-				

### Course Objectives

1. To understand biosynthesis of RNA in prokaryotes and eukaryotes.
2. To gain knowledge in RNA splicing.
3. To know biosynthesis of proteins and mechanism of its degradation process.
4. To understand regulation of gene expression in prokaryotes and eukaryotes.

### Course Outcomes

On completion of this course, the students will be able to

- CO1** explain and categorize biosynthesis of RNA in prokaryotes and eukaryotes.
- CO2** discuss RNA splicing.
- CO3** discuss about Principles of molecular cell biology of cancer.
- CO4** demonstrate and illustrate Biosynthesis of proteins and mechanism of its degradation process.
- CO5** explain and categorize regulation of gene expression in prokaryotes and eukaryotes.

### Catalog Description

To provide an understanding of i) the regulation of transcription in eukaryotic organisms; ii) post-transcriptional regulation; iii) the structure, formation and function of microRNAs; iv) how the process of translation is controlled.

### Course Content

#### 1. Unit I Biosynthesis of RNA in prokaryotes [20 Lecture Hours]

RNA polymerases, transcription cycle in bacteria, sigma factor, bacterial promoters, identification of DNA binding sites by DNA footprinting, the three stages of RNA synthesis, initiation, elongation and termination, rho-dependent and rho-independent termination. Inhibitors of transcription and applications as anti-microbial drugs.

#### 2. Unit II Biosynthesis of RNA in eukaryotes [20 Lecture Hours]

Comparison between prokaryotic and eukaryotic transcription. Transcription by RNA polymerase II, RNA polymerase II core promoters, general transcription factors, various types of RNA processing, transcription by RNA polymerase I and III. Inhibitors of eukaryotic transcription and their applications. Comparison of fidelity of transcription and replication.

### 3. Unit III RNA splicing [20 LectureHours]

Chemistry of RNA splicing, the spliceosome machinery, splicing pathways, group I introns

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

### Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and Pos		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	<b>Explain</b> and categorize biosynthesis of RNA in prokaryotes and eukaryotes.	PO1, PO2
CO2	<b>Discuss</b> RNA splicing	PO1, PO2, PO3,
CO3	<b>Discuss</b> about Principles of molecular cell biology of cancer.	PO1, PO3, PO4,
CO4	<b>Demonstrate</b> and illustrate Biosynthesis of proteins and mechanism of its degradation process.	PO6, PO7, PO10
CO5	<b>Explain</b> and categorize regulation of gene expression in prokaryotes and eukaryotes	PO8, PO9, PO11, PO10, PO12

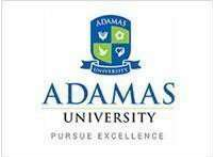
Course Code	Course Title	PO	PO	PO	PO	PO	PO	PO	PO	PO	PO	PO	PO	PO	PO
<b>BIC1102</b> <b>8</b>	Gene expression and regulation N	1	Fundamental Knowledge	2	Critical Thinking	3	Skill	4	Technical Knowledge	5	Logical Thinking Ability	6	Problem Identification	7	Analytical Knowledge
		8	Career Goals	9	Team Work	0	Sustainable Development to	1	Development to Society	2	Development to Humanity				

1=weakly mapped

2= moderately mapped

3=strongly mapped

## Model Question Paper

<b>Name:</b>			
<b>Enrolment No:</b>			
<b>Course: BIC11028 Gene expression and regulation</b>			
<b>Program: B.SC Biochemistry</b>		<b>Time: 03Hrs.</b>	
<b>Semester: Even 2019-20</b>		<b>Max. Marks: 40</b>	
<b>Instructions:</b>			
Attempt any three questions from Section A (each carrying 4 marks); any four Questions from Section B (each carrying 7 marks).			
<b>Section A ( Attemptany Three)</b>			
1.	In eukaryotes, which RNA polymerase makes rRNA?.(U)	4	CO1
2.	Which subunits of RNA polymerase is solely required for initiation of transcription?. (U)	4	CO2
3.	Which codons is the mRNA start codon that initiates translation?? (R)	4	CO3
4.	Which describes the key function of helicases during transcription?(U)	4	CO3
<b>SECTION B (Attempt any FOUR Questions)</b>			
5.	Transcription factors that are the first to bind DNA in heterochromatin regions, often promote euchromatin formation, and recruit other transcriptional machinery to promote transcription are best known as which of the following? (R)	7	CO4
1.	Before RNA polymerase can initiate transcription, this protein must bind to it, creating the RNA polymerase holoenzyme and allowing for the initiation of transcription. (Ap)	7	CO3,CO5
7.	Which is not dependent on the C-terminal domain (CTD) of RNA polymerase II?(U)	7	CO4

8.	<b>Describe</b> some common chromosomal mutations: inversions, deletions, duplications, fusions, fissions, and translocations. (U),	7	<b>CO5</b>
9.	<b>Name</b> the two basic kinds of point mutations. (U),	7	<b>CO1</b>

<b>BIC12029</b>	<b>Gene expression and regulation Lab</b>	L	T	P	C
<b>Version 1.0</b>	Contact Hours:45	0	0	3	2
<b>Pre-requisites/Exposure</b>	Basic understanding of Biochemistry				
<b>Co-requisites</b>	-				

### Course Objectives

1. To isolate and show total nucleic acids from plant tissue.
2. To isolate total RNA.
3. To develop the concept and differentiate between parent and recombinant plasmid containing bacteria.
4. To understand protein modeling.
5. To conceptualize transcription factor identification through in silico analysis.

### Course Outcomes

On completion of this course, the students will be able to

- CO1** isolate and **analyze** of total nucleic acids from plant tissue.
- CO2** **Interpret** the results of total RNA isolation.
- CO3** **categorize** between parent and recombinant plasmid containing bacteria.
- CO4** **demonstrate** protein modelling.
- CO5** **examine** transcription factor identification through in silico analysis.

### Catalog Description

How science is done through the practice of experimental inquiry. Under the guidance of the instructor and teaching assistant, students work in small teams to design experiments and test their designs in a fully equipped, state-of-the-art laboratory. A number of technical skills are utilized, including gene cloning, DNA amplification and mutagenesis by PCR, in vitro transcription and translation, and purification and analysis of proteins. Students assemble synthetic genes from parts and analyze the contribution of these parts in the regulation of gene expression, from transcription to translation.

## Course Content

- 1 Isolation of mRNA by affinity chromatography. (5Hours)
- 2 Pore plate technique (5Hours)
- 3 Plate streaking technique (5Hours)
- 4 Competent Cell preparation (5Hours)
- 5 Bacterial Transformation (10 Hours)
- 6 Replica Plating (10 Hours)
- 7 Spread plate technique (10 Hours)
- 8 MIC Assay (10 Hours)

## Reference Books

1. Molecular Biology of the Gene (2008) 6th ed., Watson, J.D., Baker, T.A., Bell, S.P., Gann, A., Levine, M. and Losick, R., Cold Spring Harbor Laboratory Press, Cold spring Harbor (New York), ISBN:0-321-50781 /ISBN:978-0-321-50781-5.
2. Lehninger: Principles of Biochemistry (2013) 6th ed., Nelson, D.L. and Cox, M.M., W. H. Freeman & Company (New York), ISBN:13: 978-1-4292-3414-6 / ISBN:10-14641-0962-1.

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

## Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

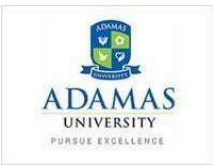
Mapping between COs and Pos		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	isolate and <b>analyse</b> of total nucleic acids from plant tissue.	PO1, PO11
CO2	<b>Interpret</b> the results of total RNA isolation.	PO1, PO2, PO3,
CO3	<b>categorize</b> between parent and recombinant plasmid containing bacteria.	PO1, PO2, PO3, PO4, PO5, PO11
CO4	<b>demonstrate</b> protein modelling.	PO1, PO5, PO12,
CO5	<b>examine</b> transcription factor identification through in silico analysis.	PO1, PO6, PO8, PO9







## Model Question Paper

<b>Name:</b>  <b>Enrolment No:</b>		
<b>Course: BIC12029 – Gene expression and regulation Lab</b> <b>Program: B.SC Biochemistry</b> <span style="float: right;"><b>Time: 03Hrs.</b></span> <b>Semester: Even 2019-20</b> <span style="float: right;"><b>Max. Marks: 40</b></span>		
1.	<b>What</b> kind of cells is used for extraction of DNA in the experiment?. (U)	<b>CO1</b>
2.	Isolation of genomic DNA follows the same principles as that of obtaining plasmid from E. coli. Which of the following is not included in it?. (U)	<b>CO2</b>
3.	How many methods are there for obtaining the plasmid DNA from the bacteria? (R)	<b>CO4</b>
4.	Proteins can be removed via by which treatment? (Ap)	<b>CO3, CO5</b>
5.	The nucleic acid remaining in the solution can be precipitated by addition of sodium or ammonium acetate and ethanol? (Ap)	<b>CO3</b>
6.	Adsorption onto a solid phase support followed by elution is used as an alternative for separation of which component?? (Ap)	<b>CO2</b>
7.	<b>Which</b> components bind to the solid column made of silica, under high salt concentration?? (U)	<b>CO4</b> <b>CO2</b> <b>CO3</b>
8.	Purification of DNA by using silica derivatized groups by DEAE is <b>termed</b> as? (U)	<b>CO2</b>

<b>BIC11030</b>	<b>BIOINFORMATICS (THEORY)</b>	<b>L</b>	<b>T</b>	<b>P</b>	<b>C</b>
<b>Version 1.0</b>	<b>Contact Hours - 60</b>	3	1	0	4
<b>Pre-requisites/Exposure</b>	Basic understanding of biochemistry and DNA, RNA				
<b>Co-requisites</b>	--				

### Course Objectives

1. To provide students with apt understanding of informatics for biological data.
2. It will also provide in depth knowledge of data bases, sequence analysis, alignment and phylogeny.
3. It will also offer an outline of basic prediction methods for biological system.
4. In depth understanding of structure prediction for proteins and RNA.

### Course Outcomes

On completion of this course, the students will be able to

**CO 1 explain** various components of data base, data transfer and data bases & organize them accordingly.

**CO 2 illustrate**, relate and interpret various biological data bases and file formats

**CO 3 perceive** alignment, perform phylogenetic analysis and build the same

**CO 4 organize** and deduct the omics data, primarily genomics and proteomics

**CO 5 design**, set up experiments for structural modelling also summarize drug designing protocols.

### Catalog Description

The course of 'bioinformatics' will help to understand the basic concept and application of computational biology. This course includes comprehensive approach through studying databases, sequence alignment and phylogenetic analysis. Furthermore, the application of computation structure prediction will also be elaborated. All the lectures will be devoted on discussion of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as power point presentation, audio visual virtual lab session as per requirement. The tutorials will enable the students with problem-solving ability led by the course coordinator. Students will perceive the basic concepts of the subject via exercise and discussions with the coordinator.

## Course Content

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### **Unit 1 Introduction to Computer Fundamentals**[12 Lecture Hours]

RDBMS - Definition of relational database Mode of data transfer (FTP, SFTP, SCP), advantage of encrypted data transfer

### **Unit 2 Introduction to Bioinformatics and Biological Databases Biological databases** [12 Lecture Hours]

- nucleic acid, genome, protein sequence and structure, gene expression databases, Database of metabolic pathways, Mode of data storage – File formats-FASTA, BLAST, Genbank and Uniprot, Data submission & retrieval from NCBI, EMBL, DDBJ, Uniprot, PDB

### **Unit 3** [12 Lecture Hours]

Sequence Alignments, Phylogeny and Phylogenetic trees- Local and Global Sequence alignment, pairwise and multiple sequence alignment. Scoring an alignment, Gap penalties, ClustalW, scoring matrices, PAM & BLOSUM series of matrices Types of phylogenetic trees, Different approaches of phylogenetic tree construction - UPGMA, Neighbour joining, Maximum Parsimony, Maximum likelihood

### **Unit 4 Genome organization and analysis**[12 Lecture Hours]

Diversity of Genomes: Viral, prokaryotic & eukaryotic genomes, Genome, transcriptome, proteome, 2-D gel electrophoresis, MaldiToff spectroscopy, Major features of completed genomes: E.coli, S.cerevisiae, Arabidopsis, Human

### **Unit 5 Protein Structure Predictions**[12 Lecture Hours]

Hierarchy of protein structure - primary, secondary and tertiary structures, modeling Structural Classes, Motifs, Folds and Domains Protein structure prediction in presence and absence of structure template Energy minimizations and evaluation by Ramachandran plot Protein structure and rational drug design

#### ***Text Book:***

1. Essential Bioinformatics, Jin XIONG, CAMBRIDGE

#### ***Reference Books:***

1. Sanjay S (2003). A First Course in Computers, Vikas Publishing House
2. Pradeep and Sinha Preeti (2007). Foundations of Computing, 4th ed., BPB Publications
4. Rastogi SC., Mendiratta N. and Rastogi P. (2007). Bioinformatics: methods and applications, genomics, proteomics and drug discovery, 2nd ed. Prentice Hall India Publication
5. Primrose and Twyman (2003). Principles of Genome Analysis & Genomics. Blackwell

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

<b>Components</b>	<b>Mid Term</b>	<b>Attendance</b>	<b>Class Assessment</b>	<b>End Term</b>
<b>Weightage (%)</b>	<b>20</b>	<b>10</b>	<b>30</b>	<b>40</b>

**Relationship between the Course Outcomes (COs) and Program Outcomes (POs)**

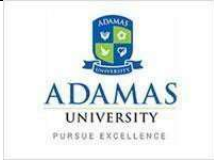
<b>Mapping between COs and POs</b>		
	<b>Course Outcomes (COs)</b>	<b>Mapped Program Outcomes</b>
<b>CO1</b>	<b>explain</b> various components of data base	<b>PO1, PO3</b>
<b>CO2</b>	<b>Illustrate</b> , relate and interpret various biological data bases and file formats	<b>PO1, PO2, PO4</b>
<b>CO3</b>	<b>perceive</b> alignment, perform phylogenetic analysis and build the same	<b>PO1, PO2, PO4</b>
<b>CO4</b>	<b>organize</b> and deduct the omics data, primarily genomics and proteomics	<b>PO1, PO2, PO5, PO8</b>
<b>CO5</b>	design set up experiments for structural modelling also summarize drug designing protocols.	<b>PO1, PO2, PO3, PO5, PO8</b>

Course Code	Course Title	Fundamental Knowledge	Critical thinking	Skill	Technical Knowledge	Logical Thinking	Problem identification ability	Analytical Knowledge	Career goals	Team Work	Sustainable Development to environment	Development to society	Development to humanity
		PO 1	PO 2	PO 3	PO 4	PO 5	PO 6	PO 7	PO 8	PO 9	PO 10	PO 11	PO 12
<b>BIC11 030</b>	BIOINFORMATICS	3	3	2	2	2	-	-	2	-	-	-	-

1=weakly mapped

2= moderately mapped

3=strongly mapped

<b>Name:</b>		 <b>ADAMAS</b> UNIVERSITY <small>PURSUE EXCELLENCE</small>	
<b>Enrolment No:</b>			
<b>Course: BIC11030- BIOINFORMATICS (THEORY)</b>		<b>Time: 03Hrs.</b>	
<b>Program: B.Sc. Biochemistry</b>		<b>Max. Marks: 40</b>	
<b>Semester: Odd 2019-20</b>			
<b>Instructions:</b>			
Attempt any <b>four</b> questions from <b>Section A</b> (each carrying 5 marks); any <b>two</b> questions from <b>Section B</b> (each carrying 10 marks).			
<b>SECTION A ( Attempt any Four questions)</b>			
1.	What are <i>tuples</i> and <i>attributes</i> in a relational data base? How are they organized to build up the data base?(U)	<b>2+3</b>	<b>CO1</b>
2.	Mention a scenario with the rationale when you will chose PAM matrices over BLOSUM for scoring an alignment.(U)	<b>5</b>	<b>CO2</b>
3.	Write a brief note on <i>PRALINE</i> , mentioning its applicability(Ap)	<b>5</b>	<b>CO3</b>
4.	Outline the features of eukaryotic genes. Name two gene prediction tools.(An)	<b>3+2</b>	<b>CO4</b>
5	Define <i>motif</i> and <i>domain</i> of a protein.(U)	<b>2.5X2</b>	<b>CO5</b>
<b>SECTION B (Attempt any Two questions)</b>			
6.	Compare <i>rooted</i> and <i>unrooted</i> phylogenetic trees. What is the significance of out-group in an evolutionary tree? What are the differences in phyllogram and cladogram?(U)	<b>4+2+4</b>	<b>CO3</b>
7.	How gene phylogeny differs with species phylogeny? Discuss the basis of building up a NJ tree?(Ap)	<b>5+5</b>	<b>CO1</b> <b>CO2</b>
8.	What can be the possible criterion for <i>ab initio</i> operon prediction? Fix scoring criterion for operon prediction considering <i>lac</i> operon as a model.(U)	<b>5+5</b>	<b>CO1</b> <b>CO2</b>
9	What are the advantages of cryoEM over X-ray crystallography in determining protein structure? How a coordinate file is built in PDB format? How can you visualize structure from PDB file?(U)	<b>3+3+4</b>	<b>CO5</b>

<b>BIC11032</b>	<b>DSE I Basic Microbiology (Theory)</b>	L	T	P	C
<b>Version 1.0</b>	<b>Contact Hours: 60</b>	3	1	0	4
<b>Pre-requisites/Exposure</b>	Knowledge of Biomolecules				
<b>Co-requisites</b>	-				

### Course Objectives

1. To understand different type of microorganisms.
2. To provide knowledge about structures of different types of bacteria.
3. To understand the different types of virus.
4. To be able to know different types of protozoa.
5. To gain knowledge about different types of fungus.

### Course Outcomes

On completion of this course, the students will be able to

- CO1. Classify and **illustrate** importance of viruses, viroids and prions.
- CO2. **Categorize** and discuss importance of bacteria.
- CO3. Classify and **discuss** importance of algae.
- CO4. **Categorize** the importance of fungus.
- CO5. Classify and **discuss** importance of protozoa.

### Catalog Description

Microbiology is the study of microorganisms, those being unicellular, multicellular, or acellular. Microbiology encompasses numerous sub-disciplines including virology, bacteriology, protistology, mycology, immunology and parasitology. Eukaryotic microorganisms possess membrane-bound organelles and include fungi and protists, whereas prokaryotic organisms—all of which are microorganisms—are conventionally classified as lacking membrane-bound organelles and include Bacteria and Archaea. Microbiologists traditionally relied on culture, staining, and microscopy. However, less than 1% of the microorganisms present in common environments can be cultured in isolation using current means. Microbiologists often rely on molecular biology tools such as DNA sequence based identification, for example the 16S rRNA gene sequence used for bacteria identification. Viruses have been variably classified as organisms as they have been considered either as very simple microorganisms or very complex molecules. Prions, never considered as microorganisms, have been investigated by virologists, however, as the clinical effects traced to them were originally presumed due to chronic viral infections, and virologists took search—discovering "infectious proteins". The goal of this paper is to conceptualize about the characteristics of microorganism.

### Course Content



## Unit I History of Development of Microbiology [10 Lecture Hours]

Development of microbiology as a discipline, Spontaneous generation vs. biogenesis. Contributions of Anton von Leeuwenhoek, Louis Pasteur, Robert Koch, Joseph Lister, Alexander Fleming. Role of microorganisms in fermentation, Germ theory of disease, Development of various microbiological techniques and golden era of microbiology,

Establishment of fields of medical microbiology and immunology through the work of Paul Ehrlich, Elie Metchnikoff, Edward Jenner

## Unit II Diversity of Microbial world [10 Lecture Hours]

Binomial Nomenclature, Whittaker's five kingdom and Carl Woese's three kingdom classification systems and their utility. Difference between prokaryotic and eukaryotic microorganisms. General characteristics of different groups: acellular microorganisms (Viruses, Viroids, Prions) and Cellular microorganisms (Bacteria, Algae, Fungi and Protozoa) with emphasis on distribution and occurrence, morphology, mode of reproduction and economic importance.

## Unit III Viruses, viroids and prions [10 Lecture Hours]

An introduction to viruses with special reference to the structure and replication of the following: Poxvirus, Poliovirus, HIV, T4 and  $\lambda$  phage, lytic and lysogenic cycles.

## Unit IV Bacteria [10 Lecture Hours]

An account of typical eubacteria, chlamydiae & rickettsiae (obligate intracellular parasites), mycoplasma, and archaeobacteria (extremophiles). Applications of bacteria in industry, environment and food.

## Unit V Algae [10 Lecture Hours]

History of phycology; General characteristics of algae including occurrence, thallus organization, algae cell ultra structure, pigments, flagella, eyespot food reserves and vegetative, asexual and sexual reproduction. Applications of Algae in agriculture, industry, environment and food.

## Unit VI Fungi [8 Lecture Hours]

Historical developments in the field of Mycology, significant contributions of eminent mycologists. General characteristics of fungi including habitat, distribution, nutritional requirements, fungal cell ultra-structure, thallus organization and aggregation, fungal wall structure and synthesis, asexual reproduction, sexual reproduction, heterokaryosis, heterothallism and parasexual mechanism. Economic Importance of Fungi in Agriculture, environment, Industry, medicine, food, biodeterioration, mycotoxins

## Unit VII Protozoa [2 Lecture Hours]

General characteristics with special reference to Amoeba.

## Reference Books

1. Prescott, Harley, Klein's Microbiology (2008) 7th Ed., Willey, J.M., Sherwood, LM, Woolverton, C.J. McGraw Hill International Edition (New York) ISBN: 978-007-126727.
2. Mandell, Douglas and Bennett.S, Principles and practices of Infectious diseases, 7<sup>th</sup> edition, Volume, 2. Churchill LivingstoneElsevier
3. Sherris Medical Microbiology: An Introduction to Infectious Diseases by Kenneth J.Ryan, C. George Ray, Publisher:McGraw-Hill
4. Medical Microbiology by Patrick R. Murray, Ken S. Rosenthal, Michael A. Pfaller, Elsevier HealthSciences.
5. Brooks GF, Carroll KC, Butel JS and Morse SA. (2007). Jawetz, Melnick and Adelberg's Medical Microbiology. 24th edition. McGraw Hill Publication
6. Goering R, Dockrell H, Zuckerman M and Wakelin D. (2007). Mims' Medical Microbiology. 4th edition. Elsevier.

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

## Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and Pos		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	Classify and <b>illustrate</b> importance of viruses, viroids and vrions	PO1, PO3, PO4, PO5, PO8, PO11
CO2	<b>Categorize</b> and discuss importance of bacteria.	PO1, PO3, PO4, PO5, PO8, PO11
CO3	Classify and <b>discuss</b> importance of algae.	PO1, PO4, PO5, PO8, PO11
CO4	<b>Categorize</b> the importance of fungus.	PO1, PO4, PO5, PO8, PO11
CO5	Classify and <b>discuss</b> importance of protozoa.	PO1, PO4, PO5, PO8,

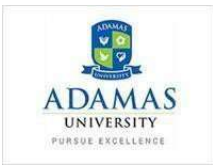
Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
<b>BIC11032</b>	DSE I Basic Microbiology	3	-	2	3	3	-	-	3	-	-	3	-

1=weakly mapped

2= moderately mapped

3=strongly mapped

## Model Question Paper

<b>Name:</b>  <b>Enrolment No:</b>	 <p style="font-size: small; margin: 0;">ADAMAS UNIVERSITY PURSUE EXCELLENCE</p>		
<b>Course: BIC11032 –DSE I Basic Microbiology</b>			
<b>Program: B.Sc Biochemistry</b> <b>Semester: Odd2020-21</b>	<b>Time: 03 Hrs.</b> <b>Max. Marks:40</b>		
<b>Instructions:</b> Attempt any three questions from <b>Section A</b> (each carrying 4 marks); any <b>Two Questions</b> from <b>Section B</b> (each carrying 10 marks). <b>Section C</b> is Compulsory (carrying 8 marks).			
<b>Section A ( Attemptany Three)</b>			
1.	Mycoplasmas are considered atypical, rather than gram positive or gram negative. <b>Why</b> is it not possible to use a standard Gram stain technique to classify these bacteria? <b>What</b> color would they appear if stained using a Gram stain technique?(U)	<b>3</b> <b>1</b>	<b>CO1</b>
2.	If Spallanzani had unknowingly poked a hole in the top of his flask of meat broth, <b>what</b> would this have implied about the theory of spontaneous generation?(An)	<b>4</b>	<b>CO2</b>
3.	Both prokaryotes and eukaryotes have ribosomes, but some antibiotics specifically target prokaryotic ribosomes. <b>How</b> is it possible to use these antibiotics to treat human diseases and <b>how</b> do they affect eukaryotic cells?(Ap)	<b>2</b> <b>2</b>	<b>CO4</b>
4.	Would pili be more advantageous to bacteria in a rapidly changing or in an unchanging environment? Please <b>explain</b> your answer.(U)	<b>1</b> <b>3</b>	<b>CO3</b> <b>CO4</b>
<b>SECTION B (Attempt any Two Questions)</b>			
5.	<b>Describe</b> the structure of the cell wall of E.coli. Mention the effect of penicillin on this cell wall(U)	<b>5</b> <b>5</b>	<b>CO1</b> <b>CO2</b>
6.	Compare between the lysogenic and lytic cycle of bacteriophage at	<b>5</b> <b>5</b>	<b>CO4</b> <b>CO2</b>

	the light of molecular dimension(U)		<b>CO3</b>
7.	Compare the structures of mycoplasma and protozoa with proper description. Write down the sterilization principle of autoclave and laminar air flow.(Ap)	4 3 3	<b>CO3</b> <b>CO5</b>
	<b>SECTION C is Compulsory</b>		
8.	All microbial cells in a bacterial culture are not instantaneously killed when they are exposed to a lethal chemical agent. Why? Explain briefly. Blood agar is both differential and enriched media. Explain. Contrast between 'Chemostat' and 'Turbidostat'.(Ap)	2 2 2 2	<b>CO2</b>

<b>BIC11034</b>	<b>DSE I Molecular Basis of Noninfectious Human Disease (Theory)</b>	L	T	P	C
<b>Version 1.0</b>	<b>Contact Hours: 60</b>	3	1	0	4
<b>Pre-requisites/Exposure</b>	Basic Knowledge of Biochemistry and Microbiology				
<b>Co-requisites</b>	-				

### Course Objectives

1. To gain knowledge about different type of nutritional disorders.
2. To study different types of metabolic disorders.
3. To provide knowledge about the different types of cancer and multifactorial disorders.
4. To gain insight about different types of protein misfolding disorders.
5. Outline different types of monogenic diseases.

### Course Outcomes

On completion of this course, the students will be able to

1. **Summarize** different nutritional disorders.
2. **Illustrate** different metabolic and lifestyle disorders.
3. **Summarize** importance of multi factorial complex disorders and Cancer.
4. **Illustrate** different protein misfolding disorders.
5. **Discuss** different monogenic diseases.

### Catalog Description

A noninfectious disease is a disease that is not transmissible directly from one person to another. This category include Parkinson's disease, autoimmune diseases, strokes, most heart diseases, most cancers, diabetes, chronic kidney disease, osteoarthritis, osteoporosis, Alzheimer's disease, cataracts, and others. NCDs may be chronic or acute. Most are non-infectious, although there are some non-communicable infectious diseases, such as parasitic diseases in which the parasite's life cycle does not include direct host-to-host transmission.

Noninfectious diseases are the leading cause of death globally. In 2012, they caused 68% of all deaths (38 million) up from 60% in 2000. About half were under age 70 and half were women. Risk factors such as a person's background, lifestyle and environment increase the likelihood of certain NCDs. Every year, at least 5 million people die because of tobacco use and about 2.8 million die from being overweight. High cholesterol accounts for roughly 2.6 million deaths and 7.5 million die because of high blood pressure. In this paper the molecular mechanism behind these kinds of disease will be investigated.

## Course Content

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- 1. Nutritional disorders:** Overview of major and minor nutrient components in the diet. Balanced diet and the concept of RDA. Nutrient deficiencies; Kwashiorkor and Marasmus, Scurvy, beriberi, pellagra and B12 deficiency, Xerophthalmia and Night blindness, Vitamin D deficiency, Vitamin K deficiency. Discuss with relation to biochemical basis for symptoms. **[8 Lecture Hours]**
- 2. Metabolic and Lifestyledisorders:** Obesity and eating disorders like Anorexia nervosa and Bulimia. Diabetes mellitus A metabolic syndrome and the relationship with hypertension, obesity, hypothyroidism and stress. Cardio vascular disorders and Atherosclerosis-defining the broad spectrum of ailments that fall in this category, understanding the factors that contribute to the syndrome, stages of disorder and the management of the condition. Irritable bowel syndrome- biochemistry behind the disorder and the influence of diet, stress and environment on the condition. **[8 Lecture Hours]**
- 3. Multifactorial complex disorders and Cancer:** Understanding the definition of multifactorial diseases. Polygenic diseases and the relationship of environmental factors and genetic makeup in the onset of diseases. **[6 Lecture Hours]**
- 4. Cancer:** characteristics of a transformed cell, causes and stages of Cancer, molecular basis for neoplastic growth and metastasis, Proto-oncogenes and tumor suppressor genes; Cancer causing mutations; Tumor viruses; Biochemical analysis of cancer; Molecular approaches to cancer treatment. **[8 Lecture Hours]**
- 5. Disorders of mood :** Schizophrenia, dementia and anxiety disorders. **[6 Lecture Hours]**
- 6. Polycystic ovarian syndrome, Parkinson's disease, ALS. [8 Lecture Hours]**
- 7. Diseases due to misfolded proteins:** Introduction to protein folding and proteasome removal of misfolded proteins; etiology and molecular basis for Alzheimer's, Prion diseases, Huntington's Chorea, sickle cell anemia, Thalassemia. **[8 Lecture Hours]**
- 8. Monogenic diseases:** Inborn errors in metabolism: PKU, Alkaptonuria, Maple syrup urine disease; Receptor and transport defects: Cystic fibrosis, Long QT syndrome, familial hypercholesterolemia, Achondroplasia. Hemoglobinopathies and clotting disorders. **[8 Lecture Hours]**

## Reference Books

1. Textbook of Biochemistry with Clinical Correlations (2011) Devlin, T.M. John Wiley & Sons, Inc. (New York), ISBN:978-0-4710-28173-4.
2. Introduction to Human Physiology (2013) 8th edition; Lauralee Sherwood. Brooks/Cole, Cengage Learning. The World of the cell, 7th edition (2009)
3. Genetics (2012) Snustad and Simmons, 4. Cooper, G.M. and Hausman, R.E. 2009
5. The Cell: A Molecular Approach. 5th edition. ASM Press & Sunderland, Washington, D.C.; Sinauer Associates, MA.

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

<b>Components</b>	<b>Mid Term</b>	<b>Attendance</b>	<b>Class Assessment</b>	<b>End Term</b>
<b>Weightage (%)</b>	<b>20</b>	<b>10</b>	<b>30</b>	<b>40</b>

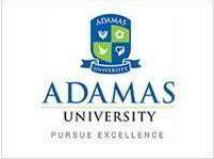
**Relationship between the Course Outcomes (COs) and Program Outcomes (POs)**

<b>Mapping between COs and Pos</b>		
	<b>Course Outcomes (COs)</b>	<b>Mapped Program Outcomes</b>
<b>CO1</b>	<b>Summarize</b> different nutritional disorders.	<b>PO1, PO2, PO3, PO5, PO6, PO7, PO8, PO11</b>
<b>CO2</b>	<b>Illustrate</b> different metabolic and lifestyle disorders.	<b>PO1, PO2, PO3, PO5, PO6, PO7, PO9, PO11</b>
<b>CO3</b>	<b>Summarize</b> importance of multi factorial complex disorders and Cancer.	<b>PO1, PO2, PO3, PO5, PO6, PO7, PO8, PO11</b>
<b>CO4</b>	<b>Illustrate</b> different protein misfolding disorders.	<b>PO1, PO2, PO3, PO5, PO6, PO7, PO11</b>
<b>CO5</b>	<b>Discuss</b> different monogenic diseases.	<b>PO1, PO2, PO3, PO5, PO6, PO7, PO11</b>





## Model Question Paper

<b>Name:</b>  <b>Enrolment No:</b>			
<p style="text-align: center;"><b>Course: BIC11034 – DSE IMolecular Basis of Noninfectious Human Disease</b></p> <p><b>Program: B.Sc Biochemistry</b> <span style="float: right;"><b>Time: 03Hrs.</b></span></p> <p><b>Semester: Odd2020-21</b> <span style="float: right;"><b>Max. Marks: 40</b></span></p>			
<p><b>Instructions:</b>          Attempt any three questions from <b>Section A</b> (each carrying 4 marks); any <b>Two Questions</b> from <b>Section B</b> (each carrying 10 marks). <b>Section C</b> is Compulsory (carrying 8 marks).</p>			
<b>Section A ( Attemptany Three)</b>			
1.	<b>Discuss</b> the physiological disorders related to VitaminB12 deficiency at molecularlevel.(U)	4	CO1
2.	<b>Describe</b> the classification of cancers briefly.(U)	4	CO3
3.	Write down the differences between simple goiter and exophthalmic goiter along with their treatment procedure.	4	CO2
4.	<b>Illustrate</b> the causes and symptoms of maple syrup urine disease and phenylketonuria.(Ap)	2 2	CO5
<b>SECTION B (Attempt any Two Questions)</b>			
5.	<b>Describe</b> the signalling pathway for establishing the cancer by mutated k-ras gene. <b>How</b> the checkpoints of cell cycle regulates the occurrence of cancer?(An)	5 5	CO3
6.	<b>Discuss</b> the role of mutant IRS-1 in the occurrence of type II diabetes mellitus. Which types of drugs are designed by tracing the specific signalling pathways? State the possible mechanism of action.(Ap)	5 3 2	CO2

7.	<b>Discuss</b> the signalling mechanism behind the cardiac failure due to the hypercholesterolemia and the preventive measures and treatments against it.(U)	6 2 2	<b>CO3</b> <b>CO5</b>
	<b>SECTION C is Compulsory</b>		
8.	<b>Describe</b> the different causes of Parkinson's disease. Write down the names of the drugs designed against it and mention the mechanism of action of any one of them.(Ap)	5 2 3	<b>CO2</b>

<b>BIC12031</b>	<b>BIOINFORMATICS (LAB)</b>	<b>L</b>	<b>T</b>	<b>P</b>	<b>C</b>
<b>Version 1.0</b>	<b>Contact Hours - 45</b>	0	0	3	2
<b>Pre-requisites/Exposure</b>	Basic Knowledge of biochemistry and computers				
<b>Co-requisites</b>	--				

### Course Objectives

1. Developing knowledge of computational biology
2. To have general perception of data base and data structure
3. To get accustomed with basic bioinformatics methods
4. To obtain the broader perspective of application of bioinformatics

### Course Outcomes

On completion of this course, the students will be able to

**CO1 utilize** various operating systems

**CO2 survey** and combine information from biological data bases

**CO3 deduce** evolutionary relationship through sequence alignment and build phylogenetic tree

**CO4 interpret** ORFs from DNA sequence data and design specific primers for performing PCR

**CO5 construct** structural models of proteins and DNA; perceive the basis of drug designing protocols.

### Catalog Description

The core-course of 'bioinformatics' will help to understand the basic concept and application of computational biology. This course includes comprehensive approach through studying data

bases, sequence alignment and phylogenetic analysis. Furthermore, the application of computation structure prediction will also be elaborated. All the lectures will be devoted to discussion of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as power point presentation, audio visual virtual lab session as per requirement. The tutorials will enable the students with problem-solving ability led by the course coordinator. Students will perceive the basic concepts of the subject via exercise and discussions with the coordinator.

### Course Content

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1. Database Mining
2. Pairwise Alignment
3. Multiple Sequence Alignment
4. Translate and Backtranslate
5. BLAST
6. Gene and ORF prediction
7. Protein Motif & Function Prediction
8. Molecular Phylogeny
9. Homology Modelling
10. Primer Designing

### Text Book:

1. Essential Bioinformatics, Jin XIONG, CAMBRIDGE

### Reference Books:

1. Sanjay S (2003). A First Course in Computers, Vikas Publishing House
2. Pradeep and Sinha Preeti (2007). Foundations of Computing, 4th ed., BPB Publications
4. Rastogi SC., Mendiratta N. and Rastogi P. (2007). Bioinformatics: methods and applications, genomics, proteomics and drug discovery, 2nd ed. Prentice Hall India Publication
5. Primrose and Twyman (2003). Principles of Genome Analysis & Genomics. Blackwell

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

### Relationship between the Course Outcomes (COs) and Program Outcomes (POs)


Mapping between COs and POs		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	<b>utilize</b> various operating systems	PO1, PO3
CO2	<b>survey</b> and combine information from biological data bases	PO1, PO2, PO4
CO3	<b>deduce</b> evolutionary relationship through sequence alignment and build phylogenetic tree	PO1, PO2, PO4
CO4	<b>interpret</b> ORFs from DNA sequence data and design specific primers for performing PCR	PO1, PO2, PO5, PO8
CO5	<b>construct</b> structural models of proteins and DNA; perceive the basis of drug designing protocols.	PO1, PO2, PO3, PO5, PO8

Course Code	Course Title	Fundamental Knowledge	Critical thinking	Skill	Technical Knowledge	Logical Thinking	Problem identification ability	Analytical Knowledge	Career goals	Team Work	Sustainable Development to environment	Development to society	Development to humanity
		PO 1	PO 2	PO 3	PO 4	PO 5	PO 6	PO 7	PO 8	PO 9	PO 10	PO 11	PO 12
<b>BIC12031</b>	BIOINFORMATICS (LAB)	3	3	2	2	2	-	-	1	-	-	-	-

1=weakly mapped

2= moderately mapped

3=strongly mapped

<b>Name:</b>			
<b>Enrolment No:</b>			
<b>Course: BIC12031 - BIOINFORMATICS LAB</b> <b>Program: B.Sc. BIOCHEMISTRY</b> <b>Time: 03 Hrs. Semester: Odd2019-20</b> <b>Max. Marks:40</b>			
<b>Instructions:</b> Attempt any <b>four</b> questions from <b>Section A</b> (each carrying 5 marks); any <b>two</b> questions from <b>Section B</b> (each carrying 10 marks).			
<b>SECTION A ( Attemptall questions)</b>			
1.	Design primer for cloning <i>Leishmania infantum</i> AQP1 gene in pET16b vector for expressing an N-terminally His-tagged LiAQP1.(Ap)	<b>20</b>	<b>CO1</b>
2.	Predict conserved motifs and domains for LiAQP1.(Ap)	<b>10</b>	<b>CO2</b>
3.	Lab note book.(U/Ap/An)	<b>5</b>	<b>CO1, CO2, CO3, CO4</b>
4.	Viva-voce.(U/Ap/An)	<b>5</b>	<b>CO1, CO2, CO3, CO4</b>

<b>BIC12033</b>	<b>DSE I Basic Microbiology Lab (Practical)</b>	L	T	P	C
<b>Version 1.0</b>	<b>Contact Hours: 45</b>	0	0	3	2
<b>Pre-requisites/Exposure</b>	Basic understanding of Biochemistry and Microbiology				
<b>Co-requisites</b>	-				

### Course Objectives

1. To gain the knowledge about different type of microorganisms
2. To distinguish between gram positive and gram negative bacteria.
3. To characterize the different kinds of fungi.
4. To characterize different kinds of virus.
5. To gain the knowledge about different types of protozoa.

### Course Outcomes

On completion of this course, the students will be able to

**CO1 Classify** and discuss identify of viruses, viroids and prions through microscope.

**CO2 Develop** concept basic principle and applications of important instruments (biological safety cabinets, autoclave, incubator, BOD incubator, hot air oven, light microscope, pH meter)

**CO3 Test** the proper sterilization of culture media for bacterial cultivation.

**CO4 Examine** different shapes of bacteria will be able to characterize them using permanent slides/ pictographs

**CO5 Examine** the fungus by cotton blue staining.

### Catalog Description

Microbiology is the study of microorganisms, those being unicellular, multicellular, or acellular. Microbiology encompasses numerous sub-disciplines including virology, bacteriology, protistology, mycology, immunology and parasitology. Eukaryotic microorganisms possess membrane-bound organelles and include fungi and protists, whereas prokaryotic organisms—all of which are microorganisms—are conventionally classified as lacking membrane-bound organelles and include Bacteria and Archaea. Microbiologists traditionally relied on culture, staining, and microscopy. However, less than 1% of the microorganisms present in common environments can be cultured in isolation using current means. Microbiologists often rely on molecular biology tools such as DNA sequence based identification, for example the 16S rRNA gene sequence used for bacteria identification. Viruses have been variably classified as organisms as they have been considered either as very simple microorganisms or very complex molecules. Prions, never considered as microorganisms, have been investigated by virologists, however, as the clinical effects traced to them were originally presumed due to chronic viral infections, and virologists took search—discovering "infectious proteins". The goal of this paper to identify, analyze and characterize the microorganism.

### Course Content



1. Microbiology Laboratory Practices and Biosafety. [10 LectureHours]
2. To study the principle and applications of important instruments (biological safety cabinets, autoclave, incubator, BOD incubator, hot air oven, light microscope, pH meter). [10 LectureHours]
3. Preparation and sterilization of culture media for bacterial cultivation[10 Lecture Hours]
4. Study of different shapes of bacteria, fungi, algae, protozoa using permanent slides/pictographs [10 LectureHours]
5. Staining of bacteria using Gram stain [10 LectureHours]
6. Isolation of pure cultures of bacteria by streaking method. [5 LectureHours]
7. Estimation of CFU count [5 LectureHours]

### Reference Books

- 1.Prescott, Harley, Klein's Microbiology (2008) 7th Ed., Willey, J.M., Sherwood, L.M., Woolverton, C.J. McGraw Hill International Edition (New York) ISBN: 978-007-126727.
- 2.Mandell, Douglas and Bennett.S, Principles and practices of Infectious diseases, 7th edition, Volume, 2. Churchill LivingstoneElsevier
- 3.Sherris Medical Microbiology: An Introduction to Infectious Diseases by Kenneth J.Ryan, C. George Ray, Publisher:McGraw-Hill
4. Medical Microbiology by Patrick R. Murray, Ken S. Rosenthal, Michael A. Pfaller,Elsevier HealthSciences.
- 5.Brooks GF, Carroll KC, Butel JS and Morse SA. (2007).Jawetz,
- 6.Melnick and Adelberg's Medical Microbiology. 24th edition. McGraw Hill Publication
- 7.Goering R, Dockrell H, Zuckerman M and Wakelin D. (2007). Mims' Medical Microbiology. 4th edition.Elsevier.

### Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

### Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and Pos		
	Course Outcomes (COs)	Mapped Program Outcomes
<b>CO1</b>	<b>classify</b> and <b>discuss</b> identify of viruses, viroids and prions through microscope.	<b>PO1, PO2, PO3, PO4, PO7, PO8</b>
<b>CO2</b>	<b>develop</b> concept basic principle and applications of important instruments (biological safety cabinets, autoclave, incubator, BOD incubator, hot air oven, light microscope, pH meter)	<b>PO1, PO2, PO3, PO4, PO7, PO8</b>
<b>CO3</b>	<b>test</b> the proper sterilization of culture media for bacterial cultivation.	<b>PO1, PO2, PO3, PO4, PO7, PO8</b>
<b>CO4</b>	<b>illustrate</b> different shapes of bacteria will be able to characterize them using permanent slides/ pictographs	<b>PO1, PO2, PO3, PO4, PO7, PO8</b>
<b>CO5</b>	<b>illustrate</b> the fungus by cotton blue staining	<b>PO1, PO2, PO3, PO4, PO7, PO8</b>

		Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking	Problem Identification Ability	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to Environment	Development to Society.	Development to Humanity
		PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
<b>BIC12033</b>	DSE I Basic Microbiology Lab	3	3	3	3	-	-	3	3	-	-	-	-

1=weakly mapped

2= moderately mapped

3=strongly mapped

## Model Question Paper

<b>Name:</b>  <b>Enrolment No:</b>			
<p style="text-align: center;"><b>Course: BIC12033– DSE I Basic Microbiology Lab</b></p> <p><b>Program: B.Sc Biochemistry</b> <span style="float: right;"><b>Time: 03Hrs.</b></span></p> <p><b>Semester: Odd2020-21</b> <span style="float: right;"><b>Max. Marks:40</b></span></p>			
<p><b>Instructions:</b>          Attempt any two questions from <b>Section A</b> (each carrying 10 marks); any <b>Two Questions</b> from <b>Section B</b> (each carrying 10 marks) <b>is Compulsory.</b></p>			
<b>Section A ( Attemptany Three)</b>			
1.	Identify the specimen of the two given slides (Ap) ,write down their specific characteristics and their relation with infectious disease.(An)	<b>5 5</b>	<b>CO1,CO2, CO3,CO4,CO5</b>
2.	Perform the Gram staining procedure with the supplied bacterial sample(Ap),write down the principle, procedure, observation and inference.(An)	<b>10</b>	<b>CO4</b>
3.	Perform the streak plate to grow the single colony of E.coli on nutrient agar plate(Ap),write down the principle, procedure, observation and inference.(An)	<b>10</b>	<b>CO3</b>
<b>SECTION B (Attempt any Two Questions)</b>			
5.	<b>Viva-Voce (U/An/Ap/R)</b>	<b>10</b>	<b>CO1, CO2,CO3, CO4,CO5</b>
6.	<b>Laboratory Note Book(U/An/Ap/Ev)</b>	<b>10</b>	<b>CO1, CO2,CO3, CO4,CO5</b>

<b>BIC12035</b>	<b>DSE I Molecular Basis of Noninfectious Human Disease Lab (Practical)</b>	L	T	P	C
<b>Version 1.0</b>	<b>Contact Hours: 45</b>	0	0	3	2
<b>Pre-requisites/Exposure</b>	Knowledge of basic Biochemistry and microbiology				
<b>Co-requisites</b>	-				

### Course Objectives

1. To understand the strategy of different type of nutritional disorders
2. To provide knowledge of different types of metabolic disorders.
3. To understand the strategy of the different types of cancer and multifactorial disorders.
4. To provide the concept of different types of protein misfolding disorders.
5. To summarize the strategy of different types of monogenic diseases.

### Course Outcomes

On completion of this course, the students will be able to

**CO1 Identify** sample of Kwashiorkor, Marasmus and Obesity.

**CO2 Determine** Homocystiene.

**CO3 Determine** glycosylated hemoglobin.

**CO4 Assess** of CVS using case studies.

**CO5 Assess** of Diabetes mellitus using case studies.

### Catalog Description

A noninfectious disease is a disease that is not transmissible directly from one person to another. This category include Parkinson's disease, autoimmune diseases, strokes, most heart diseases, most cancers, diabetes, chronic kidney disease, osteoarthritis, osteoporosis, Alzheimer's disease, cataracts, and others. NCDs may be chronic or acute. Most are non-infectious, although there are some non-communicable infectious diseases, such as parasitic diseases in which the parasite's life cycle does not include direct host-to-host transmission.

Noninfectious diseases are the leading cause of death globally. In 2012, they caused 68% of all deaths (38 million) up from 60% in 2000. About half were under age 70 and half were women. Risk factors such as a person's background, lifestyle and environment increase the likelihood of certain NCDs. Every year, at least 5 million people die because of tobacco use and about 2.8 million die from being overweight. High cholesterol accounts for roughly 2.6 million deaths and million die because of high blood pressure. In this paper the molecular mechanism behind these kinds of disease will be analyzed.

## Course Content

1. Anthropometric measurements for normal and high risk individuals and identifications for Kwashiorkor, Marasmus and Obesity [10 Lecture Hours]
2. Estimation of homocysteine levels in serum [10 Lecture Hours]
3. Estimation of glycosylated hemoglobin [10 Lecture Hours]
4. Permanent slides for different types of cancer [10 Lecture Hours]
5. Diagnostic profile for assessment of CVS and Diabetes mellitus using case studies.[10 Lecture Hours]
6. Bone densitometry test demonstration (visit to a nearby clinic) [10 Lecture Hours]

### Reference Books

1. Textbook of Biochemistry with Clinical Correlations (2011) Devlin, T.M. John Wiley & Sons, Inc. (New York), ISBN:978-0-4710-28173-4.
2. Introduction to Human Physiology (2013) 8th edition; Lauralee Sherwood. Brooks/Cole, Cengage Learning. The World of the cell, 7th edition(2009)
3. Genetics (2012) Snustad and Simmons,4.
- Cooper, G.M. and Hausman, R.E. 2009
5. The Cell: A Molecular Approach. 5th edition. ASM Press & Sunderland, Washington, D.C.; Sinauer Associates,MA.

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

### Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and Pos		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	Identify sample of Kwashiorkor, Marasmus and Obesity.	PO1, PO2,PO3, PO4 PO5, PO6, PO7,PO8,PO9, PO11
CO2	Determine Homocystiene.	PO1, PO2,PO3, PO4 PO5, PO6, PO7,PO8,PO9, PO11
CO3	Determine glycosylated hemoglobin.	PO1, PO2,PO3, PO4 PO5, PO6, PO7,PO8,PO9, PO11

<b>CO4</b>	<b>Assess</b> of CVS using case studies.	<b>PO1, PO2,PO3, PO4 PO5, PO6, PO7,PO8,PO9, PO11</b>
<b>CO5</b>	<b>Assess</b> of Diabetes mellitus using case studies.	<b>PO1, PO2,PO3, PO4 PO5, PO6, PO7,PO8,PO9, PO11</b>

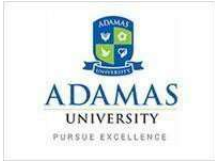
		Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking	Problem Identification Ability	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to Environment	Development to Society.	Development to Humanity
Course Code	Course Title	PO 1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
<b>BIC12035</b>	DSE I Molecular Basis of Non infectious HumanDisease	3	3	3	3	3	3	3	3	3	-	3	-

1=weakly mapped

2= moderately mapped

3=strongly mapped

### Model Question Paper

<b>Name:</b>  <b>Enrolment No:</b>			
<p style="text-align: center;"><b>Course: BIC 12035–DSEI Molecular Basis of Noninfectious Human Disease Lab</b></p> <p><b>Program: B.Sc Biochemistry</b> <span style="float: right;"><b>Time: 03Hrs.</b></span></p> <p><b>Semester: Odd 2020-21</b> <span style="float: right;"><b>Max. Marks: 40</b></span></p>			
<p><b>Instructions:</b>          Attempt any two questions from <b>Section A</b> (each carrying 10 marks); any <b>Two Questions</b> from <b>Section B</b> (each carrying 10 marks) <b>is Compulsory.</b></p>			
<b>Section A ( Attempt any Three)</b>			
1.	<b>Identify</b> the tissue sections of the two given slides (Ap) , write down their specific characteristics and their relation with noninfectious disease.(An)	<b>5</b> <b>5</b>	<b>CO1,CO2, CO3,CO4,CO5</b>
2.	<b>Estimate</b> the homocysteine content from the supplied serum sample (Ap),write down the principle, procedure, observation and inference.(An)	<b>10</b>	<b>CO2</b>
3.	Determine the concentration of glycosylated hemoglobin from the supplied blood sample(Ap),write down the principle, procedure, observation and inference.(An)	<b>10</b>	<b>CO3</b>
<b>SECTION B (Attempt any Two Questions)</b>			
5.	<b>Viva-Voce</b> (U/An/Ap/R)	<b>10</b>	<b>CO1, CO2,CO3, CO4,CO5</b>
6.	<b>Laboratory Note Book</b> (U/An/Ap/Ev)	<b>10</b>	<b>CO1, CO2,CO3, CO4,CO5</b>

<b>BIC11036</b>	<b>DSE-II Molecular Basis of Infectious Human Disease(Theory)</b>	L	T	P	C
<b>Version 1.0</b>	<b>Contact Hours: 60</b>	3	1	0	4
<b>Pre-requisites/Exposure</b>	Knowledge of Biology at 10+2 Level				
<b>Co-requisites</b>	-				

### Course Objectives

1. To gain knowledge about different type of pathogens.
2. To gain the knowledge structures of different types of bacterial diseases.
3. To acquire the knowledge about the different types of viral diseases.
4. To gain the knowledge about different types of protozoan diseases.
5. To gain the knowledge about different types of fungal diseases.

### Course Outcomes

On completion of this course, the students will be able to

- CO1 Describe** the role of different pathogens in causing diseases.  
**CO2 Understand** the molecular basis of bacterial diseases.  
**CO3 Illustrate** the molecular basis of viral diseases.  
**CO4 Understand** the molecular basis of protozoan diseases.  
**CO5 Illustrate** the molecular basis of fungal diseases.

### Catalog Description

Infectious disease is a typical category of disease, which occurs in human and other mammals due to invading of bacteria, virus, fungus, protozoa and results in different pattern of pathogenesis. This course consists of the detailed studies on molecular basis of these kinds of pathogenesis, that will inspire the students to investigate and design the drug against the different molecular targets of respective pathogens.



## Course Content

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### **Unit I Classification of infectious agent [12 lecture hours]**

Bacteria, Viruses, protozoa and fungi. Past and present emerging and re-emerging infectious diseases and pathogens. Source, reservoir and transmission of pathogens, Antigenic shift and antigenic drift. Host parasite relationship, types of infections associated with parasitic organisms. Overview of viral and bacterial pathogenesis. Infection and evasion.

### **Unit II Overview of diseases caused by bacteria [12 lecture hours]**

Detailed study of tuberculosis: History, causative agent, molecular basis of host specificity, infection and pathogenicity, Diagnostics, Therapeutics, inhibitors and vaccines. Drug resistance and implications on public health. Other bacterial diseases including Typhoid, Diphtheria, Pertussis, Tetanus, Typhoid and Pneumonia.

### **Unit III Overview of diseases caused by Viruses [12 lecture hours]**

Detailed study of AIDS, history, causative agent, pathogenesis, Diagnostics, Drugs and inhibitors. Other viral diseases including hepatitis, influenza, rabies, chikungunya and polio.

### **Unit IV Overview of diseases caused by Parasites [12 lecture hours]**

Detailed study of Malaria, history, causative agents, Vectors, life cycle, Host parasite interactions, Diagnostics, Drugs and Inhibitors, Resistance, Vaccine development. Other diseases including leishmaniasis, amoebiasis.

### **Unit V Overview of diseases caused by other organisms [12 lecture hours]**

Fungal diseases, General characteristics. Medical importance of major groups, pathogenesis, treatment.

## Reference Books

1. Prescott, Harley, Klein's Microbiology (2008) 7th Ed., Willey, J.M., Sherwood, L.M., Woolverton, C.J. McGraw Hill International Edition (New York) ISBN: 978-007-126727.
2. Mandell, Douglas and Bennett. S, Principles and practices of Infectious diseases, 7th edition, Volume, 2. Churchill Livingstone Elsevier
3. Sherris Medical Microbiology: An Introduction to Infectious Diseases by Kenneth J. Ryan, C. George Ray, Publisher: McGraw-Hill
4. Medical Microbiology by Patrick R. Murray, Ken S. Rosenthal, Michael A. Pfaller, Elsevier Health Sciences.
5. Brooks GF, Carroll KC, Butel JS and Morse SA. (2007). Jawetz, Melnick and Adelberg's Medical Microbiology. 24th edition. McGraw Hill Publication
6. Goering R, Dockrell H, Zuckerman M and Wakelin D. (2007). Mims' Medical Microbiology. 4th edition. Elsevier.

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination  
Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

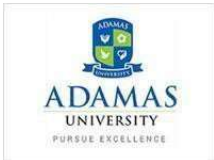
**Relationship between the Course Outcomes (COs) and Program Outcomes (POs)**

Mapping between COs and Pos		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	<b>Describe</b> the role of different pathogens in causing diseases.	PO1, PO2, PO3, PO5, PO6, PO7, PO8, PO9
CO2	<b>Understand</b> the molecular basis of bacterial diseases.	PO1, PO2, PO3, PO5, PO6, PO7, PO8, PO9
CO3	<b>Illustrate</b> the molecular basis of viral diseases.	PO1, PO2, PO3, PO5, PO6, PO7, PO8, PO9
CO4	<b>Understand</b> the molecular basis of protozoan diseases.	PO1, PO2, PO3, PO5, PO6, PO7, PO8, PO9
CO5	<b>Illustrate</b> the molecular basis of fungal diseases.	PO1, PO2, PO3, PO5, PO6, PO7, PO8, PO9

Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
<b>BIC11036</b>	DSE II Molecular Basis of Infectious Human Disease	Fundamental Knowledge											
		Critical Thinking											
		Skill											
		Technical Knowledge											
		Logical Thinking											
		Problem Identification Ability											
		Analytical Knowledge											
		Career Goals											
		Team Work											
		Sustainable Development to Environment											
		Development to Society.											
		Development to Humanity											

1=weakly mapped  
2= moderately mapped  
3=strongly mapped

## Model Question Paper

<b>Name:</b>  <b>Enrolment No:</b>			
<b>Course: BIC11036 –DSE II Molecular Basis of Infectious HumanDisease</b> <b>Program: B.Sc Biochemistry</b> <span style="float: right;"><b>Time: 03 Hrs.</b></span> <b>Semester: Odd2020-21</b> <span style="float: right;"><b>Max. Marks: 40</b></span>			
<b>Instructions:</b> Attempt any three questions from <b>Section A</b> (each carrying 4 marks); any <b>Two Questions</b> from <b>Section B</b> (each carrying 10 marks). <b>Section C</b> is Compulsory (carrying 8 marks).			
<b>Section A ( Attemptany Three)</b>			
1.	<b>How</b> the process of translation in host cell is affected during the infection by <i>Corynebacterium diphtheriae</i> ? <b>Relate</b> the phenomenon of vasodilation upon endotoxin administration. (Ap)	<b>2</b> <b>2</b>	<b>CO2</b>
2.	‘ELISA is the most preferable way to detect the infection of Dengue virus’ - <b>Justify</b> and also mention the limitations of this method in this context. (Ev)	<b>3</b> <b>1</b>	<b>CO3</b>
3.	‘Loss of Balance of T <sub>h</sub> 1 and T <sub>h</sub> 2 cell proliferation make the host system more prone to leishmaniasis’— <b>explain</b> . (U)	<b>4</b>	<b>CO4</b>
4.	<b>Why</b> azidothymidine (AZT) designed against HIV became obsolete for its treatment? Let, you have obtained the serum of a patient with influenza, you have screened it with anti H1N1 antibody and obtained negative result, but still viral particle is present in his lung tissue- how can you <b>interpret</b> this situation? (An)	<b>1</b> <b>3</b>	<b>CO3</b>
<b>SECTION B (Attempt any Two Questions)</b>			
5.	‘Drug designing against bacteria is much more easier than drug designing against protozoa and fungus’ - explain. Name different types of bacterial enzymes and their role in arising of symptoms of diseases. Define iatrogenic infection and LD <sub>50</sub> . (R)	<b>3</b> <b>5</b> <b>2</b>	<b>CO1</b> <b>CO2</b> <b>CO4</b> <b>CO5</b>
6.	Relate the ‘avidity hypothesis’ (U)	<b>3</b>	<b>CO1</b>

	with the signalling mediated by superantigen? What do you mean by 'shut down phase' during initiation of inflammation? Write down the role of leukotrienes (LTB <sub>4</sub> and LTE) and thromboxane A <sub>2</sub> during inflammation. Name one steroidal and one nonsteroidal anti-inflammatory drug.	2 3 2	
7.	Aspergillosis in lung may be life threatening ----- <b>discuss</b> at molecular level. What kind of drug will you use to abolish the physiological symptoms and why? What do you mean by A/Puerto Rico/8/34 influenza virus? Influenza virus is the RNA virus, but not retrovirus- why?	4 3 1 2	<b>CO3</b> <b>CO5</b>
	<b>SECTION C is Compulsory</b>		
8.	How the protozoa causing leishmaniasis protects itself from altered pH in host environment? What is the novelty of miltefosine over amphotericin B and sodium stibogluconate for the 95% success rate of curing of leishmaniasis. Mention the course and route of administration of miltefosine.	2 4 2	<b>CO4</b>

<b>BIC11038</b>	<b>DSE II Medical Microbiology (Theory)</b>	L	T	P	C
<b>Version 1.0</b>	<b>Contact Hours: 60</b>	3	1	0	4
<b>Pre-requisites/Exposure</b>	Knowledge of Biochemistry				
<b>Co-requisites</b>	-				

### Course Objectives

1. To gain the knowledge about different type of pathogens.
2. To gain the knowledge structures of different types of bacterial diseases.
3. To acquire the knowledge about the different types of viral diseases.
4. To gain the knowledge about different types of protozoan diseases.
5. To gain the knowledge about different types of fungal diseases.

### Course Outcomes

On completion of this course, the students will be able to

**CO1 Illustrate** and discuss Morphology, pathogenesis, symptoms, laboratory diagnosis, preventive measures and chemotherapy of gram positive bacteria.

**CO2 Explain** and discuss Morphology, pathogenesis, symptoms, laboratory diagnosis, preventive measures and chemotherapy of gram negative bacteria.

**CO3 Elaborate** and discuss disease caused by viruses.

**CO4 Explain** and discuss disease caused by Fungal infections.

**CO5 Illustrate** and discuss disease caused by Protozoal infections.

### Catalog Description

Infectious disease is a typical category of disease, which occurs in human and other mammals due to invading of bacteria, virus, fungus, protozoa and results in different pattern of pathogenesis. This course consists of the detailed studies on molecular basis of these kinds of pathogenesis, that will inspire the students to investigate and design the drug against the different molecular targets of respective pathogens.

## Course Content

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1. Normal microflora of human body, nosocomial infections, carriers, septic shock, septicemia, pathogenicity, virulence factors, toxins, biosafety levels. Morphology, pathogenesis, symptoms, laboratory diagnosis, preventive measures and chemotherapy of gram positive bacteria: *S.aureus*, *S.pyogenes*, *B.anthraxis*, *C.perferinges*, *C.tetani*, *C.botulinum*, *C.diphtheriae*, *M.tuberculosis*, *M.leprae*. [15 LectureHours]
2. Morphology, pathogenesis, symptoms, laboratory diagnosis, preventive measures and chemotherapy caused by gram negative bacteria: *E.coli*, *N. gonorrhoea*, *N. meningitidis*, *P. aeruginosa*, *S. typhi*, *S. dysenteriae*, *Y. pestis*, *B. abortus*, *H. influenzae*, *V. cholerae*, *M. pneumoniae*, *T. pallidum*, *M. pneumoniae*, *Rickettsiaceae*, *Chlamydiae*. [15 LectureHours]
3. Diseases caused by viruses- Picornavirus, Orthomyxoviruses, Paramyxoviruses, Rhabdoviruses, Reoviruses, Pox virus, Herpes virus, Papova virus, Retro viruses (including HIV/AIDS) and Hepatitis viruses. [15 LectureHours]
4. Fungal and Protozoan infections. Dermatophytes (*Trichophyton*, *Microsporun* and *Epidermophyton*) Subcutaneous infection (*Sporothrix*, *Cryptococcus*), systemic infection (*Histoplasma*, *Coccidioides*) and opportunistic fungal infections (*Candidiasis*, *Aspergillosis*), Gastrointestinal infections (Amoebiasis, Giardiasis), Blood-borne infections (Leishmaniasis, Malaria)[15 LectureHours]

### Reference Books

1. Prescott, Harley, Klein's Microbiology (2008) 7th Ed., Willey, J.M., Sherwood, L.M., Woolverton, C.J. McGraw Hill International Edition (New York) ISBN: 978-007-126727.
2. Mandell, Douglas and Bennett.S, Principles and practices of Infectious diseases, 7th edition, Volume, 2. Churchill LivingstoneElsevier
3. Sherris Medical Microbiology: An Introduction to Infectious Diseases by Kenneth J.Ryan, C. George Ray, Publisher:McGraw-Hill
4. Medical Microbiology by Patrick R. Murray, Ken S. Rosenthal, Michael A. Pfaller, Elsevier HealthSciences.
5. Brooks GF, Carroll KC, Butel JS and Morse SA. (2007). Jawetz, Melnick and Adelberg's Medical Microbiology. 24th edition. McGraw HillPublication
6. Goering R, Dockrell H, Zuckerman M and Wakelin D. (2007). Mims' Medical Microbiology. 4th edition.Elsevier.

### Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

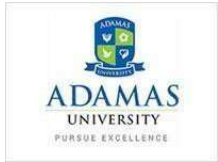
**Relationship between the Course Outcomes (COs) and Program Outcomes (POs)**

<b>Mapping between COs and Pos</b>		
	<b>Course Outcomes (COs)</b>	<b>Mapped Program Outcomes</b>
<b>CO1</b>	<b>Illustrate</b> and discuss Morphology, pathogenesis, symptoms, laboratory diagnosis, preventive measures and chemotherapy of gram positive bacteria.	<b>PO1, PO2, PO3, PO5, PO6, PO7, PO8, PO9</b>
<b>CO2</b>	<b>Explain</b> and discuss Morphology, pathogenesis, symptoms, laboratory diagnosis, preventive measures and chemotherapy of gram negative bacteria.	<b>PO1, PO2, PO3, PO5, PO6, PO7, PO8, PO9</b>
<b>CO3</b>	<b>Elaborate</b> and discuss disease caused by viruses.	<b>PO1, PO2, PO3, PO5, PO6, PO7, PO8, PO9</b>
<b>CO4</b>	<b>Explain</b> and discuss disease caused by Fungal infections.	<b>PO1, PO2, PO3, PO5, PO6, PO7, PO8, PO9</b>
<b>CO5</b>	<b>Illustrate</b> and discuss disease caused by Protozoal infections	<b>PO1, PO2, PO3, PO5, PO6, PO7, PO8, PO9</b>





## Model Question Paper

<b>Name:</b>  <b>Enrolment No:</b>			
<b>Course: BIC11038 – DSE II Medical Microbiology</b> <b>Program: B.Sc Biochemistry</b> <span style="float: right;"><b>Time: 03Hrs.</b></span> <b>Semester: Odd2020-21</b> <span style="float: right;"><b>Max. Marks:40</b></span>			
<b>Instructions:</b> Attempt any three questions from <b>Section A</b> (each carrying 4 marks); any <b>Two Questions</b> from <b>Section B</b> (each carrying 10 marks). <b>Section C</b> is Compulsory (carrying 8 marks).			
<b>Section A ( Attemptany Three)</b>			
1.	<b>How</b> the process of translation in host cell is affected during the infection by <i>Corynebacterium diphtheriae</i> ? <b>Relate</b> the phenomenon of vasodilation upon endotoxin administration. (R)	2 2	CO2
2.	‘ELISA is the most preferable way to detect the infection of Dengue virus’- <b>Justify</b> and also mention the limitations of this method in this context. (Ap)	3 1	CO3
3.	‘Loss of Balance of T <sub>h</sub> 1 and T <sub>h</sub> 2 cell proliferation make the host system more prone to leishmaniasis’— <b>explain</b> . (Ap)	4	CO4
4.	<b>Why</b> azidothymidine (AZT) designed against HIV became obsolete for its treatment? Let, you have obtained the serum of a patient with influenza, you have screened it with anti H1N1 antibody and obtained negative result, but still viral particle is present in his lung tissue-how can you <b>interpret</b> this situation? (U, Ap)	1 3	CO3

<b>SECTION B (Attempt any Two Questions)</b>			
5.	‘Drug designing against bacteria is much more easier than drug designing against protozoa and fungus’-explain.Name different types of bacterial enzymes and their role in arising of symptoms of diseases.Define iatrogenic infection and LD <sub>50</sub> .(R)	3 5 2	<b>CO1 CO2 CO4 CO5</b>
6.	Relate the ‘avidity hypothesis’ with the signalling mediated by superantigen?What do you mean by ‘shut down phase’during initiation of inflammation? Write down the role of leukotrienes(LT B <sub>4</sub> and LTE) and thromboxane A <sub>2</sub> during inflammation.Name one steroidal and one nonsteroidal anti-inflammatory drug.(R)	3 2 3 2	<b>CO1</b>
7.	Aspergillosis in lung may be life threatening ----- <b>discuss</b> at molecular level. What kind of drug will you use to abolish the physiological symptoms and why? What do you mean by A/Puerto Rico/8/34 influenza virus? Influenza virus is the RNA virus, but not retrovirus- why?(An)	4 3 1 2	<b>CO3 CO5</b>
<b>SECTION C is Compulsory</b>			
8.	How the protozoa causing leishmaniasis protects itself from altered pH in host environment? What is the novelty of miltefosine over amphotericin B and sodium stibogluconate for the 95% success rate of curing of leishmaniasis. Mention the course and route of administration of miltefosine.(R)	2 4 2	<b>CO4</b>

<b>BIC 11040</b>	<b>DSE II Nutritional Biochemistry (Theory)</b>	L	T	P	C
<b>Version 1.0</b>	<b>Contact Hours: 60</b>	3	1	0	4
<b>Pre-requisites/Exposure</b>	Knowledge of Biomolecules				
<b>Co-requisites</b>	-				

### Course Objectives

1. To gain a deeper understanding of nutrition and energy metabolism.
2. To acquire the knowledge about the significance of dietary carbohydrates in health.
3. To acquire the knowledge about the significance of dietary proteins in health.
4. To gain the knowledge about significance of dietary fats in health.
5. To understand the significance about dietary vitamins and minerals in health..

### Course Outcomes

On completion of this course, the students will be able to

**CO1 Explain** and categorize basics of Nutrition and Energy Metabolism.

**CO2 Discuss** importance of dietary carbohydrates in health.

**CO3 Illustrate** importance of dietary proteins in health.

**CO4 Discuss** importance of dietary fats in health.

**CO5 Explain the** importance of dietary vitamin and minerals in health.

### Catalog Description

Nutritional biochemistry is one of the academic foundations that make up nutritional sciences, a discipline that encompasses the knowledge of nutrients and other food components with emphasis on their range of function and influence on mammalian physiology, health, and behavior. Nutritional biochemistry is a subdiscipline that is made up of the core knowledge, concepts, and methodology related to the chemical properties of nutrients and other dietary constituents and to their biochemical, metabolic, physiological, and epigenetic functions. A primary focus of this paper is the scientific establishment of optimal dietary intakes for every nutrient and food component throughout the lifecycle.

## Course Content

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- 1. Introduction to Nutrition and Energy Metabolism:** Defining Nutrition, role of nutrients. Unit of energy, Biological oxidation of foodstuff. measurement of energy content of food, Physiological energy value of foods, SDA. Measurement of energy expenditure. Direct and Indirect Calorimetry, factors affecting thermogenesis, energy utilization by cells, energy output – Basal and Resting metabolism, physical activity, factors affecting energy input - hunger, appetite, energy balance Energy expenditure in man. Estimating energy requirements, BMR factors Recommended Nutrient Intakes (RNI) and Recommended Dietary Allowances for different age groups. **[10 Lecture Hours]**
- 2. Dietary carbohydrates and health:** Review functions of carbohydrates. Digestion, absorption, utilization and storage, hormonal regulation of blood glucose. Dietary requirements and source of carbohydrates, Dietary fiber, role of fibre in lipid metabolism, colon function, blood glucose level and GI tract functions. **[10 Lecture Hours]**
- 3. Dietary lipid and health:** Review of classification, sources, functions, digestion, absorption, utilization and storage. Essential Fatty Acids; Functions of EFA, RDA, –excess and deficiency of EFA. Lipotropic factors, role of saturated fat, cholesterol, lipoprotein and triglycerides. Importance of the following: a) Omega – fatty acids. Omega 3/ omega 6 ratio b) Phospholipids c) Cholesterol in the body d) Mono, Polyunsaturated and Saturated Fatty Acids. Dietary implications of fats and oils, Combination ratios of n6 and n3, MUFA, PUFA and SFA. **[10 Lecture Hours]**
- 4. Dietary Proteins and health:** Review of functions of proteins in the body, Digestion and absorption. Essential and Non-essential amino acids. Amino Acid Availability Antagonism, Toxicity and Imbalance, Amino acid Supplementation. Effects of deficiency. Food source and Recommended Dietary Allowances for different age group. Amino acid pool. NPU, Biological Value, Nitrogen balance. PEM and Kwashiorkor. **[10 Lecture Hours]**
- 5. Fat and water soluble Vitamins:** Vitamin A, C, E, K and D Dietary sources, RDA, Adsorption, Distribution, Metabolism and excretion (ADME), Deficiency. Role of Vitamin A as an antioxidant, in Visual cycle, dermatology and immunity. Role of Vitamin K in Gamma carboxylation. Role of Vitamin E as an antioxidant. Extra-skeletal role of Vitamin D and its effect on bone physiology. Hypervitaminosis. Vitamin C role as cofactor in amino acid modifications. Niacin- Metabolic interrelation between tryptophan, Niacin and NAD/ NADP. Vitamin B6- Dietary source, RDA, conversion to Pyridoxal Phosphate. Role in metabolism, Biochemical basis for deficiency symptoms. Vitamin B12 and folate; Dietary source, RDA, absorption, metabolic role Biochemical basis for deficiency symptoms. **[10 Lecture Hours]**
- 6. Minerals:** Calcium, Phosphorus and Iron- Distribution in the body digestion, Absorption, Utilization, Transport, Excretion, Balance, Deficiency, Toxicity, Sources, RDA. Calcium: Phosphorus ratio, Role of iron in prevention of anemia. Iodine and iodine cycle. Iodine, Fluoride, Mg, Cu, Zn, Se, Manganese, Chromium, Molybdenum Distribution in the human body, Physiology, Function, deficiency, Toxicity and Sources **[4 Lecture Hours]**
- 7. Assessment of Nutritional status:** Anthropometric measurements; Z scores, BMI, skinfold, circumference ratios. Biochemical assessment; Basal metabolic panel, Comprehensive metabolic panel, CBC, Urine Analysis, Assessment of Anemia, ROS assessment, GTT and glycosylated Hb, Differential diagnosis of B12 and folate. **[3 Lecture Hours]**

- 8. Food and drug interactions and Nutraceuticals:** Nutrient interactions affecting ADME of drugs, Alcohol and nutrient deficiency, Anti-depressants, psychoactive drugs and nutrient interactions, Appetite changes with drug intakes and malnutrition. Food as medicine. [3 LectureHours]

**Reference Books**

1. Textbook of Biochemistry with Clinical Correlations (2011) Devlin, T.M. Jhn Wiley & Sons, Inc. (New York), ISBN:978-0-4710-28173-4.
2. Nutrition for health, fitness and sport (2013) ;Williams.M.H,Anderson,D.E, Rawson,E.S.McGrawHillinternationaledition.ISBN-978-0-07-131816-7.
3. Krause's Food and Nutrition Care process.(2012); Mahan, L.K Strings,S.E, Raymond,J. Elsevier's Publications. ISBN-978-1-4377-2233-8

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

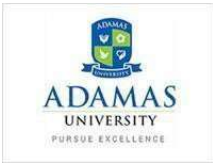
**Relationship between the Course Outcomes (COs) and Program Outcomes (POs)**

Mapping between COs and Pos		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	<b>Explain</b> and categorize basics of Nutrition and Energy Metabolism.	PO1, PO2, PO3, PO6, PO7, PO8, PO10, PO11
CO2	<b>Discuss</b> importance of dietary carbohydrates in health.	PO1, PO2, PO3, PO6, PO7, PO8, PO10, PO11
CO3	<b>Illustrate</b> importance of dietary proteins in health.	PO1, PO2, PO3, PO6, PO7, PO8, PO10, PO11
CO4	<b>Discuss</b> importance of dietary fats in health.	PO1, PO2, PO3, PO6, PO7, PO8, PO10, PO11
CO5	<b>Explain the</b> importance of dietary vitamin and minerals in health.	PO1, PO2, PO3, PO6, PO7, PO8, PO10, PO11

Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
<b>BIC 11040</b>	DSE II Nutritional Biochemistry	3	3	3	-	-	3	3	3	-	3	3	-
		Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking	Problem Identification Ability	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to Environment	Development to Society.	Development to Humanity

1=weakly mapped  
2= moderately mapped  
3=strongly mapped

## Model Question Paper

<b>Name:</b>  <b>Enrolment No:</b>			
<b>Course: BIC 11040– DSE II Nutritional Biochemistry</b> <b>Program: B.Sc Biochemistry</b> <span style="float: right;"><b>Time: 03Hrs.</b></span> <b>Semester: Odd2020-21</b> <span style="float: right;"><b>Max. Marks:40</b></span>			
<b>Instructions:</b> Attempt any three questions from <b>Section A</b> (each carrying 4 marks); any <b>Two Questions</b> from <b>Section B</b> (each carrying 10 marks). <b>Section C</b> is Compulsory (carrying 8 marks).			
<b>Section A ( Attemptany Three)</b>			
1.	<b>Discuss</b> role of Vitamin B12 in the maintenance of healthy erythrocytes.(R)	4	CO5
2.	<b>Define</b> biological value of a protein. How it is evaluated?(U)	2 2	CO1 CO3
3.	<b>What</b> are the importance raw vegetables during daily meal?(U)	4	CO2
4.	<b>Explain</b> the different strategy of N balance in human body.(Ap)	4	CO1 CO3
<b>SECTION B (Attempt any Two Questions)</b>			
5.	<b>How</b> the Vitamin E scavengesthe reactive oxygen species with the help of vitamin C? How vitamin C is involved in collagen biosynthesis?Write down the daily requirement of both of the vitamins.(U)	5 3 2	CO5
6.	What do you mean by Basal Metabolic Rate(BMR)?Writedown itsnormalvalueinhumanmaleand female. <b>Explain</b> thephysiological factors for controlling of BMR.(Ap)	2 2 6	CO1



7.	<b>What</b> do you mean by niacin index? <b>Explain</b> down the role of pyridoxal phosphate in transaminase reaction. <b>Why</b> the value of R.Q. of lipid is less than dietary carbohydrate?(U)	2 5 3	CO4 CO5
	<b>SECTION C is Compulsory</b>		
8.	How will you <b>determine</b> the protein score of a fish protein? <b>Explain-</b> “The Specific dynamic action(SDA) value of protein is highest.”(Ap)		CO3

<b>BIC 12037</b>	<b>DSE II Molecular Basis of Infectious Human Disease Lab (Practical)</b>	L	T	P	C
<b>Version 1.0</b>	<b>Contact Hours: 45</b>	0	0	3	2
<b>Pre-requisites/Exposure</b>	Basic knowledge of Biochemistry and microbiology				
<b>Co-requisites</b>	-				

### Course Objectives

1. To provide the knowledge about different type of microorganisms
2. To distinguish between gram positive and gram negative bacteria
3. To characterize the acid fast bacteria
4. To characterize bacterial spores
5. To provide the knowledge about different types diagnostic approach for several diseases

### Course Outcomes

On completion of this course, the students will be able to

**CO1 Distinguish** between different pathogens in causing diseases.

**CO2 Distinguish** between gram positive and gram negative bacteria.

**CO3 Characterize** the acid fast bacteria and bacterial endospore.

**CO4 Characterize** the fungus.

**CO5 Understand** about detection and diagnosis of infectious disease.

### Catalog Description

Infectious disease is a typical category of disease, which occurs in human and other mammals due to invading of bacteria, virus, fungus, protozoa and results in different pattern of pathogenesis. This course consists of the detailed microscopic studies of those disease causing organisms, staining of the microorganism and identification of microbe infected human tissue.

### Course Content

1. Identification of gram positive and gram negative bacteria by gram staining procedure. (5 Hours)
2. Antibiotic susceptibility test by cup plate and paper disc method. (5 Hours)
3. Simple staining/ Negative staining. (5 Hours)
4. Acid fast staining process for identification of Mycobacterium sp. (5 Hours)
5. Endospore & Capsule staining. (10 Hours)
6. Isolation and characterization of coliform bacteria from river in selective media (water potability test). (10 Hours)
7. Widal test. (10 Hours)
8. Permanent Slides of Pathogens. (10 Hours)



## Reference Books

1. Prescott, Harley, Klein's Microbiology (2008) 7th Ed., Willey, J.M., Sherwood, L.M., Woolverton, C.J. McGraw Hill International Edition (New York) ISBN: 978-007-126727.
2. Mandell, Douglas and Bennett.S, Principles and practices of Infectious diseases, 7th edition, Volume, 2. Churchill LivingstoneElsevier
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5. Brooks GF, Carroll KC, Butel JS and Morse SA. (2007).Jawetz,
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**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

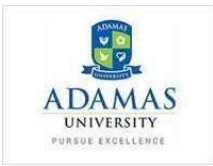
### Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and Pos		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	<b>distinguish</b> between different pathogens in causing diseases.	PO1, PO2, PO5, PO6, PO7,PO8
CO2	<b>distinguish</b> between gram positive and gram negative bacteria.	PO1, PO2, PO5, PO6, PO7,PO8
CO3	<b>characterize</b> the acid fast bacteria and bacterial endospore	PO1, PO2, PO5, PO6, PO7,PO8
CO4	<b>characterize</b> the fungus	PO1, PO2, PO5,PO6, PO7,PO8
CO5	<b>summarize</b> the diagnosis of different infectious disease	PO1, PO2, PO5, PO6,

Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
<b>BIC 12037</b>	<b>DSE II</b> Molecular Basis of Infectious Human Disease	3	3	2	3	3	3	3	3	2	1	1	-
		Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking	Problem Identification Ability	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to Environment	Development to Society.	Development to Humanity

1=weakly mapped  
2= moderately mapped  
3=strongly mapped

## Model Question Paper

<b>Name:</b>  <b>Enrolment No:</b>			
<p style="text-align: center;"><b>Course: BIC 12037–DSEIIMolecularBasisofInfectiousHumanDiseaseLab</b></p> <p><b>Program: B.Sc Biochemistry</b> <span style="float: right;"><b>Time: 03Hrs.</b></span></p> <p><b>Semester: Odd2020-21</b> <span style="float: right;"><b>Max. Marks: 40</b></span></p>			
<p><b>Instructions:</b>          Attempt any two questions from <b>Section A</b> (each carrying 10 marks); any <b>Two Questions</b> from <b>Section B</b> (each carrying 10 marks) <b>is Compulsory.</b></p>			
<b>Section A ( Attemptany Three)</b>			
1.	<b>Identify</b> the specimen of the two given slides (Ap) , write down their specific characteristics and their relation with infectious disease.(An)	5 5	<b>CO1,CO2, CO3,CO4,CO5</b>
2.	Perform the Gram staining procedure with the supplied bacterial sample(Ap),write down the principle, procedure, observation and inference.(An)	10	<b>CO2</b>
3.	Perform the Cotton blue staining procedure with the supplied fungus sample(Ap),write down the principle, procedure, observation and inference.(An)	10	<b>CO4</b>
<b>SECTION B (Attempt any Two Questions)</b>			
5.	<b>Viva-Voce</b> (U/An/Ap/R)	10	<b>CO1, CO2,CO3, CO4,CO5</b>
6.	<b>Laboratory Note Book</b> (U/An/Ap/Ev)	10	<b>CO1, CO2,CO3, CO4,CO5</b>

<b>BIC 12039</b>	<b>DSE II Medical Microbiology Lab (Pactical)</b>	L	T	P	C
<b>Version 1.0</b>	<b>Contact Hours: 45</b>	0	0	3	2
<b>Pre-requisites/Exposure</b>	Knowledge of Biochemistry				
<b>Co-requisites</b>	-				

### Course Objectives

1. To provide the knowledge about different type of microorganisms
2. To distinguish between gram positive and gram negative bacteria
3. To characterize the acid fast bacteria
4. To characterize bacterial spores
5. To provide the knowledge about different types diagnostic approach for several diseases

### Course Outcomes

On completion of this course, the students will be able to

**CO1 Identify** pathogenic bacteria based on cultural, morphological characteristics

**CO2 Illustrate** growth curve of a bacterium.

**CO3 Examine** temporary mounts of *Aspergillus* by appropriate staining.

**CO4 Examine** the bacteria by Gram staining

**CO5 Identify** the bacteria by acid fast staining

### Catalog Description

Infectious disease is a typical category of disease, which occurs in human and other mammals due to invading of bacteria, virus, fungus, protozoa and results in different pattern of pathogenesis. This course consists of the detailed microscopic studies of those disease causing organisms, staining of the microorganism and identification of microbe infected human tissue.

### Course Content

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1. Identification of pathogenic bacteria (any two) based on cultural, morphological and biochemical characteristics. [12 Lecture Hours]
2. Growth curve of a bacterium. [12 Lecture Hours]
3. To perform antibacterial testing by Kirby-Bauer method. [12 Lecture Hours]
4. To prepare temporary mounts of *Aspergillus* and *Candida* by appropriate staining. [12 Lecture Hours]
5. Staining methods: Gram's staining permanent slides showing Acid fast staining, Capsule staining and spore staining. [12 Lecture Hours]

## Reference Books

1. Prescott, Harley, Klein's Microbiology (2008) 7th Ed., Willey, J.M., Sherwood, L.M., Woolverton, C.J. McGraw Hill International Edition (New York) ISBN: 978-007-126727.
2. Mandell, Douglas and Bennett.S, Principles and practices of Infectious diseases, 7th edition, Volume, 2. Churchill LivingstoneElsevier
3. Sherris Medical Microbiology: An Introduction to Infectious Diseases by Kenneth J.Ryan, C. George Ray, Publisher:McGraw-Hill
4. Medical Microbiology by Patrick R. Murray, Ken S. Rosenthal, Michael A. Pfaller, Elsevier HealthSciences.
5. Brooks GF, Carroll KC, Butel JS and Morse SA. (2007).Jawetz,
6. Melnick and Adelberg's Medical Microbiology. 24th edition. McGraw Hill Publication7. Goering R, Dockrell H, Zuckerman M and Wakelin D. (2007). Mims' Medical Microbiology. 4th edition.Elsevier.

## Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

## Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and Pos		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	Identify pathogenic bacteria based on cultural, morphological characteristics	PO1, PO2, PO5, PO6, PO7, PO8
CO2	Illustrate growth curve of a bacterium.	PO1, PO2, PO5, PO6, PO7, PO8
CO3	Examine temporary mounts of Aspergillus by appropriate staining.	PO1, PO2, PO5, PO6, PO7, PO8



<b>CO4</b>	<b>Examine</b> the bacteria by Gram staining	<b>PO1, PO2, PO5, PO6, PO7,PO8</b>
<b>CO5</b>	<b>Identify</b> the bacteria by acid fast staining	<b>PO1, PO2, PO5, PO6, PO7,PO8</b>

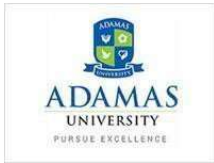
		Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking	Problem Identification Ability	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to Environment	Development to Society.	Development to Humanity
Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
<b>BIC 12039</b>	DSE II Medical Microbiology Lab	3	3	-	-	3	3	3	3	-	-	-	-

1=weakly mapped

2= moderately mapped

3=strongly mapped

## Model Question Paper

<b>Name:</b>  <b>Enrolment No:</b>		 <p style="font-size: small; margin: 0;">ADAMAS UNIVERSITY PURSUE EXCELLENCE</p>	
<b>Course: BIC 12039–DSE II Medical Microbiology Lab</b> <b>Program: B.Sc Biochemistry</b> <span style="float: right;"><b>Time: 03Hrs.</b></span> <b>Semester: Odd2020-21</b> <span style="float: right;"><b>Max. Marks:40</b></span>			
<b>Instructions:</b> Attempt any two questions from <b>Section A</b> (each carrying 10 marks); any <b>Two Questions</b> from <b>Section B</b> (each carrying 10 marks) <b>is Compulsory.</b>			
<b>Section A ( Attemptany Three)</b>			
1.	<b>Identify</b> the specimen of the two given slides (Ap) , write down their specific characteristics and their relation with infectious disease.(An)	<b>5</b> <b>5</b>	<b>CO1,CO2, CO3,CO4,CO5</b>
2.	Perform the Acid fast staining procedure with the supplied bacterial sample(Ap),write down the principle, procedure, observation and inference.(An)	<b>10</b>	<b>CO2</b>
3.	Perform the Cotton blue staining procedure with the supplied fungus sample(Ap),write down the principle, procedure, observation and inference.(An)	<b>10</b>	<b>CO4</b>
<b>SECTION B (Attempt any Two Questions)</b>			
5.	<b>Viva-Voce (U/An/Ap/R)</b>	<b>10</b>	<b>CO1, CO2,CO3, CO4,CO5</b>
6.	<b>Laboratory Note Book(U/An/Ap/Ev)</b>	<b>10</b>	<b>CO1, CO2,CO3, CO4,CO5</b>

<b>BIC12041</b>	<b>DSE II Nutritional Biochemistry Lab(Practical)</b>	L	T	P	C
<b>Version 1.0</b>	<b>Contact Hours: 45</b>	0	0	3	2
<b>Pre-requisites/Exposure</b>	Basic Knowledge of Biomolecules				
<b>Co-requisites</b>	-				

### Course Objectives

1. To provide knowledge about analysing the nutrition and energy metabolism.
2. To measure the dietary carbohydrates from food samples.
3. To measure the dietary proteins from food samples.
4. To measure the dietary lipids from food samples.
5. To measure the dietary vitamins and minerals from food samples.

### Course Outcomes

On completion of this course, the students will be able to

**CO1 Explain** assay of vitamin B12/B1.

**CO2 Discuss** importance of dietary carbohydrates in health.

**CO3 Explain** assay of serum/ urine MMA

**CO4 Identify** sample of Kwashiorkor, Marasmus and Obesity

**CO5 Explain** assay of Vitamin A/E in serum

### Catalog Description

Nutritional biochemistry is one of the academic foundations that make up nutritional sciences, a discipline that encompasses the knowledge of nutrients and other food components with emphasis on their range of function and influence on mammalian physiology, health, and behavior. Nutritional biochemistry is a subdiscipline that is made up of the core knowledge, concepts, and methodology related to the chemical properties of nutrients and other dietary constituents and to their biochemical, metabolic, physiological, and epigenetic functions. A primary focus of this paper is the qualitative and quantitative analysis of some food components.

## Course Content

1. Bioassay for vitamin B12/B1. [10 LectureHours]
2. Homocystiene estimation. [10 LectureHours]
3. Serum/ urine MMA estimation. [10 LectureHours]
4. Anthropometric identifications for Kwashiorkor, Marasmus and Obesity. [10 Lecture Hours]
5. Determination of oxidative stress: TBARS, antioxidant enzymes in hemolysate. [10 Lecture Hours]
6. Vitamin A/E estimation in serum. [5 LectureHours]
7. Bone densitometry /bone ultrasound test demonstration (visit to a nearby clinic) [5 Lecture Hours]

## Reference Books

1. Textbook of Biochemistry with Clinical Correlations (2011) Devlin, T.M. John Wiley & Sons, Inc. (New York), ISBN:978-0-4710-28173-4.
2. Nutrition for health, fitness and sport (2013) ; Williams.M.H, Anderson,D.E, Rawson,E.S. McGraw Hill international edition. ISBN-978-0-07-131816-7.
3. Krause's Food and Nutrition Care process.(2012); Mahan, L.K Strings,S.E, Raymond,J. Elsevier's Publications. ISBN-978-1-4377-2233-8.
4. The vitamins, Fundamental aspects in Nutrition and Health (2008); G.F. Coombs Jr. Elsevier's Publications. ISBN-13- 978-0-12- 183493-7.
5. Principles of Nutritional Assessment (2005) Rosalind Gibson. Oxford University Press.

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

## Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and Pos		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	Explain assay of vitamin B12/B1..	PO1, PO2, PO3, PO4, PO5, PO6, PO7, PO8, PO10, PO11

<b>CO2</b>	<b>Discuss</b> importance of dietary carbohydrates in health.	<b>PO1, PO2, PO3, PO4, PO5, PO6, PO7, PO8, PO10, PO11</b>
<b>CO3</b>	<b>Explain</b> assay of serum/ urine MMA	<b>PO1, PO2, PO3, PO4, PO5, PO6, PO7, PO8, PO10, PO11</b>
<b>CO4</b>	<b>Identify</b> sample of Kwashiorkor, Marasmus and Obesity	<b>PO1, PO2, PO3, PO4, PO5, PO6, PO7, PO8, PO10, PO11</b>
<b>CO5</b>	<b>Explain</b> assay of Vitamin A/E in serum	<b>PO1, PO2, PO3, PO4, PO5, PO6, PO7, PO8, PO10, PO11</b>

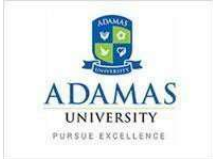
		Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking	Problem Identification Ability	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to Environment	Development to Society.	Development to Humanity
Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
<b>BIC12041</b>	DSE II Nutritional Biochemistry Lab	3	3	3	3	3	3	3	3	2	3	3	2

1=weakly mapped

2= moderately mapped

3=strongly mapped

## Model Question Paper

<b>Name:</b>			
<b>Enrolment No:</b>			
<b>Course: BIC12041–DSE II Nutritional Biochemistry Lab</b> <b>Program: B.Sc Biochemistry</b> <span style="float: right;"><b>Time: 03Hrs.</b></span> <b>Semester: Odd2020-21</b> <span style="float: right;"><b>Max. Marks:40</b></span>			
<b>Instructions:</b> Attempt any two questions from <b>Section A</b> (each carrying 10 marks); <b>Section B</b> (each carrying 10 marks) is Compulsory.			
<b>Section A ( Attemptany Two)</b>			
1.	<b>Identify</b> the nutrients from the given sample A and B.(Ap) and Write the results along with the experimental procedure in detail.(An)	4 4 2	<b>CO1, CO2, CO3,CO4</b>
2.	<b>Estimate</b> the quantity of Vitamin B1 from the given sample (Ap)and detail the experimental procedure , observation and inference.(An)	4 4 2	<b>CO1</b>
3.	<b>Estimate</b> the quantity of Vitamin A from the given serum sample(Ap) and detail the experimental procedure , observation and inference.(An)	4 4 2	<b>CO5</b>
<b>SECTION B (Compulsory)</b>			
4.	<b>Viva-Voce</b> (U/An/Ap/R)	<b>10</b>	<b>CO1, CO2,CO3, CO4,CO5</b>
5.	<b>Laboratory Note Book</b> (U/An/Ap/Ev)	<b>10</b>	<b>CO1, CO2,CO3, CO4,CO5</b>

<b>BIC 14042</b>	<b>Industry Internship (Practical)</b>	<b>L</b>	<b>T</b>	<b>P</b>	<b>C</b>
<b>Version 1.0</b>					2
<b>Pre-requisites/Exposure</b>	Basic knowledge of Biochemistry				
<b>Co-requisites</b>	--				

### Course Objectives

1. To provide students basic idea about work habits and attitudes necessary for job success.
2. To illustrate the career alternatives prior to graduation.
3. To develop communication, interpersonal and other critical skills in the job interview process.
4. To provide students the ability to analyze interests and abilities in their field of study.

### Course Outcomes

On completion of this course, the students will be able to:

- CO1 Develop** work habits and attitudes necessary for job success.
- CO2 Evaluate** career alternatives prior to graduation.
- CO3 Develop** communication, interpersonal and other critical skills in the job interview process.
- CO4 Analyze** interests and abilities in their field of study.
- CO5 Compile** employment contacts leading directly to a full-time job following graduation from college.

### Catalog Description

The practical course of 'Industry Internship' will help to develop work habits and attitudes necessary for job success. This course includes comprehensive approach to develop communication, interpersonal and other critical skills in the job interview process. Furthermore, interests and abilities in their field of study will also be illuminated. The practical experience will enable students to enrich in real-life scenario. The tutorials will enable the students with problem-solving ability led by the course coordinator. Students will perceive the basic concepts of the subject via exercise and discussions with the coordinator.

### Course Content

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## Industry Internship (PRACTICALS)

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination**  
**Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

### Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and POs		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	Students will be able to <b>develop</b> work habits and attitudes necessary for job success.	PO1, PO3, PO8
CO2	Students will be able to <b>evaluate</b> career alternatives prior to graduation.	PO1, PO2
CO3	Students will be able to <b>develop</b> communication, interpersonal and other critical skills in the job interview process.	PO1, PO3, PO8
CO4	Students will be able to <b>outline</b> interests and abilities in their field of study.	PO1, PO2, PO3, PO4, PO7
CO5	Students will be able to <b>compile</b> employment contacts leading directly to a full-time job following graduation from college.	PO3, PO8



Course Code	Course Title	1=weakly mapped 2= moderately mapped 3=strongly mapped	
		PO	PO
<b>BIC 14042</b>	Industry Internshi p (THEOR Y)	1	Fundamental Knowledge
		2	Critical Thinking
		3	Skill
		4	Technical Knowledge
		5	Logical Thinking
		6	Problem Identification Ability
		7	Analytical Knowledge
		8	Career Goals
		9	Team Work
		PO1 0	Sustainable Development to nvironment
		PO1 1	Development to Society.
		PO1 2	Development to Humanity
		3	
		1	
3			
2			
-			
-			
1			
3			
-			
-			
-			

# Semester 6

<b>BIC11043</b>	<b>Genetic Engineering and Biotechnology(Theory)</b>	L	T	P	C
<b>Version 1.0</b>	Contact Hours: 60	3	1	0	4
<b>Pre-requisites/Exposure</b>	Basic Knowledge of Molecular Biology				
<b>Co-requisites</b>	-				

## Course Objectives

1. To understand Restriction and modification systems, ligation systems in bacteria and viruses and apply their appropriate use in recombinant DNA technology.
2. To provide knowledge in cloning vectors for prokaryotes and eukaryotes and apply their appropriate use in recombinant DNA technology
3. To discuss PCR, DNA sequencing and choose their appropriate use in recombinant DNA technology.
4. To understand recombinant DNA technology in gene expression.
5. To conceptualize structures of porphyrin, their classification, several types of reaction and biological significance.

## Course Outcomes

On completion of this course, the students will be able to

- CO1** **explain** Restriction and modification systems, ligation systems in bacteria and viruses and apply their appropriate use in recombinant DNA technology.
- CO2** **explain** Cloning vectors for prokaryotes and eukaryotes and apply their appropriate use in recombinant DNA technology.
- CO3** **discuss** PCR, DNA sequencing and choose their appropriate use in recombinant DNA technology.
- CO4** **summarize** recombinant DNA technology in gene expression.
- CO5** **build** the concept about the structures of porphyrin, their classification, several types of reaction and biological significance

## Catalog Description

Students will study the theoretical and applied aspects of basic biotechnology techniques for the study of DNA and proteins. In the laboratory students will apply theory and practical skills from this and previous courses to perform standard molecular biology techniques for the isolation, manipulation and analysis of DNA as well as the expression and purification of protein.

## Course Content

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### 1. Unit I Introduction to recombinant DNA technology [6 LectureHours]

Overview of recombinant DNA technology. Restriction and modification systems, restriction endonucleases and other enzymes used in manipulating DNA molecules, separation of DNA by gel electrophoresis. Extraction and purification of plasmid and bacteriophage DNA.

### 2. Unit II Cloning vectors for prokaryotes and eukaryotes [6 LectureHours]

Plasmids and bacteriophages as vectors for gene cloning. Cloning vectors based on E. coli plasmids, pBR322, pUC8, pGEM3Z. Cloning vectors based on M13 and  $\lambda$  bacteriophage. Vectors for yeast, higher plants and animals.

### 3. Unit III Joining of DNA fragments [6 LectureHours]

Ligation of DNA molecules. DNA ligase, sticky ends, blunt ends, linkers and adapters. Synthetic oligonucleotides, synthesis and use.

### 4. Unit IV Introduction of DNA into cells and selection for recombinants [6 Lecture Hours]

Uptake of DNA by cells, preparation of competent cells. Selection for transformed cells. Identification for recombinants - insertional inactivation, blue-white selection. Introduction of phage DNA into bacterial cells. Identification of recombinant phages. Introduction of DNA into animal cells, electroporation.

### 5. Unit V Methods for clone identification [6 LectureHours]

The problem of selection, direct selection, marker rescue. Gene libraries, identification of a clone from gene library, colony and plaque hybridization probing, methods based on detection of the translation product of the cloned gene.

### 6. Unit VI Polymerase chain reaction [6 LectureHours]

Fundamentals of polymerase chain reaction, designing primers for PCR. Studying PCR products. Cloning PCR products. Real time PCR.

### 7. Unit VII DNA sequencing [6 LectureHours]

DNA sequencing by Sanger's method, modifications based on Sanger's method. Automated DNA sequencing. Pyrosequencing.

### 8. Unit VIII Expression of cloned genes [6 LectureHours]

Vectors for expression of foreign genes in E. coli, cassettes and gene fusions. Challenges in producing recombinant protein in E. coli. Production of recombinant protein by eukaryotic cells. Fusion tags and their role in purification of recombinant proteins.

### 9. Unit IX Applications of genetic engineering in Biotechnology [12 Lecture Hours]

Site-directed mutagenesis and protein engineering. Applications in medicine, production of recombinant pharmaceuticals such as insulin, human growth hormone, factor VIII., Recombinant vaccines. Gene therapy. Applications in agriculture - plant genetic engineering, herbicide resistant crops, problems with genetically modified plants, safety concerns.

## Reference Books

1. Gene Cloning and DNA Analysis (2010) 6<sup>th</sup> ed., Brown, T.A., Wiley-Blackwell publishing (Oxford, UK), ISBN:978-1-4051-8173-0.
2. Principles of Gene Manipulation and Genomics (2006) 7<sup>th</sup> ed., Primrose, S.B., and Twyman, R. M., Blackwell publishing (Oxford, UK) ISBN:13: 978-1-4051-3544-3.

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

## Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and Pos		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	<b>Explain</b> Restriction and modification systems, ligation systems in bacteria and viruses and apply their appropriate use in recombinant DNA technology.	PO1, PO3
CO2	<b>Explain</b> Cloning vectors for prokaryotes and eukaryotes and apply their appropriate use in recombinant DNA technology	PO1, PO2, PO3
CO3	<b>Discuss</b> PCR, DNA sequencing and choose their appropriate use in recombinant DNA technology	PO4, PO6, PO7, PO9
CO4	<b>Summarize</b> recombinant DNA technology in gene expression.	PO1, PO5, PO12,
CO5	<b>Build</b> the concept about the structures of porphyrin, their classification, several types of reaction and biological significance	PO1, PO6, PO8, PO9, PO12,

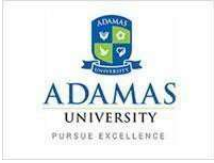
		Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking Ability	Problem Identification	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to Environment	Development to Society	Development to Humanity
Course Code	Course Title	PO 1	PO 2	PO 3	PO 4	PO 5	PO 6	PO 7	PO 8	PO 9	PO1 0	PO1 1	PO1 2
<b>BIC11043</b>	Genetic Engineering and Biotechnology	3	1	2	1	1	2	1	1	2	1	-	2

1=weakly mapped

2= moderately mapped

3=strongly mapped

### Model Question Paper

<b>Name:</b>			
<b>Enrolment No:</b>			
<b>Course: BIC11043– Genetic Engineering and Biotechnology</b> <b>Program: B.SC Biochemistry</b> <b>Semester: Even 2019-20</b>			
<b>Time: 03Hrs.</b> <b>Max. Marks: 40</b>			
<b>Instructions:</b>			
Attempt any three questions from <b>Section A</b> (each carrying 4 marks); any <b>four Questions</b> from <b>Section B</b> (each carrying 7 marks).			
<b>Section A ( Attemptany Three)</b>			
1.	<b>Which</b> of these restriction enzymes	<b>4</b>	<b>CO1</b>

	produce blunt ends?. (U)		
2.	The RP13 gene of chromosome 17 codes for protein_____.(U)	4	CO2
3.	Isoschizomers recognize <b>what?</b> (R)	4	CO4
4.	Restriction enzymes are <b>named</b> for? (U)	4	CO3
<b>SECTION B (Attempt any four Questions)</b>			
5.	Restriction enzymes are extensively used in molecular biology. Below are the recognition sites of two of these enzymes, BamHI and BclI. a) BamHI, cleaves after the first G. Does cleavage by BamHI result in a 5' or 3' overhang? What is the sequence of this overhang? 5' GGATCC 3' 3' CCTAGG 5' (Ap)	7	CO2
6.	B cII cleaves after the first T. Does cleavage by BclI result in a 5' or 3' overhang? What is the sequence of this overhang? 5' TGATC 3' 3' ACTAGT 5'(Ap)	7	CO4 CO2 CO3
7.	You can ligate the smaller restriction fragment produced in (c, i) to the smaller restriction fragment produced in (c, ii). Write out the sequence of the resulting recombinant fragment	7	CO2
	Millions of children that depend primarily on rice as a food staple become blind each year due to vitamin A deficiency. Humans cannot synthesize vitamin A and must have a supply of vitamin A or vitamin A precursor like beta-carotene in their diet. You want to create a strain of yeast that has the		

	<p>complete beta-carotene pathway so you can easily study the biochemistry of beta-carotene and vitamin A synthesis. You have a yeast strain that has 5 of the required 7 enzymes for beta-carotene synthesis. You need to provide these yeast cells with the two missing enzymes, crt1 from bacteria and psy2 from daffodils. To clone the gene encoding crt1 into yeast, you plan to</p> <ol style="list-style-type: none"> <li>1. Cut bacterial genomic DNA.</li> <li>2. Clone it into an appropriate expression vector (vector 1) to create a library.</li> <li>3. Transform your yeast cells with the library.</li> <li>4. Select for yeast cells that have obtained any vector.</li> <li>5. Screen theselected colonies for production of the crt1 protein using an antibody.</li> </ol>		
8.	<p>a) The cloning vector chosen for this experiment has the gene leu1+ , which encodes an enzyme needed to synthesize the amino acid leucine and can be used as a selectable marker. To successfully complete step 4 above, what genotype and phenotype would your yeast strain be prior to transformation?</p>	7	<b>CO5</b>
9.	<p>List the minimum features, in addition to the leu1+ gene, required in the vector for all steps 1-5 outlined above to be successful. For any DNA sequence(s) listed, indicate what type of organism it would come from.</p>	7	<b>CO3</b>

<b>BIC 12044</b>	<b>Genetic Engineering and Biotechnology Lab(Practical)</b>	L	T	P	C
<b>Version 1.0</b>	Contact Hours: 45	0	0	3	2
<b>Pre-requisites/Exposure</b>	Basic knowledge in Molecular Biology				
<b>Co-requisites</b>	-				

### Course Objectives

6. To understand the Restriction and modification systems, ligation systems in bacteria and viruses and apply their appropriate use in recombinant DNA technology
7. To conceptualize about isolation of cloning vectors for prokaryotes and eukaryotes and summarize their appropriate use in recombinant DNA technology
8. To provide knowledge of PCR, DNA sequencing and choose their appropriate use in recombinant DNA technology
9. Able to understand recombinant DNA technology in gene expression through blue-white screening
10. To **demonstrate** and apply RDT in Recombinant vaccines Gene therapy, agriculture- plant genetic engineering, herbicide resistant crops etc.

### Course Outcomes

On completion of this course, the students will be able to

- CO1** **demonstrate** the Restriction and modification systems, ligation systems in bacteria and viruses and apply their appropriate use in recombinant DNA technology.
- CO2** **develop** the concept about isolation of cloning vectors for prokaryotes and eukaryotes and **apply** their appropriate use in recombinant DNA technology.
- CO3** **demonstrate** PCR, DNA sequencing and choose their appropriate use in recombinant DNA technology.
- CO4** **apply** recombinant DNA technology in gene expression through blue-white screening.
- CO5** **demonstrate** and apply RDT in Recombinant vaccines Gene therapy, agriculture - plant genetic engineering, herbicide resistant crops etc.

### Catalog Description

Construction Management is the overall planning, coordination and control of a project from inception to completion aimed at meeting a client's requirements in order to produce a functionally and financially viable project. This course includes specific activities like defining the responsibilities and management structure of the project management team, planning methods and implementing it in project controls (time and cost), defining roles and



responsibilities of personnel in the organization, equipments and safety measures in construction. All the lectures will be devoted on discussions of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as power point presentation, audio visual virtual labs session as per requirement. The tutorials will familiarize the students with practical problem-solving techniques led by the course coordinator. Students will strongly grab the basic concepts of the subject via exercise and discussions with the coordinator.

### Course Content

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1. Restriction Digestion of DNA. (10 Hours)
2. Ligation of DNA. (10 Hours)
3. Primer Designing. (10 Hours)
4. PCR. (10 Hours)
5. Growth Kinetics Study by using Bioreactor. (15 Hours)
6. Estimation of Alcohol production using Bioreactor. (15 Hours)

#### Reference Book:

1. Gene Cloning and DNA Analysis (2010) 6<sup>th</sup> ed., Brown, T.A., Wiley- Blackwell publishing (Oxford, UK), ISBN:978-1-4051-8173-0.
2. Gene Cloning Manual: Sambrook and Russell
3. Principles of Gene Manipulation and Genomics (2006) 7<sup>th</sup> ed., Primrose, S.B., and Twyman, R. M., Blackwell publishing (Oxford, UK) ISBN:13: 978- 1-4051-3544-3.

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

#### Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and Pos		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	<b>Demonstrate</b> the Restriction and modification systems, ligation systems in bacteria and viruses and apply their appropriate use in recombinant DNA technology	PO1, PO2

<b>CO2</b>	<b>Develop</b> the concept about isolation of cloning vectors for prokaryotes and eukaryotes and <b>apply</b> their appropriate use in recombinant DNA technology	<b>PO1,PO2, PO3,</b>
<b>CO3</b>	<b>Demonstrate</b> PCR, DNA sequencing and choose their appropriate use in recombinant DNA technology	<b>PO1, PO2, PO3, PO4</b>
<b>CO4</b>	<b>Apply</b> recombinant DNA technology in gene expression through blue-white screening	<b>PO1, PO5, PO10</b>
<b>CO5</b>	<b>Demonstrate</b> and apply RDT in Recombinant vaccines Gene therapy, agriculture - plant genetic engineering, herbicide resistant crops etc	<b>PO1, PO6, PO8</b>

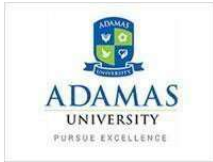
Course Code	Course Title	PO 1	PO 2	PO 3	PO 4	PO 5	PO 6	PO 7	PO 8	PO 9	PO10	PO11	PO12
BIC1204 4	Genetic Engineering and Biotechnology Lab	3	3	2	1	1	1	-	1	-	1	-	-

1=weakly mapped

2= moderately mapped

3=strongly mapped

## Model Question Paper

<b>Name:</b> <b>Enrolment No:</b>			
<b>Course: BIC 12044– Genetic Engineering and BiotechnologyLab</b> <b>Program: B.SC Biochemistry</b> <b>Semester: Even 2019-20</b> <b>Time: 03Hrs.</b> <b>Max. Marks: 40</b>			
<b>Instructions:</b> Attempt all questions			
1	Isolate the Genomic DNA from bacterial cell.(Ap)	10	CO3
2	Estimate the DNA concentration by spectrophotometer.(An)	10	CO4
3.	Note Book	10	CO1 CO2 CO3
4	Viva Voce	10	CO1, CO2, CO3

<b>BIC 11045</b>	<b>Hormones: Biochemistry and Function (Theory)</b>	L	T	P	C
<b>Version 1.0</b>	<b>Contact Hours: 60</b>	3	1	0	4
<b>Pre-requisites/Exposure</b>	Knowledge of Biology at 10+2 level				
<b>Co-requisites</b>	-				

### Course Objectives

To provide a deeper understanding about the organization of endocrine system of human body.

1. To acquire the knowledge relation between neural and endocrinesystem.
2. To acquire the knowledge about the establishment of physiological homeostasis by hormones.
3. To provide the knowledge about biochemical basis of regulation of hormone secretion.
4. To understand the biochemical disorder due to hypo and hyper secretion of hormones. .

### Course Outcomes

On completion of this course, the students will be able to

**CO1 Explain** the process of endocrine system and Functions of hormones and illustrate their regulation.

**CO2 Discuss** hormone mediated signalling mechanisms and illustrate their application in modern biology.

**CO3 Explain** the structure of different endocrine glands and hormones secretes by them.

**CO4 Discuss** the biosynthesis and mechanism of their action of different hormones.

**CO5 Discuss** the disorders related to hyper and hypo secretion of hormones.

## Catalog Description

Endocrinology is a branch of biology and medicine dealing with the endocrine system, its diseases, and its specific secretions known as hormones. It is also concerned with the integration of developmental events proliferation, growth, and differentiation, and the psychological or behavioral activities of metabolism, growth and development, tissue function, sleep, digestion, respiration, excretion, mood, stress, lactation, movement, reproduction, and sensory perception caused by hormones. Specializations include behavioral endocrinology and comparative endocrinology. The endocrine system consists of several glands, all in different parts of the body, that secrete hormones directly into the blood rather than into a duct system. Therefore, endocrine glands are regarded as ductless glands. Hormones have many different functions and modes of action; one hormone may have several effects on different target organs, and, conversely, one target organ may be affected by more than one hormone. So to conceptualize the coordination between the physiological functions mediated by hormones by their special biochemical features is the goal of this paper.

## Course Content

- 1. Introduction to endocrinology:** Functions of hormones and their regulation. Chemical signaling - endocrine, paracrine, autocrine, intracrine and neuroendocrine mechanisms. Chemical classification of hormones, transport of hormones in the circulation and their half-lives. Hormone therapy. General introduction to Endocrine methodology. **[15 Lecture Hours]**
- 2. Hormone mediated signaling:** Hormone receptors - extracellular and intracellular. Receptor - hormone binding, Scatchard analysis. G protein coupled receptors, G proteins, second messengers - cAMP, cGMP, IP<sub>3</sub>, DAG, Ca<sup>2+</sup>, NO. Effector systems - adenylylase,

guanylcyclase, PDE, PLC. Protein kinases (PKA, PKB, PKC, PKG). Receptor tyrosine kinases - EGF, insulin, erythropoietin receptor; ras - MAP kinase cascade, JAK - STAT pathway. Steroid hormone/ thyroid hormone receptor mediated gene regulation. Receptor regulation and cross talk. Growth factors: PDGF, EGF, IGF-II, and erythropoietin. **[15 Lecture Hours]**

3. **Hypothalamic, pituitary and thyroid hormones.** Hypothalamic - pituitary axis. Study the physiological and biochemical actions of hypothalamic hormones, pituitary hormones - GH, prolactin, TSH, LH, FSH, POMC peptide family, oxytocin and vasopressin, feedback regulation cycle. Endocrine disorders - gigantism, acromegaly, dwarfs, pigmies and diabetes insipidus. Thyroid hormone: Thyroid gland. Biosynthesis of thyroid hormone and its regulation; its physiological and biochemical action. Pathophysiology - Goiter, Graves disease, cretinism, myxedema, Hashimoto's disease. Hormones regulating Ca<sup>2+</sup> homeostasis: PTH, Vitamin D and calcitonin. Mechanism of Ca<sup>2+</sup> regulation and pathways involving bone, skin, liver, gut and kidneys. Pathophysiology - rickets, osteomalacia, osteoporosis. **[15 Lecture Hours]**
4. **Pancreatic, GI tract, adrenal and reproductive hormones:** Regulation of release of insulin, glucagon, gastrin, secretin, CCK, GIP, adipolectin, leptin and ghrelin. Summary of hormone metabolite control of GI function. Physiological and biochemical action. Pathophysiology - diabetes type I and type II. Aldosterone, renin angiotensin system, cortisol, epinephrine and norepinephrine. Fight or flight response, stress response. Pathophysiology – Addison's disease, Conn's syndrome, Cushing syndrome. Reproductive hormones: Male and female sex hormones. Interplay of hormones during reproductive cycle, pregnancy, parturition and lactation. Hormone based contraception. **[15 Lecture Hours]**

### Reference Books

1. Lehninger: Principles of Biochemistry (2013) 6th ed., Nelson, D.L. and Cox, M.M. W.H. Freeman & Company (New York), ISBN: 13: 978-1-4641-0962-1 / ISBN: 10-14641-0962-1.
2. Vander's Human Physiology (2008) 11th ed., Widmaier, E.P., Raff, H. and Strang, K.T. McGraw Hill International Publications, ISBN: 978-0-07-128366-3.
3. Endocrinology (2007) 6th ed., Hadley, M.C. and Levine, J.E. Pearson Education (New Delhi), Inc. ISBN: 978-81-317-2610-5.
4. Fundamental of Anatomy and Physiology (2009), 8th ed., Martini, F.H. and Nath, J.L., Pearson Publications (San Francisco), ISBN: 10: 0-321-53910-9 / ISBN: 13: 978-0321-53910-6.

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

### Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and Pos		
	Course Outcomes (COs)	Mapped Program Outcomes
<b>CO1</b>	Student will be able to <b>explain</b> the process of endocrine system and Functions of hormones and illustrate their regulation.	<b>PO1, PO2, PO3, PO4, PO5, PO6, PO7, PO8, PO12</b>
<b>CO2</b>	Student will be able to <b>discuss</b> Hormone mediated signalling mechanisms and illustrate their application in modern biology.	<b>PO1, PO2, PO3, PO4, PO5, PO6, PO7, PO8, PO12</b>
<b>CO3</b>	Student will be able to <b>explain</b> the structure of different endocrine glands and hormones secretes by them.	<b>PO1, PO2, PO3, PO4, PO5, PO6, PO7, PO8, PO12</b>
<b>CO4</b>	Student will be able to <b>discuss</b> the biosynthesis and mechanism of their action of different hormones.	<b>PO1, PO2, PO3, PO4, PO5, PO6, PO7, PO8, PO12</b>
<b>CO5</b>	Student will be able to <b>discuss</b> the disorders related to hyper and hypo secretion of hormones.	<b>PO1, PO2, PO3, PO4, PO5, PO6, PO7, PO8, PO12</b>

		Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking	Problem Identification Ability	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to Environment	Development to Society.	Development to Humanity
Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
<b>BIC 11045</b>	Hormones: Biochemistry and Function	3	3	3	3	3	3	3	3	-	-	-	3

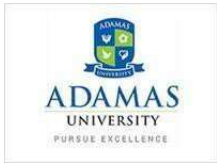


1=weakly mapped

2= moderately mapped

3=strongly mapped

## Model Question Paper

<b>Name:</b>  <b>Enrolment No:</b>		 <p style="font-size: small; margin: 0;">ADAMAS UNIVERSITY PURSUE EXCELLENCE</p>	
<b>Course: BIC 12045 Hormones: Biochemistry and Function</b> <b>Program: B.Sc Biochemistry</b> <span style="float: right;"><b>Time: 03Hrs.</b></span> <b>Semester: Even 2020-21</b> <span style="float: right;"><b>Max. Marks: 40</b></span>			
<b>Instructions:</b> Attempt any three questions from <b>Section A</b> (each carrying 4 marks); any <b>Two Questions</b> from <b>Section B</b> (each carrying 10 marks). <b>Section C</b> is Compulsory (carrying 8 marks).			
<b>Section A ( Attemptany Three)</b>			
1.	Write down the “Two cell hypothesis” of estrogen biosynthesis. <b>Why</b> placenta is dependent on adrenal cortex of mother and fetus for estrogen biosynthesis?(U)	2 2	CO1 CO4
2.	<b>Find out</b> the similarities between the mechanism of actions of adrenaline and glucagon with special reference to induction of glycogenolysis. <b>Why</b> FSH, LH and hCG belongs to the same family?(AP)	4	CO3
3.	Write down the roles of the STAR and C20-C22 desmolase in steroidogenesis? <b>How</b> the counteraction of growth hormone and insulin was established?(R)	2 2	CO4
4.	<b>Derive</b> the numerical formula of the ratio of the affinity of hormone and its competitive agonist with a specific receptor molecule, when both hormone and antagonists are present in the medium and competing with each other for the binding with receptor.(AP)	4	CO2
<b>SECTION B (Attempt any Two Questions)</b>			
5.	<b>How</b> adrenaline exerts its chronotropic and ionotropic effect through its $\beta_1$ receptors? Peptide bond hydrolysis is a major factor for the regulation of blood pressure by the hormones produced by kidney- <b>prove</b> .	3 5 2	CO2 CO5 CO1

	<b>What</b> do you mean by CRE?(R)		
6.	<b>How</b> the rhythm of oestrogen and progesterone is altered throughout the monthly ovarian and uterine cycle of human female?( <b>explain</b> with the events). <b>Why</b> only one ovum is fertilized in one month? <b>Oestrogen</b> is also important in male for regulated gonad hormone secretion- <b>justify</b> . <b>Write</b> down the normal level of hCG during pregnancy.(U)	5 2 2 1	CO2 CO3 CO5
7.	<b>How</b> the Ca homeostasis is established in human body by counteraction of two different hormones?(explain). <b>Explain</b> the V1 and V2 receptor mediated actions of antidiuretic hormone(ADH)? <b>How</b> autoimmune diabetes mellitus occurs? <b>How</b> endogenous analgesics are formed during processing of hormones?(R)	3 4 1 2	CO1 CO2 CO3 CO5
<b>SECTION C is Compulsory</b>			
8.	<b>How</b> the blood glucose homeostasis is established in human body by counteraction of different hormones? If the Na-K balance is disturbed in blood plasma, then agonist of <b>which</b> hormone you will apply and why?(U)	4 4	CO3 CO2 CO5

## Course Objectives

<b>BIC 12046</b>	<b>Hormones Biochemistry and Function Lab (Practical)</b>	L	T	P	C
<b>Version 1.0</b>	<b>Contact Hours: 45</b>	0	0	3	2
<b>Pre-requisites/Exposure</b>	Knowledge in Biochemistry				
<b>Co-requisites</b>	-				

1. To observe the microscopic organization of endocrine system of human body.
2. To analyze relation between neural and endocrine system.
3. To analyze the establishment of physiological homeostasis by hormones.
4. To examine biochemical basis of regulation of hormone secretion.
5. To identify the cause of the biochemical disorder due to hypo and hyper secretion of hormones.

## Course Outcomes

On completion of this course, the students will be able to

**CO1 Explain** serum glucose level and analyse the data.

**CO2 Examine** serum Ca<sup>2+</sup> level and analyse the data.

**CO3 Illustrate** serum T4 level and analyse the data.

**CO4 Examine** of serum electrolytes level and analyse the data.

**CO5 Case study of** different disorders and categorize them accordingly.

## Catalog Description

Endocrinology is a branch of biology and medicine dealing with the endocrine system, its diseases, and its specific secretions known as hormones. It is also concerned with the integration of developmental events proliferation, growth, and differentiation, and the psychological or behavioral activities of metabolism, growth and development, tissue function, sleep, digestion, respiration, excretion, mood, stress, lactation, movement, reproduction, and sensory perception caused by hormones. Specializations include behavioral endocrinology and comparative endocrinology. The endocrine system consists of several glands, all in different parts of the body, that secrete hormones directly into the blood rather than into a duct system. Therefore, endocrine glands are regarded as ductless glands. Hormones have many different functions and modes of action; one hormone may have several effects on different target organs, and, conversely, one target organ may be affected by more than one hormone. So to analyze the coordination between the physiological functions mediated by hormones by their special biochemical features is the goal of this paper.

## Course Content

1. Glucose Tolerance Test.[5Hours]
2. Pregnancy Test.[5Hours]
3. Ovulation Test.[5Hours]
4. *In silico* Analysis of Hormone-Receptor Interaction.[5Hours]
5. Estimation of serum Ca<sup>2+</sup>. [10 LectureHours]
6. Estimation of serum T4. [15 LectureHours]
7. Estimation of serum electrolytes. [15 LectureHours]

Case studies.

## Reference Books

1. Vander's Human Physiology (2008) 11<sup>th</sup> ed., Widmaier, E.P., Raff, H. and Strang, K.T., McGraw Hill International Publications (New York), ISBN:978-0-07-128366-3.
2. Harper's Biochemistry (2012) 29<sup>th</sup> ed., Murray, R.K., Granner, D.K., Mayes and P.A., Rodwell, V.W., Lange Medical Books/McGraw Hill. ISBN:978-0-07-176-576-3.
3. Textbook of Medical Physiology (2011) 10<sup>th</sup> ed., Guyton, A.C. and Hall, J.E., Reed Elsevier India Pvt. Ltd. (New Delhi). ISBN:978-1-4160-4574-8.
4. Fundamental of Anatomy and Physiology (2009), 8<sup>th</sup> ed., Martini, F.H. and Nath, J.L., Pearson Publications (San Francisco), ISBN: 10:0-321-53910-9 / ISBN: 13:978-0321-53910-6. Chemistry of Nucleic acids, Adams.

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

## Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and POs		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	<b>Explain</b> serum glucose level and analyse the data.	PO1, PO2, PO3, PO5, PO6, PO7, PO8, PO11
CO2	<b>Examine</b> serum Ca <sup>2+</sup> level and analyse the data.	PO1, PO2, PO3, PO5, PO6, PO7, PO8, PO11
CO3	<b>Illustrate</b> serum T4 level and analyse the data.	PO1, PO2, PO3, PO5, PO6, PO7, PO8, PO11



<b>CO4</b>	<b>Examine</b> of serum electrolytes level and analyse the data.	<b>PO1, PO2, PO3, PO5, PO6, PO7, PO8, PO11</b>
<b>CO5</b>	<b>Case study of</b> different disorders and categorize them accordingly.	<b>PO1, PO2, PO3, PO5, PO6, PO7, PO8, PO11</b>

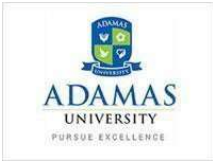
		Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking	Problem Identification Ability	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to Environment	Development to Society.	Development to Humanity
Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
<b>BIC 12046</b>	Hormones Biochemistry and Function Lab	3	3	3	-	3	3	3	3	-	-	3	-

1=weakly mapped

2= moderately mapped

3=strongly mapped

**Model Question Paper**

<b>Name:</b>			
<b>Enrolment No:</b>			
<b>Course: BIC BIC 12046 –:Hormones Biochemistry and Function</b> <b>Lab Program: B.Sc Biochemistry</b> <span style="float: right;"><b>Time: 03Hrs.</b></span> <b>Semester: Even 2020-21</b> <span style="float: right;"><b>Max. Marks: 40</b></span>			
<b>Instructions:</b> Attempt any two questions from <b>Section A</b> (each carrying 10 marks); any <b>Two Questions</b> from <b>Section B</b> (each carrying 10 marks) is <b>Compulsory</b> .			
<b>Section A ( Attemptany Three)</b>			
1.	<b>Stain</b> with haematoxylin-eosin and <b>Identify</b> the specimen of the two given slides (Ap) , write down their specific characteristics (An)	5 5	CO5
2.	<b>Estimate</b> the Ca from the supplied serum sample using the standard curve of Ca with calceinreagent(Ap), write down the principle, procedure, observation and inference.(An)	10	CO2
3.	<b>Estimate</b> the glucose content from the supplied plasma sample (Ap), write down the principle, procedure, observation and inference.(An)	10	CO1
<b>SECTION B (Attempt any Two Questions)</b>			
5.	<b>Viva-Voce</b> (U/An/Ap/R)	10	CO1, CO2, CO3, CO4, CO5
6.	<b>Laboratory Note Book</b> (U/An/Ap/Ev)	10	CO1, CO2, CO3, CO4, CO5



<b>BIC11047</b>	<b>DSE III PLANT BIOCHEMISTRY (theory)</b>	<b>L</b>	<b>T</b>	<b>P</b>	<b>C</b>
<b>Version 1.0</b>	<b>Contact Hours - 60</b>	3	1	0	4
<b>Pre-requisites/Exposure</b>	Basic Knowledge of Biochemistry & Botany				
<b>Co-requisites</b>	--				

### Course Objectives

1. To provide students with hands-on training in the field of plant biochemistry.
1. To provide in depth knowledge of modern plant science research.
2. Students will become more proficient with different practical applications of plant biochemistry (e.g. plant tissue culture).

### Course Outcomes

On completion of this course, the students will be able to

**CO1 Explain** the basic principles of plant biochemistry.

**CO2 Illustrate** fundamentals of plant physiological processes like germination, senescence etc.

**CO3 Demonstrate** different assay systems of plant enzymes.

**CO4 Demonstrate** tissue culture techniques.

**CO5 Summarize** the knowledge of plant biochemistry to understand important physiological processes like photosynthesis.

### Catalog Description

The discipline specific course “plant biochemistry lab” is a practical paper which has been designed to provide the knowledge of different aspects of plant biochemistry. It will provide understanding of important physiological processes in plants. Students will be able to understand of tissue culture and other techniques and will practice hands-on all of them.. All the lectures will be devoted on discussions of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as power point presentation, audio visual virtual lab session as per requirement. The tutorials will familiarize the students with practical problem-solving techniques led by the course coordinator. Students will strongly grab the basic concepts of the subject via exercise and discussions with the coordinator.

## Course Content

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### Unit I Introduction to Plant cell structure

Plasma membrane, Vacuole and tonoplast membrane, cell wall, plastids and peroxisomes.

### Unit II Photosynthesis and Carbon assimilation

Structure of PSI and PSII complexes, Light reaction, Cyclic and non cyclic photophosphorylation, Calvin cycle and regulation; C4 cycle and Crassulacean acid metabolism (CAM), Photorespiration.

### Unit III Respiration

Overview of glycolysis, Alternative reactions of glycolysis, Regulation of plant glycolysis, Translocation of metabolites across mitochondrial membrane, TCA cycle, Alternative NAD(P)H oxidative pathways; Cyanide resistant respiration.

### Unit IV Nitrogen metabolism

Biological Nitrogen fixation by free living and in symbiotic association, structure and function of enzyme Nitrogenase. Nitrate assimilation: Nitrate and Nitrite reductase. Primary and secondary ammonia assimilation in plants; ammonia assimilation by Glutamine synthetase- glutamine oxoglutarate amino transferase (GS-GOGAT) pathway. Seed storage proteins in legumes and cereals.

### Unit V Regulation of plant growth

Introduction to plant hormones and their effect on plant growth and development, Regulation of plant morphogenetic processes by light.

### Unit VI Secondary metabolites

Antioxidants. Representatives alkaloid group and their amino acid precursors, function of alkaloids, Examples of major phenolic groups; simple phenylpropanoids, Coumarins, Benzoic acid derivatives, flavonoids, tannins and lignin, biological role of plant phenolics, Classification of terpenoids and representative examples from each class, biological functions of terpenoids.

### Unit VI Plant tissue culture

Cell and tissue culture techniques, types of cultures: organ and explants culture, callus culture, cell suspension culture and protoplast culture. Plant regeneration pathways: organogenesis and somatic embryogenesis. Applications of cell and tissue culture and somoclonal variation.

## SUGGESTED READINGS

1. Plant Biochemistry (2008), Caroline Bowsher, Martin steer, Alyson Tobin, Garland science ISBN978-0-8153-4121-5

2. Biochemistry and molecular Biology of plant- Buchanan. (2005) 1 edition. Publisher: I

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination**  
**Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

**Relationship between the Course Outcomes (COs) and Program Outcomes (POs)**

<b>Mapping between COs and Pos</b>		
	<b>Course Outcomes (COs)</b>	<b>Mapped Program Outcomes</b>
<b>CO1</b>	<b>Explain</b> the basic principles of plant biochemistry.	<b>PO1, PO2, PO3</b>
<b>CO2</b>	<b>Illustrate</b> fundamentals of plant physiological processes like germination, senescence etc.	<b>PO1, PO10, PO11</b>
<b>CO3</b>	<b>Demonstrate</b> different assay systems of plant enzymes.	<b>PO1, PO2, PO3, PO4, PO8</b>
<b>CO4</b>	<b>Demonstrate</b> tissue culture techniques.	<b>PO1, PO3, PO6, PO8</b>
<b>CO5</b>	<b>Summarize</b> the knowledge of plant biochemistry to understand important physiological processes like photosynthesis.	<b>PO1, PO4, PO8,</b>

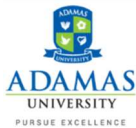
Course Code	Course Title	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
<b>BIC11047</b>	<b>DSE III PLANT BIOCHEMISTRY</b>	Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking	Problem Identification Ability	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to Environment	Development to Society.	Development to Humanity
		3	2	3	2	-	1	-	3	-	1	1	-

1=weakly mapped

2= moderately mapped

3=strongly mapped

**Model Question  
Paper**

 <b>ADAMAS UNIVERSITY</b> PURSUE EXCELLENCE	<b>ADAMAS UNIVERSITY</b> <b>MID-SEMESTER EXAMINATION MARCH 2022</b> (Academic Session: 2021 – 22)		
<b>Name of the Program:</b>	<b>BSc. Biochemistry</b>	<b>Semester:</b>	VI
<b>Paper Title:</b>	PLANT BIOCHEMISTRY	<b>Paper Code:</b>	BIC11047
<b>Maximum Marks:</b>	20	<b>Time Duration:</b>	2 Hrs
<b>Total No. of Questions:</b>	3	<b>Total No of Pages:</b>	1
<i>(Any other information for the student may be mentioned here)</i>	<ol style="list-style-type: none"> <li>1. At top sheet, clearly mention Name, Univ. Roll No., Enrolment No., Paper Name &amp; Code, Date of Exam.</li> <li>2. All parts of a Question should be answered consecutively. Each Answer should start from a fresh page.</li> <li>3. Assumptions made if any, should be stated clearly at the beginning of your answer.</li> </ol>		

<b>Group A</b> <b>Answer all the Questions (2 x 5 = 10)</b>			
Sl No:		Knowledge Level	CO
1	a) During day time, CAM plants procure CO <sub>2</sub> for photosynthesis from which compound? b) Summarize Hill reactions and its significance. c) Maximum photosynthesis in green plants takes place in which light? d) Draw the diagram of cyclic photophosphorylation.(1+2+1+1)	<b>Apply</b>	<b>CO2</b>
2	Illustrate the role of vacuoles in plant cells. What are tonoplast, plasmodesmata and lignin? (2+3)	<b>Remember</b>	<b>CO1</b>
<b>Group B</b> <b>Answer all the Questions (1 x 10 = 10)</b>			
3	a) Discuss the reactions of Calvin cycle. b) Illustrate the role of PS I & PS II in photosynthesis. (c) What are C <sub>4</sub> plants? Draw the diagram of CAM pathway.(4+3+3)	<b>Understanding,</b> <b>Apply</b>	<b>CO3, CO4</b>

<b>BIC12048</b>	<b>DSE III PLANT BIOCHEMISTRY LAB (PRACTICAL)</b>	<b>L</b>	<b>T</b>	<b>P</b>	<b>C</b>
<b>Version 1.0</b>	<b>Contact Hours - 45</b>	0	0	3	2
<b>Pre-requisites/Exposure</b>	Basic Knowledge of Biochemistry & Botany				
<b>Co-requisites</b>	--				

### Course Objectives

3. To provide students with hands-on training in the field of plant biochemistry.
1. To provide in depth knowledge of modern plant science research.
4. Students will become more proficient with different practical applications of plant biochemistry (e.g. plant tissue culture).

### Course Outcomes

On completion of this course, the students will be able to

**CO1 Explain** the basic principles of plant biochemistry.

**CO2 Illustrate** fundamentals of plant physiological processes like germination, senescence etc.

**CO3 Demonstrate** different assay systems of plant enzymes.

**CO4 Demonstrate** tissue culture techniques.

**CO5 Summarize** the knowledge of plant biochemistry to understand important physiological processes like photosynthesis.

### Catalog Description

The discipline specific course “plant biochemistry lab” is a practical paper which has been designed to provide the knowledge of different aspects of plant biochemistry. It will provide understanding of important physiological processes in plants. Students will be able to understand of tissue culture and other techniques and will practice hands-on all of them.. All the lectures will be devoted on discussions of basic theories and advanced topics, focusing on practical implementation of knowledge. Classes will be conducted by lecture as well as power point presentation, audio visual virtual lab session as per requirement. The tutorials will familiarize the students with practical problem-solving techniques led by the course coordinator. Students will strongly grab the basic concepts of the subject via exercise and discussions with the coordinator.

## Course Content

### PLANT BIOCHEMISTRY LAB

1. Qualitative/Quantitative analysis of carbohydrates from plant sources. [5 Hours]
2. Isolation of photosynthetic pigment from plant leaves and their separation by TLC. [5 Hours]
- 3 Qualitative/Quantitative analysis of proteins from plant sources [5 Hours]
4. Qualitative/Quantitative analysis of lipids from plant sources [5 Hours]
5. Estimation of secondary metabolites from plant sources. [10 Hours]
6. Extraction of caffeine from tea leaves. [10 Hours]
7. Biochemical analysis of seeds. [10 Hours]
8. Basics of Plant Tissue Culture (PTC). [10 Hours]

### SUGGESTED READINGS

3.Plant Biochemistry (2008), Caroline Bowsher, Martin steer, Alyson Tobin, Garland science ISBN978-0-8153-4121-5

4.BiochemistryandmolecularBiologyofplant-Buchanan.(2005)1edition.Publisher:I

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

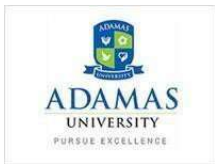
### Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and Pos		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	<b>Explain</b> the basic principles of plant biochemistry.	PO1, PO2, PO3
CO2	<b>Illustrate</b> fundamentals of plant physiological processes like germination, senescence etc.	PO1, PO10, PO11
CO3	<b>Demonstrate</b> different assay systems of plant enzymes.	PO1, PO2, PO3, PO4, PO8
CO4	<b>Demonstrate</b> tissue culture techniques.	PO1, PO3. PO6, PO8
CO5	<b>Summarize</b> the knowledge of plant biochemistry to understand important physiological processes like photosynthesis.	PO1, PO4, PO8,





## Model Question Paper

<b>Name:</b>  <b>Enrolment No:</b>			
<b>Course: BIC12048 – DSE III PLANT BIOCHEMISTRY LAB (PRACTICAL)</b> <b>Program: B.Sc. Biochemistry</b> <span style="float: right;"><b>Time: 03 Hrs.</b></span> <b>Semester: Even 2019-20</b> <span style="float: right;"><b>Max. Marks: 40</b></span>			
<b>Instructions:</b> Attempt any <b>two</b> questions from <b>Section A</b> (each carrying 10 marks); <b>Section B</b> is Compulsory (carrying 10 marks).			
<b>Section A ( Attemptany Two)</b>			
1.	a) Write the principle of extraction of urease from jack bean.(U) b) Determine the effect of temperature on a membrane.(Ap)	4 6	CO1 CO2
2.	a) Explain the basic theory of tissue culture.(U) b) Determine the rate of oxygen evolution with respect to light intensity.(Ap)	4 6	CO3 CO4
3.	a) Write the principle behind column chromatography.(U) b) Demonstrate the presence of amylase in germinating seed with a simple experiment.(Ap)	4 6	CO3 CO2
<b>SECTION B is compulsory</b>			
4.	<b>Viva-voce</b> (U/An/Ap/R/Ev)	<b>10</b>	CO1 CO2 CO3 CO4 CO5
5.	<b>Practical copy</b> (U/Ap/Ev)	<b>10</b>	CO1 CO2 CO3

<b>BIC11049</b>	<b>DSE III Advanced cell biology(Theory)</b>	L	T	P	C
<b>Version 1.0</b>	Contact hours: 60	3	1	0	4
<b>Pre-requisites/Exposure</b>	Basic Knowledge in cell biology				
<b>Co-requisites</b>	-				

### Course Objectives

1. To **recall** and **extend** the basic idea of structure and function of prokaryotic and eukaryotic cells
2. To **discuss** detailed perspective of cell including physiological properties, cell composition, growth, metabolic processes, signalling pathways, lifecycle.
3. To **interpret** the applications of different microscopy as tools for understanding cell biology.
4. To **discuss** about cell cycles, cell division and apoptosis
5. To **apply** and **assess** the cell biology in microscopic and molecular level to understand human health and disease.

### Course Outcomes

On completion of this course, the students will be able to

CO1: **recall** and extend the basic idea of structure and function of prokaryotic and eukaryotic cells

CO2: **discuss** detailed perspective of cell including physiological properties, cell composition, growth, metabolic processes, signalling pathways, life cycle.

CO3: **interpret** the applications of different microscopy as tools for understanding cell biology.

CO4: **discuss** about cell cycles, cell division and apoptosis.

CO5: **apply** and assess the cell biology in microscopic and molecular level to understand of human health and disease.

### Catalog Description

This course deals with the biology of cells of higher organisms: The structure, function, and biosynthesis of cellular membranes and organelles; cell growth and oncogenic transformation; transport, receptors, and cell signaling; the cytoskeleton, the extracellular matrix, and cell movements; chromatin structure and RNA synthesis.

## Course Content

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### 1. Unit I Introduction to cell biology [10 Lecture Hours]

Prokaryotic (archaea and eubacteria) and eukaryotic cell (animal and plant cells), cells as experimental models.

### 2. Unit II Tools of cell biology [10 Lecture Hours]

Light microscopy, phase contrast microscopy, fluorescence microscopy, confocal microscopy, electron microscopy, FACS. Centrifugation for subcellular fractionation.

### 3. Unit III Structure of different cell organelles

Structure of nuclear envelope, nuclear pore complex. ER structure. Organization of Golgi, Lysosome. Structure and functions of mitochondria, chloroplasts and peroxisomes. Zellweger syndrome.

### 4. Unit IV Protein trafficking [10 Lecture Hours]

Selective transport of proteins to and from the nucleus. Regulation of nuclear protein import and export. Targeting proteins to ER, smooth ER and lipid synthesis. Export of proteins and lipids from ER and into ER. Lipid and polysaccharide metabolism in Golgi. Protein sorting and export from Golgi. Mechanism of vesicular transport, cargo selection, coat proteins and vesicle budding, vesicle fusion. Protein import and mitochondrial assembly, protein export from mitochondrial matrix. Import and sorting of chloroplast proteins.

### 5. Unit V Cytoskeletal proteins [10 Lecture Hours]

Structure and organization of actin filaments. Treadmilling and role of ATP in microfilament polymerization, organization of actin filaments. Non-muscle myosin. Intermediate filament proteins, assembly and intracellular organization. Assembly, organization and movement of cilia and flagella.

### 6. Unit VI Cell wall and extracellular matrix [5 Lecture Hours]

Prokaryotic and eukaryotic cell wall, cell matrix proteins. Cell-matrix interactions and cell-cell interactions. Adherence junctions, tight junctions, gap junctions, desmosomes, hemidesmosomes, focal adhesions and plasmodesmata.

### 7. Unit VII Cell cycle, cell death and cell renewal [5 Lecture Hours]

Eukaryotic cell cycle, restriction point, and checkpoints. Cell division. Apoptosis and necrosis - brief outline. Salient features of a transformed cell.

## Reference Books

1. The Cell: A Molecular Approach (2009) 5th ed., Cooper, G.M. and Hausman, R.E., ASM Press & Sunderland (Washington DC), Sinauer Associates, MA, ISBN:978-0-87893-300-6.
2. Molecular Cell Biology (2012) 7th ed., Lodish, H., Berk, A., Zipursky, S.L., Matsudaira, P., Baltimore, D. and Darnell, J., W.H. Freeman & Company (New York), ISBN:13:978-1-4641-0981-2 / ISBN:10:1-4641-0981-8.

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

## Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and Pos		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	<b>recall</b> and extend the basic idea of structure and function of prokaryotic and eukaryotic cells	PO1, PO6
CO2	<b>discuss</b> detailed perspective of cell including physiological properties, cell composition, growth, metabolic processes, signalling pathways, life cycle.	PO4, PO2, PO3, PO1, PO5
CO3	<b>interpret</b> the applications of different microscopy as tools for understanding cell biology.	PO1, PO2, PO3, PO4, PO5, PO7
CO4	<b>discuss</b> about cell cycles, cell division and apoptosis.	PO1, PO5, PO7, PSO2
CO5	<b>apply</b> and assess the cell biology in microscopic and molecular level to understand of human health and disease.	PO10, PO11, PO12

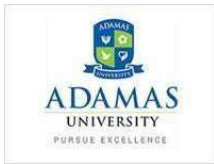
		Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking Ability	Problem Identification	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to Environment	Development to Society	Development to Humanity
Course Code	Course Title	PO 1	PO 2	PO 3	PO 4	PO 5	PO 6	PO 7	PO 8	PO 9	PO1 0	PO1 1	PO1 2
<b>BIC110 49</b>	<b>DSE III Advanced cell biology</b>	3	2	2	1	2	1	2	1	-	1	1	1

1=weakly mapped

2= moderately mapped

3=strongly mapped

### Model Question Paper

<b>Name:</b>	
<b>Enrolment No:</b>	
<b>Course: BIC11049 – DSE III Advanced cell biology</b> <b>Program: B.SC Biochemistry</b> <b>Semester: Even 2019-20</b>	
<b>Time: 03Hrs.</b> <b>Max. Marks: 40</b>	
<b>Instructions:</b> Attempt any three questions from <b>Section A</b> (each carrying 4 marks); any <b>four Questions</b> from <b>Section B</b> (each carrying 7 marks).	
<b>Section A ( Attemptany Three)</b>	
1.	Pinocytosis refers to the uptake of membrane and solutes by the cell in small vesicles. True/False? <b>Justify.</b> (U)
4	<b>CO1</b>
2.	A "signal patch" is the
4	<b>CO2</b>

	receptor to which a "signal peptide" binds. True/False? <b>Justify.</b> (U)		
3.	<b>What</b> is apoptosis? (R)	4	CO5
4.	<b>Prove</b> that lipid molecules can move through the membranes? (Ap)	4	CO3
	<b>SECTION B</b> (Attempt any <b>four Questions</b> )		
5.	a) <b>Compare</b> and contrast the 'vesicular transport' and 'cisternal maturation' models of golgi function (Ap). b) Why SNARES are necessary for the process of "cotranslational transport"?	4+3	CO2
6.	a) <b>Where</b> Glycosylation is a form of protein covalent modifications that occurs in cell? Describe the process briefly (U) b) Signal peptides are permanent structural components of a protein how can you use it for protein purification?(Ap)	1+3+3	CO4 CO2 CO3
7.	a) <b>Why</b> sulfation required for cell?(U) b) <b>What</b> do you know about membrane liquidity?(R) c) <b>What</b> is the structural role of ER in cell?(U)	2+3+2	CO2
8.	a) <b>Describe</b> the connection between single membrane bound organelles in cell?(Ap) b) <b>Difference</b> between primary and secondary lysosome.(Ap)	5+2	CO5

9.	a) <b>What</b> are the hall marks of cancer?(R) b) <b>How</b> can you prove velocity gradient centrifugation is inferior to density gradient centrifugation?(U)	3+4	CO2
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<b>BIC12050</b>	<b>DSE III Advanced Cell Biology Lab(Practical)</b>	L	T	P	C
<b>Version 1.0</b>	Contact Hours: 45	0	0	3	2
<b>Pre-requisites/Exposure</b>	Basic Knowledge of Cell Biology				
<b>Co-requisites</b>	-				

### Course Objectives

1. To **examine** animal and plant cells
2. To **demonstrate** micrographs of different cell components
3. To **illustrate** the interactions of the cells and how cellular components are used to generate and utilize energy in cells
4. To **apply** their knowledge of cell biology to select examples of changes or losses in cell function and compare and contrast the events of cell cycle and its regulation.
5. To **appraise** the laws of heredity with practical emphasis on inheritance..

### Course Outcomes

On completion of this course, the students will be able to

CO1: **examine** animal and plant cells.

CO2: **demonstrate** micrographs of different cell components.

CO3: **illustrate** the interactions of the cells and how cellular components are used to generate and utilize energy in cells.

CO4: **apply** their knowledge of cell biology to select examples of changes or losses in cell function and compare and contrast the events of cell cycle and its regulation.

CO5: **appraise** the laws of heredity with practical emphasis on inheritance.

## Catalog Description

Course provides hand-on training in current cell biological methods. A discovery-based component of this course focuses on application of the mentioned methods to study changes in protein expression and cytoskeleton organization in cells exposed to micro environmental stress stimuli.

## Course Content

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1. Visualization of animal and plant cell by methylene blue. [10 LectureHours]
2. Identification of different stages of mitosis in onion root tip. [10 LectureHours]
3. Identification of different stages of meiosis in grasshopper testis. [10 LectureHours]
4. Micrographs of different cell components (dry lab). [10 LectureHours]
5. Sub-cellular fractionation. [10 LectureHours]
6. Visualization of nuclear fraction by acetocarmine stain. [5 LectureHours]
7. Staining and visualization of mitochondria by Janus green stain. [5 LectureHours]

## Reference Books

1. The Cell: A Molecular Approach (2009) 5th ed., Cooper, G.M. and Hausman, R.E., ASM Press & Sunderland (Washington DC), Sinauer Associates, MA, ISBN:978-0-87893-300-6.
2. Molecular Cell Biology (2012) 7th ed., Lodish, H., Berk, A., Zipursky, S.L., Matsudaira, P., Baltimore, D. and Darnell, J., W.H. Freeman & Company (New York), ISBN:13:978-1-4641-0981-2 / ISBN:10:1-4641-0981-8.

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40



### Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

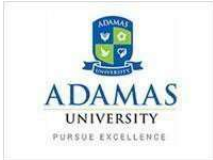
<b>Mapping between COs and Pos</b>		
	<b>Course Outcomes (COs)</b>	<b>Mapped Program Outcomes</b>
<b>CO1</b>	<b>Examine</b> animal and plant cells	<b>PO1, PO2</b>
<b>CO2</b>	<b>Demonstrate</b> micrographs of different cell components	<b>PO1,PO2, PO3, PO4</b>
<b>CO3</b>	<b>Illustrate</b> the interactions of the cells and how cellular components are used to generate and utilize energy in cells	<b>PO5, PO6, PO8, PO7</b>
<b>CO4</b>	<b>Apply</b> their knowledge of cell biology to select examples of changes or losses in cell function and compare and contrast the events of cell cycle and its regulation.	<b>PO10, PO11,PO12</b>
<b>CO5</b>	<b>Apprise</b> the laws of heredity with practical emphasis on inheritance.	<b>PO1, PO6, PO8, PO9, PO12, PSO2</b>

		Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking Ability	Problem Identification	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to Environment	Development to Society	Development to Humanity
Course Code	Course Title	PO 1	PO 2	PO 3	PO 4	PO 5	PO 6	PO 7	PO 8	PO 9	PO10	PO11	PO12
<b>BIC12050</b>	DSE III Advanced Cell Biology Lab	3	2	1	1	1	2	1	2	1	3	1	2

1=weakly mapped

2= moderately mapped  
3=strongly mapped

### Model Question Paper

<b>Name:</b> <b>Enrolment No:</b>			
<b>Course: BIC12050– DSE III Advanced Cell Biology Lab</b> <b>Program: B.SC Biochemistry</b> <b>Semester: Even 2019-20</b> <b>Time: 03Hrs.</b> <b>Max. Marks: 40</b>			
<b>Instructions:</b> Attempt all questions			
1	Perform the WBC identifications using Gimseastain.(Ap)	<b>10</b>	<b>CO3</b>
2	Count the number of WBS in the sample and calculate the number of WBC in per ml blood.(Ap)	<b>10</b>	<b>CO3</b>
3	Note book(U/Ap/R)	<b>10</b>	<b>CO1, CO2, CO3</b>
4	Viva Voce(U/Ap/R)	<b>10</b>	<b>CO1, CO2, CO3</b>

<b>BIC 11051</b>	<b>DSE III Research Methodology(Theory)</b>	L	T	P	C
<b>Version 1.0</b>	Contact Hours: 120	6	0	0	6
<b>Pre-requisites/Exposure</b>	Basic Knowledge of Biochemistry				
<b>Co-requisites</b>	-				

## Course Objectives

- To verify and test important facts
- To understand an event or process or phenomenon to identify the cause and effect relationship
- To develop new scientific tools, concepts and theories to solve and understand scientific and nonscientific problems
- To find solutions to scientific, nonscientific and social problems
- To overcome or solve the problems occurring in our everyday life.

## Course Outcomes

On completion of this course, the students will be able to

CO1: **verify** and test important facts

CO2: **illustrate** an event or process or phenomenon to identify the cause and correct relationship

CO3: **develop** new scientific tools, concepts and theories to solve and understand scientific and nonscientific problems

CO4: **determine** solutions to scientific and social problems

CO5: **solve** the problems occurring in our everyday life

## Catalog Description

The Objective of this course to pay attention to the most important dimension of Research i.e. Research Methodology. It will enable the Researchers to develop the most appropriate methodology for their Research Studies. The mission of the course is to impart research skills to the beginners and help improve the quality of Research by the existing researchers..

## Course Content

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- Article Writing [12 LectureHours]
- Essay [12 LectureHours]
- Research Paper [12 LectureHours]
- Book Review [12 LectureHours]
- Laboratory Research [12 LectureHours]

- Marketing Research [12 LectureHours]
- Legislative Drafting [12 LectureHours]
- Thesis; Dissertation [12 LectureHours]
- Book[12 LectureHours]
- Citation Methods and Styles[6 LectureHours]
- Research Grant Proposals[6 LectureHours]

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

Components	Mid Term	Attendance	Class Assessment	End Term
Weightage (%)	20	10	30	40

**Relationship between the Course Outcomes (COs) and Program Outcomes (POs)**

Mapping between COs and Pos		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	Verify and test important facts	PO1, PO11
CO2	Illustrate an event or process or phenomenon to identify the cause and correct relationship	PO1,PO2, PO3,
CO3	Develop new scientific tools, concepts and theories to solve and understand scientific and nonscientific problems	PO1, PO2, PO3, PO4, PO5, PO11,
CO4	Determine solutions to scientific and social problems	PO1, PO5, PO12,
CO5	Overcome or solve the problems occurring in our every day life	PO1, PO6, PO8, PO9, PO12

Course Code	Course Title	PO 1	PO 2	PO 3	PO 4	PO 5	PO 6	PO 7	PO 8	PO 9	PO1 0	PO1 1	PO1 2	
<b>BIC 11051</b>	DSE III Research Methodology	Fundamental Knowledge	3	3	2	1	2	1	-	1	1	-	1	1
		Critical Thinking												
		Skill												
		Technical Knowledge												
		Logical Thinking Ability												
		Problem Identification												
		Analytical Knowledge												
		Career Goals												
		Team Work												
		Sustainable Development to Environment												
		Development to Society												
		Development to Humanity												

1=weakly mapped

2= moderately mapped

3=strongly mapped

<b>BIC 15053</b>	<b>SEMINAR ON CONTEMPORARY RESEARCH</b>	<b>L</b>	<b>T</b>	<b>P</b>	<b>C</b>
	<b>BIOCHEMISTRY(Practical)</b>				
<b>Version 1.0</b>					<b>2</b>
<b>Pre-requisites/Exposure</b>	Concept of Biochemistry and allied subjects at previous classes				
<b>Co-requisites</b>	--				

### Course Objectives

1. This will enable students to design and evaluate scientific investigations
2. Students will learn to deduce evidence-based conclusions.
3. To develop presentation and scientific content writing.

### Course Outcomes

On completion of this course, the students will be able to

CO1 **compile** novel ideas to enrich their scientific interest

CO2 **extend** their theoretical and practical knowledge obtained to utilize biochemical and biotechnological skills for presenting a scientific work.

CO3 **compare** different biochemical studies with contemporary research and analyze the results obtained.

CO4 **design** and **evaluate** scientific investigations

CO5 **deduce** evidence-based conclusions.

### Catalog Description

The core-course of 'seminar on contemporary research in biochemistry' will enable the students to nurture their research interest by compiling basic knowledge obtained during their education together with novel ideas from contemporary research. An idea about appropriate application of biochemical and biotechnological skill for industrial and research purpose can be developed. With the potential to design and evaluate scientific investigations to the students, who will learn to comprehend conclusions based on experimental evidences. The entire literature review work and experimentation focuses on practical implementation of knowledge. Students will perceive the basic concepts of the subject via exercise and discussions with the mentor.

**Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination Examination Scheme:**

<b>Components</b>	<b>Report/Thesis submission</b>	<b>Presentation</b>
<b>Weightage (%)</b>	<b>50</b>	<b>50</b>

### Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

<b>Mapping between COs and POs</b>		
	<b>Course Outcomes (COs)</b>	<b>Mapped Program Outcomes</b>
<b>CO1</b>	<b>compile</b> novel ideas to enrich their scientific interest	<b>PO2, PO3, PO5</b>
<b>CO2</b>	use their theoretical and practical knowledge obtained to <b>apply</b> biochemical and biotechnological skills for presenting a scientific work.	<b>PO5, PO7, PO8, PO10</b>
<b>CO3</b>	<b>compare</b> different biochemical studies with contemporary research and analyze the results obtained.	<b>PO5, PO6, PO7</b>
<b>CO4</b>	<b>design</b> and <b>evaluate</b> scientific investigations	<b>PO5, PO6, PO8, PO10, PO11</b>
<b>CO5</b>	<b>deduce</b> evidence-based conclusions.	<b>PO6, PO7, PO8, PO11</b>

		Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking Ability	Problem Identification	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to environment	Development to Society	. Development to Humanity
Course Code	Course Title	PO 1	PO 2	PO 3	PO 4	PO 5	PO 6	PO 7	PO 8	PO 9	PO1 0	PO1 1	PO1 2
<b>BIC 15053</b>	Seminar on Contemporary Research in Biochemistry	-	3	1	-	3	3	3	3	-	1	2	1

1=weakly mapped

2= moderately mapped



3=strongly mapped

<b>BIC 15052</b>	<b>DISSERTATION</b>	<b>L</b>	<b>T</b>	<b>P</b>	<b>C</b>
<b>Version 1.0</b>		0	0	4	6
<b>Pre-requisites/Exposure</b>	Concept of Biochemistry and allied subjects at class XII				
<b>Co-requisites</b>	--				

### Course Objectives

1. This will enable students to design and evaluate scientific investigations
2. Students will learn to deduce evidence-based conclusions.
3. Develop presentation and scientific content writing skill..

### Course Outcomes

On completion of this course, the students will be able to

CO1 **compile** novel ideas to enrich their scientific interest

CO2 use their theoretical and practical knowledge obtained to **apply** biochemical and biotechnological skills for identification, estimating different biomolecules and their applications. CO3 **compare** different biochemical techniques and analyze the results obtained.

CO4 **design** and **evaluate** scientific investigations

CO5 **deduce** evidence-based conclusions.

### Catalog Description

The core-course of ‘dissertation’ will enable the students to nurture their research interest by compiling basic knowledge obtained during their education together with novel ideas from contemporary research. An idea about appropriate application of biochemical and biotechnological skill for industrial and research purpose can be developed. With the potential to design and evaluate scientific investigations to the students, who will learn to comprehend conclusions based on experimental evidences. The entire literature review work and experimentation focuses on practical implementation of knowledge. Students will perceive the basic concepts of the subject via exercise and discussions with the mentor.

### Modes of Evaluation: Quiz/Assignment/ presentation/ extempore/ Written Examination

#### Examination Scheme:

<b>Components</b>	<b>Report/Thesis submission</b>	<b>Presentation</b>
<b>Weightage (%)</b>	<b>50</b>	<b>50</b>

### Relationship between the Course Outcomes (COs) and Program Outcomes (POs)

Mapping between COs and POs		
	Course Outcomes (COs)	Mapped Program Outcomes
CO1	<b>compile</b> novel ideas to enrich their scientific interest	PO2, PO3, PO5
CO2	use their theoretical and practical knowledge obtained to <b>apply</b> biochemical and biotechnological skills for identification, estimating different biomolecules and their applications.	PO5, PO7, PO8, PO10
CO3	<b>compare</b> different biochemical techniques and analyze the results obtained.	PO5, PO6, PO7
CO4	<b>design</b> and <b>evaluate</b> scientific investigations	PO5, PO6, PO8, PO10
CO5	<b>deduce</b> evidence-based conclusions.	PO6, PO7, PO8

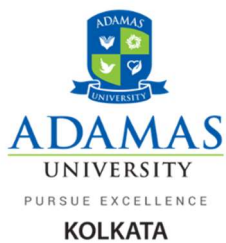
		Fundamental Knowledge	Critical Thinking	Skill	Technical Knowledge	Logical Thinking Ability	Problem Identification	Analytical Knowledge	Career Goals	Team Work	Sustainable Development to Environment	Development to Society	Development to Humanity
Course Code	Course Title	PO 1	PO 2	PO 3	PO 4	PO 5	PO 6	PO 7	PO 8	PO 9	PO10	PO11	PO12
<b>BIC 14054</b>	Dissertation	-	1	1	-	3	3	3	3	-	1	-	-

1=weakly mapped

2= moderately mapped

3=strongly mapped

Course Code	Course Title	PO 1	PO 2	PO 3	PO 4	PO 5	PO6	PO 7	PO 8	PO 9	PO 10	PO 11	PO 12
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**ADAMAS UNIVERSITY  
SCHOOL OF LIFE SCIENCE AND BIOTECHNOLOGY**

**B.Sc. (Hons) Biochemistry (2020-21)**

**CO-PO mapping:**

BIC11001	Molecules of Life	3	3	-	-	-	-	-	-	-	-	-	-
BIC12002	Molecules of Life Lab	3	3	3	3	-	3	3	-	-	-	-	-
BIC11003	PROTEIN (THEORY)	3	3	3	3	-	-	3	-	-	-	-	-
BIC12004	PROTEIN LAB (PRACTICAL)	3	-	3	-	-	-	-	-	-	-	-	-
ENG11057	English Language & Literature	-	-	-	-	-	-	-	-	-	-	-	3
DGS11001	Design Thinking	-	3	-	-	-	3	-	-	3	3	-	-
BOT11001	ELECTIVE BOTANY - I (THEORY)	3	3	-	-	-	-	-	-	-	-	-	-
ZOL11001	ELECTIVE ZOOLOGY I (THEORY)	3	3	-	-	-	-	-	-	-	-	-	-
CHM11151	Elective Chemistry I	-	-	3	-	-	-	-	-	-	-	-	-
PSG13004	General Psychology	3	3	-	-	-	-	-	-	-	-	-	-
BOT12002	Elective Botany I Lab	3	-	3	-	-	-	-	-	-	-	-	-
ZOL12002	ELECTIVE ZOOLOGY LAB I (PRACTICAL)	3	-	3	-	-	-	-	-	-	-	-	-
BIC22515	Elective Chemistry I lab	3	-	3	-	3	-	-	-	-	-	-	-
BIC11005	Enzymes (THEORY)	3	3	3	-	-	-	-	-	-	-	-	-
BIC12006	Enzymes Lab	3	-	3	3	-	-	3	-	3	-	-	-



	AMINO ACIDS AND NUCLEOTIDES LAB												
BIC11016	SEC I : TOOLS AND TECHNIQUES IN BIOCHEMISTRY	3	-	3	3	3	3	3	-	-	-	-	-
BIC11017	Clinical Biochemistry (THEORY)	3	3	-	-	-	-	3	-	-	-	-	-
SOC14100	Community Service						3			3			3
EIC11001	Venture Ideation	-	-	-	-	-	-	-	3	3	-	3	-
BIC11012	Cell Biology	3	-	3	-	-	-	-	-	-	-	-	-
BIC12013	Cell biology Lab	-	-	3	3	-	-	-	-	-	-	-	-
SDS11506	Statistics I	3	3	-	-	-	-	-	-	-	-	-	-
CHM11151	Elective Chemistry I	3	3	-	-	-	-	-	-	-	3	-	-
CHM12152	Elective Chemistry Lab III	-	-	-	-	-	-	3	-	-	-	-	-
CSE11641	Elective Computer Science I	3	-	3	-	-	-	-	-	-	-	-	-
CSE12642	Elective Computer Science I Lab	3	-	3	-	3	-	-	-	-	-	-	-

BIC11018	Gene organization, replication and repair	3	3	3	-	3	-	-	-	-	-	-	-
BIC12019	Gene organization, replication And repair Lab	-	3	-	-	3	-	3	-	-	-	-	-
BIC11020	Immunology	3	3	-	-	3	3	-	-	-	-	-	-
BIC12021	Immunology Lab	-	3	-	-	3	-	3	-	-	-	-	-
BIC11022	Membrane biology and bioenergetics	-	3	-	-	-	-	-	-	-	-	-	-
BIC12023	MEMBRANE BIOLOGY AND BIOENERGETICS  LAB (PRACTICAL)	3	-	3	-	-	-	-	3	-	-	-	-
BIC11024	Protein Purification Techniques	3	3	-	-	3	-	3	-	-	-	-	-
BIC11025	Recombinant DNA Technology	3	3	2	2	2	2	-	1	1	2	1	-
PSG11021	Human Values and Professional Ethics	-	-	-	-	-	-	-	-	-	-	3	-
SDS11507	Elective Statistics II		3			3							





	Molecular Basis of Non infectious Human Disease												
BIC11036	DSE  II Molecular Basis of Infectious Human Disease	3	3	3	-	3	3	3	3	3	-	-	-
BIC11038	Medical Microbiology	3	3	3	-	3	3	3	3	3	-	-	-
BIC11040	Nutritional Biochemistry	3	3	3	--	-	3	3	3	-	3	3	-
BIC12037	Molecular Basis of Infectious Human Disease Lab	3	3	-	3	3	3	3	3	-	-	-	-
BIC12039	DSE Medical Microbiology Lab	3	3	-	-	3	3	3	3	-	-	-	-
BIC12041	Nutritional Biochemistry Lab	3	3	3	3	3	3	3	3	-	3	3	-
BIC14057	Industry Internship	-	3	3	3	3	3	3	3	3	-	-	-
BIC11043	Genetic Engineering and Biotechnology	3	-	-	-	-	-	-	-	-	-	-	-
BIC12044	Genetic Engineering and Biotechnology Lab	3	3	-	-	-	-	-	-	-	-	-	-
BIC 11045	Hormones: Biochemistry and	3	3	3	3	3	3	3	3	-	-	-	3

	Function												
BIC 12046	Lab:Hormones Biochemistry and Function	3	3	3	-	3	3	3	3	-	-	3	-
BIC11047	PLANT BIOCHEMISTRY	3	-	3	-	-	-	-	3	-	-	-	-
BIC12048	PLANT BIOCHEMISTRY LAB	3	-	3	-	-	-	-	3	-	-	-	-
BIC11049	Advanced cell biology	3	-	-	-	-	-	-	-	-	-	-	-
BIC12050	Advanced Cell Biology Lab	3	-	-	-	-	-	-	-	-	3	-	-
BIC11051	Research Methodology	3	3	-	-	-	-	-	-	-	-	-	-
BIC 15053	Seminar on Contemporary Research in Biochemistry	-	3	-	-	3	3	3	3	-	-	-	-
BIC 15052	Dissertation	-	-	-	-	3	3	3	3	-	-	-	-
Average		2.69 5652 17	1.869 565	1.542 857	0.89 4737	1.22 7273	1.105 263	1.41 1765	0.890 625	0.67 5	0.454 545	0.685 714	0.428 571